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Divergent and Scalable Synthesis of α -Hydroxy/Keto- β -Amino Acid Analogues by Catalytic Enantioselective Addition of Glyoxylate Cyanohydrin to Imines

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ABSTRACT: The catalytic enantioselective addition of glyoxylate cyanohydrin to imines to afford α -keto- β -amino acid equivalents is reported. Sterically tuned aminobenzothiadiazine catalysts provided high yields and stereoselectivities (up to 100% yield, 99% ee, >99:1 dr) for both aromatic and aliphatic imines and the stereodivergent synthesis of both diastereomers was achieved. The optimal catalytic system was scalable, even with a low catalyst loading. The resulting adducts were converted into various chiral building blocks, including β -amino acid analogues, which are important motifs in medicinal chemistry, while maintaining a high enantiomeric excess.

KEYWORDS: asymmetric synthesis, organocatalysis, β -amino acid, cyanohydrin, diastereodivergent synthesis

β -Amino acids are important non-proteinogenic amino acid motifs that can dramatically change the physical properties and biological activity of peptides. Among them, α -oxygen-functionalized analogues, such as α -hydroxy- and α -keto- β -amino acids, are important substructures found in various bioactive natural products and pharmaceuticals (Scheme 1a).¹ Furthermore, α -keto- β -amino acids can be effectively used for decarboxylative transformations, including ketoacid-hydroxylamine (KAHA) ligation, which enables chemoselective coupling with complex fragments.^{2,3} Owing to the importance of β -amino acids, catalytic and stereoselective transformations for their rapid preparation have recently been explored.^{4–6}

The efficient synthetic approach, in which a C–C bond forms simultaneously with stereoselective introduction of functional groups at the α - and β -positions, can be accomplished by the Mannich-type addition of C2 nucleophiles to imines (Scheme 1b).⁴ Terada and coworkers reported the asymmetric addition of an α -diazoester to *N*-benzoylimine catalyzed by a BINOL-based chiral phosphoric acid.^{4a} Matsunaga, Kumagai, and Shibasaki have also described Mannich reactions using special α -hydroxyacetamides as nucleophiles.^{4b,c} Although these pioneering works have demonstrated that Mannich adducts obtained enantioselectively can be converted into α -oxygen-functionalized β -amino acids, the derivatization need relatively harsh conditions such as strong oxidants, acids, or bases. Therefore, a new efficient approach is needed.

To establish a new method that allows the divergent synthesis of α -oxygen-functionalized β -amino acids, we focused on the cyanohydrin motif, which is regarded as a carbonyl equivalent. Rawal and coworkers recently reported the enantioselective addition of masked acyl cyanide (MAC)⁷ to imines, where the adduct was transformed into α -amino acids by subsequent deprotection steps (Scheme 1c).⁸ We envisaged that glyoxylate cyanohydrin, a C2 nucleophile which has not previously been used in catalytic enantioselective reactions⁹, could be applied to the

Mannich-type addition using chiral bifunctional organocatalysts¹⁰ to afford α -keto- β -amino acid equivalents enantioselectively (Scheme 1d). The adduct is expected to undergo divergent conversion to α -hydroxy-/keto- β -amino acids by transformation of the cyanohydrin moiety. Herein, we describe the highly enantioselective addition of glyoxylate cyanohydrin to *N*-Boc-imines catalyzed by sterically-tuned aminobenzothiadiazine and the divergent synthesis of a series of β -amino acid analogues from the obtained adducts.

Based on the above concept, we initially screened reaction conditions for the asymmetric Mannich-type reaction using glyoxylate cyanohydrin **1** as a nucleophile (Table 1). First, nucleophile **1** and Boc-imine **2a** were treated with bifunctional aminothiourea catalyst **4a**^{10a} in CH_2Cl_2 at -20°C to afford the desired adduct in 100% yield as a 1.6:1 diastereomeric mixture, with major diastereomer **3a**¹¹ having 57% ee (entry 1). To improve the stereoselectivity, we screened other hydrogen-bond donors in the catalyst structure (entries 2 and 3). Squaramide **4b**¹² also afforded moderate stereoselectivity, while benzothiadiazine **4c**,^{10b} which was recently developed as a powerful hydrogen bond-donating organocatalyst, provided significantly better enantioselectivity (87% ee, entry 3). By changing the solvent from CH_2Cl_2 to toluene, the enantioselectivity for the major enantiomer was improved to 95% ee (entry 4). Furthermore, lowering the reaction temperature provided better diastereoselectivity while maintaining the high yield and enantioselectivity (entry 5). Next, we modified the chiral amine motif in the catalysts to achieve higher diastereoselectivity. Cinchona alkaloid-based catalyst **4d** provided lower enantioselectivity, while 1,2-diphenylethylenediamine structure **4e** did not significantly improve the selectivity (entries 6 and 7). In contrast, replacing one methyl substituent with a benzyl group boosted the diastereoselectivity to 9.4:1 (entry 8). Additional fine tuning of the alkyl substituents showed that secondary alkyl groups (*i*Pr and cyclopentyl) afforded better diastereoselectivity (entries 9 and 10).

Scheme 1. Strategy for catalytic enantioselective synthesis of medicinally important α -hydroxy/keto- β -amino acid motifs

a) α -Hydroxy/ α -keto- β -amino acid motifs

b) Catalytic asymmetric Mannich-type addition of C2 nucleophiles

c) Enantioselective addition of MAC to imines for α -amino acid equivalent

d) Glyoxylate cyanohydrin as C2 nucleophile for β -amino acid synthesis

Finally, the reaction using catalyst **4h** provided desired adduct **3a** as almost a single stereoisomer, with 97% ee and 31:1 dr (entry 10). This catalytic system was easily scalable, with compound **3a** prepared on a gram-scale using a catalyst loading of only 1 mol% (eq. 1). Surprisingly, substituents at the 5-position of the aromatic ring changed the diastereoselectivity in the products (entry 11). In particular, catalyst **4j**, bearing an isopropyl group, predominantly generated **3a'**¹³ with excellent yield and stereoselectivity (entry 12). The reversal of diastereoselectivity through catalyst control remains challenging, while several organocatalytic asymmetric transformations have been previously reported.^{14,15}

With optimal conditions in hand (Table 1, entries 10 and 12), we next examined the substrate scope of the diastereodivergent, enantioselective addition of glyoxylate cyanohydrin **1** to *N*-Boc imine (**2a**). As the electron density on the aromatic rings increased, the stereoselectivity of catalyst **4h** improved, with corresponding adducts **3a–c** obtained in high yields (entries 1–3). In contrast, the enantioselectivity was slightly diminished when electron-rich aromatic imines were applied using catalyst **4j** (**3a'–c'**). This trend was in agreement with the results for electron-deficient aromatic rings, while all adducts were still obtained with excellent stereoselectivity (entries 4 and 5). These results also indicated that the electronic properties of *N*-Boc-imine did not significantly affect the diastereoselectivity of the stereodivergent catalytic system using **4h** and **4j**. A methyl substituent at the *meta* or *ortho* positions relative to the imino group also resulted in a high yield and excellent stereoselectivity using both catalysts (entries 6 and 7). The reaction of

Table 1. Reaction optimization^a

entry	cat.	solvent, temp.	yield ^b (%)	dr ^c (3a : 3a')	ee ^c (%) 3a	ee ^c (%) 3a'
1	4a	CH ₂ Cl ₂ , –20 °C	100	1.6:1	57	16
2	4b	CH ₂ Cl ₂ , –20 °C	70	1.6:1	75	59
3	4c	CH ₂ Cl ₂ , –20 °C	100	2.4:1	87	43
4	4c	toluene, –20 °C	99	1.6:1	95	63
5	4c	toluene, –40 °C	94	2.3:1	97	66
6	4d	toluene, –40 °C	100	3.9:1	67 ^d	79 ^d
7	4e	toluene, –40 °C	91	2.1:1	89	89
8	4f	toluene, –40 °C	90	9.4:1	96	52
9	4g	toluene, –40 °C	93	15:1	96	33
10	4h	toluene, –40 °C	97	31:1	97	2
11	4i	toluene, –40 °C	82	1:3.6	59	85
12	4j	toluene, –40 °C	92	1:14	81	92

^a**1** (0.1 mmol), **2a** (0.15 mmol), and catalyst (0.005 mmol) in solvent (1 mL) reacted at the indicated temperature for 24 h. ^bIsolated yield. ^cEstimated by chiral HPLC analysis. ^d*ent*-**3a** and *ent*-**3a'** were obtained as major enantiomers.

naphthyl, thiophenyl and furanyl imines proceeded similarly, with corresponding adducts **3h–j** and **3h'–3j'** obtained with excellent stereoselectivity (entries 8–10). Finally, the catalytic system was applied to *N*-Boc-alkylimines derived from cyclohexanecarbaldehyde and isovaleraldehyde affording the adducts with good stereoselectivities, albeit only the enantioselectivity of the adduct **3k'** considerably diminished (entries 11 and 12).

Subsequently, we demonstrated the derivatization of Mannich adduct **3a** to obtain α -oxygen-functionalized β -amino

acids, as initially planned (Scheme 2). Compound **3a** was a stable, easy-to-handle, and storable solid, and Cbz deprotection followed by treatment with aqueous silver nitrate provided β -amino- α -ketoester **6** in 56% yield over two steps. Importantly, the optical purity of the compound was retained during this transformation, with β -amino- α -ketoester **6** obtained with 94% ee¹⁶. This is the first report of the isolation and full characterization of an enantioenriched *N*-carbamate-protected α -oxo- β -phenylalanine. In addition, the reductive transformation of **5** using L-Selectride afforded *anti*- α -hydroxy- β -amino acid derivative **7** in 66% yield over two steps. Changing the reductant to sodium borohydride resulted in the reduction of not only the cyanohydrin moiety, but also the ester, to afford aminodiols **8** in 68% yield, which is another useful motif in the pharmaceutical sciences.¹⁷ These reductive transformations also proceeded without any loss of stereochemical information in the β -amino group. Furthermore, we attempted the stereo-divergent synthesis of β -amino acid analogues. The cyano groups of **3a** and **3a'**, diastereoselectively prepared in Table 1, were hydrolyzed under the palladium-catalyzed hydration conditions reported by Naka and co-workers¹⁸ to provide complex β -amino acids **9** and **9'** as single stereoisomers, respectively.

Scheme 2. Derivatization of adducts 3a and 3a'

To gain insight into the reaction mechanism, we conducted the reaction using cyanohydrins **10** and **12** bearing different protective groups as nucleophiles (Scheme 3a). The catalytic addition of *O*-methyl carbonate **10** divergently provided **11** and **11'** with excellent enantioselectivity using catalysts **4h** and **4j**, respectively (eq 2). Furthermore, methyl ester **12** also afforded corresponding adducts **13** and **13'** with similar stereoselectivity (eq 3). These results indicated that the bulk of the nucleophile was not significantly important for stereoselectivity.

Considering the configuration of the benzothiadiazine catalyst and obtained adduct **3a**, we next conducted the computational studies to determine the transition state of the Mannich-type addition. Preliminary results suggested a transition state in which both the deprotonated nucleophile and the *N*-Boc-imine

are activated by two NH protons of benzothiadiazine and an ammonium unit in the catalyst, respectively (Scheme 3b).¹⁹ However, it remains difficult to explain the inverse diastereoselectivities of catalysts **4h** and **4j**. Therefore, further mechanistic studies to fully explain this unique catalytic system are currently underway.

Scheme 3. Preliminary mechanistic studies

(a) Steric effects of protective groups in cyanohydrin

(b) Calculated transition states to provide (2*S*,3*R*)-adduct

In summary, we applied *O*-Cbz-protected glyoxylate cyanohydrin as a C2 nucleophile in a catalytic Mannich-type addition to afford chiral α -keto- β -amino acid equivalents. An aminobenzothiadiazine catalyst bearing strong hydrogen-bond donor activity was found to be suitable for the transformation, and steric tuning of the catalyst afforded excellent stereoselectivity. The enantioenriched adducts were also readily converted into a series of chiral motifs, including useful β -amino acids, as initially expected. This approach could be applied to the efficient synthesis of various chiral building blocks, and further improvements and applications of this method are currently under investigation in our laboratory.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS publications website.

Experimental procedures and analytical data for all new compounds (PDF)

X-ray data file (CIF)

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Notes

The authors declare no competing financial interests.

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