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<td>Author(s)</td>
<td>Iwahara, Naoya; Sato, Tohru; Tanaka, Kazuyoshi</td>
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Effect of Coulomb interactions on the vibronic couplings in C\textsubscript{60}

Naoya Iwahara, Tohru Sato, and Kazuyoshi Tanaka

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Effect of Coulomb interactions on the vibronic couplings in C\textsubscript{60}

Naoya Iwashara, Tohru Sato,\textsuperscript{a)} and Kazuyoshi Tanaka

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Vibronic couplings in C\textsubscript{60} anion are discussed on the basis of the concept of the vibronic coupling density (VCD) [T. Sato, K. Tokunaga, and K. Tanaka, J. Chem. Phys. 124, 024314 (2006); K. Tokunaga, T. Sato, and K. Tanaka, J. Chem. Phys. 124, 154303 (2006); and T. Sato, K. Tokunaga, and K. Tanaka, J. Phys. Chem. A 112, 758 (2008)]. The VCD analysis clearly reveals that the coupling to the bending \( h_5(2) \) mode is weaker than the coupling to the stretching \( h_8(7) \) and \( h_8(8) \) modes. For the vibronic couplings with the stretching modes, polarizations of the electron density difference on the bonds play a crucial role in the vibronic couplings. Such a polarized electron density difference appears as a result of the Coulomb interactions between the electrons in the lowest unoccupied molecular orbital and relevant doubly-occupied orbitals. © 2012 American Institute of Physics.

[I. INTRODUCTION]

The vibronic coupling\textsuperscript{1,2} in the C\textsubscript{60} anion has been intensively investigated\textsuperscript{3–14} because this coupling is essential in the Jahn-Teller effect. Thus, reliable vibronic coupling constants (VCC) are required to reveal the Jahn-Teller effect. Among the vibronic couplings, Gunnarsson et al. has concluded that the coupling for the bending \( h_5(2) \) mode is weaker than the coupling to the stretching \( h_8(7) \) and \( h_8(8) \) modes. For the vibronic couplings with the stretching modes, polarizations of the electron density difference on the bonds play a crucial role in the vibronic couplings. Such a polarized electron density difference appears as a result of the Coulomb interactions between the electrons in the lowest unoccupied molecular orbital and relevant doubly-occupied orbitals. \( \bullet \) 2012 American Institute of Physics.

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\textsuperscript{a)}Electronic mail: tsato@scl.kyoto-u.ac.jp.

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the ground electronic state couples to two $a_g$ and eight $h_g$ modes. The VCCs of C$_{60}$ for the $\Gamma$ modes ($\Gamma = a_g, h_g$) are defined by\cite{27,28,34}

$$V_{a_g(i)} = \langle \Psi_{T_0} | \left( \frac{\partial \hat{H}}{\partial Q_{a_g(i)}} \right) | \Psi_{T_0} \rangle, \quad (1)$$

$$V_{h_g(\mu)} = \langle \Psi_{T_0} | \left( \frac{\partial \hat{H}}{\partial Q_{h_g(\mu)}} \right) | \Psi_{T_0} \rangle, \quad (2)$$

where $\hat{H}$ is the electronic Hamiltonian which includes the Coulomb potential between nuclei. $| \Psi_{T_0} \rangle$ ($\lambda = x, y, z$) denotes the electronic ground state of C$_{60}$ at the geometry of the reference system $R_0$, and $Q_{T_0(\mu)}$ is the normal coordinate of the neutral C$_{60}$ for the $\Gamma(\mu)$ mode ($\Gamma \gamma = a_g, h_g\theta, h_g\varepsilon, h_g\xi, h_g\eta, h_g\zeta$). Among the three electronic states $\lambda = x, y, z$, we choose the $T_{1u}z$ electronic state because the electronic state couples only to the $h_g\theta$ modes.\cite{29} Hereafter we will present results for the $T_{1u}z$ electronic state and the $h_g\theta$ vibrational modes. The phase of the $a_g$ and $h_g\theta$ modes are chosen so that the VCCs are negative. Applying the Hellmann–Feynman theorem, we obtain

$$V_{a_g(i)} = \sum_{A=1}^{60} \left( \frac{\partial E(R)}{\partial R_A} \right)_{R_0} \cdot \frac{u_A^{(i)}}{\sqrt{M}}, \quad (3)$$

$$V_{h_g(\mu)} = \sum_{A=1}^{60} \left( \frac{\partial E(R)}{\partial R_A} \right)_{R_0} \cdot \frac{u_A^{(\mu)\theta}}{\sqrt{M}}, \quad (4)$$

where $E(R) = \langle \Psi_{T_0} | \hat{H} | \Psi_{T_0} \rangle$. $A$ denotes an atom in C$_{60}$, $R_A$ the Cartesian coordinates of the atom $A$, $R$ the set of all coordinates of $R_A$, $u_A^{(\mu)\theta}$ the displacement of the atom $A$ of the mass-weighted $\Gamma \gamma$ vibrational vector, and $M$ the mass of carbon. For the calculation of the VCCs, we used Eqs. (3) and (4).

The VCCs are decomposed into couplings for the atoms

$$V_{a_g(i),A} = \left( \frac{\partial E(R)}{\partial R_A} \right)_{R_0} \cdot \frac{u_A^{(i)}}{\sqrt{M}}, \quad (5)$$

$$V_{h_g(\mu),A} = \left( \frac{\partial E(R)}{\partial R_A} \right)_{R_0} \cdot \frac{u_A^{(\mu)\theta}}{\sqrt{M}}, \quad (6)$$

We call $V_{a_g(i),A}$ and $V_{h_g(\mu),A}$ atomic vibronic coupling constants (AVCC).

With the use of the VCCs $V_{a_g(i)}$ and $V_{h_g(\mu)}$, the stabilization energies for the $a_g$ modes and the $h_g$ modes are written as follows:

$$E_I = \sum_{i=1}^{2} \frac{V_{a_g(i)}^2}{2\omega_{a_g(i)}}, \quad (7)$$

$$E_{TT} = \sum_{\mu=1}^{8} \frac{V_{h_g(\mu)}^2}{2\omega_{h_g(\mu)}}. \quad (8)$$

B. Vibronic coupling density

The VCCs can be rewritten as the integral form using electronic and vibrational structures.\cite{20,21} For the $h_g$ modes, the VCCs are written as

$$V_{h_g(\mu)} = \int d^3r \eta_{h_g(\mu)}(r). \quad (9)$$

The vibronic coupling density $\eta_{h_g(\mu)}$ is defined by the product of the electron density difference $\Delta \rho$ and the potential derivative $v_{h_g(\mu)}$:

$$\eta_{h_g(\mu)}(r) = \Delta \rho(r) \times v_{h_g(\mu)}(r), \quad (10)$$

where the electron density difference $\Delta \rho$ is the difference between the electron density $\rho$ of C$_{60}$ and the electron density $\rho_0$ of C$_{60}$

$$\Delta \rho(r) = \rho(r) - \rho_0(r). \quad (11)$$

The potential derivative $v_{h_g(\mu)}$ is the derivative of the one-electron potential $u(r; R)$ acting on a single electron from all the nuclei of C$_{60}$ with respect to $Q_{h_g(\mu)}$:

$$v_{h_g(\mu)}(r) = \left( \frac{\partial u(r; R)}{\partial Q_{h_g(\mu)}} \right)_{R_0}. \quad (12)$$

$$= \sum_{A=1}^{60} \frac{-Z(r - R_{0A})}{|r - R_{0A}|^3} \cdot \frac{u_A^{(\mu)\theta}}{\sqrt{M}}, \quad (13)$$

where $u$ is

$$u(r; R) = \sum_{A=1}^{60} \frac{-Z}{|r - R_{0A}|}. \quad (14)$$

Here $Z = 6$, the atomic number of carbon atom. In the present treatment, core and valence electrons are not treated separately. They are taken into account in the electron density difference $\Delta \rho$. From the VCD, we obtain a local picture of the vibronic coupling constant.
C. Contribution of each orbital to the electron density difference

Within the Kohn-Sham method, $\Delta \rho$ is written as follows:

$$\Delta \rho(r) = \sum_{\nu, \lambda, \sigma} \left( N_{\nu 0} \psi_{\nu 0, \lambda \sigma}(r) \right)^2 - N_0 \left( \psi_{0, \lambda \sigma}(r) \right)^2,$$

where $\{ \psi_{0, \lambda \sigma}(r) \}$ is a set of Kohn-Sham orbitals of the C$_{60}$ anion, $\{ \psi_{\nu 0, \lambda \sigma}(r) \}$ a set of the orbitals of the neutral C$_{60}$. $\Lambda$ the irreducible representation, $\lambda$ the component of $\Lambda$, $\sigma$ the $\zeta$ component of the electron spin, $\nu$ the quantum number except for $\Lambda$, $\lambda$, and $\sigma$, and $N_{\nu 0, \lambda \sigma}$ and $N_0$ the occupation numbers of the orbitals $\psi_{\nu 0, \lambda \sigma}$ and $\psi_{0, \lambda \sigma}$, respectively. The Kohn-Sham orbitals are chosen to be real. In order to evaluate the contribution of each orbital to the electron density difference, we expand $\Delta \rho$ in terms of $\{ \psi_{0, \lambda \sigma}(r) \}$:

$$\Delta \rho(r) = \sum_{\nu, \lambda, \sigma} A(\nu \lambda \sigma) \psi_{0, \lambda \sigma}^2(r)$$

$$+ \sum_{\nu, \lambda', \sigma'} B(\nu \lambda \lambda') \psi_{\nu, \lambda' \sigma'}(r),$$

where $A(\nu \lambda \sigma) = \sum_{\nu, \lambda} N_{\nu 0} \langle \psi_{\nu 0, \lambda \sigma} \psi_{\nu 0, \lambda' \sigma} \rangle^2 - N_0 \langle \psi_{0, \lambda \sigma} \psi_{0, \lambda' \sigma} \rangle,$

$$B(\nu \lambda \lambda') = \sum_{\nu, \lambda} N_{\nu 0} \langle \psi_{\nu 0, \lambda \sigma} \psi_{\nu 0, \lambda' \sigma} \rangle$$

$$\times \langle \psi_{\nu 0, \lambda' \sigma} \psi_{\nu 0, \lambda \sigma} \rangle.$$  

III. METHOD OF CALCULATION

To obtain the electronic states, the equilibrium geometry $R_0$, and the vibrational modes, DFT calculations were performed. In the DFT calculations, we employed the hybrid functional B3LYP (Ref. 30) with a triple-zeta basis set 6-311G(d). The gradients of $E(R)$ were calculated analytically for both a symmetrized wavefunction $|\Psi_\text{S}^T_{\text{C}}\rangle$ and a symmetry-broken wavefunction $|\Psi_\text{BS}^I_{\text{C}}\rangle$ which comes from the $\zeta$ component of the $T_{\text{C}}$ state. The vibrational modes employed are symmetrized. For the DFT calculations, we used GAUSSIAN 09. The VCCs and VCDs were calculated using our code.

IV. RESULTS AND DISCUSSION

A. Calculation of the vibronic coupling constants

Experimental and present theoretical VCCs are shown in Table I. Present theoretical couplings with the symmetry-broken wavefunction $|\Psi_\text{BS}^I_{\text{C}}\rangle$ (Table I, BS) are in line with our couplings derived from the experimental PE spectrum of Wang et al. On the other hand, in comparison with the experimental VCCs, some of the VCCs calculated with the symmetrized wavefunction $|\Psi_\text{S}^T_{\text{C}}\rangle$ (Table I, S) are qualitatively different; the VCC for the $h_4(4)$ mode is overestimated and the VCCs for the $h_4(5)$, $h_4(6)$, and $h_4(7)$ are underestimated. As we will discuss in Sec. IV B, this discrepancy occurs because the orbital relaxation is not fully included in $|\Psi_\text{S}^T_{\text{C}}\rangle$. The effect of orbital relaxation is not included correctly as long as we use a symmetrized Slater determinant. Hereafter, we concentrate on the VCCs calculated with $|\Psi_\text{BS}^I_{\text{C}}\rangle$ and analyze them.

Our theoretical VCCs for the $h_4(2)$ mode is almost the half of Gunnarsson’s constant. Consequently the contribution of the $h_4(2)$ mode to the Jahn-Teller stabilization energy $E_{\text{JT}}$ (8) is not the strongest, which is qualitatively consistent with many theoretical studies.

In order to obtain the contribution to the VCCs from each atom, we calculated the AVCCs (Eqs. (5) and (6)). The AVCCs are tabulated in Table II. In general, the AVCCs for the $h_4(2)$ mode are smaller than those for the $h_4(7)$ and $h_4(8)$ modes. In the case of the stretching $h_4(7)$ and $h_4(8)$ modes, the AVCC is the largest at the C8 atom which is one of the 6:6 carbons (Fig. 1).

B. Vibronic coupling density analysis

The strengths of the vibronic couplings can be explained in terms of the VCD (9). The difference between the AVCCs for the $h_4(2)$ and the $h_4(8)$ modes (Fig. 2) is the largest at the C8 atom (Table II), thus we focus on these two modes in the vibronic coupling density analysis.

The $T_{\text{C}}$ lowest unoccupied molecular orbital (LUMO) of C$_{60}$ and the electron density differences of C$_{60}$ are obtained using the symmetry-broken wavefunction $|\Psi_\text{BS}^I_{\text{C}}\rangle$. The symmetrized wavefunction $|\Psi_\text{S}^T_{\text{C}}\rangle$ are shown in Fig. 3. The positive (gray) area of electron density difference $\Delta \rho$ originates from the LUMO density. One should note that $\Delta \rho$ (Fig. 3(b))

| Table I. Vibronic coupling constants to the $a_g$ and $b_g$ modes ($10^{-4}$ a.u.) and the stabilization energies (meV). The VCCs obtained from photoelectron (PE) spectra are taken from Refs. 5 and 16. Experimental frequencies are taken from the data of a Raman scattering measurement of fullerene. The VCCs with BS and S were calculated using the symmetry-broken wavefunction $|\Psi_{\text{BS}}^T_{\text{C}}\rangle$ and symmetrized wavefunction $|\Psi_{\text{S}}^T_{\text{C}}\rangle$, respectively. |
|---|---|---|---|---|
| Freq. (cm$^{-1}$) | PE spectra | B3LYP |
| $a_g(1)$ | 496 | 0.537 | 0.15 | 0.301 | 0.308 |
| $a_g(2)$ | 1470 | 1.644 | 2.33 | 2.352 | 2.396 |
| $h_4(1)$ | 273 | 0.215 | 0.36 | 0.185 | 0.128 |
| $h_4(2)$ | 437 | 0.458 | 0.84 | 0.435 | 0.448 |
| $h_4(3)$ | 710 | 0.837 | 0.77 | 0.740 | 0.655 |
| $h_4(4)$ | 774 | 0.628 | 0.99 | 0.554 | 1.163 |
| $h_4(5)$ | 1099 | 0.992 | 1.15 | 0.758 | 0.211 |
| $h_4(6)$ | 1250 | 1.010 | 0.85 | 0.544 | 0.111 |
| $h_4(7)$ | 1428 | 2.283 | 1.78 | 2.096 | 1.783 |
| $h_4(8)$ | 1575 | 1.581 | 2.29 | 2.031 | 2.110 |
| $E_s$ | 15.9 | 17.0 | 19.2 | 19.9 |
| $E_{\text{JT}}$ | 57.7 | 88.2 | 49.4 | 51.0 |
| $E_s + E_{\text{JT}}$ | 73.6 | 105.2 | 68.6 | 70.9 |
TABLE II. Atomic vibronic coupling constants of the \(a_t\) and \(h_g\) modes (10\(^{-4}\) a.u.). The atomic label is shown in Fig. 1.

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<th>Atom</th>
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<th>(a_t(2))</th>
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</table>

has not only positive but also polarized negative (blue) area. This negative area appears due to the Coulomb repulsion between the electron in the \(t_{1u}\) LUMO and the other electrons in doubly-occupied orbitals. Polarization in \(\Delta \rho\) has been reported in other \(\pi\) conjugated systems.\(^{20}\)

In Fig. 4, the potential derivatives and the VCDs for the \(h_g(2)\) and \(h_g(8)\) modes are shown. The potential derivative of the bending \(h_g(2)\) mode \((\psi_{h_g(2)})\) is localized on each atom (Fig. 4 (a1)). On the C1, C3, C4, C7, C8, and C9 atoms, the displacements of the \(h_g(2)\) modes are perpendicular to the surface of the \(C_{60}\) cage, and the signs of \(\psi_{h_g(2)}\) of the inside and outside of the cage are opposite. The VCD \(\eta_{h_g(2)}\) is localized on each atom (Fig. 4 (a2)), and hence the integral of the VCD is canceled around the atom. In the case of the stretching \(h_g(8)\) mode, on the other hand, \(\psi_{h_g(8)}\) is delocalized on the C-C bonds of the 6:6 carbons (Fig. 4 (b1)). Since both \(\Delta \rho\) and \(\psi_{h_g(8)}\) are particularly delocalized on the C8-C8’ bonds, \(\eta_{h_g(8)}\) is also delocalized on the bonds (Fig. 4 (b2)). Therefore, the AVCC of the C8 atom is the largest. The delocalized \(\eta_{h_g(8)}\) on the C8-C8’ bonds appears because of the polarized electron density difference. Therefore the polarization of \(\Delta \rho\) is crucial to evaluate the vibronic coupling constants. In fact, the orbital VCC \(\langle \psi_{0,1_L} | \psi_{h_g(8)} | \psi_{0,1_L} \rangle\) is \(-0.775 \times 10^{-4}\) a.u. where \(\psi_{0,1_L}\) is the LUMO of the neutral \(C_{60}\).

The electron density difference calculated using the symmetrized electronic state \(|\Psi_{t_{1u}}^S\rangle\) (Fig. 3(c)) has small negative density. Moreover, \(\Delta \rho\) has negative density on the C1, C2, and C9 atoms despite the small LUMO coefficients on these atoms. This result suggests that the Coulomb repulsion is not fully included in the electron density difference. The distribution of \(\Delta \rho\) obtained from \(|\Psi_{t_{1u}}^S\rangle\) is small on the C8-C8’ bonds, and the VCD for the \(h_g(8)\) is also small on the bonds.\(^{31}\) The discrepancy between the experimental VCCs and theoretical ones computed with \(|\Psi_{t_{1u}}^S\rangle\) originates from the electron density difference which does not include the orbital relaxation.

Some authors have explained the order of the VCCs of \(C_{60}\). Varma et al. considered that the contributions of the couplings for the stretching \(h_g(7)\) and \(h_g(8)\) modes to \(E_{JT}\) are larger than that of the coupling for the bending \(h_g(2)\) mode because a normalized displacement the distortion along the stretching mode gives larger change in energy than that along the bending mode does.\(^{3}\) However, since they explained the VCC using the force constant, their discussion will not always give a correct result. Hands et al. assumed that the vibronic coupling is strong when the overlap between the \(t_{1u}\) LUMO and the displacements of the atoms is large.\(^{3,8}\) They concluded that the coupling for the \(h_g(2)\) mode has larger contribution to \(E_{JT}\). Nonetheless, they did not take into account the derivative of the Coulomb potential between electron and nuclei \(v(r)\), and that leads to an opposite result to ours.

C. Electron-electron interactions in the electron density difference

In order to evaluate the contribution from each Kohn-Sham orbital to \(\Delta \rho\), we decomposed \(\Delta \rho\) into orbital densities (16). Figure 5 shows the contribution of each Kohn-Sham orbital to \(\Delta \rho\) \((\sum_{\sigma} \sum_{\lambda} \psi_{t_{1u}} A(L(\nu))d_{\lambda \sigma})\). Since one of the triply degenerate \(t_{1u}\) orbitals is occupied by an electron and the positive density of \(\Delta \rho\) mainly originates from the \(t_{1u}(7)\) orbital, the contribution of the LUMO \((t_{1u}(7))\) is the largest.

FIG. 3. (a) The LUMO, (b) the electron density difference \(\Delta \rho\) calculated with the symmetry-broken wavefunction \(|\Psi_{t_{1u}}^{BS}\rangle\), and (c) the electron density difference calculated with the symmetrized wavefunction \(|\Psi_{t_{1u}}^{S}\rangle\). The gray and blue indicate positive and negative, respectively. The isosurface value is 0.035 a.u. for LUMO and 0.0008 a.u. for electron density difference. The orientation of \(C_{60}\) is the same as that of Fig. 1.
However, as we have discussed in Sec. IV B, the contributions to $\Delta \rho$ from the occupied orbitals are also important. The coefficients for the doubly occupied orbitals are negative because of the Coulomb repulsions between the electron in the LUMO and the other electrons in occupied orbitals. Since these orbitals have finite orbital coefficients of $s$ atomic orbitals and in-plane $p$ atomic orbitals, $\Delta \rho$ has negative polarized density. Among the orbitals other than the LUMO, $h_g(10)$ and $t_{1u}(7)$ orbitals (Fig. 6) have the largest negative and positive coefficients, respectively. The densities of these orbitals can mix with each other because these orbitals have common representations in their direct products. Since the density of the $h_g(10)$ orbital overlaps the LUMO density, the Coulomb repulsion between the electrons occupied in the LUMO and the $h_g(10)$ orbitals is large. The Coulomb interactions give rise to an additional negative contribution in $\Delta \rho$. Due to the decrease of the $h_g(10)$ density, the density around C1, C2, and C8 atoms where there is small singly occupied molecular orbital (SOMO) density also decreases. The negative density around C1, C2, and C8 is canceled by the $t_{1u}(3)$ density. As a result, $\Delta \rho$ is small around the C1, C2, and C8 atoms where the density of the $t_{1u}(7)$ SOMO is small, whereas there are both positive and negative areas around the atoms where the SOMO density is large.

It is desired that the wavefunction of $C_{60}$ has the correct symmetry. Moreover, for the calculation of the VCCs, we must take into account the polarization of $\Delta \rho$. To fulfill both of them, a state-averaged post Hartree–Fock method could be one of the solutions. In the post Hartree–Fock calculation, we have to include not only frontier orbitals but also other orbitals which contribute to $\Delta \rho$.

V. CONCLUSION

In this work, we analyzed the order of the VCCs using the concept of the VCD. The coupling to the bending $h_g(2)$ mode is small because the displacement of the vibrational mode is perpendicular to the surface of the $C_{60}$ cage and the VCD is localized on atom. On the other hand, the coupling to the stretching $h_g(8)$ mode is large due to the delocalization of the VCD on the C8-C8’ bond. The polarization of the electron density difference is crucial for the couplings to the stretching modes. From the analysis of $\Delta \rho$, we found that the contribution from not only the frontier orbitals but also other orbitals are large.

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22The vibronic couplings for the \( a_\pi \) modes are taken from the data in Ref. 35.
23See supplementary material at http://dx.doi.org/10.1063/1.4709611 for the VCDs for the \( h_g(2) \) and \( h_g(8) \) modes calculated with \( |\Psi 1S\rangle \).