1	β -Aryl- β -amino acid aminotransferase from <i>Variovorax</i> sp. JH2 is useful for
2	enantioselective β -phenylalanine production
3	
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16	
17	Abbreviations: BHA, glutamate-1-semialdehyde 2,1-aminomutase from Bacillus
18	halodurans; CDI, glutamate-1-semialdehyde 2,1-aminomutase from Corynebacterium
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1	diphtheriae; CNE, glutamate-1-semialdehyde 2,1-aminomutase from Cryptococcus
2	neoformans; FPLC, fast protein liquid chromatography; FRA, aminotransferase class-III
3	from <i>Frankia</i> sp.; <i>e.e.</i> , enantiomer excess; GITC,
4	2,3,4,6-tetra-O-acetyl-β-D-glucopyranosyl isothiocyanate; HPLC, high-performance
5	liquid chromatography; IP, internal peptide; PLP, pyridoxal 5'-phosphate; (S)-BAT,
6	(S) - β -phenylalanine:2-oxoglutarate aminotransferase; SDS-PAGE, sodium dodecyl
7	sulfate-polyacrylamide gel electrophoresis; SPO, glutamate-1-semialdehyde
8	2,1-aminomutase from Schizosaccharomyces pombe.

1 ABSTRACT

A bacterium, Variovorax sp. JH2, has an ability to degrade (S)-\beta-phenylalanine stereoselectively. The involved degradation, enzyme in the (S)- β -phenylalanine:2-oxoglutarate aminotransferase, was purified to homogeneity from *Variovorax* sp. JH2 and characterized. The enzyme was useful for (R)- β -phenylalanine production from racemic β-phenylalanine by enantioselective decomposition of (S)- β -phenylalanine through transamination. (S)- β -Phenylalanine and (S)-3-amino-3-(3-pyridyl)propionate served as good amino-donors in the transamination and 2-oxoglutarate, oxaloacetate, pyruvate, and 1,3-acetonedicarboxylate served as amino-acceptors. The enzyme had a molecular weight of about 72,000 and consisted of two identical subunits. Three internal amino acid sequences (54, 67, and 63 residues) were determined and showed homology with glutamate-1-semialdehyde 2,1-aminomutases.

Key words: β-Phenylalanine aminotransferase; β-Aryl-β-amino acid aminotransferase;

- 16 Variovorax sp.; Stereoselective synthesis

1. Introduction

 β -Phenylalanine (3-amino-3-phenylpropionic acid) is a type of β -amino acid present in several bioactive molecules, such as antibiotics and enzyme inhibitors [1, 2]. Because of its bioactivity, optically active β -phenylalanine and its derivatives are becoming important chiral building blocks for the synthesis of pharmaceutical agents. Enzymes catalyzing the synthesis reaction of β -phenylalanine are one of the potential methods for the preparation of chiral compounds, however, only a few enzymatic methods for the synthesis of optically active β -phenylalanine and its derivatives have been developed [3-7]. In our previous report, the microbial production of optically active β -phenylalanine through stereoselective degradation of racemic β -phenylalanine was reported [8]. Microorganisms were discovered that can catalyze stereoselective β -phenylalanine degradation available for the synthesis of optically active β -phenylalanine, and efficient β -phenylalanine production was achieved using these In particular, a potent strain Variovorax sp. JH2 isolated as a potent strains. β-phenylalanine-assimilating microorganism showed an ability degrade to (S)- β -phenylalanine stereoselectively through transamination with 2-oxoglutarate as an amino-acceptor. Some enzymes involved in β -amino acid metabolism have been reported, such as ω-amino acid aminotransferases and leucine 2,3-aminomutase [9-14].

1	However, there are a few reports on β -phenylalanine aminotransferases [6, 7]. The
2	present study investigated the aminotransferase involved in β -phenylalanine metabolism
3	of Variovorax sp. JH2. The potential of the purified enzyme for (R) - β -phenylalanine
4	production from racemic β -phenylalanine by enantioselective decomposition of
5	(S)- β -phenylalanine through transamination in the presence of 2-oxoglutarate was
6	examined and the detailed characterization of the enzyme,
7	(S)- β -phenylalanine:2-oxoglutarate aminotransferase ((S)-BAT), is described.
8	
9	2. Materials and methods
10	2.1. Chemicals
11	DL-3-Amino-3-phenylpropionic acid (racemic β -phenylalanine) was purchased from
12	Aldrich Chemical Co. (Milwaukee, WI, USA). 3-Oxo-3-phenylpropionic acid
13	(benzoylacetic acid) was a kind gift from Kaneka Co., Ltd (Osaka, Japan). All other
14	chemicals used in this work were of analytical grade and available commercially.
15	
16	2.2. Microorganisms and cultivation
17	Variovorax sp. JH2 isolated from a soil sample as a β -phenylalanine-assimilating
18	microorganism was used as a source of enzyme. The strain was cultivated at 28°C
	5

1	with shaking (120 strokes/min) for 24 h in 2-L flasks containing 500 ml of medium
2	comprising 0.1% (w/v) KH ₂ PO ₄ , 0.1% (w/v) K ₂ HPO ₄ , 0.03% (w/v) MgSO ₄ ·7H ₂ O,
3	0.01% (w/v) yeast extract, 0.2% (w/v) NH ₄ Cl, and 0.3% (w/v) racemic β -phenylalanine
4	(pH 7.0).
5	
6	2.3 Enzyme assays
7	The standard methods for the activity assay were as follows. The 200 μ l reaction
8	mixture contained 10 mM racemic β -phenylalanine, 10 mM 2-oxoglutarate, 5 mM
9	pyridoxal-5'-phosphate (PLP), 200 mM Tris/HCl buffer (pH 9.0), and an appropriate
10	amount of enzyme. The mixture was incubated at 28°C for 15 to 120 min where the
11	reaction proceeded linearly, and the reaction was terminated by adding 20 μl of 15%
12	(v/v) perchloric acid. Investigations of the effects of pH, temperature, and chemicals
13	on the enzymatic activity and kinetic parameters were carried out essentially by the
14	standard methods with slight modifications as described above. The reaction mixture
15	was centrifuged at 15,000 rpm for 3 min (MX-150; Tomy Seiko, Tokyo, Japan), and the
16	supernatant was analyzed as described below.
17	
18	2.4. Analytical methods for enzyme activity

1	To evaluate the enzymatic activity in the transamination of β -phenylalanine with
2	2-oxoglutarate, the concentration of β -phenylalanine and benzoylacetic acid in the
3	reaction mixture was analyzed by reverse-phase HPLC by using a Shimadzu LC-VP
4	system (Shimadzu, Kyoto, Japan) equipped with a Cosmosil 5C ₁₈ -AR-II column (0.46 \times
5	250 mm; Nacalai Tesque, Kyoto, Japan). The mobile phase was
6	acetonitrile/water/trifluoroacetate (4/6/0.1, v/v/v) at a flow rate of 1.0 ml/min and the
7	effluents were monitored at 210 nm by ultraviolet detection. One unit (U) of enzyme
8	activity was defined as the amount catalyzing the formation of 1 μ mol benzoylacetic
9	acid per min from racemic β -phenylalanine under the above conditions. To investigate
10	the substrate specificity for the amino-acceptor, the decrease in β -phenylalanine
11	concentration by transamination with β -phenylalanine as an amino-donor and various
12	amino-acceptors was monitored by the same method. To investigate the substrate
13	specificity for the amino-donor, the formation of L-glutamate by transamination with
14	2-oxoglutarate as an amino-acceptor and various amino-donors was monitored using an
15	amino acid analyzing system consisting of Separations module 2695 (Waters, Milford,
16	MA, USA) and Multi λ fluorescence detector 2475 (Waters). Samples were analyzed
17	after derivatization with N-hydroxysuccinimidyl-6-aminoquinolinylcarbamate. The
18	effluents were monitored by fluorescence detection (excitation, 250 nm; emission, 395

nm). Analysis of the enantiomeric purity of the remaining β -phenylalanine was performed using reverse-phase HPLC after derivatization with 2,3,4,6-tetra-*o*-acetyl-β-D-glucopyranosyl isothiocyanate (GITC) [15-17]. Methanol/water (55/45, v/v, pH 2.5 adjusted with phosphoric acid) was used as the mobile phase. Investigation of stereospecificity towards 3-amino-3-(3-pyridyl)propionate was also performed by the same method. 2.5. Analytical methods for proteins Protein concentrations were determined using a protein assay kit (Bio-Rad, Hercules, CA, USA) with bovine serum albumin as a standard [18]. To assay the relative concentration of protein eluted by liquid chromatography, the effluents were monitored at 280 nm by ultraviolet detection. The relative molecular mass of the native enzyme was determined by gel-filtration This method was carried out using a fast protein liquid liquid chromatography. chromatography (FPLC) system (Amersham Pharmacia Biotech Co., Uppsala, Sweden) equipped with Superdex 200 HR 10/30 column (Amersham Pharmacia Biotech Co.). The mobile phase comprised 200 mM NaCl and 20 mM potassium phosphate buffer

1	(pH 7.0) and the flow rate was 0.3 ml/min. Partially purified enzyme solution was
2	injected and the eluate was fractionated at 0.3 ml/tube. The molecular mass of the
3	enzyme was calculated based on the relative mobility using standard proteins (M.W.
4	marker; Oriental Yeast Co., Tokyo, Japan).
5	Sodium dodecyl sulfate-polyacrylamide gel electrophoresis (SDS-PAGE) was
6	performed in a 12.5% polyacrylamide gel using a Tris/glycine buffer system [19]. The
7	molecular mass of the subunit of the enzyme was determined from a comparison of the
8	mobility of the purified enzyme in SDS-PAGE with whose of standard proteins (M.W.
9	marker III; Daiichi Kagaku Yakuhin, Tokyo, Japan).
10	
11	2.6. Enzyme purification
12	All procedures were carried out at 0-5°C, and 20 mM potassium phosphate buffer
13	(pH 7.0) containing 0.2 mM PLP was used as a standard buffer unless otherwise
14	specified. Variovorax sp. JH2 cells (about 28 g obtained from 5 L of culture broth)
15	were suspended in 100 ml of buffer and disrupted for 10 min using an ultrasonic
16	oscillator (Kubota, Osaka, Japan). This sample was ultracentrifuged at 40,000 rpm for
17	1 h (L-80 Ultracentrifuge; Beckman, Fullerton, CA, USA) and the supernatant was used
18	as a cell-free extract. The cell-free extract was fractionated with solid ammonium

1	sulfate. The precipitate at 30-50% saturation of ammonium sulfate was obtained by
2	centrifugation at 12,000 rpm for 20 min (High Mac CR21; Hitachi, Tokyo, Japan) and
3	dissolved in the same buffer. This enzyme solution (50 ml) was dialyzed against 3 L
4	of buffer for 12 h and the dialyzed enzyme solution was applied to a DEAE-Sephacel
5	column (3.0×7.0 cm; Amersham Pharmacia Biotech Co.) equilibrated with the same
6	buffer. After washing the column with 200 ml of buffer, the enzyme was eluted with a
7	linear gradient of 0-1.0 M NaCl in 250 ml buffer. The fractions containing the enzyme
8	activity were combined (270 ml) and dialyzed for 12 h against 3 L of buffer. After the
9	addition of 4 M NaCl, the enzyme solution was applied to a Phenyl-sepharose CL-4B
10	column (3.0×7.3 cm; Amersham Pharmacia Biotech Co.) equilibrated with buffer
11	containing 4 M NaCl. After washing the column with 500 ml of buffer, the enzyme
12	was eluted with a linear gradient of 0-0.5% (v/v) Tween 20 in 200 ml of buffer. The
13	fractions containing the enzyme activity were combined (125 ml) and dialyzed for 12 h
14	against 3 L of buffer comprising 20 mM Tris/HCl buffer (pH 9.0) and 0.2 mM PLP.
15	The dialyzed enzyme solution was concentrated to 10 ml by ultrafiltration using
16	Centriprep YM-3 (MILLIPORE, Bedford, MA, USA). The enzyme solution was
17	applied to a Mono Q HR 5/5 column (Amersham Pharmacia Biotech Co.) equilibrated
18	with buffer comprising 20 mM Tris/HCl buffer (pH 9.0) and 0.2 mM PLP. After

1	washing the column with the buffer, the enzyme was eluted with a linear gradient of
2	0-1.0 M NaCl in 60 ml of the buffer. The fractions containing the enzyme activity
3	were combined (4 ml) and dialyzed for 12 h against 2 L of buffer comprising 20 mM
4	potassium phosphate buffer (pH 7.0) and 0.2 mM PLP. This sample was used as
5	purified enzyme solution for characterization of its biochemical properties.
6	
7	2.7. Preparation of internal peptides from the purified enzyme
8	Enzyme solution containing 1 mg of the purified enzyme was lyophilized and
9	dissolved in 100 μl of buffer comprising 20 mM Tris/HCl (pH 9.0) and 8 M urea.
10	After incubation at 37°C for 60 min for denaturation, the concentration of urea was
11	decreased to 4 M by adding 100 μ l of 20 mM Tris/HCl buffer (pH 9.0). After the
12	addition of 0.05 nmol lysyl endopeptidase (Wako Pure Chemicals, Osaka, Japan), the
13	solution was incubated at 30°C for 6 h for digestion. The digested sample was applied
14	to a μ RPC C2/18 SC2/10 column (Amersham Pharmacia Biotech Co.) connected to a
15	SMART system (Amersham Pharmacia Biotech Co.) and equilibrated with mobile
16	phase of 0.1% (v/v) trifluoroacetic acid in water. The peptides were eluted using a
17	linear gradient of 0-80% (v/v) acetonitrile in 10 ml of mobile phase at a flow rate of 0.1
18	ml/min. The amino acid sequences of the internal peptides were analyzed by

1	automated Edman degradation with a 491HT protein sequencer (Applied Biosystems,
2	Foster City, CA, USA).
3	
4	3. Results
5	3.1. Purification of (S)-BAT
6	(S)-BAT was purified about 17-fold by four successive steps from the cell-free extract
7	of Variovorax sp. JH2. Typical results of the purification are shown in Table 1. The
8	final preparation gave a single band on SDS-PAGE (Fig. 1) and showed specific activity
9	of 1.5 U/mg. The subunit molecular mass of purified enzyme determined by
10	SDS-PAGE was about 40,000 (Fig. 1). The native molecular mass determined by
11	gel-filtration chromatography was about 72,000.
12	
13	3.2. Kinetic parameters of (S)-BAT
14	Kinetic parameters of (S)-BAT in the transamination reaction with racemic
15	β -phenylalanine and 2-oxoglutarate were investigated as described in Materials and
16	methods except that the reaction was performed with a variety of substrate
17	concentrations. The $K_{\rm m}$ value for racemic β -phenylalanine in the reaction with 10 mM
18	2-oxoglutarate was 0.63 mM, and the V_{max} value was 1.7 U/mg. The K_{m} value for

2-oxoglutarate in the reaction with 10 mM racemic β -phenylalanine was 2.5 mM.

3.3. Stereoselectivity of (S)-BAT

To investigate the stereoselectivity of (S)-BAT in the transamination of racemic β -phenylalanine with 2-oxoglutarate for the enantioselective production of (*R*)- β -phenylalanine from racemic β -phenylalanine, the time-course of the reaction was analyzed essentially by the standard methods (Fig. 2). In a 1 h reaction, half of the 10 mM racemic β-phenylalanine was decomposed and about 5 mM benzoylacetic acid was produced. The optical purity of the remaining β -phenylalanine reached 99.6% *e.e.* for (R)- β -phenylalanine. Longer incubation did not result in a further decrease in the amount of β -phenylalanine. This result indicated that the purified enzyme catalyzed (S)- β -phenylalanine-specific transamination. In the longer incubation, benzoylacetic acid was degraded gradually by spontaneous decarboxylation.

15 3.4. Substrate specificity of (S)-BAT

The substrate specificity of (S)-BAT was investigated (Table 2). As amino-donors, 10 mM β-phenylalanine analogs, some other β-amino acids, proteinogenic amino acids, and their D-isomers were tested with 10 mM 2-oxoglutarate as an amino-acceptor.

1	Besides β -phenylalanine, 3-amino-3-(3-pyridyl)propionate served as a good substrate
2	for the enzyme and the relative activity was 56% of the enzymatic activity for racemic
3	β -phenylalanine. In the enantiomer analysis of the remaining amino acid during
4	racemic 3-amino-3-(3-pyridyl)propionate deamination, a decrease in only one
5	enantiomer was observed on HPLC analysis after GITC derivatization; indicating that
6	the enzyme had stereoselectivity for 3-amino-3-(3-pyridyl)propionate.
7	Extremely low activity was observed with L-glutamine, L-aspartate, L-asparagine,
8	β -alanine, and L-alanine, and the relative activities for these compounds were 1.1%,
9	0.57%, 0.30%, 0.28%, and 0.07% of the enzymatic activity for racemic β -phenylalanine,
10	respectively. The following compounds were judged to be inactive as amino-donors:
11	other proteinogenic amino acids (L-proline and L-glutamate were not tested), D-isomers
12	of 18 proteinogenic amino acids (D-proline and D-glutamate were not tested),
13	DL-β-aminoisobutyrate, taurine, L-phenylglycine, D-phenylglycine,
14	L-homophenylalanine, D-homophenylalanine, (<i>R</i>)-2-amino-2-phenylethanol,
15	(S)-2-amino-2-phenylethanol, (R) -1-phenylethylamine, (S) -1-phenylethylamine,
16	(R)-1-(1-naphthyl)ethylamine, (S) -1-(1-naphthyl)ethylamine, and,
17	1,2,3,4-tetrahydro-1-naphthylamine.
18	As an amino-acceptor, various oxo acids and ketones were tested at concentrations of

1	10 mM with 10 mM racemic β -phenylalanine as the amino-donor. Oxaloacetate,
2	pyruvate, and 1,3-acetonedicarboxylate served as the amino-acceptors. The relative
3	activities were 9.9%, 3.8%, and 2.0% of the enzymatic activity for 2-oxoglutarate,
4	respectively. The following compounds were judged to be inactive as
5	amino-acceptors: 3-phenylpyruvate, 2-oxovarelate, ketomalonate, 4-oxovalerate,
6	benzoylacetate, 2-heptanone, acetophenone, and ethyl benzoylacetate.
7	
8	3.5. Effects of pH and temperature on the activity and stability of the enzyme
9	The optimum pH for the enzymatic activity (β -phenylalanine:2-oxoglutarate
10	transamination) was measured at 28°C under various pH conditions. The enzyme
11	showed a maximum activity at pH 9.5 (Fig. 3A). The optimum temperature for the
12	enzymatic activity was assayed at pH 9.0 at various temperatures (0-65°C). The
13	enzyme activity was found to be maximally active at 50°C (Fig. 3B).
14	The enzyme was incubated at 28°C for 30 min under various pH conditions, and then
15	a sample of the enzyme solution was taken and the residual activity
16	$(\beta$ -phenylalanine:2-oxoglutarate transamination) was measured. The enzyme was
17	stable in the pH range 6.0-8.0 (Fig. 4A). The enzyme was preincubated for 30 min at
18	pH 9.0 under various temperatures (0-65°C) and the residual activity was assayed.

The enzyme was inactivated entirely by the incubation above 60°C (Fig. 4B).

3.6. Effects of inhibitors and metal ions on the enzymatic activity

4	The effects of various inhibitors and metal ions (2 mM) on the enzyme activity were
5	examined by the standard methods. The enzyme activity was inhibited by Ag^+ , Hg^{2+} ,
6	Zn^{2+} , dinitrophenol, and hydroxylamine and decreased to 0%, 2.4%, 52%, 46%, and
7	23%, respectively, of the original activity. The other metal ions tested (Li^+ , Na^+ , K^+ ,
8	Rb ⁺ , Cs ⁺ , Be ²⁺ , Mg ²⁺ , Ca ²⁺ , Sr ²⁺ , Ba ²⁺ , Fe ²⁺ , Co ²⁺ , Ni ²⁺ , Sn ²⁺ , Pb ²⁺ , Fe ³⁺ , Al ³⁺ , and Cr ³⁺)
9	were judged not to affect the enzymatic activity. Metal ion chelaters
10	(ethylenediamine- <i>N</i> , <i>N</i> , <i>N</i> ', <i>N</i> '-tetraacetic acid, 8-hydroxyquinoline, and
11	o-phenanthroline), serine protease inhibitors (phenylmethane sulfonylfluoride and
12	diisopropylphosphofluoride), and sulfhydryl inhibitors (5,5'-dithiobis-2-nitrobenzoate,
13	iodoacetate, p-chloromercuribenzoate, and N-ethylmaleimide) were judged to have no
14	significant influence on the activity.
15	
16	3.7. Analysis of amino acid sequences
17	The amino acid sequences of internal peptides were analyzed and three parts of the
18	amino acid sequence were determined. These amino acid sequences were named IP1

1	(54 residues), IP2 (67 residues), and IP3 (63 residues) and were applied to homology
2	searches using BLAST (Fig. 5). All three amino acid sequences showed homology
3	with aminotransferases and, especially, β - aminotransferases and
4	glutamate-1-semialdehyde 2,1-aminomutases were most homologous with them
5	[20-23].
6	
7	4. Discussion
8	
	As a result of the present investigation of substrate specificity, β-phenylalanine and
9	As a result of the present investigation of substrate specificity, β-phenylalanine and 3-amino-3-(3-pyridyl)propionate were found to serve as good amino-donors for the
9 10	As a result of the present investigation of substrate specificity, β -phenylalanine and 3-amino-3-(3-pyridyl)propionate were found to serve as good amino-donors for the (<i>S</i>)-BAT-catalyzed reaction. The enzyme catalyzed transamination with the other
9 10 11	As a result of the present investigation of substrate specificity, β -phenylalanine and 3-amino-3-(3-pyridyl)propionate were found to serve as good amino-donors for the (<i>S</i>)-BAT-catalyzed reaction. The enzyme catalyzed transamination with the other amino-donors such as L-glutamine, L-aspartate, L-asparagine, β -alanine, and L-alanine,

that the enzyme recognizes the 3-amino-3-aromatic propionate-skeleton specifically.

This result also indicated that the enzyme is different from already-known

ω-aminotransferases or aromatic aminotransferases. With regard to stereoselectivity

for amino-donors, the enzyme was found to catalyze (S)-enantiomer selective

transamination in the reactions with racemic \beta-phenylalanine and also showed

enantioselectivity in transamination with racemic 3-amino-3-(3-pyridyl)propionate.

Concerning the amino-acceptor, several oxo acids, such as 2-oxoglutarate, pyruvate, oxaloacetate and 1,3-acetonedicarboxylate, served as substrates. In particular, 2-oxoglutarate was found to be the most suitable substrate as the amino-acceptor. There are several differences between the (S)-BAT from Variovorax sp. JH2 and β-aminotransferases from *Mesorhizobium* sp. in terms of substrate specificity. The β-aminotransferases of *Mesorhizobium* sp. acted on some amino acids that are not the substrate of (S)-BAT, such as β -alkyl- β -amino acids, aromatic α -amino acids (phenylalanine and tyrosine), aliphatic α -amino acids (valine and leucine) and also acted on arylamines, such as 1-phenylethylamine. Furthermore, both 2-oxoglutarate and pyruvate have been reported to serve as good amino-acceptors for β -aminotransferases These results indicated that β -aminotransferases from from *Mesorhizobium* sp. *Mesorhizobium* sp. is an intermediate enzyme between β -aminotransferases, ω -aminotransferases, and α -amino acid aminotransferases. These facts led to the conclusion that the (S)-BAT from Variovorax sp. JH2 is a novel aminotransferase highly specific to β -aryl- β -amino acids with preference for 2-oxoglutarate as an amino-acceptor.

As the result of homology searches of the amino acid sequence, (S)-BAT showed the
highest homology with glutamate 1-semialdehyde 2,1-aminomutases as well as β-

aminotransferases [6]. These enzymes were classified into aminotransferase subgroup II, containing acetylornithine, ornithine, ω -amino acid, 4-aminobutyric acid, and diaminopelargonate aminotransferases, and it is possible that (S)-BAT is classified as a member of the subgroup [24, 25]. It was reported that glutamate 1-semialdehyde 2,1-aminomutases act in porphyrin and chlorophyll metabolism in vivo, however, (S)-BAT acts in β -phenylalanine assimilation of Variovorax sp. JH2 [25, 26]. In the β-aminotransferases of *Mesorhizobium* sp., Arg43 and Arg404 supposed to be important residues to recognize a carboxylate group [6]; however, the corresponding arginine residues are not found in (S)-BAT (Fig. 5), which indicates the difference in these enzymes. Above all, it was clarified that (S)-BAT has a unique primary structure in

11 Above all, it was clarified that (*S*)-BAT has a unique primary structure in 12 aminotransferase subgroup II and useful for optically active β -aryl- β -amino acid 13 production from racemic substrates by decomposition of specific enantiomer through 14 transamination in the presence of 2-oxoglutarate.

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1 Figure legends

Fig. 1 SDS-PAGE of the purified enzyme. Lane A is 1 mg of the purified enzyme. Lane B is molecular mass standards (from top): phosphorylase b (97 kDa), BSA (66 kDa), aldolase (42 kDa), carbonic anhydrase (30 kDa), trypsin inhibitor (20 kDa), and lysozyme (14 kDa). Fig. 2 Time-course of the transamination with purified (S)-BAT. Concentrations of β -phenylalanine (BPA) (\Box) and benzoylacetic acid (\bullet) are shown. Optical purity of β -phenylalanine in the 1 h reaction is shown with an arrow. Fig. 3 Effects of pH and temperature on the activity of (S)-BAT. The effects of pH (A) and temperature (B) on the enzymatic activity (β -phenylalanine:2-oxoglutarate transamination) are shown. Buffers (200 mM) used for pH adjustment are as follows: acetate/sodium acetate buffer (\Box) for pH 5.0-6.0, potassium phosphate buffer (\blacktriangle) for pH 6.0-8.0, Tris/HCl buffer (•) for pH 7.5-10.0, borate/NaOH buffer (°) for pH 10.0-10.5, Na₂PO₄/NaOH buffer (■) for pH 10.5-12.0.

1	Fig. 4 Effects of pH and temperature on the stability of (S)-BAT. The effects of pH
2	(A) and temperature (B) on the stability are shown. Buffers (0.5 M) used for pH
3	adjustment are as follows: acetate/sodium acetate buffer (D) for pH 4.0-6.0, potassium
4	phosphate buffer (▲) for pH 6.0-8.0, Tris/HCl buffer (●) for pH 8.0-10.0, borate/NaOH
5	buffer (\circ) for pH 9.0-11.0, Na ₂ PO ₄ /NaOH buffer (\blacksquare) for pH 11.0-12.0.
6	
7	Fig. 5 Amino acid sequence alignments of IP1-3 and proteins homologous to them.
8	Identical amino acids to those in the sequences of IP1-3 are marked in gray. MES, β -
9	aminotransferases from Mesorhizobium sp.; CNE, glutamate-1-semialdehyde
10	2,1-aminomutase from Cryptococcus neoformans; SPO, glutamate-1-semialdehyde
11	2,1-aminomutase from Schizosaccharomyces pombe; BHA, glutamate-1-semialdehyde
12	2,1-aminomutase from Bacillus halodurans; FRA, aminotransferase class-III from
13	Frankia sp.; CDI, glutamate-1-semialdehyde 2,1-aminomutase from Corynebacterium
14	diphtheriae.
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Figure1 Click here to download high resolution image



В









Figure5
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	Total protein	Total activity	Specific	
Step	(mg)	(U)	activity (U/mg)	Yield (%)
Cell-free extract	1260	109	0.087	100
Ammonium sulfate	410	83	0.20	76
fractionation			0.20	
DEAE sephacel	170	42	0.24	38
Phenyl sepharose	4.0	3.4	0.85	3.1
Mono Q HR 5/5	0.1	0.14	1.5	0.13

Table 1. Purification of (S)-BAT

(A)		
Amino donor	Structural formula	Relative activity (%)
Racemic BPA	соон	100
Racemic 3-amino- 3-(3-pyridyl)propionic acid	COOH	56.0
L-Glutamine	H2N OC	1.1
L-Aspartate	HOOC COOH	0.57
L-Asparagine	H2NOC COOH	0.30
β-Alanine	NH 2 COOH	0.28
L-Alanine	NH 2	0.07
(B)		
Amino accepter	Structural formula	Relative activity (%)
2-Oxoglutarate	ноос	100
Oxaloacetate	ноос Соон	9.9
Pyruvate	Соон	3.8
1,3-Acetonedicarboxylate	ноос	2.0

Table 2. Substrate specificity of (S)-BAT for amino donor (A) and for amino acceptor (B)