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Letters

Citation: Appl. Phys. Lett. **98**, 172901 (2011); doi: 10.1063/1.3583460 View online: http://dx.doi.org/10.1063/1.3583460 View Table of Contents: http://apl.aip.org/resource/1/APPLAB/v98/i17 Published by the American Institute of Physics.

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Electronic and structural properties of the oxygen vacancy in BaTiO₃

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(Received 16 February 2011; accepted 6 April 2011; published online 26 April 2011)

The electronic and structural properties of the oxygen vacancy (V_0) in cubic BaTiO₃ are studied using first-principles calculations based on a hybrid Hartree–Fock density functional approach. Our calculations identify the double shallow donor behavior of V_0 , indicating its contribution to the *n*-type conductivity. In addition, a metastable configuration is found, which shows an off-symmetric atomic structure around V_0 in conjunction with deep localized electronic states in the band gap. Based on the identified characteristics of V_0 , the previous experimental and theoretical findings are explained. © 2011 American Institute of Physics. [doi:10.1063/1.3583460]

Perovskite titanates have been extensively investigated owing to its abundant physics and great applications.¹⁻³ Among them, BaTiO₃ is a prototypical ferroelectric and used for thin-film capacitors.¹ Through lots of investigations, the native defect, especially, the oxygen vacancy (V_0) is known to significantly influence the physical properties of BaTiO₃. It thus needs to clarify the structural and electronic properties of $V_{\rm O}$, but there exists controversy. Experimentally, $V_{\rm O}^{+}$ was suggested to be stable in *n*-type specimens in wide range of temperature from electric conductivity measurements,4-6 whereas electron paramagnetic resonance (EPR) experiments have not clearly identified the physical feature of $V_0^{+,7-9}$ Likewise, several first-principles studies have been conducted about $V_{\rm O}$ using the supercell approach with the local density approximation (LDA) or generalized gradient approximation (GGA) and the embedded-cluster approach with MP2.^{10–12} However, the results are scattered because of the different approximations, simulation models, and/or postprocesses applied to reduce errors, e.g., the severe band-gap error of the LDA and GGA.

In this Letter, our aim is to accurately illustrate the electronic and structural properties of $V_{\rm O}$ in cubic BaTiO₃ using the Heyd–Scuseria–Ernzerhof (HSE06) hybrid functional.^{13,14} The use of HSE06 seems to be rationale, since this hybrid functional has accurately described defect physics in wide band-gap oxides.^{15–17} Through systematic calculations, we manifest that $V_{\rm O}$ plays a role as a double shallow donor under thermal equilibrium. Some experimental observations of $V_{\rm O}^+$ are suggested to be responsible for the metastable configuration of $V_{\rm O}$ with a deep, localized electronic state or other causes.

The calculations were performed using the projector augmented-wave method¹⁸ as implemented in the VASP code.¹⁹ The standard exchange mixing containing 25% of Hartree–Fock and 75% of the Perdew–Burke–Ernzerhof-GGA (Ref. 20) was employed in the HSE06 functional. The effects of spin polarization were considered for supercells with $V_{\rm O}$. The electronic wave functions were described using a plane wave basis set with an energy cutoff of 400 eV. A 135-atom supercell and the Γ -only k-point sampling were used. The atomic coordinates were relaxed until the Hellmann–Feynman force acting on each atom was reduced to less than 0.05 eV ${\rm \AA}^{-1}.$

The formation energy of $V_{\rm O}$ is evaluated as

$$\Delta E_f(V_{\rm O}^q) = E_{\rm T}(V_{\rm O}^q) - E_{\rm T}({\rm H}) + \mu_{\rm O} + q(E_{\rm VBM} + \varepsilon_{\rm F}), \qquad (1)$$

where $E_{\rm T}(V_{\rm O}^q)$ is the total energy of a supercell with one $V_{\rm O}$ in charge state q, and $E_{\rm T}({\rm H})$ is the total energy of the host supercell. μ_0 is the oxygen chemical potential, varying between the O-rich limit $[\mu_{O}=(1/2)\mu_{O_2(\text{molecule})}]$ and the O-poor limit $(\mu_{Ti} = \mu_{Ti(bulk)})$ and $\mu_{Ti} + \mu_{O} = \mu_{TiO(bulk)})$. These equilibrium conditions are taken on the basis of previous considerations.²¹ $E_{\rm VBM}$ and $\varepsilon_{\rm F}$ are the valence band maximum (VBM) with a potential alignment correction²² and the Fermi level measured from the VBM, respectively. As listed in Table I, HSE06 reproduces the lattice constant and band gap of $BaTiO_3$ much better than GGA and GGA+U. However, the HSE06 band gap is still slightly smaller than the experimental value. To complement the small band-gap error, a post-correction was applied via a rigid shift in the conduction band minimum (CBM) and defect-induced states with CB characteristics.^{3,22}

Figures 1(a) and 1(b) show the formation energy of $V_{\rm O}$ in BaTiO₃ as a function of the Fermi level. $V_{\rm O}^{2+}$ is the most stable for most position of the Fermi level, and the formation energies for three charge states become comparable with each other when the Fermi level is close to the CBM. These characteristics indicate that $V_{\rm O}$ behaves as a double shallow donor, which releases two electrons. At the O-rich limit, the formation energy of $V_{\rm O}^{2+}$ when the Fermi level is located at the CBM, $\Delta E_f^{\rm CBM}(V_{\rm O}^{2+})$, is 5.31 eV. The formation of $V_{\rm O}$ is therefore unlikely under O-rich conditions. Moving to the O-poor limit, the formation energy is lowered by 4.66 eV. A low $\Delta E_f^{\rm CBM}(V_{\rm O}^{2+})$ value of 0.65 eV indicates a substantial

TABLE I. Lattice constant (*a*) and band gap (E_g) of cubic BaTiO₃ obtained using the HSE06, GGA+U (U_{eff} =4.36 eV), (Refs. 3 and 23) and GGA. Experimental values are also shown.

	HSE06	GGA+U	GGA	Experiment
a (Å)	3.993	4.064	4.036	3.996 ^a
$E_{\rm g}~({\rm eV})$	3.11	2.14	1.67	3.4 ^b

0003-6951/2011/98(17)/172901/3/\$30.00

^aReference 24.

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FIG. 1. (Color online) Formation energy of $V_{\rm O}$ in BaTiO₃ as a function of the Fermi level under (a) the O-rich and (b) the O-poor limits. The charge states of 0, +, and 2+ correspond to the slopes. The squared wave functions $(|\psi_e|^2)$ of the electron-occupied states in the conduction band for (c) $V_{\rm O}^0$ and (d) $V_{\rm O}^*$. $|\psi_e|^2$ are illustrated on the Ti–O plane including $V_{\rm O}$, which are the cross-sections of isosurfaces at >0.007 Å⁻³.

concentration of $V_{\rm O}$ under O-poor conditions. As illustrated in Fig. 1(c), $V_{\rm O}^0$ creates a hydrogenic effective-mass electronic state occupied by two electrons. $V_{\rm O}^+$ behaves very similarly, apart from the fact that the state is occupied by one electron in this case [Fig. 1(d)]. This feature is consistent with the formation-energy diagram, indicating nearly the same formation energies for $V_{\rm O}^0$, $V_{\rm O}^+$, and $V_{\rm O}^{2+}$ at the Fermi level near the CBM.

Additionally, we find a metastable configuration of $V_{\rm O}$ with an off-symmetric atomic structure. The metastable $V_{\rm O}$ possesses C_{4v} point-group symmetry and the distances between the two nearest Ti atoms and $V_{\rm O}$ are asymmetric: for V_{Ω}^{0} , 1.85 Å, and 2.47 Å, corresponding to the Ti displacements of 0.15 Å (inward) and 0.47 Å (outward) from their ideal positions, respectively. As shown in Fig. 2, the electron-occupied states at 0.28 and 0.66 eV below the CBM are found in the case of the neutral charge state. The former is characterized by a host CBM-like orbital, while the latter, i.e., the deeper state, exhibits squared wave function $(|\psi_e|^2)$ localized in the vicinity of $V_{\rm O}$. In the + charge state, an occupied localized state of 0.57 eV below the CBM is created, similar to the deeper state in $V_{\rm O}^0$. However, it should be emphasized here that these configurations are energetically less favorable by 0.83 eV for $V_{\rm O}^0$ and 1.19 eV for $V_{\rm O}^+$ than the most stable configurations shown in Fig. 1.²⁵ Therefore, they are metastable and unlikely to form under thermal equilib-



FIG. 2. (Color online) Electron-occupied single-particle states below the CBM and the corresponding $|\psi_e|^2$ in metastable configurations of (a), (b) V_0^0 , and (c) V_0^+ . These states are for the majority spin component and any states are not found for the minority spin component. $|\psi_e|^2$ are illustrated on the Ti–O plane including V_0 , which are the cross-sections of isosurfaces at >0.007 Å⁻³ for (a) and >0.03 Å⁻³ for (b) and (c).

rium; however, they may form under specific growth conditions or treatments.

In previous supercell approaches using the LDA or GGA, $V_{\rm O}$ has been found to be a double shallow donor^{10,11} as we identified through the present hybrid functional calculations. Note, however, that the LDA and GGA in conjunction with the band-gap correction significantly overestimate ΔE_f (by ~3 eV), compared with the present hybrid functional results. Similar pictures were recently found in other oxides.¹⁵⁻¹⁷ In contrast to the supercell approaches, an embedded-cluster approach with MP2 suggested a localized characteristic,¹² which is similar to the metastable configuration shown in Fig. 2. It is noteworthy that the LDA and GGA do not stabilize the localized configuration presumably due to the well-known delocalization error. The embeddedcluster approach cannot treat delocalized states and therefore it cannot predict the relative stability of the localized and delocalized configurations. Our results indeed show the metastability of the localized configuration with a sizable energy difference from the stable configuration.

On the basis of our results, we now discuss the experimental observations relevant to $V_{\rm O}$ in BaTiO₃. In spite of the fact that many EPR measurements have been performed, the physical features and even existence of $V_{\rm O}^+$ are unclear.⁷⁻⁹ Several EPR measurements assigned their signals to a para-magnetic center, $Ti^{3+} - V_O^{2+}$, composed of V_O^{2+} with the nearest Ti⁴⁺ ion trapping one electron. Considering its total charge, this paramagnetic center corresponds to V_0^+ . However, the g-factors assigned to this center are very much different from each other, and these are observed at very low temperature $(\sim 4 \text{ K})$ to that higher than the Curie temperature $(>T_C)$ \sim 390 K). Therefore, the paramagnetic center is present in not only the tetragonal phase below the Curie temperature but also the cubic phase. Theoretically, it has been reported that $V_{\rm O}$ in the tetragonal and cubic phases has a similar characteristic in its electronic structure.²⁶ A recent experiment suggested another $Ti^{3+} - V_O^{2+}$ center in thin films.⁸ Laguta et al.8 claimed that the g-factor comes from the occupied $e_{\rm g}(d_{3z^2-r^2})$ state as the ground state of $V_{\rm O}^+$, as found in an embedded-cluster study,¹² and thereby their EPR signal originates from $V_{\rm O}^{+}$. Another EPR experiment predicted that $V_{\rm O}^{\rm +}$ possesses C_{4v} symmetry and its ground state becomes t_{2g} if an acceptor-like impurity is located nearby it.9

As seen in Figs. 1(c) and 1(d), however, V_0 has spatially delocalized electrons under thermal equilibrium. Thus, the stable V_{Ω}^{+} configuration cannot explain the paramagnetic centers mentioned above. Differently, the metastable configuration of $V_{\rm O}^{+}$ has C_{4v} symmetry and an in-gap state with a $d_{3z^2-r^2}$ and $d_{x^2-v^2}$ mixed characteristic $|\psi_e|^2$, mainly localized in the vicinity of V_{Ω} and one of its first nearest Ti atoms [Fig. 2(c)]: the electron-population of $|\psi_e|^2$ on the first nearest Ti site is three times higher than that on the second nearest Ti site. The metastable V_{Ω}^{+} is therefore possibly responsible for the observed EPR signals. In fact, the presence of Ti³⁺ $-V_{\Omega}^{2+}$ centers has been suggested in polycrystalline films synthesized using rf sputter deposition,⁸ multidomain crystals after reduction under H₂ atmosphere,⁹ and polycrystals obtained by thermal decomposition of BaTiO(C_2O_4)₂·4H₂O precursors.' On the other hand, there exists no report on the observation of the $Ti^{3+} - V_O^{2+}$ center in single crystals. Laguta et al.⁸ mentioned that the observation of the paramagnetic center in thin films but not in bulk crystals may be attributed to the specific conditions under which the thin films are processed. These lead us to a suggestion that the EPR signal of $V_{\rm O}^+$ results from the metastable configuration of $V_{\rm O}$.

According to the O₂ partial pressure dependence of the electric conductivity for polycrystalline BaTiO₃ specimens, $V_{\rm O}$ is fully ionized at high temperatures above ~873 K.⁵ An estimated ionization energy for $V_{\rm O}^+ \rightarrow V_{\rm O}^{2+}$ is ~0.5 eV, while for $V_{\rm O}^0 \rightarrow V_{\rm O}^+$, the ionization energy is only a few hundredths of electron volt.⁶ The experimentally assigned high ionization energy of ~0.5 eV is incompatible with the double shallow donor nature of $V_{\rm O}$, predicted in our study. Therefore, the deep donor level may be responsible not for isolated $V_{\rm O}$ but for other defects, complexes, or impurities.

Very recently, a relaxorlike state, which is typically caused by the formation of local polar nanoregions above paraelectric -> ferroelectric transition or the Burns temperatures, was found for the high-temperature paraelectric cubic phase in BaTiO₃ single crystals.²⁷ It is often understood that the observed relaxor state in nominally pure samples can be related to some defects that induce the inversion symmetry breaking. According to our results, thermodynamically the most stable configuration of $V_{\rm O}$ is symmetric, and thus is unlikely to be a cause. However, the metastable V_0 breaks the inversion symmetry, and thereby may produce the polarization when it can form with a high enough concentration. Lately, another perovskite titanate, SrTiO₃, was reported to exhibit a ferroelectric state, in the nonstoichiometric bulk and thin films with O-deficiency²⁸ and a composition of $Sr/Ti < 1.^{29}$ It was theoretically proposed that Ti antisite defects off-centered from the Sr site can lead to the polarization owing to their local electric dipoles and low formation energies under Ti-rich conditions.³ Analogically, one can speculate that Ti antisites, composed of a Ti atom off-centered from the Ba site, are also likely causes of the observed relaxor-like state in BaTiO₃.

To summarize, we have investigated V_0 in BaTiO₃ using the hybrid density-functional calculations. V_0 is a double shallow donor under thermal equilibrium and hence can contribute to the *n*-type conductivity. A metastable asymmetric atomic-configuration also exists and its formation probably depends on specific growth conditions. The metastable configuration can be responsible for the observed EPR signals in non-single crystalline BaTiO₃.

This work was supported by Grants-in-Aid for Scientific Research (A), Young Scientists (B), Scientific Research on Priority Areas (No. 474), and a Global COE Program.

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