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Kyoto University
Iron content and ferritin contribution in nori.

The iron content and ferritin contribution in fresh, dried and toasted nori, *Pyropia yezoensis*.

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Abstract
Iron is one of the essential trace elements for humans. In this study, the iron contents in fresh, dried and toasted nori (*Pyropia yezoensis*) were analyzed. The mean iron content of fresh, dried and toasted nori were 21.7, 23.0, 26.2 mg/100 g (dry weight), respectively. These values were superior to other food of plant origin. Furthermore, most of the iron in nori was maintained during processing, such as washing, drying, and toasting. Then, the form of iron in fresh, dried and toasted nori was analyzed. As a result, an iron storage protein ferritin contributed to iron storage in raw and dried nori, although the precise rate of its contribution is yet to be determined, while ferritin protein cage was degraded in the toasted nori. It is the first report that verified the ferritin contribution to iron storage in such edible macroalgae with commercial importance.

Key words:
Ferritin, iron, red algae, macroalgae, nori
Introduction

Iron is an essential element in almost all living kingdoms. It is recommended that approximately 10-20 mg (9-12.5 mg for a male and 20 mg for a female) iron is taken for an adult from foods everyday (Data from Food and Agriculture Organization of the United Nations (FAO); http://www.fao.org/docrep/004/y2809e/y2809e0j.htm), although the values range largely depending on the availabilities of iron in various food matrices. In general, iron deficiency is one of the most serious nutritional problem that affect huge amounts of people world-wide. A well-known iron source with good availability for human is heme iron included in animal foodstuffs. In contrast, plant foodstuffs contain little amount of heme iron. However, they show various iron contents and some can be a candidate for nutritional iron source, which can rescue people suffering from an iron deficiency all over the world. One of the iron rich plant foodstuffs is legume seed, such as soybean, pea, and common bean. Soybean seed contains approximately 10 mg iron per 100g (dry weight). In legume plants, a major part of iron is stored in ferritin, a ubiquitous multimeric iron storage protein. Ferritin, which forms spherical hollow protein shell composed of 24 subunits, can deposit thousands of iron atoms as non-toxic and bio-available form in its inner cavity. It is suggested that this type of iron, deposited in ferritin, is an iron source with good bioavailability among plant-derived iron. Therefore, attempts for the biofortification of staple food crops using the ectopically introduced ferritin gene have been performed in the last decade. Recently, it is also demonstrated that algae is a good candidate for a bio-available iron source. Garcia-Casal et al. assayed the iron contents in several types of coastal macroalgae such as Ulva sp., Sargassum sp., Porphyla sp., and Gracilariopsis sp., which are classified as a green alga, a brown alga, and red algae (the last two species), respectively. As a result, they have high iron contents compared with other food materials derived from plants and all of them are showed good iron absorption rates for human. Traditionally, Japanese and people in other eastern countries have
consumed coastal macroalgae as an indispensable ingredient. Among the edible algal,
susabinori or simply called nori (*Porphyra yezoensis*), red macroalga, is one of the most
important species, which is extensively cultured on the surface of coastal region in
Japan, Korea, and China. The thalli of nori are harvested, cut out, dried, pressed to a
sheet, and distributed as dried products (http://www.fao.org/fishery/culturedspecies/Porphyra_spp/en). In Japan, dried nori
sheets are selected and divided to some grades depending on its quality, and finally most
of them are toasted before our consumption. As shown in this FAO statistics, the global
amounts of nori production was 691 million tons in 2012, and this value tends to be
increased in recent years, partly because a trend in a health-conscious diet of Western
countries. Although iron contents in nori were reported, 14, 17, 18) the reported iron
contents in nori are highly varied (2-90 mg iron/100g of dry matter). Therefore, the iron
contents of nori should be re-evaluated, e.g., the loss amounts of iron during the food
processing and the iron contents in the various grade of nori. Furthermore, a form of
stored iron in nori remains unclear.

With respect to the mechanism for iron storage in algae, researches have been
performed mainly on prokaryotic and eukaryotic unicellular microalgae. For examples,
ferritin has a critical role in iron storage and in the proliferation of cyanobacteria. 19)  
Similarly, ferritin plays an important role in iron storage in bloom-forming marine
pennate diatoms. 20) As a unicellular Chlorophyceae, Chlamydomonas has multiple
genes encoding plant-type ferritins, and these gene expression are differently regulated
against iron deficiency and sufficiency. 21) Ferritin is also detected in a unicellular red
alga, *Cyanidium caldarium*, 22) whereas it is suggested that another iron storage
mechanism does exist in unicellular brown alga, *Ectocarpus siliculosus* which has no
orthologue of ferritin in its genome. 23) On the other hand, knowledge of iron storage in
macroalgae is still very limited. Recently, we showed that a green macroalga *Ulva
pertusa* contains high amount of iron, and ferritin protein was detected in thalli of this
alga. 24) It was the first report that demonstrates ferritin contribution for macroalgal iron
storage. In the present study, we focused on iron in various forms and grades of nori, *P. yezoensis*. At the same time, we provide the first evidence that ferritin functions as an iron storage device in red macroalga, nori.

### Materials and methods

#### Algae

Nori (*P. yezoensis*) were cultured and harvested on the coastal region in Suma (Kobe, Japan) approximately 100 m offshore from the beach in December 2012 and 2013. After harvest, nori fronds were washed, cut out, pressed to sheets and dried at 40 °C for 3 hours using automatic nori manufacturing machine. Dried nori sheets were further heated and dried at 70 °C for 3 hours. The resultant nori sheets were designated as ‘dried nori’. Dried nori were selected and divided to some classes, such as the first, second, third, fourth grade, and out of the grade. These are the general procedures in processing of nori. Subsequently, dried nori sheets were toasted at 300 °C for 3 seconds to generate ‘toasted nori’ sheets. The samples harvested in December 2012 were all judged as the first grade, and samples were mainly used in this study unless specially mentioned.

#### Metal content measurement

A few kilograms of the harvested thalli of nori (*P. yezoensis*) were used as a non-processed sample, designated as ‘raw’. Raw nori thalli were washed with distilled water and lyophilize for three days, followed by air dried at 120 °C for 5 hours. Dried and toasted nori sheets were also air dried at 120 °C similarly. 0.5 g (dry weight) of each sample was wet-ashed with a solution of 14 ml of HNO₃ (60%, Nacalai tesque, Kyoto, Japan) and 0.5 ml of H₂O₂ (Wako, Tokyo, Japan) for 5 hours at 250 °C using a graphite block acid digestion system ‘Ecopre’ (Actac, Tokyo, Japan). The resulting digested solution of each sample was diluted to 40 ml with distilled water (HPLC grade, Nacalai tesque). Then, each sample was diluted 200 folds, followed by the measurement of iron.
concentration using an atomic absorption spectrophotometer (AAS) AA-6800 (Shimadzu, Kyoto, Japan) equipped with graphite furnace atomizer. The iron concentration of each sample was calculated by standard addition method and/or calibration curve method. To ensure the results of iron concentration analyses, the iron contents of dried nori samples harvested in 2012 winter were analyzed by Uv-visual (Uv/Vis) spectrophotometric analysis described below. Each 40 ml-sample was mixed with equal volume of 0.5% potassium ferrocyanide followed by the incubation at room temperature for 35 minutes in the dark place. Then, the absorbance at 690 nm was measured using a Uv/Vis spectrophotometer (UV-2550, Shimadzu). The concentration of each sample was determined from an average of 3-10 independent samples. ANOVA with Tukey-Kramer’s multiple comparison test was used to compare iron contents among various forms and grades of nori. Mean differences were considered significant at $P < 0.01$.

**Cloning of P. yezoensis ferritin cDNA (PyFer) and similarity analysis**

Total RNA was extracted using the Sepasol reagent (Nacalai) according to the manufacture’s instructions. Gene specific primer set, (5′-cgtcttacatgacgatg-3′) and (5′-ccagaaactgacatgggag-3′), was designed according to the sequence deposited in GenBank (Accession No. JX293834). The similarity analysis of the nucleotide sequence was carried out using BLAST2.0 at DDBJ. Theoretical isoelectric point ($p_I$) and molecular weight ($M_w$) of protein was calculated from deduced amino acid sequence of PyFer by pI/Mw tool [http://au.expasy.org/tools/](http://au.expasy.org/tools/). Multiple alignment of the ferritin sequences were performed by ClustalW [http://clustalw.ddbj.nig.ac.jp/top-j.html](http://clustalw.ddbj.nig.ac.jp/top-j.html) program. The phylogenetic tree was created by ClustalW and viewed by tree view program.

**Preparation of recombinant PyFer (rPyFer)**

To construct the pET vector based expression plasmid, cDNA fragment of PyFer
was amplified by the primer set (5′-gactaccatgccgctatgacgttttcg-3′) and (5′-gactaggatccattacgcctgggcatcctc-3′). The resulting PCR fragment was digested by NcoI and BamHI, and inserted to the NcoI/BamHI site of pET21d (Novagen, San Diego, CA). Protein expression was performed at 30 °C after isopropyl-β-D-thiogalactopyranoside induction. The cells were harvested and disrupted by sonication, followed by protein extraction with phosphate buffered saline containing protease inhibitor cocktail (Nacalai tesque). PyFer was expressed as soluble protein, and further purified by ammonium sulfate precipitation (50 % saturation), anion exchange chromatography by Q-sepharose (GE-healthcare, Piscataway, NJ) and size exclusion chromatography by Superdex 200pg 16/60 column (GE healthcare).

Detection of ferritin in various forms of nori

To enable specific detection of ferritin in nori, a rabbit polyclonal antibody was raised against the recombinant PyFer (rPyFer), and used for the western blot analysis. Prior to western blot analysis, SDS-PAGE were performed with or without a reducing reagent, that enable the detection of ferritin monomer and oligomer (24-mer) form, respectively. Generally, oligomeric ferritin can be detected by non-reducing-PAGE without heat denaturation, because ferritin oligomer has high stability against the treatment of heat and denaturant. Protein samples were prepared by extraction from 0.1 g of raw, dried, and toasted nori using 5 ml of PBS. So, each loaded sample contained soluble protein extracted from 200 µg of each form of nori. The horse-radish peroxidase labeled anti rabbit IgG (Promega) was used as the secondary antibody. Signal was visualized by using Chemilumi-one (Nacalai) and laser imager (LAS-4000, GE-healthcare).

Iron containing proteins were detected by non-reducing SDS-PAGE followed by Prussian blue staining of the gel. The gel separated above mentioned protein extracts was dipped in the mixture of 2% (w/v) Potassium Ferrocyanide and 2% (w/v) hydrochloride.
Results and Discussion

Iron contents of nori and their variation depending on the forms, grades, and years

The iron contents of raw, dried, and toasted nori, which were harvested in December, 2012 and judged as the first grade, are shown in Figure 1A. The mean values of iron contents of raw, dried, and toasted nori were 19.0, 22.6, and 26.2 mg/100 g dry weight, respectively (Fig. 1A). These values were measured by AAS analysis combined with standard addition method. According to the results, the iron contents of edible forms, dried and toasted nori, are quite high among all plant-derived food stuffs such as legumes, cereals, vegetables, and fruits (http://fooddb.mext.go.jp/). In addition, iron content of raw nori was comparable with that of U. pertusa, which had highest iron content among coastal green, red, and brown macroalgae, we analyzed in previous study.24) Together with the good bio-availability of iron in red alga Porphyra sp. (a native species of Venezuela),15, 16) nori can be considered as a good nutritional iron source. Since the dried and toasted nori samples were made from the raw thalli used in this study, these results reflect the true transition of iron concentration during the food processing.

To ensure these values in Fig. 1A, iron concentrations were further measured by two different methods, AAS combined with calibration curve method and colorimetric analysis using Uv/Vis spectrophotometer. The iron contents of dried nori were calculated to 22.6±1.21, 22.9±0.520, and 25.6±0.205 mg/100 g dry weight by AAS with standard addition method, AAS with calibration curve method, and Uv/Vis spectrophotometric method, respectively (Fig. 1B). These values indicated that AAS with calibration curve method, which is simpler than the standard addition method, can be adoptable for the measurements of iron contents of dried nori samples. Colorimetric analysis with Uv/Vis spectrophotometric method may slightly overestimate the values. Accordingly, the following iron content measurements were performed by AAS data with standard curve method using dried nori as samples. Dried nori sheets are usually
selected and divided to different grades depending on their appearance, such as colors and glaze, which reflect the quality of nori. The iron concentrations in various grades of dried nori are shown in Figure 1C. The samples were harvested in winter of 2012 and 2013. The mean iron contents of the first grade nori harvested in 2012 and 2013 were 22.9 and 27.0 mg/100g dry weight, respectively (Fig. 1C). The value of 2013 was significantly higher than that of 2012, suggesting that the iron content of nori product varies among the harvested years. Since the variation of the concentration of iron and other minerals among culture locations were pointed out by Yoshie et al., the values may vary among the years of harvest to some extent. The variation in iron content among different grades can be evaluated by the comparison between the first and fourth grade of 2013, and among the first, second and out of the grade of 2012 (Figure 1C). The comparisons in iron contents of different grades of nori have significant only when compared among grades harvested in a same year, because the grade is judged relatively every year. In both cases of 2012 and 2013 samples, the iron contents of the samples in the grade of good quality are higher in general. For examples, in 2012 samples, the mean iron content of 'out of the grade', which is the worse quality and not to be sold in the market, is 11.3 mg/100g dry weight, while that of the first and second grade were 22.9 and 15.9 mg/100g dry weight, respectively (Fig. 1C). Similarly, in 2013 samples, mean iron contents of the first and fourth grade 27.0 and 18.7 mg/100g dry weight, respectively (Fig. 1C). These results indicate that there is a clear correlation between iron contents and grades of nori, although previous study suggested no such tendency was observed. The reason for the data confliction is not clear. However, the iron content values of preceding studies tend to be lower (2-12 mg/100 g dry weight) than that of the present results, available database, and another study measuring the iron contents in various algae, including red macroalga (Porphyra sp.). To clarify the relationship between the grades of nori and iron contents, further encompassing study is required.
cDNA cloning of ferritin from P. yezoensis (PyFer)

The open reading frame of PyFer cDNA composed of 795 bp, which encoded 264 amino acid residues (Fig. 2A). The calculated pl and Mw were 4.52 and 28,135, respectively. The cDNA sequence has been submitted to the NCBI database (GenBank ID: AB918149). The amino acid sequence of PyFer shows 38% and 35% sequence identity with ferritin from *Ulva pertusa*24, 26) (green macroalga) and soybean ferritin subunit No.4.28) Further, PyFer and CmFer, a ferritin cDNA of unicellular red alga *Cyanidioschyzon merolae*,29) share 38.9% identity in their amino acid sequences. The phylogenetic tree of various ferritin cDNA indicates that PyFer belongs to the same cluster as the plant-type ferritins, including higher plants (soybean), green algae (*Ulva* sp., *Chlamydomonas reinhardtii*, and *Volvox carteri*) and red algae (*P. yezoensis* and *C. merolae*). This tree further supports the close relationship between PyFer and CmFer of a unicellular micro-redalga *C. merolae*, (Fig. 2B).

The putative secondary structure deduced from the three dimensional structure of algal26) and higher-plant ferritin28) was shown in Fig. 2A. According to the sequence alignment, the central part of PyFer forms 4-helix bundle, which is the conserved motif among all the identified ferritin from bacteria to mammals.30) The amino acid residues forming the iron oxidation site (Ferroxidase site)31, 32) are completely conserved in PyFer sequence (Fig. 2A). Similar to the other plant-derived ferritin, PyFer possesses a putative transit peptide (TP), which is responsible for the targeting to a plastid, although this putative TP has no similarity to other plant ferritin sequences. However, the TP sequences are very highly divergent.33, 34) Together with the fact that almost all the identified plant-type ferritin are targeted to a plastid, PyFer also can be considered as a chloroplast protein. The additional N-terminal region, which positions downstream of the TP, is generally designated as the extension peptide (EP) in plant type ferritin.3) This region is the N-terminus of mature plant ferritin and forms an α-helix unique to plant ferritins.26, 28) The EP region also presents in PyFer sequence in the downstream of putative TP region. Since the core region forming the 4-helix bundle is highly conserved
among PyFer, UpFer, and other plant type ferritins, PyFer is also supposed to forms
4-helix bundle subunit, which assembles to spherical 24-mer.

Detection of PyFer in Thallus and various forms of nori

To clarify whether ferritin (PrFer) contributes to the iron storage in nori, we
performed western blotting using specific antibody raised against recombinant PyFer.
First, we detected the monomeric PyFer after reduced SDS-PAGE. Figure 3B shows
that monomeric PyFer (approximately 27 kDa) are present in raw, dried, and toasted
nori. As described above, ferritin usually functions as an iron storage protein by forming
a multimeric protein shell composed of 24 subunits. To detect functional 24-mer of
PyFer in each form of nori, we performed another western blotting after non-reducing
condition SDS-PAGE without heat treatment. Figure 4A shows that PyFer were detected
as the bands, whose apparent molecular masses are much larger than 250 kDa marker
(Fig. 4A). Subsequently, to detect the iron containing protein, the gel of non-reducing
SDS-PAGE was treated by Prussian blue staining (Fig. 4B). In Figure 4B, iron
containing soybean ferritin purified from dry soybean seeds was loaded as positive
control of iron containing plant ferritin (Fig. 4B, lane C). Figure 4A and 4B show that
bands of multimeric and iron containing PyFer are detected in raw and dried nori,
whereas not in toasted one. Thus, these results suggest that PyFer functions as
multimeric (24-mer) iron storage protein in raw and dried nori, that is similar to the case
of higher plants. In contrast, there was no PyFer 24-mer in toasted nori, although
monomeric one was present in comparable amounts with raw and dried nori (Fig. 3B).
Thus, PyFer functioned as a multimeric iron storage protein, which was tolerant of the
drying process at 70°C. However, the precise rate of ferritin contribution in iron content
of nori is still unknown. Further characterizations of PyFer in various forms of nori are
required. As shown in Fig. 3A, few soluble proteins were detected in toasted nori,
indicating that almost soluble proteins were denaturated or aggregated during toasting
process at 300 °C. In contrast, PyFer was still detectable in toasted nori, even though it
wasn’t native 24meric form. Monomeric or dimeric ferritin may still contribute to iron binding in toasted nori, because the various forms of ferritin oligomer were seen in pea seed,\(^{35}\) and the iron containing dimeric ferritin was detected in soymilk (Masuda T. Unpublished data).

It has been demonstrated that ferritin plays a crucial roles in iron storage in higher plants\(^{35-37}\) and unicellular microalgae.\(^ {19,38-40}\) On the other hand, ferritin contribution to iron storage in macroalgae or seaweeds is explored only recently,\(^{24}\) although a ferritin gene is described as a stress-induced gene in a green seaweed \textit{Ulva} sp.\(^ {41}\) Algae have been developed sophisticated mechanisms for iron acquisition and storage, because one-third of the ocean is assumed to be deficient in iron due to its extremely low solubility in the oxidized state. Recently, genome project of nori, \textit{P. yezoensis}, has just been completed.\(^ {42-44}\) Hence, nori can be one of the candidates for a model in exploring a mechanism for iron acquisition and storage in macroalgae in addition to its economical and nutritional significance.

**Acknowledgements**

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**Figure legends**

Figure 1

Iron contents of various forms and grades of nori (\textit{P. yezoensis})

(A) Iron contents (mg/100 g dry weight) of raw, dried, and toasted nori. Iron contents of these nori samples harvested in 2012 winter were analyzed by AAS with standard
addition method. (B) Evaluation of the iron content values measured by AAS with
standard addition (AAS_1), AAS with calibration curve method (AAS_2), and Uv/Vis
spectrometry. (C) Iron contents of various grades of dried nori were assayed by AAS
combined with calibration curve method. Iron concentrations were determined from
three to eight independent experiments. Each data point represents the average of
replicates and bars indicate standard deviations. The differences in means were
compared with ANOVA with Tukey-Kramer’s multiple comparison test. Means not
sharing identical letters are significantly different (P < 0.01).

Figure 2
Sequence description of PyFer.

(A) Multiple alignment of PyFer (GenBank accession number AB918149) with other
ferritins: Ulva pertusa (UpFer, AB691549), Glycine max (GmFer1, M64337), and G.
max (GmFer4, AB062756). The amino acid residues forming the ferroxidase site are
black highlighted. Putative transit peptide (TP) and extension peptide (EP), and helix A-E
of PyFer are indicated by bars on the sequences. Strictly conserved amino acid residues
among 6 members are indicated by asterisks, while similar residues are by colons.
(B) Unrooted phylogenetic tree of various ferritins generated using the neighbor-joining
(NJ) method. Bar indicates p-distances. Ferritins are PyFer, UpFer, GmFer1, GmFer4
(Genbank IDs of them are shown above), Cyanidioschyzon merolae (CmFer,
XP_005537881), Ulva fasciata (UfFer, EF437243), Chlamydomonas reinhardtii
(CrFer1, AF503338), Volvox carteri f. Nagariensis (VaFer, XP_002951031),
Pseudo-nitzschia australis (PaFer, ACI30661), Pseudo-nitzschia multiseries (PmFer,
ACI30660), Synechocystis sp. PCC6803 (CyanobacFer, AGF53187), Escherichia coli
Figure 3
Reducing and heat-denaturing SDS-PAGE (A) and Western-blot (B) analysis of protein extracted from raw, dried, and toasted nori.

Lane 1, raw; lane 2, dried; lane 3, toasted nori. (A) Protein extract from 100 µg of raw, dried, and toasted nori were loaded to 12.5% polyacrylamide gel and stained with Coomassie Brilliant Blue R-250. (B) The 50 times diluted above extracts were separated by 12.5% gel, followed by electro-blotted to the PVDF membrane. Anti-recombinant PyFer anti-serum was used as primary antibody. Approximately 1 ng of recombinant PyFer was loaded as a control (lane C).

Figure 4
Non-reducing and non-heat-denaturing SDS-PAGE analysis of oligomeric state of PyFer in the raw, dried, and toasted nori extract.

Protein extract from 100 µg of raw, dried, and toasted nori were loaded to 7.5% polyacrylamide gel without reducing reagents and heat treatment. Lane 1, raw; lane 2, dried; lane 3, toasted nori. (A) Western-blot analysis of non-reducing nori extracts. Anti-recombinant PyFer anti-serum was used as primary antibody. (B) The gel was
stained by Prussian blue stain. The iron containing protein on the gel was stained as light blue bands. Approximately 50 ng of native soybean ferritin purified from dry soybean seeds was loaded as a control (lane C).
References


20) Marchetti A, Parker MS, Moccia LP, Lin EO, Arrieta AL, Ribalet F, Murphy
17 36) Lescure AM, Proudhon D, Pesey H, Ragland M, Theil EC, and Briat JF, *Proc
Figure 1 Masuda et al.
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**A-helix**

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**B-helix**

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**C-helix**

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<td>DD-KGEALYAMELSLSEKLNFQKQLQALQAIAKDAHDKDAALCDVEGGLLSEQYDAVKEHA</td>
</tr>
<tr>
<td>GmFer1</td>
<td>VE-KGDALYAMELSLSEKLNVNEKLHNNVHDRCNMDPDVESEFSEGEVESIKIS</td>
</tr>
<tr>
<td>GmFer4</td>
<td>AD-KGDALHAMELSLSEKLTLNEKLHNNVATKNGDVLQADFVETELYGEQVEAIKRI</td>
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**D-helix**

<table>
<thead>
<tr>
<th>Transit peptide</th>
<th>Extension peptide</th>
</tr>
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<tbody>
<tr>
<td>PyFer</td>
<td>VYVSQLRRVGKGVLYLDQELGEEAA---</td>
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<tr>
<td>UpFer</td>
<td>DMVAQLKRVGVTPHGVWHDQEQIVLGEDAAA</td>
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<tr>
<td>GmFer1</td>
<td>EYVAQLRVRGKGVGHGWHDQQLL--------</td>
</tr>
<tr>
<td>GmFer4</td>
<td>EYVAQLRVRGKGVGHGWHDQMLLEGDAAA</td>
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</tbody>
</table>

**E-helix**

Figure 2A Masuda et al.
Figure 2B  Masuda et al.
Figure 3  Masuda et al.
Figure 4  Masuda et al.