

**New diagnostic method for Alzheimer's disease based on the toxic  
conformation theory of amyloid  $\beta$**

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This review was written in response to the author's receipt of the Japan Society for  
Bioscience, Biotechnology, and Agrochemistry Award in 2019.

## Abstract

Recent investigations suggest that soluble oligomeric amyloid  $\beta$  ( $A\beta$ ) species may be involved in early onset of Alzheimer's disease (AD). Using systematic proline replacement, solid-state NMR, and ESR, we identified a toxic turn at position 22 and 23 of  $A\beta$ 42, the most potent neurotoxic  $A\beta$  species. Through radicalization, the toxic turn can induce formation of the C-terminal hydrophobic core to obtain putative  $A\beta$ 42 dimers and trimers. Synthesized dimer and trimer models showed that the C-terminal hydrophobic core plays a critical role in formation of high molecular weight oligomers with neurotoxicity. Accordingly, an anti-toxic turn antibody (24B3) that selectively recognizes a toxic dimer model of E22P- $A\beta$ 42 was developed. Sandwich enzyme-linked immunosorbent assay with 24B3 and 82E1 detected a significantly higher ratio of  $A\beta$ 42 with a toxic turn to total  $A\beta$ 42 in cerebrospinal fluid of AD patients compared with controls, suggesting that 24B3 could be useful for early onset of AD diagnosis.

**Keywords:** Alzheimer's disease; amyloid  $\beta$ ; antibody; protein kinase C; solid-phase peptide synthesis

## **The opportunity to start this research**

First, I would like to explain the background to this study. My first original paper [1] was on the isolation and Epstein–Barr virus-inducing activity of (–)-indolactam-V (IL-V) [2], the basic ring structure of teleocidins [3-5], from *Streptomyces blastmyceticum*. IL-V is a tumor promoter and activator of protein kinase C (PKC) isozymes [6,7], the key enzyme family involved in signal transduction on the cell surface [8]. Since *Streptomyces blastmyceticum* produces a large amount of IL-V, detailed structure–activity relationship studies were performed to clarify the structural requirements of its tumor-promoting activity [9,10]. Consequently, the next step was to identify the binding mode of IL-V with PKC isozymes, whose binding sites are two C1 domains: C1A and C1B [11-13]. For this purpose, it was necessary to prepare PKC isozymes in sufficient quantity. However, there were difficulties in obtaining PKC isozymes with potent binding affinity to various PKC ligands; namely, phorbol 12,13-dibutyrate (PDBu), IL-V, and teleocidins. Specifically, the C1 domains were highly sensitive to oxidation because of their cysteine-rich sequence with a zinc-finger-like structure (ring-finger).

In 1992, I was in Stanford University as a visiting scholar, when Professor Paul A. Wender advised me to synthesize each PKC C1 domain by solid-phase peptide synthesis. Based on his advice, two PKC $\gamma$  C1 domain peptides (PKC $\gamma$ -C1A and PKC $\gamma$ -C1B) of approximately 50 amino acid residues were custom synthesized at Stanford University. Fortunately, we detected weak but significant [ $^3$ H]PDBu binding against each of them, although their purity and binding constants ( $K_d$ ) were not satisfactory [14]. After returning home in 1993, I started to establish a synthetic method of long peptides without fragment condensation in collaboration with Dr. Hiroyuki Fukuda at Applied Biosystems, Japan. After struggling for five years, we eventually succeeded in synthesizing all PKC isozyme C1 domains at high purity, and were able to precisely determine their  $K_d$  values against

[<sup>3</sup>H]PDBu [15-17]. The representative example (PKC $\delta$ -C1B) is shown in Figure 1. This data provided the basis for rational design of new medicinal leads with PKC isozyme selectivity. Moreover, we identified diacylglycerol kinase (DGK)  $\beta$  and  $\gamma$  as new PDBu receptors using synthetic C1 peptides of all DGK isozymes [18]. We were even able to synthesize a 116-mer PKC $\gamma$ -C1A-C1B peptide without fragment condensation [19]. The core method of this synthesis consists of three points: polyethylene glycol polystyrene support as a resin, hexafluorophosphate azabenzotriazole tetramethyl uronium (HATU) developed by Carpino [20] as a highly efficient activator, and the continuous flow-type peptide synthesizer, Pioneer<sup>TM</sup> supplied by Applied Biosystems. Unfortunately, this machine has been out of service since 2008, and we have recently introduced a microwave-type peptide synthesizer, Biotage Initiator+ Alstra<sup>TM</sup> (Biotage). The opportunity to start my research was the establishment of this peptide synthesis technology, and also my encounter with Dr. Fukuda, who collaborated with Dr. Takuji Shirasawa at Tokyo Metropolitan Institute of Gerontology, a world-renowned scientist in amyloid  $\beta$  research.

## **Toxic conformation theory of amyloid $\beta$**

Alzheimer's disease (AD) is one of the most common neurodegenerative disorders, characterized by extracellular amyloid fibrils and intracellular neurofibrillary tangles (Figure 2) [21]. The former consist mainly of 40- or 42-mer amyloid  $\beta$  protein (A $\beta$ 40 and A $\beta$ 42), while the latter are composed of hyperphosphorylated tau protein. AD proceeds along a sequence of aggregation (oligomerization) of A $\beta$ , hyperphosphorylation of tau protein, and ultimately, loss of nerve cells (amyloid hypothesis, Figure 3) [22]. More recently, accumulated evidence suggests that soluble A $\beta$  oligomers, rather than A $\beta$  fibrils, play a more important role in the pathogenesis of AD because of their more potent



neurotoxicity [23-25]. In 1999, when I began my research on A $\beta$ , the molecular mechanism of A $\beta$  aggregation and neurotoxicity had not been determined, and even the secondary structure of the “toxic oligomers” remained unknown. The main reasons for this were because of difficulties in the synthesis of A $\beta$ 42, the most toxic and highly aggregative species, and its highly aggregative character, making high-resolution X-ray crystallography and liquid-phase nuclear magnetic resonance (NMR) analysis almost impossible. Although solid-phase NMR analysis on aggregates of A $\beta$ 40, the less neurotoxic and less aggregative species, were first reported by Tycko’s group in 2002 [26], there were few reports on the precise structure of A $\beta$ 42 aggregates until 2015 [27-29].

First, a systematic proline replacement approach was adopted to identify the secondary structure of A $\beta$ 42 responsible for its potent aggregative ability and neurotoxicity. Prolines are not present in  $\beta$ -sheet structures but are easily accommodated in turn structures. Wood *et al.*, [30] reported the first proline replacement on A $\beta$  fragments (A $\beta$ 15–23 and A $\beta$ 12–26), and suggested that the residues at position 17–23 are involved in intermolecular  $\beta$ -sheets of A $\beta$  fibrils. However, such replacements should be performed using full-length A $\beta$ 42, whose solid-phase synthesis is difficult because of its last 14 C-terminal hydrophobic and bulky amino acid residues. As described, my colleagues and I had developed a practical and efficient method to synthesize hydrophobic and bulky peptides of over 50 amino acid residues without fragment condensation [15], enabling the flexible replacement of each amino acid residue of A $\beta$ 42 with proline [31,32].

Approximately 40 proline-substituted mutants of A $\beta$ 42 were synthesized to examine their aggregative velocity, thermodynamic stability of their aggregates, and neurotoxicity against rat pheochromocytoma (PC12) cells. The resultant data suggested that the

residues at positions 15–21 and 24–32 were involved in the intermolecular  $\beta$ -sheet structure of A $\beta$ 42 aggregates, and that the turn at position 22 and 23 plays a critical role in aggregation and neurotoxicity of A $\beta$ 42. Notably, mutation sites in familial AD are concentrated at this position [33,34]. Although the N-terminal 13 residues did not adopt any solid structure in A $\beta$ 42 aggregates, the C-terminal eight residues participated in their intramolecular  $\beta$ -sheet structure [35]. Based on these findings, we proposed toxic dimer and trimer models, as shown in Figure 4 [36].

Several reports suggested that radicalization of both Tyr10 and Met35 was important for inducing aggregation and neurotoxicity of A $\beta$ 42 [37-39]. Radicalization of Tyr10 by coordinated Cu(II) at His residues at positions 6, 13, and 14 of A $\beta$ 42 is considered to be an initial event for A $\beta$ 42 to aggregate and induce neurotoxicity [40]. Accordingly, Tyr10 could be converted to a phenoxy radical by the reaction of Cu(II) to Cu(I) to obtain hydrogen peroxide. In our model, phenoxy radical can efficiently oxidize Met35 by formation of a turn at position 22 and 23, bringing Tyr10 and Met35 closer together [41,42]. The resultant cation radical of Met35 can be ionically stabilized by the C-terminal carboxylate of Ala42, forming a hydrophobic core that accelerates aggregation to form a dimer and trimer. Since carboxyl radicals are incorporated in aggregates, liberation of these radicals will cause oxidative stress against neuronal cells for extended periods of time. This is our “toxic conformation theory”, and the turn at position 22 and 23 is called a “toxic turn” (Figure 4).

To confirm the presence of the toxic turn, solid-state NMR of partially  $^{13}\text{C}$ - and  $^{15}\text{N}$ -labeled A $\beta$ 42 at positions 21–24 or 25–27 was measured after aggregation using  $^{13}\text{C}$ -1H dipolar assisted rotational resonance [43,44], in collaboration with Professor Kiyonori Takegoshi at Kyoto University. The data clearly showed the presence of turns at position 22 and 23 and position 25 and 26 in aggregates of wild-type A $\beta$ 42 and E22K-A $\beta$ 42

(Italian mutation for familial AD [45]) [46,47]. Moreover, conformationally-fixed A $\beta$ 42 with a lactam ring at position 22 and 23 was neurotoxic, while a lactam ring at position 25 and 26 was virtually non-toxic [47]. A $\beta$ 42 is produced from amyloid precursor protein (APP) by two secretases [48]. A $\beta$ 42 monomer is in equilibrium between two conformers with a turn at positions 22 and 23 or 25 and 26, with the former aggregating to form toxic oligomers (Figure 5).

Systematic proline replacement of A $\beta$ 40 was also reported by Wetzel's group in 2004 [49]. Although the C-terminal structure of A $\beta$ 40 is different from A $\beta$ 42, the turn at position 22 and 23 was also present in A $\beta$ 40 aggregates. The turn mimic peptide, E22P-A $\beta$ 40, was more aggregative and neurotoxic compared with wild-type A $\beta$ 40, suggesting that the turn at position 22 and 23 in A $\beta$ 40 was pathological [31,32]. However, the position of the turn determined by solid-state NMR was different in A $\beta$ 40 aggregates [26] (Figure 6). The turn was present near position 25 and 26, which is similar to the nontoxic conformer in A $\beta$ 42 aggregates [47], because of the presence of a salt bridge between Asp23 and Lys28.

Recently, the structure of highly homogeneous A $\beta$ 42 aggregates determined by solid-state NMR was independently reported by three groups [27-29]. Surprisingly, the turn position in A $\beta$ 42 aggregates was quite different from that in A $\beta$ 40 aggregates: in the former, it was near position 22 and 23, while it was near position 25 and 26 in the latter. Both of these positions correlate with our proposed turn structures in A $\beta$ 42 aggregates (Figure 6). The presence of Ile41 and Ala42 at the C-terminus of A $\beta$ 42 could create a salt bridge between the primary ammonium ion of Lys28 and carboxylate anion of Ala42 to fix the molecule with a turn at position 22 and 23, unlike A $\beta$ 40. This clearly demonstrates

that A $\beta$ 42 is more aggregative and neurotoxic than A $\beta$ 40 since the turn at position 22 and 23 is important to induce aggregation and neurotoxicity of A $\beta$ .

### **Synthesis and characterization of dimer models of A $\beta$**

As described, my colleagues and I identified the key structure (toxic turn) of A $\beta$ 42 that induces aggregation and neurotoxicity. Recent investigations suggest that A $\beta$  oligomers are more neurotoxic than mature A $\beta$  fibrils [23-25]. However, A $\beta$  oligomers are labile and in complex equilibrium among monomers, several oligomers, and mature fibrils (Figure 7). For example, 12-mers and 24-mers might form from dimers and/or trimers as a minimum unit. Further, some A $\beta$  oligomers are unable to form amyloid fibrils to exist as quasi-stable oligomers (off-pathway), while others directly aggregate into fibrils (on-pathway). It is difficult to separate pure oligomeric species for biological evaluation. To solve this problem, two approaches have been attempted: stabilization of A $\beta$  oligomers by photo-induced cross-linking, and synthesis of chemically pure and stable A $\beta$  oligomer models.

Teplov's group isolated dimers, trimers, and tetramers of A $\beta$ 40 using photo-induced cross-linking after incubation, and showed that their neurotoxicity was higher than A $\beta$ 40 monomers [50]. However, each oligomeric species might be a complex mixture of compounds with different three-dimensional structures since various positions of cross-linking are possible. It is therefore difficult to determine which three-dimensional structure is responsible for the toxicity observed in neuronal cells. Alternatively, there are several reports on the synthesis of dimer models of A $\beta$ 40 [51-55]. Nonetheless, in all cases, toxic conformation of A $\beta$ 40 was not considered. Moreover, the linker positions were almost within N-terminal regions rather than C-terminal ones which are involved in

the formation of a hydrophobic core to induce oligomerization (Figure 4). Most of these models formed aggregative amyloid fibrils, and some were neurotoxic [52,54,55].

Based on our toxic dimer model (Figure 4), three dimer models were synthesized: two with linkers at positions 30 or 38 of A $\beta$ 40 (**1–3**), and one with a linker at position 40 of A $\beta$ 42 (**4**) [56,57] (Figure 8). L,L-2,6-Diaminopimelic acid (DAP) or L,L-2,6-diaminoazelaic acid (DAZ) were used as linkers. Although DAP was previously used by Kok *et al.*, [51] for the synthesis of a dimer model with a DAP linker at position 10 to mimic the dityrosine-linked dimer of A $\beta$ 40, there have been no reports using DAZ as a linker for A $\beta$  dimers. Since the distance between the two methyl groups of Ala30 in E22K-A $\beta$ 42 aggregates was estimated by solid-state NMR to be 5–6 Å [58], the DAZ linker was preferable for simulating the intramolecular  $\beta$ -sheet at position 31–36 of the A $\beta$ 40 dimer model. To mimic the toxic turn at position 22 and 23, the E22P-mutant was used.

Synthesis of **1–4** was started by preparation of optically pure di-Fmoc-L,L-DAP and di-Fmoc-L,L-DAZ. The method of Paradisi *et al.*, [59] was adopted, albeit with slight modifications [56]. Solid-phase synthesis on a continuous flow-type peptide synthesizer (Pioneer<sup>TM</sup>) was performed using di-Fmoc-L,L-DAP or di-Fmoc-L,L-DAZ (0.5 equivalent with respect to the loading peptide to avoid formation of a monocoupled peptide), in accordance with Kok's study [51]. The purity of the dimers (**1–4**) was > 98%, as determined by high performance liquid chromatography (HPLC) analysis and electrospray ionization-quadrupole time-of-flight-mass spectrometry (ESI-qTOF-MS) measurements. Further, dimer yields were 5.6, 3.1, 11.6, and 6.0%, respectively [56,57].

Although dimer **2**, with a DAZ linker, had higher  $\beta$ -sheet content than **1** with a DAP linker as expected, the neurotoxicity of **2** as well as **1** against SH-SY5Y neuroblastoma cell lines was absent, even at 10  $\mu$ M [56]. In contrast, dimer **3**, with a DAP linker in the

C-terminal hydrophobic core at position 38, exhibited more potent neurotoxicity than the corresponding monomer, E22P-A $\beta$ 40. All these dimer models exhibited weak thioflavin-T (Th-T) fluorescence after 48 h incubation, which reflected the amount of A $\beta$  aggregates, suggesting they exist as quasi-stable oligomeric species even after 24–48 h incubation [56].

However, characterization of the A $\beta$  oligomers responsible for AD pathogenesis has been controversial. Conventional techniques such as sodium dodecyl sulfate–polyacrylamide gel electrophoresis (SDS-PAGE) and size-exclusion chromatography (SEC) can be useful for this purpose, but SDS induces A $\beta$  oligomerization [60]. Additionally, SEC cannot accurately determine molecular weight since there are no ideal calibration proteins for A $\beta$ . Recently, ion mobility-mass spectrometry (IM-MS) combined with ESI as a native ionization technique, has enabled separation of various A $\beta$  oligomers to accurately determine their molecular weight without using organic solvents that disrupt noncovalent interactions within A $\beta$  oligomers [61–63]. In fact, 2–12-mer aggregates of A $\beta$ 40 and A $\beta$ 42 have been assigned by IM-MS. However, larger oligomers could not be detected because of the high aggregative velocity of A $\beta$  monomers.

In collaboration with Dr. Kenji Hirose and Mr. Taiji Kawase at Japan Waters Ltd., IM-MS measurements of **1–3** were performed after 4 h incubations [56]. Similar spectra were obtained even after 24 h incubations. The representative two-dimensional heat maps with  $m/z$  domains and drift times are shown in Figure 9. Dimers **1** and **2**, with a DAP or DAZ linker at position 30, existed predominantly as hexamers in monomer conversion ( $n = 3$ ), with oligomer distribution of 2–12-mer ( $n = 1–6$ ). In contrast, dimer **3**, with a DAP linker at position 38, existed as 12–24-mers ( $n = 6–12$ ). The fact that only **3** exhibited more potent neurotoxicity than E22P-A $\beta$ 40 monomer, suggests that hydrophobic interaction of the C-terminal region of A $\beta$  is indispensable for formation of larger toxic

oligomers, and that the oligomer size necessary to induce neurotoxicity is around 20-mer in monomer conversion.

According to our “toxic-conformation theory”, formation of the C-terminal hydrophobic core may be the initial event for A $\beta$ 42 to induce oligomer formation and exhibit neurotoxicity through radicalization (Figure 4) [36]. Although the C-terminal hydrophobic core could not easily form in A $\beta$ 40, a DAP linker at position 38 enabled A $\beta$ 40 to form toxic oligomers such as A $\beta$ 42. In fact, the corresponding A $\beta$ 42 version of **3**, E22P,V40DAP-A $\beta$ 42 dimer (**4**), also existed as a 12–24-mer in monomer conversion ( $n = 6–12$ ) (unpublished results), and was significantly neurotoxic [57]. Dimers **3** and **4** may thus be one of the practical toxic dimer models of A $\beta$  since they form quasi-stable protofibrillar aggregates with potent neurotoxicity, and consequently can be regarded as “off-pathway” aggregates.

### **Synthesis and characterization of trimer models of A $\beta$**

A $\beta$  oligomers originated from A $\beta$  dimers may be involved in the pathogenesis of AD [64-66]. However, several investigations have suggested involvement of A $\beta$  trimers, which show more potent synapse toxicity such as inhibition of long-term potentiation (LTP) than A $\beta$  dimers or tetramers [67,68]. There were no reports on the synthesis and characterization of full-length A $\beta$  trimer models, although trimer models of A $\beta$  fragments were synthesized [69,70]. As my colleagues and I had experienced previously with systematic proline replacement [31,32], such models should be made for full-length A $\beta$ . There are two possible trimer models: one consists of two intermolecular parallel  $\beta$ -sheets, with an additional A $\beta$  monomer bound to the dimer model, while the other is a propeller-type model, as shown in Figure 4. Recently, this type of trimer was identified by solid-

state NMR in 150 kDa oligomers of A $\beta$ 42 [71]. Since the propeller-type model is quite different from the dimer models with intermolecular parallel  $\beta$ -sheets, we attempted to mimic this trimer structure.

After molecular modeling studies, 1,3,5-phenyltris-L-alanine (PtA) was adopted as a trimer linker to reproduce a stable trimer structure (Figure 10A) [72]. Synthesis of tri-Fmoc-PtA was performed based on the approach of Ritzén *et al.* [73]. With asymmetric hydrogenation, Imamoto's Rh-(*S,S*)-QuinoxP [74] produced excellent results (> 98% ee, > 98% de). Three trimer models of E22P-A $\beta$ 40 with the PtA linker at position 34, 36, or 38 (**5–7**) were synthesized on a microwave peptide synthesizer (Initiator+ Alstra™, Biotage) using tri-Fmoc-PtA (0.33 equivalent was used to avoid formation of partially coupled PtA) (Figure 10B). Using C4 and C18 columns, trimer models with E22P mutation of A $\beta$ 40 (**5–7**) were obtained at yields of 0.58, 1.1, and 0.45% after HPLC purification. Purity was checked by HPLC analysis and molecular formulae confirmed by ESI-qTOF-MS measurements [72]. Unexpectedly, continuous flow-type peptide synthesizer (Pioneer™) without heating and microwave irradiation did not produce these target molecules, possibly because the molecular motion might be strongly restricted by the trimer linker, thereby suppressing coupling reactions by steric hinderance at room temperature.

With the trimer models (**5–7**), neurotoxicity against SH-SY5Y cells was far less compared with dimer model **3** and E22P-A $\beta$ 40 monomer [72]. Only **7**, with a PtA linker at position 38, exhibited significant but weak neurotoxicity compared with E22P-A $\beta$ 40. IM-MS measurements suggested that an oligomer size of **7** was 9–21-mer in monomer conversion ( $n = 3–7$ ), which is comparable to **3** (12–24-mer). In contrast, oligomer sizes of **5** and **6**, both without neurotoxicity, were 3–6-mer ( $n = 1–2$ ) and 3–12-mer ( $n = 1–4$ ), respectively. The 150 kDa oligomer of A $\beta$ 42 [71] contained a trimer structure with an



anti-parallel  $\beta$ -sheet at the C-terminus. However, the corresponding trimer model **6** did not exhibit any neurotoxicity, even at 10  $\mu$ M. In general, neurotoxicity of A $\beta$  is considered to be ascribable to the aggregation process, not specific oligomers [75]. Since both **3** and **7** formed amyloid fibrils over 48 h incubation, this suggests that tertiary structure as well as oligomer size can contribute to neurotoxicity. The concentration of oligomers might also be important. Neurotoxicity (cell death) by aggregation at the cell surface requires high oligomer concentration ( $>1$   $\mu$ M), while synapse toxicity of oligomers was observed at a low concentration ( $< 1$   $\mu$ M) [24].

Hitherto, several correlations between oligomer size and neurotoxicity have been reported [24]. For example, ADDL (3–24-mer) [76] and A $\beta$ O (15–20-mer) [77] are neurotoxic, while synapse toxicity such as LTP was induced by A $\beta$ \*56 (12-mer) [78] and even a dimer or trimer of A $\beta$ . This indicates that the oligomer size necessary for neurotoxicity and synapse toxicity can be quite different. In addition, the presence of larger oligomers, such as protofibrils (36–700-mer) [79] and amylospheroids ( $\sim 150$ -mer) [80], suggests that specific oligomers alone cannot explain neurotoxicity. Our data suggests that at least A $\beta$  oligomers  $< 12$ -mer are not neurotoxic. Synapse toxicity of the trimer models is thus worth investigating. However, caution should be applied since synthetic oligomers do not always reflect the heterogeneous A $\beta$  mixture present *in vivo*.

### **Development of an anti-toxic-turn antibody, 24B3, and its application towards AD diagnosis**

To identify the molecular species of A $\beta$  oligomers that show neurotoxicity, it is necessary to extract and purify applicable A $\beta$  oligomers from human brain or cerebrospinal fluid (CSF). However, structure–function analysis of A $\beta$  oligomers is problematic because A $\beta$  oligomers are in complex equilibrium among monomers,

oligomers, and fibrils. To address this problem, anti-A $\beta$  oligomer antibodies that recognize the specific tertiary structure of each oligomer would be useful. Several anti-A $\beta$  oligomer antibodies have been developed, and some are widely used for A $\beta$  research. For example, A11 polyclonal antibody reacts with toxic oligomers of A $\beta$ 42 more strongly than its fibrillar aggregates [81]. Attention has also been given to NU1 as an A $\beta$  oligomer-specific antibody [82]. Moreover, anti-oligomeric (anti-protofibrillar) A $\beta$  antibody (BAN2401) has been developed for the treatment of AD [83]. The weak point of these antibodies is that their epitopes are not clarified, and thus their usage in oligomer analyses has several limitations.

My colleagues and I focused on the toxic turn of A $\beta$ 42 at position 22 and 23, which could be mimicked by proline substitution. At first, we obtained 11A1 antibody by immunization to mice with E22P-A $\beta$ 10-35 bound to a carrier protein [84]. 11A1 antibody reacted not only with extracellular A $\beta$  aggregates (senile plaques) but also with intracellular A $\beta$  aggregates, some of which were considered to be toxic oligomers. Notably, intracellular A $\beta$  oligomers were clearly detected in neurons differentiated from induced pluripotent stem cells of AD patients [85]. However, reactivity of 11A1 against senile plaques and monomeric A $\beta$  was also high, making it difficult to use this antibody as a diagnostic tool for AD.

Next, we focused on 24B3 antibody [57], which was obtained along with 11A1 by screening using E22P-A $\beta$ 42. This reacted strongly with E22P-A $\beta$ 42 and the dimer models, **3** and **4**, but unlike 11A1, hardly bound to wild-type A $\beta$ 42 [57]. It was notable that preincubated wild-type A $\beta$ 42 was recognized by 24B3, indicating that toxic oligomers derived from wild-type A $\beta$ 42 could also bind to 24B3. These results prompted us to generate a sandwich enzyme-linked immunosorbent assay (ELISA) in collaboration

with Immuno-Biological Laboratories (IBL) Co., Ltd., for potential application in early AD diagnosis, since formation of toxic A $\beta$  oligomers is considered to be an initial event of AD [23-25]. After several trials, 82E1 [86], which recognizes the N-terminal-end of A $\beta$ , was fixed to a solid-phase for capture, while 24B3 conjugated with horseradish peroxidase for detection was used for quantification of toxic oligomers with a toxic A $\beta$  turn in human CSF. To prove the concept of the “toxic-conformer theory”, human CSF was analyzed using this ELISA in collaboration with Professor Takahiko Tokuda at Kyoto Prefectural University of Medicine [57]. As shown in Figure 11, the ratio of toxic conformer to total A $\beta$ 42 in AD/mild cognitive impairment (MCI) patients was significantly higher compared with age-matched non-AD controls. Nonetheless, a significant difference in the amount of toxic conformer between AD/MCI and control groups was not observed, possibly because of the small number of samples. Indeed, Akiba *et al.*, [87] reported significantly high levels of the toxic conformer in AD patients using our ELISA kit. They also suggested that the ratio of toxic conformer to total A $\beta$ 42 in patients with idiopathic normal pressure hydrocephalus (iNPH) was significantly higher than control groups. This indicates that the toxic conformer ratio could be a reliable biomarker for predicting the likelihood of patients with iNPH progressing into AD.

The 24B3 and 11A1 antibodies significantly suppressed neurotoxicity induced by A $\beta$ 42 and E22P-A $\beta$ 42, whereas 82E1 and 4G8 (sequence-specific antibodies against region 18–23 of A $\beta$ ) [88] did not [57]. The protective effect of 24B3 was higher than 11A1. Based on these results, passive immunization of 24B3 against AD model mice (Tg2576) was examined in collaboration with Dr. Takahiko Shimizu at Chiba University [89]. Intraperitoneal administration for 3 months (10 mg/kg/week) improved cognitive impairment, although the number of senile plaques did not change. Moreover, even single intravenous administration (20 mg/kg) suppressed the memory deficit. Masking of toxic

oligomers by 24B3 could lead to effective AD therapies with little adverse effects if 24B3 was reproduced for human use.

To investigate the contribution of the toxic conformer with the toxic turn to AD pathogenesis, in collaboration with Dr. Shimizu, we recently developed a new AD mouse model with E22P-A $\beta$  mutation using a knock-in technique to avoid the artificial phenotype observed in transgenic-type model mice [90]. Interestingly, a trimer band and high molecular-weight oligomer bands (but not a monomer band) were detected in a Tris buffer-soluble fraction of E22P-A $\beta$  knock-in mice at six months of age, when cognitive impairment occurred in the novel object recognition test. These data suggest that the toxic conformer of A $\beta$  induced cognitive dysfunction mediated by toxic oligomer formation. However, loss of neurons was not observed in these model mice, indicating that A $\beta$  oligomers induced only synapse toxicity such as inhibition of LTP without severe neurotoxic effects. Onset of AD might require accumulation of toxic A $\beta$  oligomers, followed by hyper-phosphorylation of tau proteins. Toxic oligomers might act as mediators for tau phosphorylation to result in neuronal loss. In this context, E22P-A $\beta$  knock-in mice would be a useful model for evaluating oligomer-induced cognitive impairment in AD. In addition, synapse toxicity of the trimer models (5–7) should also be evaluated.

## **Summary and future perspectives**

As described, the antibody 24B3 developed on the basis of the “toxic conformation theory” could be a useful tool for early diagnosis of AD. The next step is to optimize this antibody based on X-ray crystallographic analysis of the hapten peptide and its Fab domain complex. Structural optimization of the toxic dimer and trimer models for second generation antibodies against toxic oligomers of A $\beta$  is also a pressing need. Recently,

attention has been given to AD diagnosis using human plasma samples instead of CSF, which is more invasive. Yanagisawa and colleagues proposed new analytical methods to quantify the amount of A $\beta$  peptides in plasma samples using mass spectrometry to predict accumulation of A $\beta$  in the brain [91]. Our own approach is to use a Simoa<sup>TM</sup> (single molecule array) technology. Using this technology, Tokuda and colleagues established the method to quantify the amount of phosphorylated-tau (p-tau) in human plasma, and showed that p-tau levels were significantly higher in AD patients than non-AD controls [92]. In collaboration with Professor Tokuda and Professor Juan C. Troncoso at the Johns Hopkins University School of Medicine, we are measuring the amount of toxic conformer in human plasma samples using Simoa<sup>TM</sup> technology (unpublished results).

Prevention and treatment of AD are indispensable after diagnosis of early onset. It has been reported that many natural products (mainly flavonoids) suppress aggregation and neurotoxicity of A $\beta$ <sub>42</sub> [93,94]. However, their mechanism of action remains unknown. Recently, we proposed three structural features required for suppression of A $\beta$ <sub>42</sub> aggregation [95]. These features are: a catechol moiety that reacts with Lys residues of A $\beta$ <sub>42</sub> after oxidation [96], planarity due to  $\alpha,\beta$ -unsaturated carbonyl groups that ensure intercalation of the molecule into the intermolecular  $\beta$ -sheet region in A $\beta$  aggregates [97,98], and carboxy groups of triterpenoids or anthraquinoids that form salt bridges with Lys16 of A $\beta$ <sub>42</sub> oligomers [99,100]. These structural features would be useful for predicting new anti-A $\beta$ <sub>42</sub> aggregative compounds from natural sources, which might become promising lead compounds for protection against early onset of AD.

Treatment of AD is also a pressing need. Immunotherapy using anti-A $\beta$  antibodies has been extensively investigated [101,102] after pioneering work by Schenk and colleagues in 1999 [103]. However, most attempts gave disappointing results. Possible reasons for this might be that the antibodies used were not optimized for toxic species of

A $\beta$  aggregates, and that administration time of the antibodies to AD patients was too late. Although application of our 24B3 antibody [57] might be promising, considerable money is required for drug development and antibody therapy. In addition,  $\gamma$ -secretase and  $\beta$ -secretase inhibitors have also failed in clinical trials [104].

Another therapeutic approach has been performed using PKC activators. Cumulative evidence suggests that inhibition of PKC $\alpha$  decreases the amount of A $\beta$ <sub>42</sub> and A $\beta$ <sub>40</sub>, and that activation of PKC $\epsilon$  enhances degradation of A $\beta$  through endothelin converting enzyme [105-107]. Bryostatin 1, isolated from bryozoan [108], is one of the most promising PKC activators because of its low adverse effects such as tumor promotion and proinflammatory activity [107,109]. However, low availability from natural sources and structural complexity has hampered the supply for medicinal leads. To overcome these issues, synthesis of simplified analogs of bryostatin 1 that potently activate PKC [110,111], and practical synthetic methods of bryostatin 1 have been reported [112,113]. Our own approach was to develop new medicinal leads from naturally occurring PKC ligands, phorbol esters, ingenol esters, teleocidins, and aplysiatoxins [114]. After extensive structure–activity studies, aplysiatoxins were found to be the most promising leads [115] since their hydrophobicity is relatively low, and their structure is regarded as a conformationally-fixed analogue of diacylglycerol, an endogenous ligand for PKC isozymes. Moreover, we have developed 10-Me-Aplog-1 [116], a simplified analogue of aplysiatoxin isolated from sea hare [117], as a new PKC activator without tumor-promoting and proinflammatory activities (Figure 12). Since 10-Me-Aplog-1 did not show any tumor-promoting and proinflammatory activities, it might be effective for decreasing toxic oligomers of A $\beta$ .

I felt something like my destiny when I began to synthesize a simplified analogue of aplysiatoxin as a surrogate of bryostatin 1 with Dr. Yu Nakagawa in 2007. My research

on the chemistry and biology of A $\beta$  originated from the structural study of PKC C1 domains through peptide synthesis. Additionally, research on the treatment of AD connected again with research on naturally occurring PKC ligands. That is why research is interesting generally.

## **Acknowledgements**

I would like to express many thanks to the staff and graduate students who worked with me in the laboratory of Organic Chemistry in Life Sciences, Division of Food Science & Biotechnology, Graduate School of Agriculture, Kyoto University, and to all the collaborators in this study, most of them mentioned in the text and shown in the references.

I would like to express my sincere thanks to Dr. Koichi Koshimizu, emeritus professor of Kyoto University, for giving me the opportunity to study teleocidins, which was the starting point for this study. At the beginning of this study, Dr. Hideo Hayashi, emeritus professor of Osaka Prefectural University, provided much help and encouragement. I would also like to thank Dr. Hajime Ohigashi, emeritus professor of Kyoto University, for continuous support of my research. Finally, I would like to thank the Japan Society for Bioscience, Biotechnology, and Agrochemistry (JSBBA) for awarding me the JSBBA Award 2019.

I thank Rachel James, Ph.D., from Edanz Group ([www.edanzediting.com/ac](http://www.edanzediting.com/ac)) for editing a draft of this manuscript.

## **Disclosure statement**

No potential conflict of interest was reported by the author.

## **Funding**

This work was supported in part by Grants-in-Aid for Scientific Research (S) (26221202), (A) (19H00921, 21258015, 18208011), (B) (16380080, 13460048), (C) (11660109), Scientific Research on Innovative Areas “Frontier Research on Chemical Communications (17H06405), and “Chemical Biology of Natural Products” (23102011) from JSPS. This work was also supported in part by funds from The Uehara Memorial Foundation, The Naito Science & Engineering Foundation, The Takeda Science Foundation, and Asahi Group Foundation.



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**Figure captions**

**Figure 1.** PKC $\delta$ -C1B domain peptide<sup>15)</sup> and its crystal structure.<sup>13)</sup>

**Figure 2.** Senile plaques and neurofibrillary tangles in brain slices of AD patients, along with A $\beta$ 42 and A $\beta$ 40 sequences.

**Figure 3.** The amyloid hypothesis.<sup>22)</sup> MCI: mild cognitive impairment; AD: Alzheimer's disease.

**Figure 4.** Putative structure of dimers and trimers of A $\beta$ 42 induced by radicalization.<sup>36,42)</sup> Green arrows show intra- or intermolecular  $\beta$ -sheet regions in A $\beta$ 42 aggregates.

**Figure 5.** Toxic or less toxic conformers of A $\beta$ 42.<sup>47)</sup>

**Figure 6.** Structure of each monomer unit of aggregates of A $\beta$ 42<sub>27-29)</sub> and A $\beta$ 40<sub>26)</sub> revealed by solid-state NMR.

**Figure 7.** Putative structures of A $\beta$  oligomers. It is not clear which oligomers are on-pathway, leading to amyloid fibrils, stable and less toxic species, and which oligomers are off-pathway, not resulting in fibril formation.

**Figure 8.** Structure of A $\beta$ 40 dimer models with a linker at position 30 or 38 and A $\beta$ 42 dimer with a linker at position 40. The A $\beta$ 40 dimer structure is based on proline scanning by Williams *et al.*<sup>49)</sup> Green arrows show intermolecular  $\beta$ -sheet regions in A $\beta$ 40 aggregates.

**Figure 9.** Oligomer analysis of dimer and trimer models (**2**, **3**, **7**) by IM-MS after 4 h incubation at 12.5  $\mu$ M for **2** and **3**, and 8  $\mu$ M for **7** at 37 °C. *n* = number of dimers or trimers.

**Figure 10.** (A) Synthesis of (*S,S,S*)-tri-Fmoc-PtA. (a) 1,1,3,3-Tetramethyl guanidine, THF, 0 °C to room temperature, 4 h, 76%. (b) Rh-(*S,S*)-QuinoxP,<sup>74)</sup> H<sub>2</sub> (4 atm), EtOAc/MeOH, room temperature, 4 h, 92% (>98%ee, >98%de). (c)

858 NaOHaq, MeOH, 1 h, 97%. (d) H<sub>2</sub>, Pd/C, MeOH, overnight, 75%. (e) Fmoc-  
859 OSu, Na<sub>2</sub>CO<sub>3</sub>, MeCN, H<sub>2</sub>O, 3 d, 50%.

860 (B) Structure of trimer models (**5–7**) of E22P-Aβ<sub>40</sub>.

861 **Figure 11.** Sandwich ELISA of cerebrospinal fluid from AD patients.<sup>57)</sup>

862 **Figure 12.** Structure of aplysiatoxin and its simplified analogue, 10-Me-Aplog-1.<sup>116)</sup>

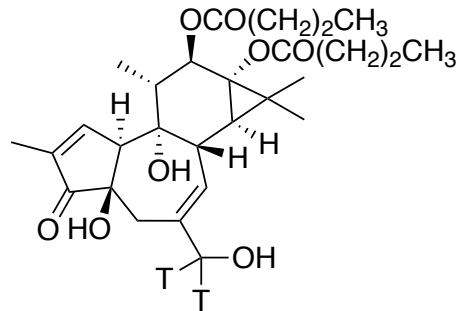
863





H R F K V Y N Y M S P T F **C** D H **C** G S L L W G L V K Q G L K **C** E D **C** G M N V **H** H K **C** R E K V A N L **C** G

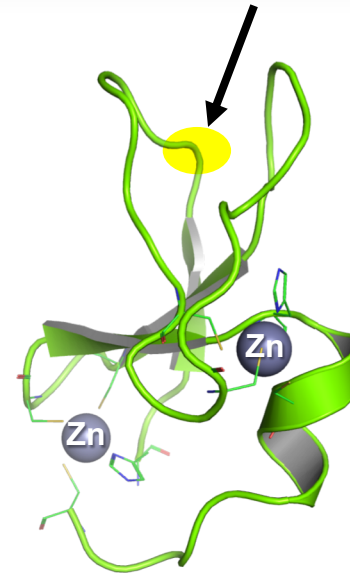
- 1) Solid-phase synthesis
- 2) Zinc-folding



- 3) [ $^3\text{H}$ ]PDBu binding assay

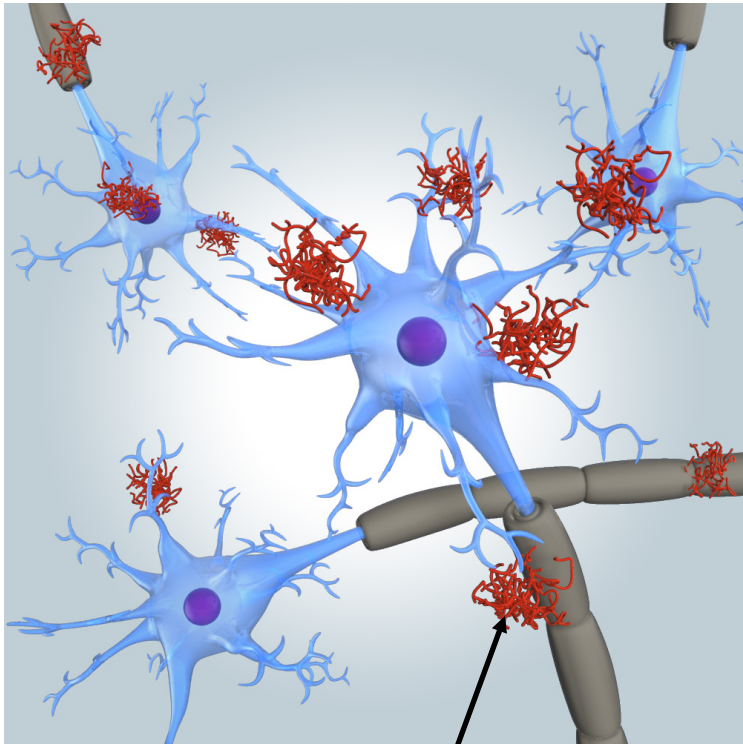
$$K_d = 0.53 \text{ nM}$$

PKC ligand binding site



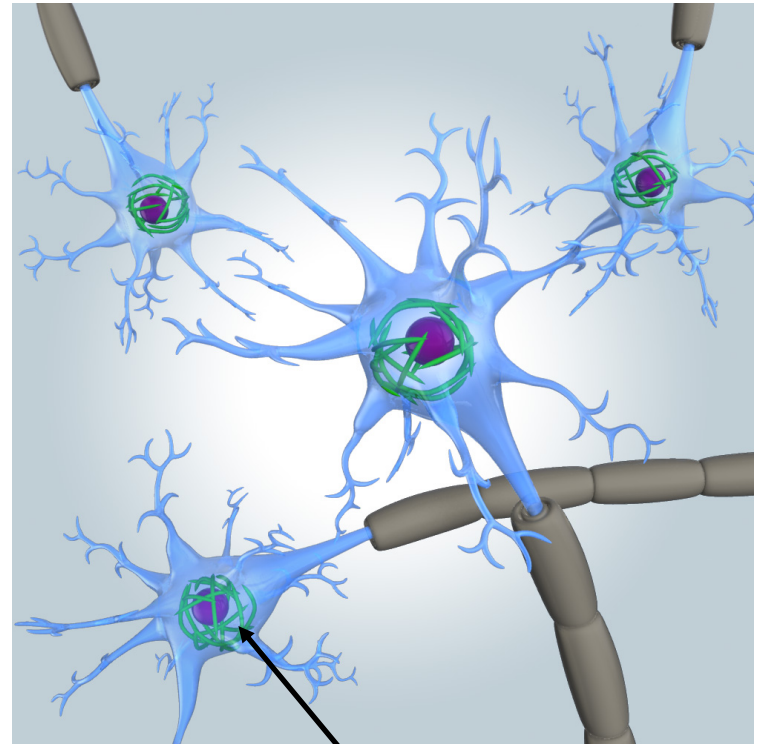
Crystal structure of  $\delta$ -C1B

Figure 1



**Senile plaques  
(A $\beta$  aggregates)**

**A $\beta$ 42:** DAEFRHDSGY EVHHQKLVFF  
**A $\beta$ 40:** DAEFRHDSGY EVHHQKLVFF



**Neurofibrillary tangles**

**AEDVGSNKGA IIGLMVGGVV IA**  
**AEDVGSNKGA IIGLMVGGVV**

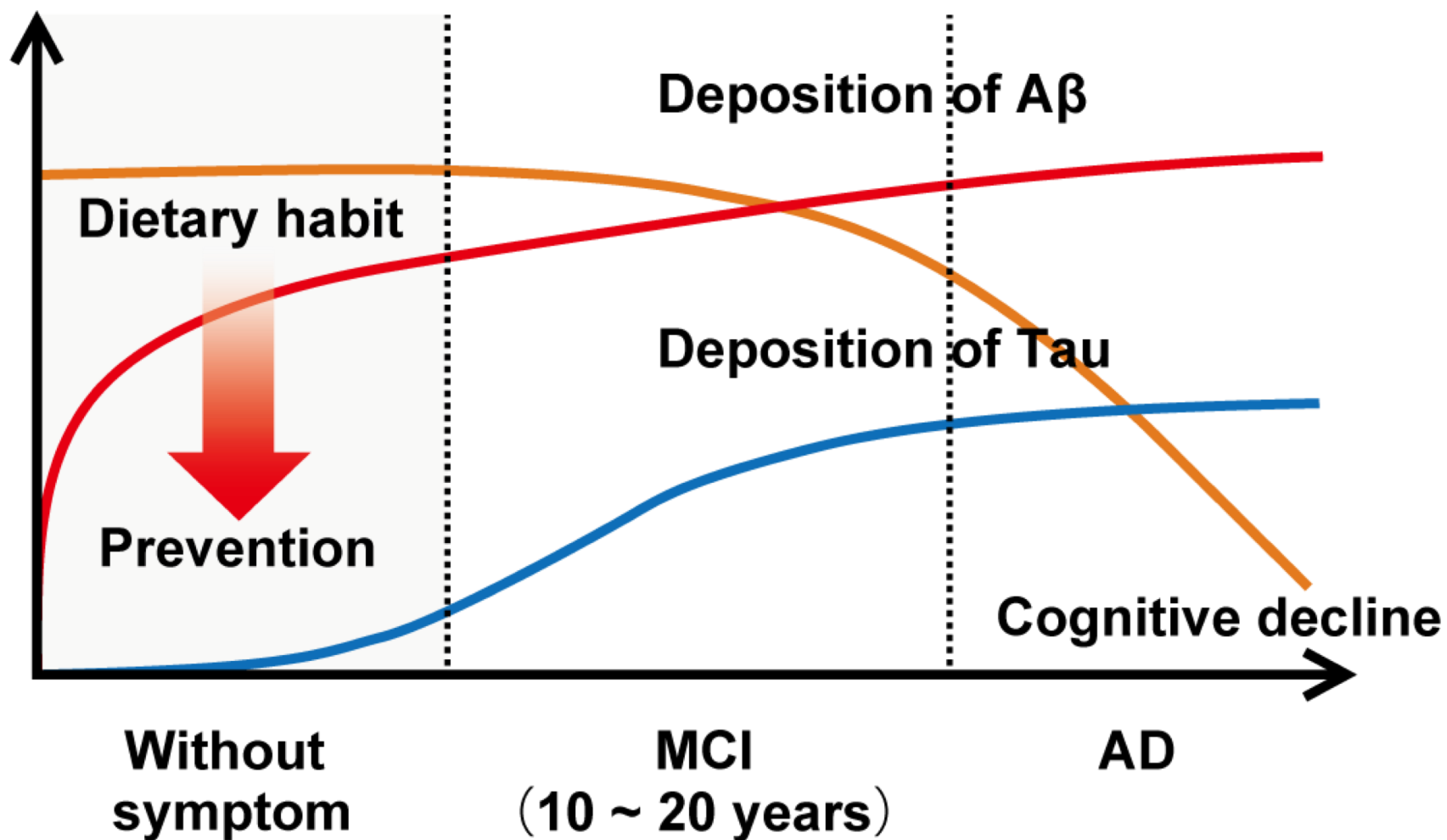


Figure 3

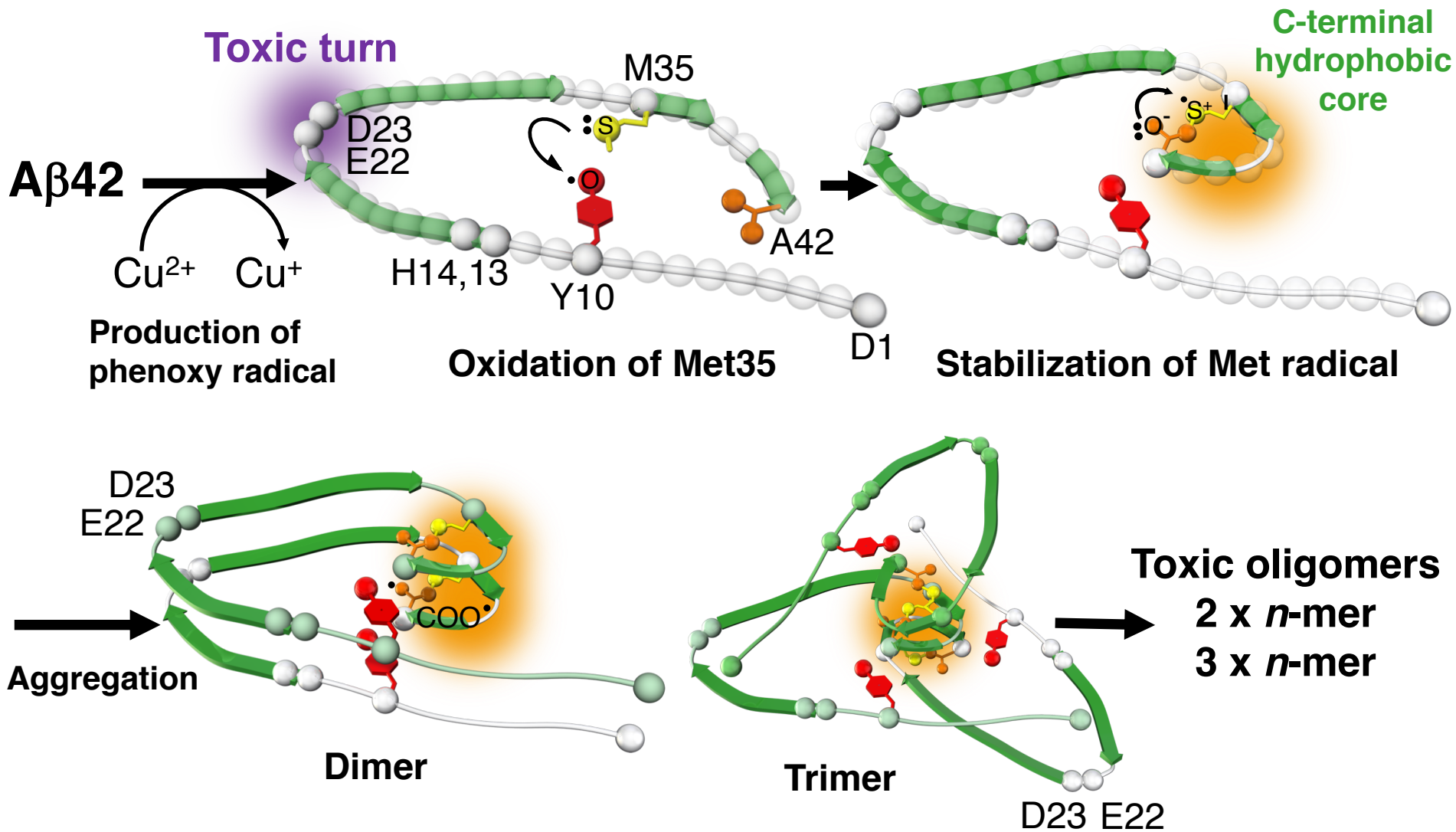


Figure 4

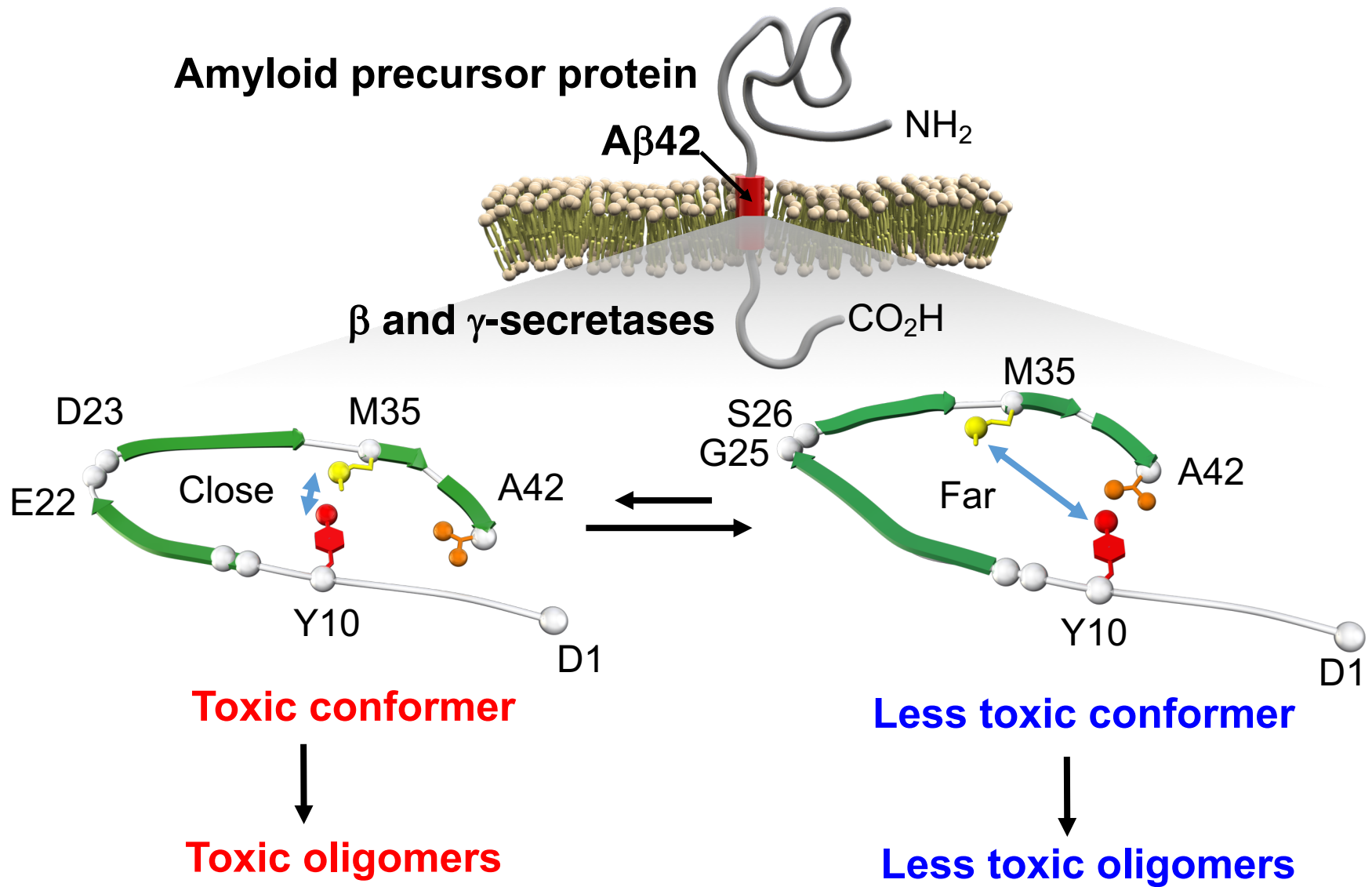
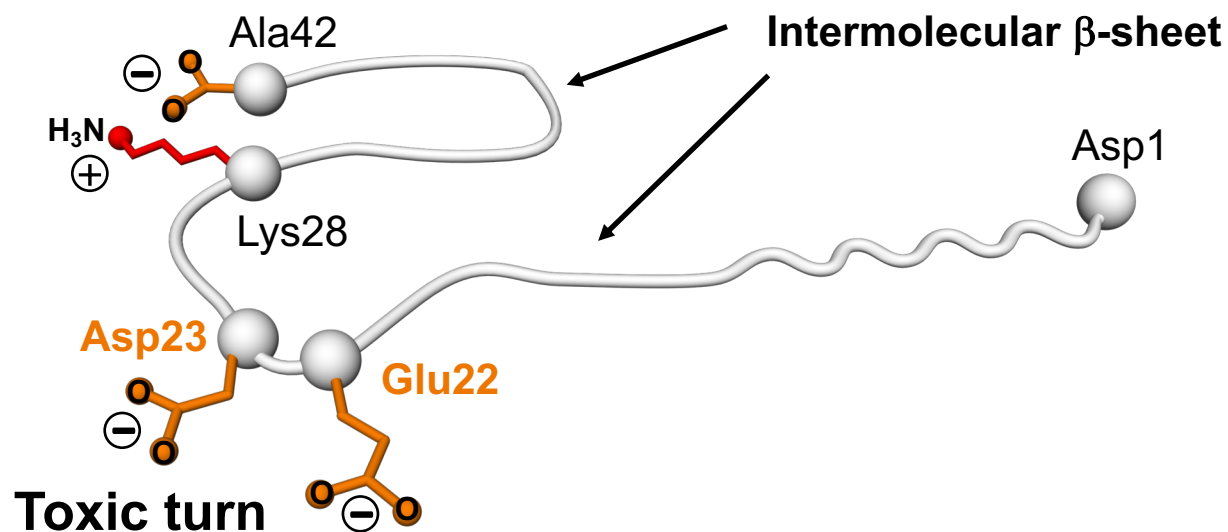


Figure 5

**A $\beta$ 42**



**A $\beta$ 40**

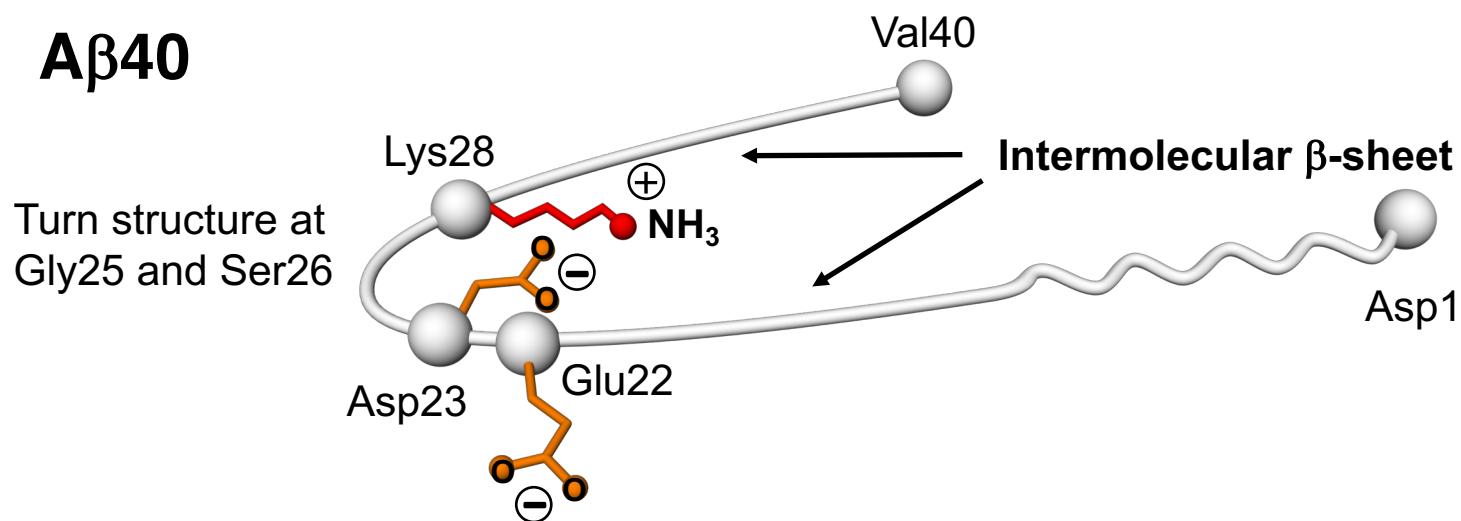


Figure 6

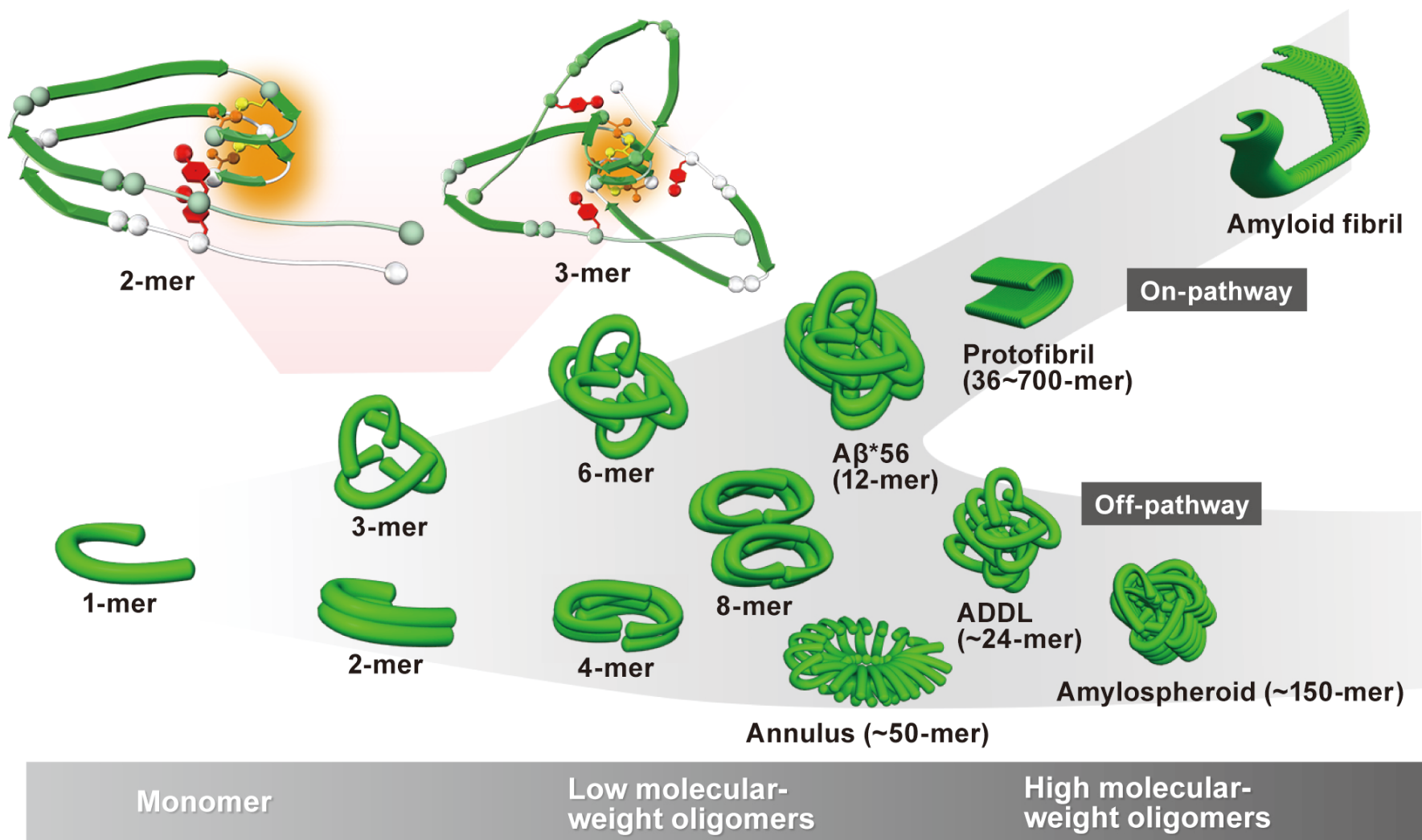
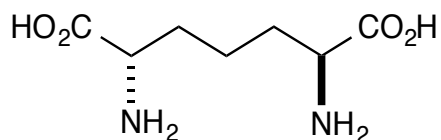
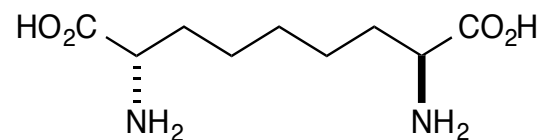


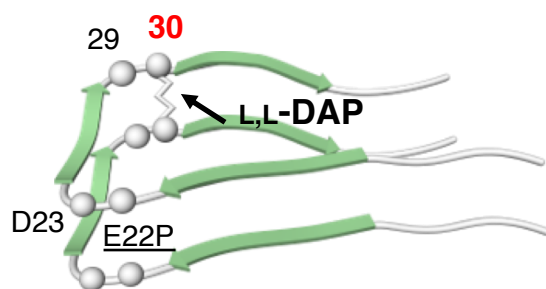
Figure 7



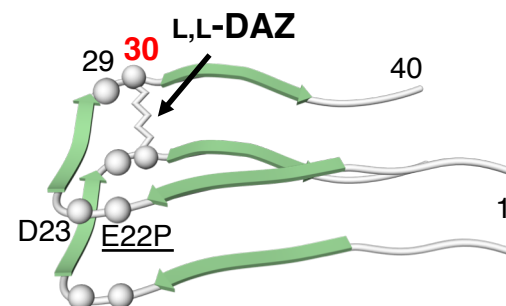
L,L-Diaminopimelic acid (DAP)



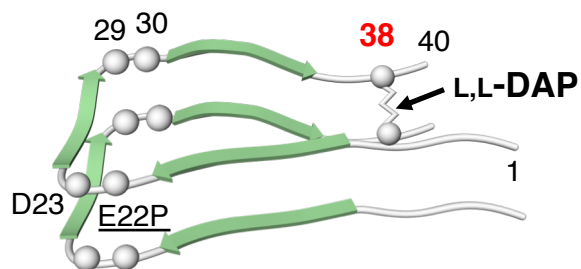
L,L-Diaminoazelaic acid (DAZ)



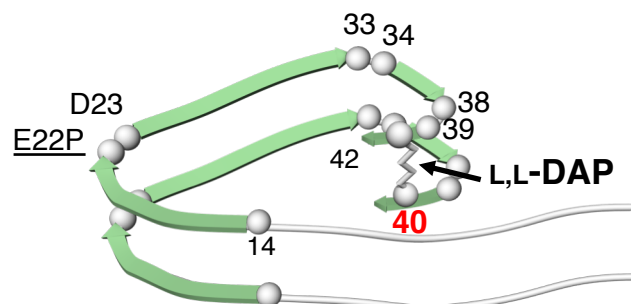
E22P,A30DAP-A $\beta$ 40 dimer (1)



E22P,A30DAZ-A $\beta$ 40 dimer (2)



E22P,G38DAP-A $\beta$ 40 dimer (3)



E22P,V40DAP-A $\beta$ 42 dimer (4)



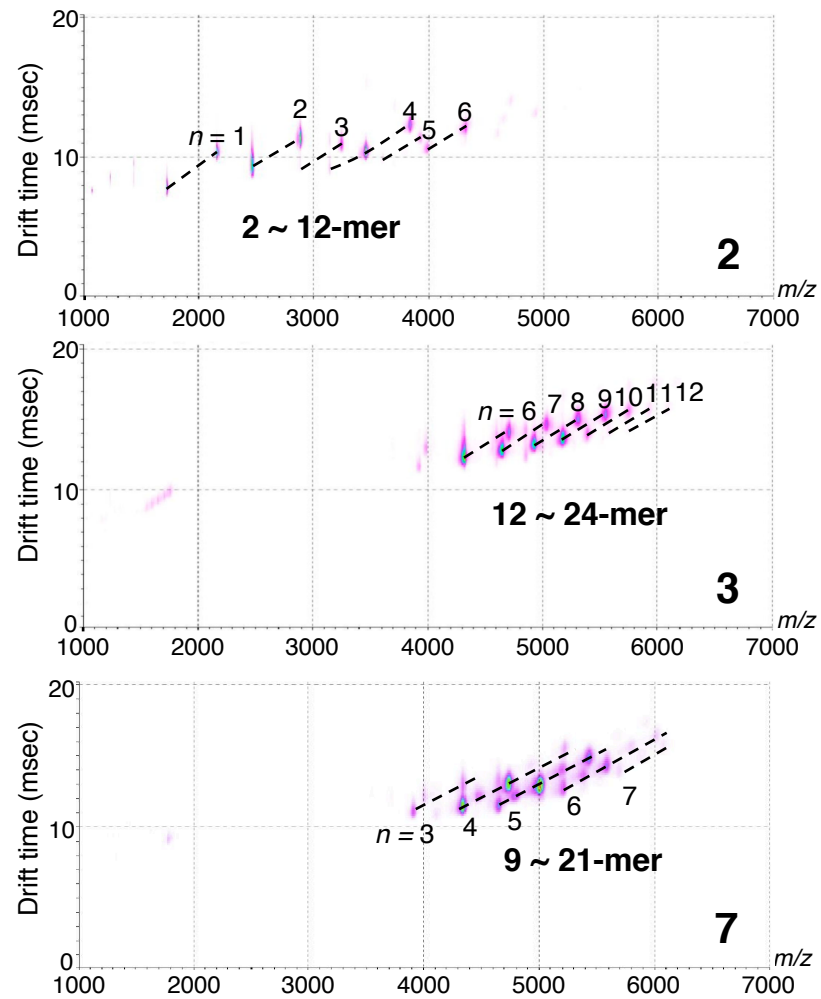


Figure 9

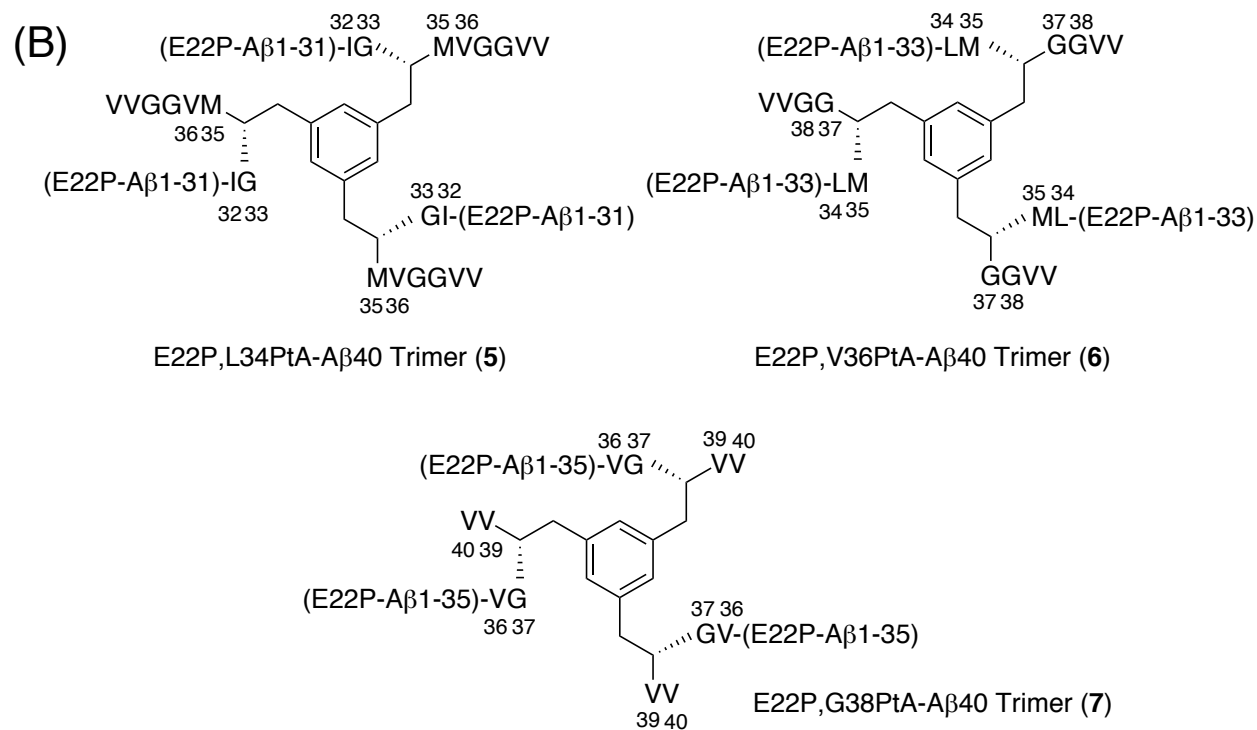
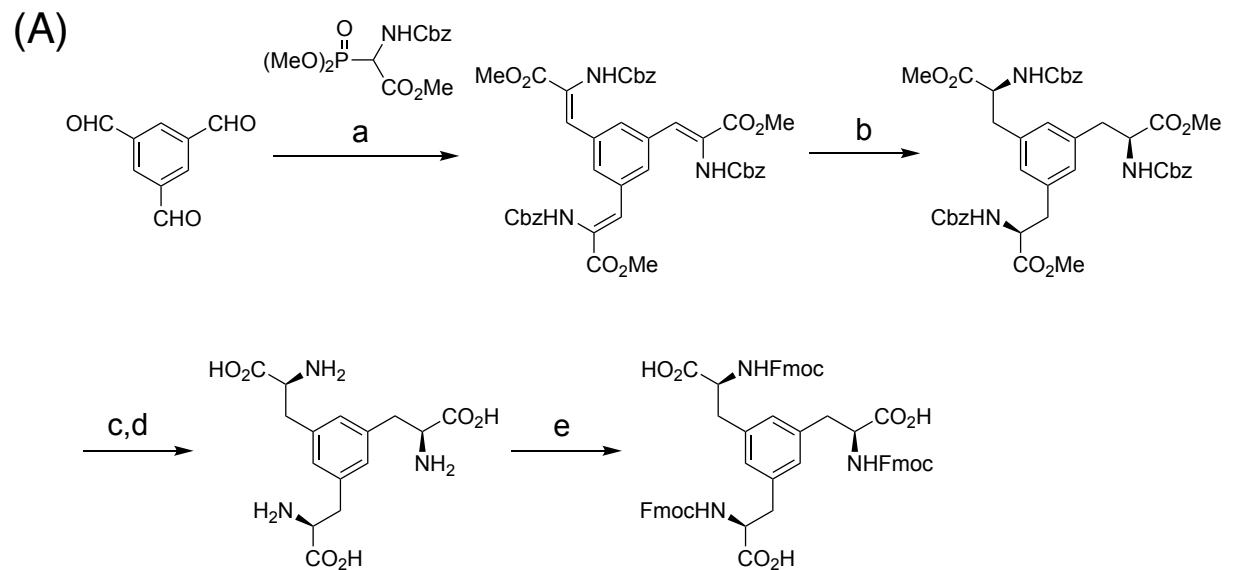


Figure 10

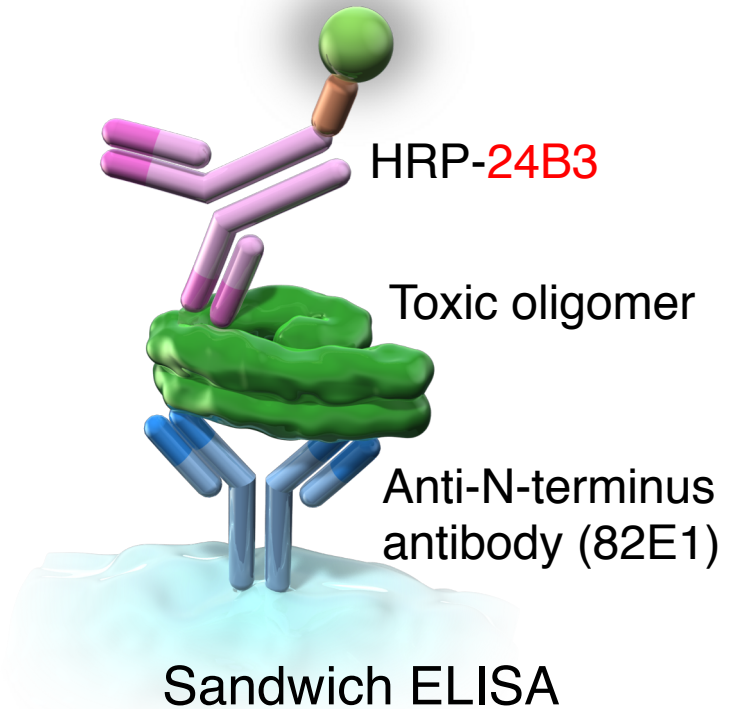
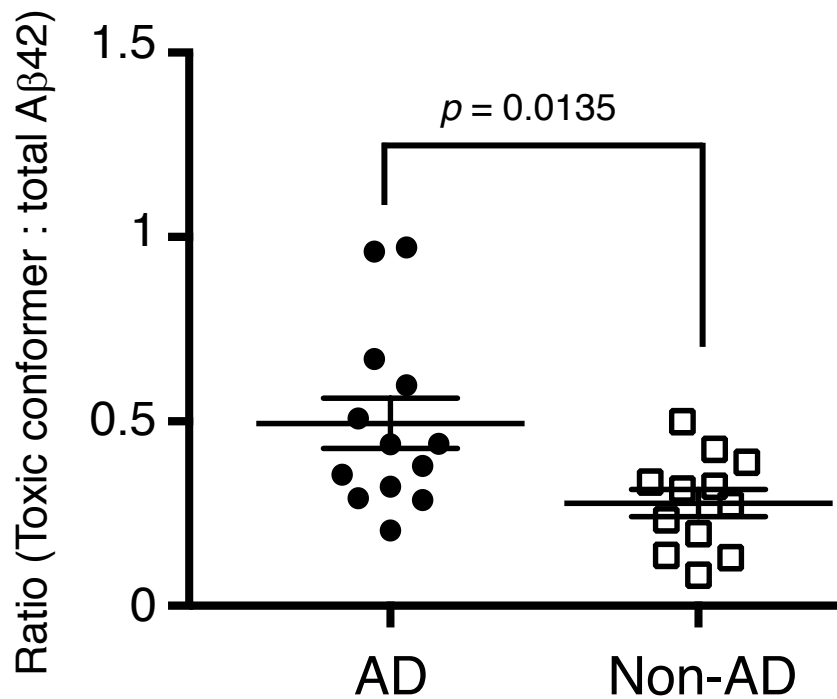
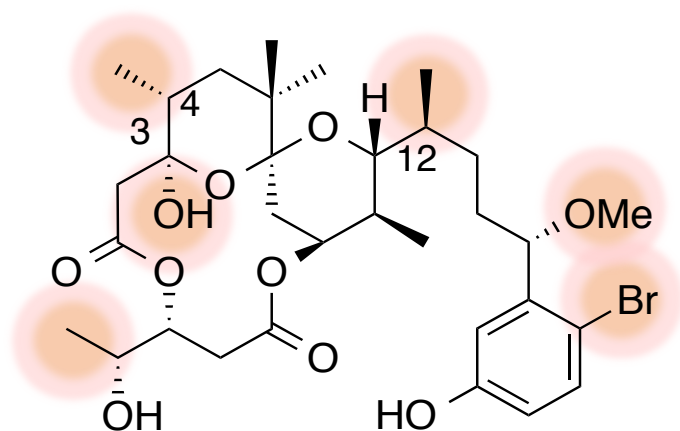
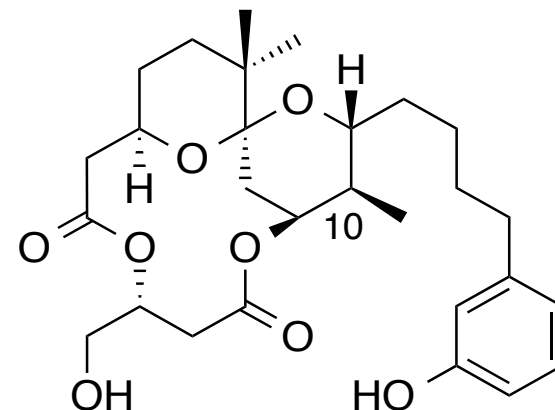


Figure 11



**Extraction of  
anti-proliferative  
activity**



“Master key”

## Aplysiatoxin (ATX)

**Tumor promoting**  
**Proinflammatory**

**Potent PKC ligand**  
**Anti-proliferative activity**

**High toxicity *in vivo***



“Special key”

## 10-Me-aplog-1

**Non-tumor promoting**  
**Non-proinflammatory**

**Potent PKC ligand**  
**Anti-proliferative activity**

**Low toxicity *in vivo***

Figure 12