Synthesis and characterization of the amyloid  $\beta 40$  dimer model with a linker at position 30 adjacent to the intermolecular  $\beta$ -sheet region

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# Abstract

Amyloid fibrils in senile plaque mainly consist of the 40-mer and 42-mer amyloid  $\beta$ -proteins (A $\beta$ 40 and A $\beta$ 42). Although A $\beta$ 42 plays more important role in the pathogenesis of Alzheimer's disease (AD), A $\beta$ 40 could be involved in the progression of AD pathology because of its large amount. Recent studies revealed that variable sizes of A $\beta$  oligomers contributed to the neuronal death and cognitive dysfunction. However, how large oligomeric species are responsible for AD pathogenesis remains unclear. We previously proposed a toxic dimer model of A $\beta$  with turn structure at positions 22 and 23 using solid-state NMR and systematic proline replacement. Based on this model, we herein show the synthesis and biological activities of an E22P-A $\beta$ 40 dimer at position 30, which was connected to L,L-2,6-diaminopimeric acid. The E22P-A $\beta$ 40 dimer formed stable 6~8-mer oligomers without amyloid fibrils, but was not neurotoxic on human neuroblastoma cells. On the other hand, E22P-A $\beta$ 40 generated high molecular-weight oligomers into fibrils, and showed the neurotoxicity. These results suggest that such kind of A $\beta$ 40 dimer with a parallel  $\beta$ -sheet might not be pathological.

# Highlights

- E22P-A $\beta$ 40 dimer with turn at positions 22 and 23 was synthesized.
- The cross-linkage at position 30 by optically active L,L-DAP was adopted in the dimer.
- E22P-A $\beta$ 40 dimer formed 6~8-mer oligomers without fibrils, but was not neurotoxic.
- A $\beta$ 40 dimer with a parallel  $\beta$ -sheet might not be related to Alzheimer's pathology.

# **Keywords**

Alzheimer's disease, amyloid  $\beta$ -protein, dimer, oligomer, neurotoxicity.

# Abbreviations

A $\beta$ , amyloid  $\beta$ ; AD, Alzheimer's disease; L,L-DAP, L,L-2,6-diaminopimeric acid; HPLC, high performance liquid chromatography; MALDI-TOF-MS, matrix-assisted laser desorption/ionization time-of-flight mass spectrometry; NMR, nuclear magnetic resonance; PBS, phosphate buffered saline; TEM, transmission electron microscopy; Th-T, thioflavin-T.

### 1. Introduction

In Alzheimer's disease (AD), amyloid fibrils form senile plaques that mainly consist of 40- and 42-residue amyloid  $\beta$ -proteins (A $\beta$ 40 and A $\beta$ 42) [1]. A $\beta$ 42 is regarded as the leading cause of AD because its aggregative ability and neurotoxicity are more potent than those of A $\beta$ 40 [2]. In contrast, A $\beta$ 40 could be involved in the progression of AD pathology because of its large amount. Accumulating evidence has shown that the soluble oligomeric assembly of A $\beta$  is more exclusively implicated in neuronal death and cognitive dysfunction than insoluble fibrils and protofibrils [3,4]. "Aggregation" in this context is defined as the change from A $\beta$  monomers into amyloid fibrils *via* oligomers and protofibrils. Since A $\beta$ 42 only weakly associates with A $\beta$ 40 [5,6], homogeneous oligomers of A $\beta$ 42 or A $\beta$ 40 form more easily than the corresponding heterogeneous oligomers. The minimum unit of these oligomers, which have been divided into low molecular-weight oligomers (2~12-mer) and high molecular-weight oligomers (24~100-mer), is considered to be a dimer or trimer (2 or 3 x *n*-mer) [7]. However, how large oligomeric species of A $\beta$  are responsible for the pathogenesis of AD remains unknown.

Our previous study using solid-state NMR [8] and systematic proline replacement [9] identified a toxic conformer of A $\beta$ 42 with a turn structure at Glu22 and Asp23, in which the residues at positions Gln15~Ala21 and Val24~Ile32 took part in the intermolecular parallel  $\beta$ -sheet of A $\beta$ 42 aggregates. Moreover, the C-terminal hydrophobic core formed by another turn at Gly38 and Val39 together with the intramolecular  $\beta$ -sheet (Met35~Gly37 and Val40~Ala42) accelerated the aggregation (oligomerization) of A $\beta$ 42 [10]. Based on these findings, we proposed a structural model for a toxic dimer of A $\beta$ 42 with a turn at Glu22 and Asp23 as shown in Figure 1A. Regarding the aggregates of A $\beta$ 40, the most abundant species among A $\beta$  [2], both A $\beta$ 40 and A $\beta$ 42 would have the turn structure at positions 22 and 23 together with the intermolecular parallel  $\beta$ -sheet, whereas the C-terminal hydrophobic core is absent in the A $\beta$ 40 aggregates because of its lower hydrophobicity based on Wetzel's model [11].

Practical synthesis of A $\beta$  dimers have thus far been limited to A $\beta$ 40 dimers due to the

intrinsic potent aggregative ability of A $\beta$ 42 dimers during the synthesis and preparation. For example, Kok *et al.* [12] reported the synthesis of a dityrosine cross-linked A $\beta$ 42 dimer at position 10; however, its biological activity was not tested due to the insufficiently low amounts obtained. Based on the elevated levels of dityrosine-linked A $\beta$  dimers in the brains and blood of AD patients, Kok *et al.* synthesized two dimer models of A $\beta$ 40, in which dityrosine [12] or bifunctional 2,6-diaminopimeric acid (DAP) as "a diastereomeric mixture" [13] was incorporated at position 10 instead of Tyr10. Both models facilitated the production of fibrils in spite of the formation of globular aggregates [12,13] associated with neurotoxicity of the dityrosine-linked A $\beta$ 40 dimer [12]. The dityrosine-linked A $\beta$ 40 dimer synthesized by Walsh and co-workers also formed typical amyloid fibrils, even though it impaired long-term potentiation in rats *in vivo* [14]. The disulfide-linked A $\beta$ 40 dimer at Ser26 inhibited long-term potentiation [15,16], but did not appear to be physiologically relevant to the pathology of AD.

Since A $\beta$ 40 dimers connected adjacent to the intermolecular  $\beta$ -sheet region have not yet been examined, we synthesized a new dimer model of E22P-A $\beta$ 40 (Fig. 1B) with the toxic turn, in which position 30 was linked to L,L-2,6-diaminopimeric acid (L,L-DAP: Fig. 1C), not to the diastereomeric mixture [13], in order to investigate the effects of dimerization on its biological activity. Wild-type A $\beta$ 40 was negligibly neurotoxic, whereas E22P-A $\beta$ 40 was neurotoxic, albeit 10-fold less than wild-type A $\beta$ 42 [17]. Ala30 was selected as the linking position because our previous solid-state NMR studies using rotational resonance (R2) revealed that this residue in E22K-A $\beta$ 42 aggregates (Italian mutant) was proximal (< 6 Å) [18].

### 2. Material and methods

### 2.1. Synthesis of the E22P-A $\beta$ 40 dimer.

L,L-DAP (Fig. 1C) was synthesized according to the protocol established by Paradisi *et al.* [19] with slight modifications [20]. L,L-DAP was obtained from (S)- $\alpha$ -phenylethylamine in 6 steps (2.5 % yield). The structure of L,L-DAP was confirmed

by EI-MS, HR-FABMS (JMS-600H; JEOL, Tokyo, Japan), and <sup>1</sup>H NMR (AVANCE III 400 and 500; Bruker, Rheinstetten, Germany, ref. DHO = 4.79 ppm, D<sub>2</sub>O) :  $\delta$  1.46 (2H, m), 1.91 (4H, m), 3.77 (2H, t, *J* = 5.9 Hz) ppm. The specific rotation (P-2200; Jasco, Tokyo, Japan) of L,L-DAP was similar to that reported previously;  $[\alpha]_D$ : + 37.6°(*c* 0.96, 1N HCl, 28 °C), ref.  $[\alpha]_D$ : + 42.1°(*c* 0.96, 1N HCl) [19]. The amino groups of L,L-DAP were protected with Fmoc for the solid-phase synthesis of the E22P-A $\beta$ 40 dimer; <sup>1</sup>H NMR (ref. CD<sub>2</sub>HOD = 3.31 ppm, CD<sub>3</sub>OD):  $\delta$  1.54 (2H, m), 1.74 (2H, m), 1.88 (2H, m), 4.16 (4H, m), 4.30 (4H, m), 7.28 (4H, m), 7.34 (4H, t, *J* = 7.0 Hz), 7.63 (4H, m), 7.75 (4H, d, *J* = 7.4 Hz) ppm.

The E22P-A $\beta$ 40 dimer (Fig. 1B) was synthesized in a stepwise fashion on 0.1 mmol of preloaded Fmoc-L-Val-PEG-PS resin by Pioneer<sup>TM</sup> (Applied Biosystems; Foster City, CA, USA) using the Fmoc method, as described elsewhere [21]. Briefly, after the completion of chain elongation and cleavage from the resin, the crude peptide was precipitated by diethylether, followed by purification using HPLC on a YMC-Pack PROTEIN-RP column (20 mm i.d. x 150 mm; YMC, Kyoto, Japan) with elution at 8.0 mL/min by a 70-min linear gradient of 20-60% CH<sub>3</sub>CN containing 0.1% trifluoroacetic acid. Subsequent purification was carried out using YMC-Pack ODS-A (20 mm i.d. x 150 mm; YMC) with elution at 8.0 mL/min by an 80-min linear gradient of 20-60% CH<sub>3</sub>CN containing 0.1% trifluoroacetic acid. Lyophilization gave a pure peptide, the purity of which was confirmed by HPLC (>98%, 2.8% yield). The molecular weight of the E22P-Aβ40 dimer was confirmed by matrix-assisted desorption/ionization time-of-flight laser mass spectrometry (MALDI-TOF-MS, Bruker Ultraflex); m/z, calculated: 8608.80; found: 8609.55 [MH]<sup>+</sup>.

# 2.2 Thioflavin-T (Th-T) assay.

The aggregative ability of each A $\beta$  was evaluated by thioflavin-T (Th-T; Sigma, St. Louis, MO, USA) fluorescence assay developed by Naiki and Gejyo [22]. The procedure was described elsewhere [21]. Each A $\beta$  was dissolved in 0.1% NH<sub>4</sub>OH at 250  $\mu$ M, followed by 10-fold dilution with phosphate buffered saline (PBS: 50 mM sodium

phosphate and 100 mM NaCl, pH 7.4) at the final concentration of 25  $\mu$ M. After incubation at 37 °C for the desired period, 2.5  $\mu$ L of the reaction solution was added to 250  $\mu$ L of 5.0  $\mu$ M Th-T in 5.0 mM Gly-NaOH (pH 8.5), followed by the measurement of fluorescence at 430 nm excitation and 485 nm emission using a microplate reader (Fluoroskan Ascent; Thermo Scientific, Rockford, IL, USA).

### 2.3. Transmission electron microscopy (TEM).

The aggregates of each  $A\beta$  were examined under a H-7650 electron microscope (Hitachi, Ibaraki, Japan). The experimental procedure was described elsewhere [21]. After the supernatant was removed from pellets obtained in Th-T assay, resultant aggregates were then suspended in water (20 µL) by gentle vortex mixing, and centrifuged 13,000 rpm for 3 min. These suspensions were applied to a 200 mesh formvar-coated copper grid (Nissin EM, Tokyo, Japan), and allowed to dry in air for 5 min before being negatively stained for 2 min with 2% uranyl acetate, and subsequently subjected to microscope.

#### 2.4. Size exclusion chromatography.

Each A $\beta$  solution, as prepared in Th-T assay, was incubated at 37°C. After the solution was collected temporally and centrifuged at 15,000 rpm at 4°C for 10 min, 60 µL of the supernatant was analyzed by size exclusion chromatography on the Superdex75 10/300GL column (10 mm i.d. x 300 mm; GE Healthcare, Fairfield, CT, USA) with elution at 0.6 mL/min by filtered- and degassed-PBS, attached to a Waters LC system with a 2489 UV/Visible detector and 1525 binary HPLC pump controlled by Empower<sup>TM</sup>3 software (Waters, Taunton, MA, USA). The peptide was detected by absorbance at 220 nm. Calibration curves of size exclusion columns were constructed by dextran standards (Mp: mean peak molecular weight, 43,500; 21,400; 9,890; 4,440 Da) (Sigma) together with Blue dextran 2000 (GE Healthcare) as an indicator of the void volume (V<sub>o</sub>).

### 2.5. MTT assay.

SH-SY5Y cells, maintained in Dulbecco's modified eagle medium (DMEM; Wako, Osaka, Japan) containing 10% fetal bovine serum, were used as one of the typical neuronal cell models to estimate the neurotoxicity of each A $\beta$  with slight modifications to the described method [21]. In brief, each A $\beta$  was dissolved in 0.1% NH<sub>4</sub>OH to 220  $\mu$ M. The resultant peptide solution (10  $\mu$ L) diluted with 0.1% NH<sub>4</sub>OH to appropriate final concentrations (10<sup>-7</sup>, 10<sup>-6</sup>, 10<sup>-5.5</sup>, 10<sup>-5</sup> M) in the medium was added to 100  $\mu$ L of the culture medium of near-confluent cells (10<sup>4</sup> cells/well). After being treated at 37°C for 48 hr, 10  $\mu$ L of 5 mg/mL MTT (Sigma) was added to cells, followed by an incubation for 4 h at 37°C. After the removal of 70  $\mu$ L of the culture medium, the cell lysis buffer (100  $\mu$ L/well; 10% SDS, 0.01 M NH<sub>4</sub>Cl) was subsequently added to the cells. The cell lysate was then incubated overnight in the dark at room temperature before measurements at 595 nm with a microplate reader (MultiScan JX; Thermo Scientific, Waltham, MA, USA). Absorbance obtained by the addition of vehicle (0.1% NH<sub>4</sub>OH) was taken as 100%.

### 2.6. Statistical analysis.

All data are presented as the mean  $\pm$  SEM. The differences were analysed with one-way analysis of variance (ANOVA), followed by Bonferroni's test or unpaired Student's *t*-test. These tests were implemented within GraphPad Prism software (version 5.0d). *p* values <0.05 were considered significant.

### 3. Results

#### 3.1. Synthesis of the E22P-AB40 dimer

After several trials and errors, we adopted L,L-DAP (Fig. 1C) as a covalent cross-linker of E22P-A $\beta$ 40 with significant neurotoxicity, albeit 10-fold less than wild-type A $\beta$ 42 [17]. Solid-phase synthesis using Fmoc-L,L-DAP at position 30, adjacent to the intermolecular  $\beta$ -sheet region of A $\beta$ 40 aggregates, provided a pure and sufficient amount of the E22P-A $\beta$ 40 dimer as shown in Fig. 1B. (purity: >98%, 2.8% yield). Given E22P mutant as a toxic conformer surrogate, a monoclonal antibody (11A1) [23], previously developed against the toxic conformer of  $A\beta$  with turn at positions 22 and 23, reacted with the E22P-A $\beta$ 40 dimer (data not shown). We further attempted to synthesize a E22P-A $\beta$ 40 dimer doubly cross-linked at positions 17 and 30 using L,L-DAP in order to enhance the stability of the toxic conformation of E22P-A $\beta$ 40 dimer, but such an attempt was disappointing (data not shown).

# 3.2. The ability of the E22P-A $\beta$ 40 dimer to form fibrils

The aggregative ability of the E22P-A $\beta$ 40 dimer with a linker at position 30 was estimated using Th-T, a reagent that fluoresces when bound to A $\beta$  aggregates, and transmission electron microscopy (TEM). As shown in Figure 2A, E22P-A $\beta$ 40 aggregated with a lag time of ~8 h and a maximum fluorescence value after being incubated for 24 h. The wild-type A $\beta$ 40 did not aggregate after a 24-h incubation under the same conditions, as reported previously [21]. In contrast, the fluorescence of the E22P-A $\beta$ 40 dimer remained almost unchanged even after a 336-h incubation (14 days) (Fig. 2A). These results were consistent with TEM results showing that globular aggregates, not fibrils, were predominantly detected in the E22P-A $\beta$ 40 dimer (Fig. 2B). On the other hand, the typical amyloid fibrils together with high molecular-weight oligomeric aggregates were formed in E22P-A $\beta$ 40 even after 2 days (Fig. 2B). These results indicated that the E22P-A $\beta$ 40 dimer had the ability to form a soluble assembly and globular oligomers, but not fibrils.

## 3.3. The ability of the E22P-A $\beta$ 40 dimer to form oligomers

Oligomerization of the E22P-A $\beta$ 40 dimer was analysed further using size exclusion chromatography. As a control reference, Figure 3A showed that the soluble peak corresponding to the monomer of E22P-A $\beta$ 40 time-dependently decreased, and almost disappeared after an incubation for 16 h; this result implied that E22P-A $\beta$ 40 directly formed insoluble fibrils, as observed in Figure 2B. On the other hand, the E22P-A $\beta$ 40 dimer formed stable oligomers of 6~8-mer during incubation for 4~48 hr (Fig. 3B).

Although the assembly size of the predominant oligomers after 168-h and 336-h incubations exceeded 10-mer, it was difficult to calculate this correctly because of the limitation of the size exclusion column (Superdex75 10/300GL) used in this experiment.

### 3.4. The neurotoxicity of the E22P-Aβ40 dimer on SH-SY5Y cells

The neurotoxicity of the E22P-A $\beta$ 40 dimer in human neuroblastoma SH-SY5Y cells, one of the typical models of neuronal cell cultures, was measured using the MTT assay. After being incubated for 48 h, the viability of cells treated with E22P-A $\beta$ 40 (3.2  $\mu$ M = 10<sup>-5.5</sup> M) significantly decreased, which indicated E22P-A $\beta$ 40 to be neurotoxic as reported previously in PC12 cells [21] (Fig. 4A). In contrast, the neurotoxicity of the E22P-A $\beta$ 40 dimer was not observed, even at 10  $\mu$ M (10<sup>-5.5</sup> M), as shown in Fig. 4B. Ono *et al.* prepared a mixture of dimers, trimers, and tetramers of wild-type A $\beta$ 40, using photo-induced cross-linking of unmodified protein (PICUP) technology that mainly bound covalently at Tyr10 [24], and demonstrated that each oligomer of A $\beta$ 40 exhibited the neurotoxicity in a size-dependent manner [25]. However, the marginal toxicity of the PICUP-induced A $\beta$ 40 dimer did not seem to basically contradict our results in Figure 4B.

# 4. Discussion

We demonstrated that the dimer model of E22P-A $\beta$ 40 (Fig. 1B) linked at position 30 adjacent to the intermolecular  $\beta$ -sheet region in the aggregates formed stable low molecular-weight oligomers, mainly consisting of 6~8-mers, that did not extend to fibrils. However, this dimer model was less neurotoxic than the corresponding E22P-A $\beta$ 40 monomer with the toxic turn at positions 22 and 23, suggesting that such kind of the dimer model in Figure 1B, which is the minimum structure of non-toxic A $\beta$ 40 fibrils [26], could not be crucially involved in the pathogenesis of AD. Given the significant toxicity of E22P-A $\beta$ 40, the formation of high molecular-weight oligomers in the case of A $\beta$ 40 may be required for AD pathology. Although the insignificant toxicity of the E22P-A $\beta$ 40 dimer does not necessarily deter further investigations on the position of cross-linking, the

synthesis and characterization of dimer models of "more toxic A $\beta$ 42" appear to be more important. Whether high molecular-weight oligomers or low molecular-weight oligomers derived from A $\beta$ 42 can contribute to the neurotoxicity will be an attractive issue in the future. Since this study validated optically active L,L-DAP as a useful linker near the intermolecular  $\beta$ -sheet region of A $\beta$ 40, this strategy may also be applicable to the synthesis of various A $\beta$ 42 dimer models.

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### **Figure legends**

**Fig. 1.** Toxic dimer models of Aβ42 and Aβ40 with turns at positions 22 and 23. (A) A toxic dimer model of Aβ42 based on our previous structural studies using solid-state NMR and systematic proline replacement. (B) The E22P-Aβ40 dimer covalently cross-linked at Ala30 by L,L-DAP, whose structure was based on Wetzel's model [11]. (C) L,L-DAP as a molecular linker

Fig. 2. Analyses of the aggregative abilities of the E22P-A $\beta$ 40 monomer and dimer to form fibrils. (A) The Th-T assay of each A $\beta$  derivative after incubation of the indicated period. •, E22P-A $\beta$ 40;  $\bigcirc$ , the E22P-A $\beta$ 40 dimer. Data were presented as the mean ± SEM (n = 8). (B) A TEM analysis of A $\beta$  aggregates formed from E22P-A $\beta$ 40 and the E22P-A $\beta$ 40 dimer after incubation of the indicated period. Scale bar = 100 nm. *Left*, E22P-A $\beta$ 40; *Right*, the E22P-A $\beta$ 40 dimer.

Fig. 3. Analyses of the abilities of the E22P-A $\beta$ 40 monomer and dimer to form soluble oligomers. The ability of each A $\beta$  derivative to form soluble oligomers was evaluated by size exclusion chromatography after incubation of the indicated period. (A) E22P-A $\beta$ 40; (B) the E22P-A $\beta$ 40 dimer. The peptide was detected by absorbance at 220 nm. The size of molecular marker (kDa) is shown. V<sub>o</sub>: void volume. \*: artifact.

**Fig. 4.** Neurotoxicity of the E22P-Aβ40 monomer and dimer in SH-SY5Y cells. The near-confluent SH-SY5Y cells (10<sup>4</sup> cells/well) were treated with the E22P-Aβ40 monomer or dimer (10<sup>-7</sup>, 10<sup>-6</sup>, 10<sup>-5.5</sup>, 10<sup>-5</sup> M) in DMEM medium containing 10% fetal bovine serum before incubation at 37°C for 48 hr. Data were presented as the mean ± SEM (n = 3). Absorbance obtained by the addition of vehicle (0.1% NH<sub>4</sub>OH) was taken as 100%. \*p< 0.05 *versus* vehicle (Veh).

Fig. 1. Murakami, K. et al.

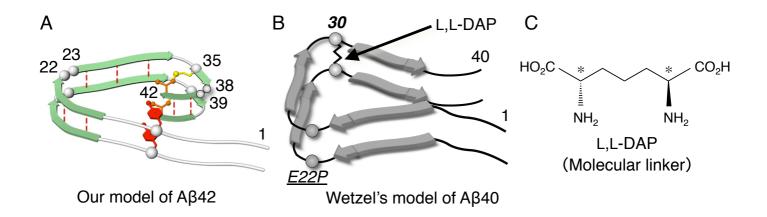


Fig. 2. Murakami, K. et al.

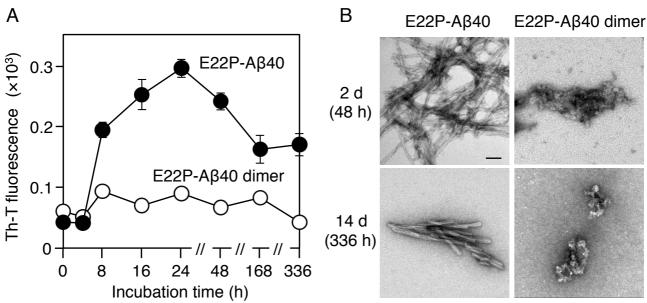


Fig. 3. Murakami, K. et al.

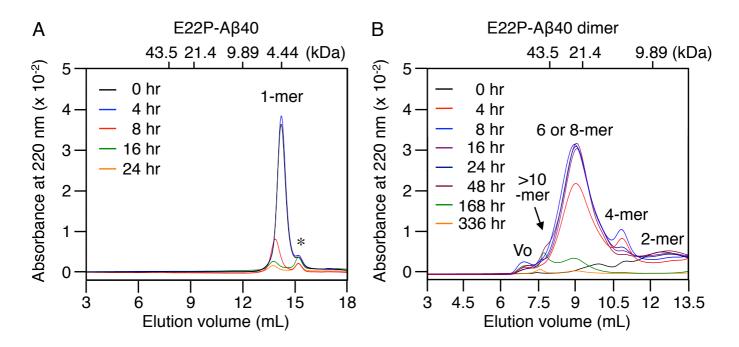


Fig. 4. Murakami, K. et al.

