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Displacement Speed statistics in an Open Turbulent Jet Spray Flame

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18 **ABSTRACT**

19 In this study, a three-dimensional Direct Numerical Simulation of an open turbulent jet spray
20 flame has been used to investigate the statistical behaviour of displacement speed S_d and its
21 components and provide physical explanations for the observed behaviours at different axial
22 locations. The open turbulent jet spray flame exhibits fuel-lean conditions close to the jet inlet
23 but fuel-rich conditions have been observed further downstream due to the evaporation of fuel
24 droplets. Furthermore, for the axial locations considered, combustion takes place under low
25 Damköhler number conditions. The displacement speed of reaction progress variable
26 isosurfaces shows qualitatively similar behaviour for all axial locations considered –
27 predominantly positive across the major part of the flame but with small, potentially negative,
28 values towards the burned gas side. The components of displacement speed arising from
29 chemical reaction rate and flame normal molecular diffusion remain leading order contributors
30 and the competition between these determines the mean behaviour of displacement speed.
31 These observations are consistent with previous studies of turbulent spray flames in a canonical
32 configuration and low Damköhler number turbulent premixed and stratified flames. This
33 suggests that the flow geometry in the absence of mean curvature might not be important in
34 determining the mean behaviour of displacement speed and its components. As a result, the
35 modelling methodologies employed for turbulent stratified flames have the potential to be
36 extended for turbulent spray flames. However, the modelling methodologies, which implicitly
37 assume equality between the surface-weighted values of density-weighted displacement speed
38 and the local laminar burning velocity, might be rendered invalid for turbulent spray flames.

39

40 **Keywords:** Turbulent droplet combustion; Open turbulent jet; Spray flame; Mixture Fraction;
41 Displacement Speed

42 1. INTRODUCTION

43 The combustion of droplet-laden mixtures plays an important role in a number of engineering
44 applications, ranging from Internal Combustion engines (e.g. Compression Ignition and Direct
45 Injection engines) [1,2] to aero gas turbines [2,3] to explosion hazards [4]. In such engineering
46 applications, the fuel is typically delivered into the combustion chamber as a cloud of liquid
47 droplets or as a spray, and the properties of this cloud or spray will have significant influence
48 on the efficiency of combustion, power output and the formation of pollutants. Despite the
49 applicability of the combustion of droplet-laden mixtures, it has been given relatively limited
50 consideration in comparison to the vast body of literature on premixed, non-premixed,
51 partially-premixed and stratified flames. However, a greater level of understanding of turbulent
52 spray combustion is essential to the development of future generations of higher-efficiency,
53 lower-emission combustion devices and to ensure greater control of industrial processes.

54 Significant insights into the behaviour of the combustion of turbulent droplet-laden mixtures
55 through both experimental [5-14] and numerical [14-27] investigations have been obtained.
56 Furthermore, recently several Direct Numerical Simulations (DNS) analyses [19-21,24-27]
57 focussed on the flame propagation statistics in turbulent droplet-laden mixtures in canonical
58 configurations. In these analyses [19-21,24-27], the statistical behaviours of the displacement
59 speed of the reaction progress variable c have been analysed, providing important information
60 with respect to modelling methodologies in turbulent spray flames. These statistics of
61 displacement speed are fundamentally important for flame surface area evolution [28] and both
62 level-set [29], and Flame Surface Density (FSD) [28] based approaches of turbulent reaction
63 rate closure. However, the effects of mean shear were absent in the configurations analysed in
64 [19-21,24-27] and thus it is worthwhile to analyse the flame propagation in a configuration
65 with mean shear, which is typical of laboratory and industrial scale burners. Several recent
66 analyses concentrated on displacement speed statistics of turbulent premixed flames in
67 laboratory-scale burner [30-32] and canonical configurations [33] based on high-fidelity
68 simulations but such an analysis is yet to be carried out for turbulent spray flames. To the best
69 of the authors' knowledge, the analysis of the flame propagation behaviour in the combustion
70 of turbulent droplet-laden mixtures is yet to be considered in detail for an open turbulent jet
71 spray flame [22,23], which is representative of a laboratory-scale experimental configuration
72 [14]. Such an analysis would offer important insights with regards to the propagation behaviour
73 of turbulent spray flames in realistic configurations, which are currently not available.

74 The current analysis builds upon the existing literature of flame propagation into droplet-laden
75 mixtures [19-21,24-27] by considering an open turbulent jet spray flame [22,23], analysing the
76 behaviour of the density-weighted displacement speed S_d^* and its components at different axial
77 locations of the spray flame. The main objectives of the current study are:

- 78 (i) To identify and provide explanations for the statistical behaviours of the density-
79 weighted displacement speed S_d^* of the reaction progress variable c , and its components
80 in the context of an open turbulent jet spray flame.
- 81 (ii) To provide modelling implications for displacement speed statistics in turbulent droplet
82 combustion.

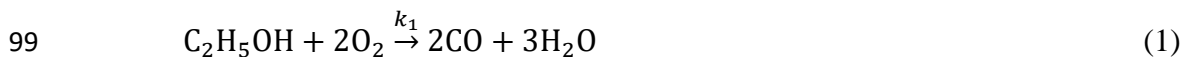
83 The remainder of the paper is organised as follows. The next section discusses the relevant
84 mathematical background and numerical implementation for the current study. Following this,
85 the results are presented and, subsequently, discussed. Finally, the main findings are
86 summarised, and conclusions are drawn.

87

88 **2. MATHEMATICAL BACKGROUND & NUMERICAL IMPLEMENTATION**

89 **2.1 Relevant mathematical background**

90 In the current analysis, the DNS data analysed has been obtained by Pillai and Kurose [22,23]
91 using the FK³ code [22,23,34-42]. The liquid spray fuel is Ethanol (C₂H₅OH) and a two-step
92 global reaction mechanism with 6 species (C₂H₅OH, O₂, CO₂, H₂O, N₂ and CO) is considered
93 for representing combustion process [43] to ensure computational economy. This reaction
94 mechanism was developed by modifying the reaction rate parameters and provides good
95 reproducibility of experimentally measured flame speeds in fuel-air mixtures whilst being able
96 to predict lean and rich flammability limits, flame temperature and burned gas composition
97 with good accuracy across a range of equivalence ratios [43]. The two-step global chemistry
98 can be represented in the following manner:



101 where k_1 is the reaction rate of Ethanol oxidisation in Eq. 1 and k_2 is the rate of forward
 102 reaction for CO oxidisation in Eq. 2. The reaction rates are given as modified Arrhenius
 103 expressions [43] in the following manner [43]:

$$104 \quad k_1 = 1.8 \times 10^{12} \cdot \exp\left(\frac{-30}{RT}\right) [C_2H_5OH]^{0.15} [O_2]^{1.6} \quad (3)$$

$$105 \quad k_2 = 10^{14.6} \cdot \exp\left(\frac{-40}{RT}\right) [CO]^1 [H_2O]^{0.5} [O_2]^{0.25} \quad (4)$$

106 In Eq. 2, the reverse reaction k_{-2} is defined as [43]:

$$107 \quad k_{-2} = 5 \times 10^8 \cdot \exp\left(\frac{-40}{RT}\right) [CO_2]^1 \quad (5)$$

108 The terms in the square brackets of Eqs. 3-5 represent the molar concentrations ($moles \cdot m^{-3}$)
 109 of different chemical species. The molar concentration of $[X_k]$ of the k^{th} species is given as:

$$110 \quad [X_k] = [\rho Y_k / W_k] \quad (6)$$

111 where W_k is the molecular weight of the k^{th} species. The two-step global reaction mechanism
 112 used in the current analysis provides a more accurate representation of the flame parameters
 113 compared to a one-step global reaction mechanism [43].

114 In the current study, the carrier gas-phase is treated as a Eulerian continuum and the dispersed
 115 fuel droplets are tracked as Lagrangian mass points. The gas phase is solved using a Eulerian
 116 framework and the governing equations considered for mass, momentum, energy and species
 117 mass fraction in the following manner:

$$118 \quad \frac{\partial \rho}{\partial t} + \frac{\partial(\rho u_i)}{\partial x_i} = S_\rho \quad (7)$$

$$119 \quad \frac{\partial(\rho u_i)}{\partial t} + \frac{\partial(\rho u_i u_j)}{\partial x_j} = -\frac{\partial p}{\partial x_i} + \frac{\partial \tau_{ij}}{\partial x_j} + S_{\rho u} \quad (8)$$

$$120 \quad \frac{\partial \rho h}{\partial t} + \frac{\partial(\rho h u_j)}{\partial x_j} = \frac{\partial p}{\partial t} + u_j \frac{\partial p}{\partial x_j} + \frac{\partial}{\partial x_j} \left(\rho D_h \frac{\partial h}{\partial x_j} \right) + \tau_{ij} \frac{\partial u_i}{\partial x_j} + S_{rad} + S_{\rho h} \quad (9)$$

$$121 \quad \frac{\partial \rho Y_k}{\partial t} + \frac{\partial(\rho Y_k u_j)}{\partial x_j} = \frac{\partial}{\partial x_j} \left(\rho D_k \frac{\partial Y_k}{\partial x_j} \right) + S_{comb,k} + S_{\rho Y_k} \quad (10)$$

122 Equations 7-10 are considered alongside the equation of state for an ideal gas. In Eqs. 7-10, ρ
 123 is the density, u is the gas-phase velocity, p is the pressure, τ is the stress tensor, h is the
 124 specific enthalpy, Y_k is the mass fraction of the k^{th} chemical species, S_{rad} is the source term
 125 for radiative heat loss, $S_{comb,k}$ is the source term due to combustion reaction, and D_h and D_k

126 are the gaseous thermal diffusivity and mass diffusion coefficient of k^{th} species, respectively,
 127 which are defined as:

$$128 \quad D_h = \frac{\lambda}{\rho c_p} \quad (11)$$

$$129 \quad D_k = \frac{\lambda}{\rho c_p} \quad (12)$$

130 where λ is the thermal conductivity, c_p is the specific heat and unity Lewis number conditions
 131 (i.e. $Le = 1.0$) are considered. It should be noted that the phase coupling between the carrier
 132 gas-phase and dispersed-phase (i.e. fuel droplets) is achieved using the Particle-Source-In-Cell
 133 (PSI-Cell) approach [44]. The PSI-Cell approach considers each computational cell as a control
 134 volume and each fuel droplet is considered as a source of mass, momentum and energy to the
 135 gas-phase. As the fuel droplets evaporate and pass through the cell, the change in their mass,
 136 momentum and energy are considered as a source/sink to the gas-phase mass, momentum and
 137 energy, respectively. This is achieved through the source terms S_ρ , $S_{\rho u}$, $S_{\rho h}$ and $S_{\rho Y_k}$ found in
 138 Eqs. 7-10 and these represent the interactions between the gas-phase and dispersed-phase,
 139 allowing two-way coupling between the two phases. The source terms S_ρ , $S_{\rho u}$, $S_{\rho h}$ and $S_{\rho Y_k}$
 140 are defined in the following manner:

$$141 \quad S_\rho = -\frac{1}{\Delta V} \sum_N \frac{dm_d}{dt} \quad (13)$$

$$142 \quad S_{\rho u} = -\frac{1}{\Delta V} \sum_N \frac{dm_d \vec{u}_d}{dt} \quad (14)$$

$$143 \quad S_{\rho h} = -\frac{1}{\Delta V} \sum_N \frac{dm_d h_d}{dt} \quad (15)$$

$$144 \quad S_{\rho Y_k} = -\frac{1}{\Delta V} \sum_N \frac{dm_d}{dt} \text{ for fuel } (k = F), 0 \text{ for other species } (k \neq F) \quad (16)$$

145 In Eqs. 13-16, ΔV is the volume of each control volume (i.e. each computational grid cell) for
 146 the gas phase calculation, m_d is the fuel droplet mass, \vec{u}_d is the droplet velocity, h_d is the
 147 specific enthalpy of a fuel droplet and N is the number of fuel droplets within a control volume.

148 A non-equilibrium Langmuir-Knudsen evaporation model [45-48] is considered for the
 149 evaporation of the fuel droplets as the non-equilibrium effects become significant for droplet
 150 diameter $d_d < 50\mu m$ [47]. In the current study, the spray flame is dilute as the volumetric
 151 loading of droplets is small and, therefore, the collisions and coalescence of droplets is
 152 neglected. A Lagrangian framework [34,35,46-48] is considered which individually tracks the

153 evaporating fuel droplets of the dispersed phase by solving the equations for droplet position
 154 \vec{x}_d , droplet mass m_d , droplet velocity \vec{u}_d and droplet temperature T_d in the following manner:

$$155 \quad \frac{d\vec{x}_d}{dt} = \vec{u}_d \quad (17)$$

$$156 \quad \frac{dm_d}{dt} = - \left(\frac{Sh}{3Sc} \right) \frac{m_d}{\tau_d} \ln(1 + B_M) \quad (18)$$

$$157 \quad \frac{d\vec{u}_d}{dt} = \frac{f_1}{\tau_d} (\vec{u}(\vec{x}_d, t) - \vec{u}_d) + g \quad (19)$$

$$158 \quad \frac{dT_d}{dt} = \left(\frac{Nu}{3Pr} \right) \left(\frac{c_p}{c_{p,d}} \right) \left(\frac{f_2}{\tau_d} \right) (T - T_d) + \frac{1}{m_d} \left(\frac{dm_d}{dt} \right) \frac{L_V}{c_{p,d}} \quad (20)$$

159 where Sh is the Sherwood number, Sc is the Schmidt number, B_M is the Spalding mass transfer
 160 number, Nu is the Nusselt number, Pr is the Prandtl number, T is the gas-phase temperature,
 161 c_p is the specific heat of the gas mixture, $c_{p,d}$ is the specific heat of a fuel droplet, g is the
 162 gravitational acceleration and the latent heat of vaporisation L_V at T_d is calculated as:

$$163 \quad L_V = L_{V,T_{BL,atm}} \left(\frac{T_{CL} - T_d}{T_{CL} - T_{BL,atm}} \right)^{0.38} \quad (21)$$

164 where $L_{V,T_{BL,atm}}$ is the latent heat of vaporisation at atmospheric pressure, T_{CL} is the critical
 165 temperature of the fuel and $T_{BL,atm}$ is the boiling point of fuel at atmospheric pressure. In Eq.
 166 18, the droplet response time τ_d is calculated by:

$$167 \quad \tau_d = \frac{\rho_d d_d^2}{18\mu} \quad (22)$$

168 where d_d is the droplet diameter, ρ_d is the fuel droplet density and μ is the gas-phase dynamic
 169 viscosity. The gas-phase Prandtl and Schmidt numbers, Nusselt and Sherwood numbers are
 170 defined in the following manner:

$$171 \quad Pr = \frac{\mu c_p}{\lambda} \quad (23)$$

$$172 \quad Sc = \frac{\mu}{\rho D_k} \quad (24)$$

$$173 \quad Nu = 2.0 + 0.552 Re_{sl}^{1/2} \cdot Pr^{1/3} \quad (25)$$

$$174 \quad Sh = 2.0 + 0.552 Re_{sl}^{1/2} \cdot Sc^{1/3} \quad (26)$$

175 where Re_{sl} is the droplet Reynolds number which is based on the slip velocity $U_{sl} =$
 176 $|\vec{u}(\vec{x}_d, t) - \vec{u}_d|$ and is given as:

177
$$Re_{sl} = \frac{\rho U_{sl} d_d}{\mu} \quad (27)$$

178 In Eqs. 19 and 20, the quantities of f_1 and f_2 are the corrections for Stokes drag and heat transfer
 179 for evaporating fuel droplets, respectively [15,16,36,47]. In Eq. 18, the Spalding mass transfer
 180 number B_M is given by:

181
$$B_M = \frac{Y_{F,S} - Y_F}{1 - Y_{F,S}} \quad (28)$$

182 where Y_F is the mass fraction of the fuel vapor on the far-field condition for the droplets (N.B.
 183 the same condition is used for u_i and T) and $Y_{F,S}$ is the fuel vapour mass fraction on the droplet
 184 surface given as:

185
$$Y_{F,S} = \frac{X_{F,S}}{X_{F,S} + (1 - X_{F,S}) W_{avg} / W_F} \quad (29)$$

186
$$X_{F,S} = \frac{p_{sat}}{p} - \left(\frac{2L_k}{d_d} \right) \beta \quad (30)$$

187 where p_{sat} is the saturated vapour pressure, W_{avg} is the average molecular weight of the gas
 188 mixture, W_F is the molecular weight of the fuel vapour, $X_{F,S}$ is the fuel vapour mole fraction at
 189 the droplet surface, for which the non-equilibrium effects are accounted using the Langmuir-
 190 Knudsen evaporation law [45-47]. In Eq. 30, L_K is the Knudsen layer thickness and β is the
 191 nondimensional evaporation parameter given as [46,47]:

192
$$L_K = \frac{\mu [2\pi T_d (R/W_F)]^{1/2}}{Sc.p} \quad (31)$$

193
$$\beta = - \left(\frac{\rho_d Pr}{8\mu} \right) \frac{d(d_d^2)}{dt} \quad (32)$$

194 where R is the universal gas constant ($R = 8.314 J.mol^{-1}.K^{-1}$). It should be noted that the
 195 source term S_{rad} in Eq. 9 accounts for the radiative heat loss rate per unit volume. It is modelled
 196 using an optically thin approximation [49,50] of radiative heat transfer between a fluid element
 197 in the flame and the cold surroundings. The radiative loss S_{rad} is approximated as:

198
$$S_{rad} = 4\sigma (T^4 - T_b^4) [\sum_k p_k a_{pk}] \quad (33)$$

199 where $\sigma = 5.669 \times 10^{-8} W.m^{-2}.K^{-4}$ is the Stefan-Boltzmann constant, T is the gas-phase
 200 temperature, T_b is the background temperature and is assumed to be $300K$, p_k is the partial
 201 pressure of the k^{th} species and $a_{p,k}$ is the Planck mean absorption coefficient of the k^{th}
 202 species. The Planck mean absorption coefficient have been calculated using RADCAL [49]

203 and the curve fits for $a_{p,k}$ for the radiating species considered in this model (i.e. CO_2 , H_2O and
 204 CO) are given as polynomial functions of temperature [49].

205

206 The evaporation of droplets leads to the creation of mixture inhomogeneities that can be
 207 characterised by the mixture fraction ξ , which, for the current study, can be defined as [51]:

$$208 \quad \xi = \frac{\beta - \beta_O}{\beta_f - \beta_O} \quad (34)$$

209 where $\beta_f = 6.0/W_{C_2H_6O}$, $\beta_O = -Y_{O_\infty}/W_O$ and $\beta = 2Y_C/W_C + 0.5Y_H/W_H - Y_{O_\infty}/W_O$; Y_m is
 210 the mass fraction of species m and W_α is the molar mass of element α . It is possible to define
 211 a reaction progress variable c that is based on the oxidiser mass fraction following several
 212 previous analyses [19-21,24-27,52,53]:

$$213 \quad c = \frac{(1-\xi)Y_{O_2\infty} - Y_{O_2}}{(1-\xi)Y_{O_2\infty} - Y_{O_2}^{Eq}} \quad (35)$$

214 where Y_{O_2} is the oxygen mass fraction, $Y_{O_2\infty}$ is the oxygen mass fraction in the pure oxidiser
 215 stream and $Y_{O_2}^{Eq}$ is the equilibrium oxidiser mass fraction (i.e. $Y_{O_2}^{Eq} = f(Y_{O_2}, \xi)$).

216 From Eq. 35, it is possible to derive a transport equation of the reaction progress variable c
 217 based on the transport equations of the oxygen mass fraction Y_{O_2} and the mixture fraction ξ as
 218 [20,21,27]:

$$219 \quad \rho \frac{\partial c}{\partial t} + \rho u_j \frac{\partial c}{\partial x_j} = \nabla \cdot (\rho D \nabla c) + \dot{\omega}_c + \dot{S}_{ev} + \dot{S}_c \quad (36)$$

220 where the first term on the right-hand-side arises due to molecular diffusion, the second term
 221 represents the reaction rate, the third term is the source/sink term arising due to droplet
 222 evaporation, and the final term is the cross-scalar dissipation term arising due to reactant
 223 inhomogeneity [20,21,27,54,55]. The cross-scalar dissipation term \dot{S}_c in Eq. 36 arises due to
 224 mixture inhomogeneity, which in the current case exists due to droplet evaporation [20,27].
 225 According to the definition of c (see Eq. 35), the definitions of $\dot{\omega}_c$, \dot{S}_{ev} and \dot{S}_c depend on the
 226 local value of mixture fraction ξ . The reaction rate of the reaction progress variable $\dot{\omega}_c$ can be
 227 expressed as [20,21,24-27]:

$$\dot{\omega}_c = \begin{cases} -\frac{\xi_{st}\dot{\omega}_{O_2}}{[\xi(1-\xi_{st})Y_{O_2\infty}]} , & \xi \leq \xi_{st} \\ -\frac{\dot{\omega}_{O_2}}{[(1-\xi)Y_{O_2\infty}]} , & \xi > \xi_{st} \end{cases} \quad (37)$$

The expressions for \dot{S}_{ev} and \dot{S}_c are given as [20,21,27]:

$$\dot{S}_{ev} = \begin{cases} \frac{-\xi_{st}}{[\xi^2(1-\xi_{st})Y_{O_2\infty}]} (\xi\dot{S}_O + (Y_{O_2\infty} - Y_{O_2})\dot{S}_\xi) , & \xi \leq \xi_{st} \\ \frac{-1}{[(1-\xi_{st})^2Y_{O_2\infty}]} ((1-\xi)\dot{S}_O + Y_{O_2}\dot{S}_\xi) , & \xi > \xi_{st} \end{cases} \quad (38)$$

$$\dot{S}_c = \begin{cases} \frac{2\rho D}{\xi} \nabla c \cdot \nabla \xi , & \xi \leq \xi_{st} \\ \frac{-2\rho D}{(1-\xi)} \nabla c \cdot \nabla \xi , & \xi > \xi_{st} \end{cases} \quad (39)$$

where $\dot{S}_\xi = (\dot{S}_F - \dot{S}_O/s)/(Y_{F\infty} - Y_{O_2\infty}/s)$ is the droplet source/sink term in the mixture fraction transport equation and $\dot{S}_F = (1 - Y_F)S_\rho$ and $\dot{S}_O = -Y_{O_2}S_\rho$ are the droplet source/sink terms in the fuel and oxygen transport equations, respectively.

The molecular diffusion term (i.e. the first term on the right-hand-side) in Eq. 36 can be split into its normal and tangential components to give the following [56,57]:

$$\nabla \cdot (\rho D \nabla c) = \vec{N} \cdot \nabla (\rho D \vec{N} \cdot \nabla c) - 2\rho D \kappa_m |\nabla c| \quad (40)$$

where D is the progress variable diffusivity, $\vec{N} = -\nabla c/|\nabla c|$ is the flame normal vector, $\kappa_m = 0.5(\nabla \cdot \vec{N})$ is the arithmetic mean of the two principal curvatures of a given iso-surface $c = c^*$. It should be noted that the first term on the right-hand-side of Eq. 40 provides the component of the molecular diffusion normal to the flame front and the second term gives the tangential molecular diffusion component.

The transport equation of c can be rewritten in kinematic form as [20,21,27]:

$$\frac{\partial c}{\partial t} + u_j \frac{\partial c}{\partial x_j} = S_d |\nabla c| \quad (41)$$

where S_d is the displacement speed which is the speed at which a given reaction progress variable c iso-surface moves normal to itself with respect to an initially coincident material surface. Comparing Eqs. 36 and 41 gives [20,21,27]:

$$S_d = \frac{\nabla \cdot (\rho D \nabla c) + \dot{\omega}_c + \dot{S}_c + \dot{A}_c}{\rho |\nabla c|} = \frac{\vec{N} \cdot \nabla (\rho D \vec{N} \cdot \nabla c) - 2\rho D \kappa_m |\nabla c| + \dot{\omega}_c + \dot{S}_c + \dot{A}_c}{\rho |\nabla c|} \quad (42)$$

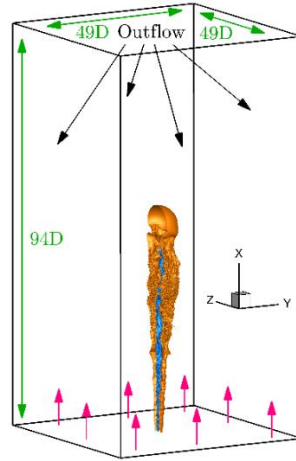
249 This can be rewritten in the following manner [20,21,27]:

$$250 \quad S_d = \underbrace{\frac{\vec{N} \cdot \nabla (\rho D \vec{N} \cdot \nabla c)}{\rho |\nabla c|}}_{S_n} - \underbrace{\frac{2\rho D \kappa_m |\nabla c|}{\rho |\nabla c|}}_{S_t} + \underbrace{\frac{\dot{\omega}_c}{\rho |\nabla c|}}_{S_r} + \underbrace{\frac{\dot{S}_{ev}}{\rho |\nabla c|}}_{S_{ev}} + \underbrace{\frac{\dot{S}_c}{\rho |\nabla c|}}_{S_c} \quad (43)$$

251 Furthermore, as displacement speed is affected by thermal expansion through its density
 252 dependence, it is worthwhile to consider the density-weighted displacement speed S_d^* as it is
 253 often needed for the modelling purposes [20,21,27,56,57]:

$$254 \quad S_d^* = \underbrace{\frac{\vec{N} \cdot \nabla (\rho D \vec{N} \cdot \nabla c)}{\rho_0 |\nabla c|}}_{S_n^*} - \underbrace{\frac{(2\rho D \kappa_m)}{\rho_0}}_{S_t^*} + \underbrace{\frac{\dot{\omega}_c}{\rho_0 |\nabla c|}}_{S_r^*} + \underbrace{\frac{\dot{S}_{ev}}{\rho_0 |\nabla c|}}_{S_{ev}^*} + \underbrace{\frac{\dot{S}_c}{\rho_0 |\nabla c|}}_{S_c^*} \quad (44)$$

255 where ρ_0 is the unburned reactant density. Accordingly, the statistical behaviours of the terms
 256 $\vec{N} \cdot \nabla (\rho D \vec{N} \cdot \nabla c)$, $(-2\rho D \kappa_m |\nabla c|)$, $\dot{\omega}_c$, \dot{S}_{ev} and \dot{S}_c and their combined contributions will be
 257 discussed in detail in Section 3 of this paper.



258
 259 **Fig. 1:** Schematic of the computational domain and Direct Numerical Simulation set-up. The central
 260 blue surface represents the spray droplet injection whereas the orange represents the annular
 261 pilot. The pink arrows represent the co-flow.

262

263 2.2 Considered DNS case and computational configuration

264 In the current analysis, the DNS configuration corresponds to the experimental study of the
 265 Ethanol spray EtF3 flame of Gounder et al. [14]. The configuration of the EtF3 flame is shown
 266 schematically in Fig. 1. The spray and carrier gas are injected from a central jet nozzle ($D_j =$
 267 10.5mm) with a bulk velocity $U_j = 24\text{m} \cdot \text{s}^{-1}$ surrounded by a coaxial pilot annulus ($U_p =$
 268 $11.6\text{m} \cdot \text{s}^{-1}$ and $T_p = 2493\text{K}$) and an air co-flow ($U_c = 4.5\text{m} \cdot \text{s}^{-1}$). The pilot is composed of
 269 the fully burned products of a stoichiometric mixture of 5.08% Acetylene (C_2H_2), 10.17%

270 Hydrogen (H₂) and 84.75% air by volume. This pilot provides the heat necessary for the
 271 evaporation of the liquid fuel droplets. The flame is stabilised in the shear layer that is formed
 272 between the inner jet and pilot streams. The mass flow rate of liquid Ethanol in the jet is 45
 273 g/min. However, amongst the polydisperse droplets formed by the nebulizer, some of the
 274 droplets evaporate before reaching the exit of the nozzle and, thus, explains the presence of
 275 partially gaseous fuel in the jet. The Ethanol mass flow rates at the nozzle exit are 14.3 g/min
 276 for the gaseous phase and 30.7 g/min for the liquid droplets, giving a gaseous equivalence ratio
 277 of 0.85. These parameters are summarised in Tables 1 and 2 for both the inner jet, pilot and co-
 278 flow jet streams.

279 **Table 1:** Flow parameters for central jet at burner exit [21-24]

| Flame Designation | Etf3 |
|--|------------------------|
| Fuel | Ethanol |
| Jet Diameter (D_j) [mm] | 10.5 |
| Bulk Jet Velocity (U_j) [$m \cdot s^{-1}$] | 24 |
| Bulk Coflow Stream Velocity (U_c) [$m \cdot s^{-1}$] | 4.5 |
| Carrier air mass flow rate [$g \cdot min^{-1}$] | 150 |
| Liquid Fuel Injection Rate [$g \cdot min^{-1}$] | 45 |
| Measured liquid flow at exit [$g \cdot min^{-1}$] | 30.7 |
| Vapour fuel flow rate at exit [$g \cdot min^{-1}$] | 14.3 |
| Kinematic viscosity (ν) [$m^2 \cdot s^{-1}$] | 1.279×10^{-5} |
| Jet Reynolds number, $Re = U_j D_j / \nu$ [-] | 19,700 |
| Jet Mach number, $M = U_j / c_\infty$ [-] | 0.07 |
| Equivalence ratio at jet exit, ϕ_j [-] | 0.85 |
| Initial droplet and ambient temperature (T_0) [K] | 293.15 |

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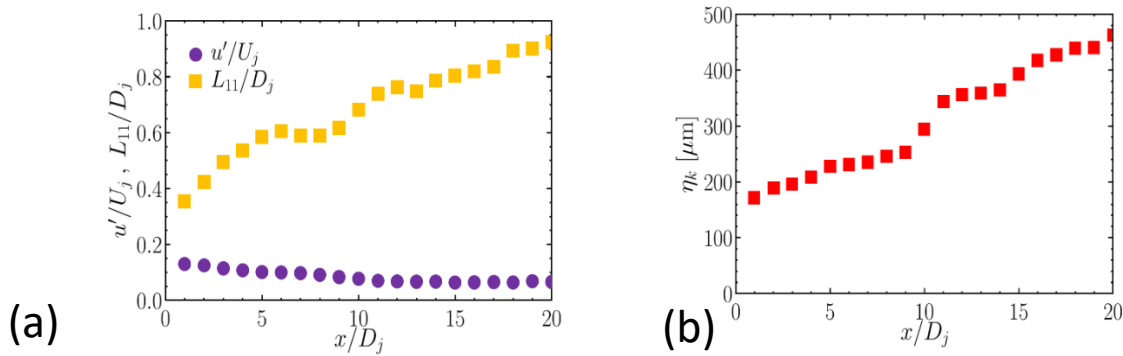
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Table 2: Flow parameters for annular pilot at burner exit [21-24]

| | |
|--|---|
| Flame Designation | Etf3 |
| Fuel | Acetylene (C_2H_2) + Hydrogen (H_2) + Air |
| Pilot Diameter (D_p) [mm] | 25 |
| Bulk Pilot Velocity, burned (U_p) [$m \cdot s^{-1}$] | 11.6 |
| Pilot temperature (T_p) [K] | 2493 |
| Pilot composition ($Y_{CO_2} : Y_{H_2O} : Y_{N_2}$) | (0.1722 : 0.10575 : 0.722) |

286



287

Fig. 2: The evolution of (a) the turbulence intensity u'/U_j and normalised integral length scale L_{11}/D_j , and (b) Kolmogorov length scale η_k along the shear layer.

289

290

291 For the considered simulation, the particles are modelled as spherical point masses which
 292 exchange mass, heat, and momentum with the carrier gas. A polydisperse spray with a
 293 distribution of diameters matching that of the experiment [14] is injected with droplet diameters
 294 ranging from 1 μm to 80 μm , with the most probable diameter being about 20 μm . In the current
 295 study, both the collisions and break-up have been neglected, since it is a dilute spray flame,
 296 with an inflow droplet volume fraction of about 5×10^{-4} . The droplet spray is generated by
 297 an ultrasonic nebulizer situated inside the burner, 215mm upstream of the exit plane [14].
 298 Therefore, it is likely that the secondary atomization occurs inside the central jet tube of the
 299 burner, in which case the probability density function (PDF) of droplet size distribution
 300 imposed as the inflow boundary condition at the exit plane in the DNS should be sufficient,
 301 since further secondary atomization effects can be neglected. Moreover, no evidence of
 302 secondary atomization has been provided experimentally, hence it was not accounted for in the
 303 DNS. A recent analysis [58] also compared the combustion of polydisperse droplets in a two-
 304 dimensional free jet simulated using either a carrier phase DNS with point source or a fully

305 Eulerian phase-DNS where good agreement was found when comparing the gaseous fuel mass
306 fraction fields.

307

308 For the simulation considered in this work, a domain of $94D_j \times 49D_j \times 49D_j$ (where D_j is the
309 nozzle diameter) is used and is discretised by a non-uniform Cartesian grid of size $1160 \times$
310 400×400 . A large stretching is applied in all directions towards the boundaries to form
311 absorbing zones that minimize reflection and contamination of the acoustic field near the jet
312 [22,23]. The minimum cell size needs to be larger than the droplet size to capture evaporation
313 accurately which is due to the coupling strategy between the Eulerian and Lagrangian phases.
314 In order to guarantee an appropriate resolution of both the turbulence and the premixed flame
315 front, the smallest cell size at the nozzle exit is $\Delta x = 150\mu m$. For interested readers, further
316 details on the boundary conditions and computational grid can be found in [22,23,39]. The
317 integral length-scale and velocity fluctuations are evaluated within the shear layer and are
318 reported in Fig. 2a. As expected, the turbulence intensity decreases and the integral length scale
319 increases as turbulence decays in the downstream direction. The evolution of Kolmogorov
320 length scale η_k is also reported in Fig. 2b, where it can be seen that the Kolmogorov length
321 scale η_k increases continuously downstream from $\eta_k \approx 170\mu m$ at the nozzle lip. The largest
322 value of the ratio $\Delta x/\eta_k$ is thus about $\Delta x/\eta_k \approx 1.35$ at the lip, which is within the range
323 recommended by Pope [59]. It should be noted that the DNS simulation results have been
324 compared against the experimental data of Gounder et al. [14] at different axial distances from
325 the nozzle [23]. Good agreement has been found between the experimental and
326 computational results, and the interested readers are referred to [23,39] for a detailed discussion
327 of these issues which will not be repeated here for the sake of brevity.

328

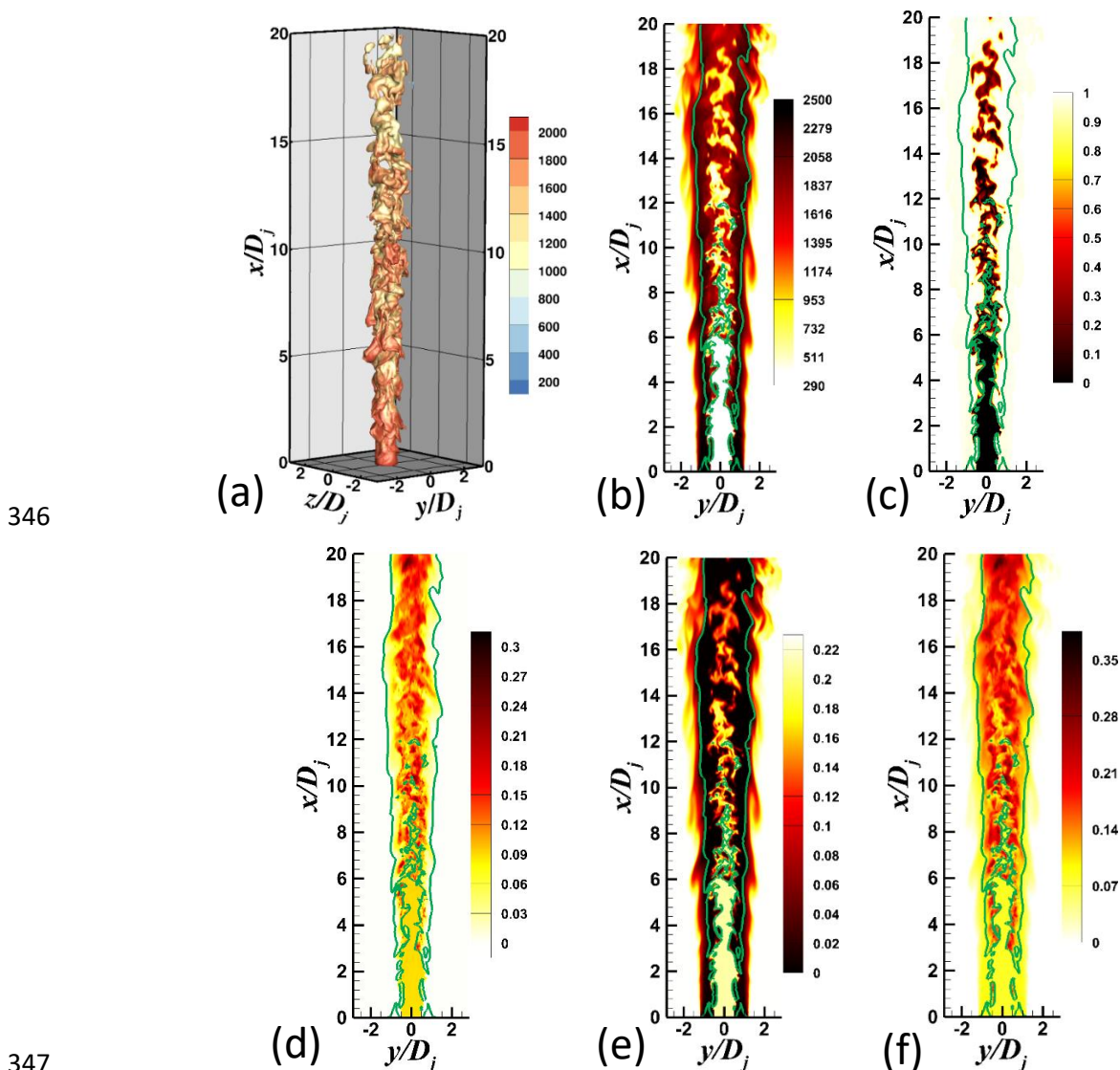
329 **3. RESULTS & DISCUSSION**

330 **3.1 Flame behaviour**

331 Figure 3a shows the instantaneous iso-surface of reaction progress variable $c = 0.8$ coloured
332 with temperature T . It is evident from Fig. 3a that significant wrinkling of the jet flame occurs
333 due to flame-turbulence and flame-droplet interactions. Furthermore, that the variations in
334 temperature are indicative of the changes in burning rates due to the variations in equivalence
335 ratio caused by droplet evaporation. The instantaneous fields on the central x-y plane of
336 temperature T , reaction progress variable c , fuel mass fraction $Y_{C_2H_5OH}$, oxygen mass fraction

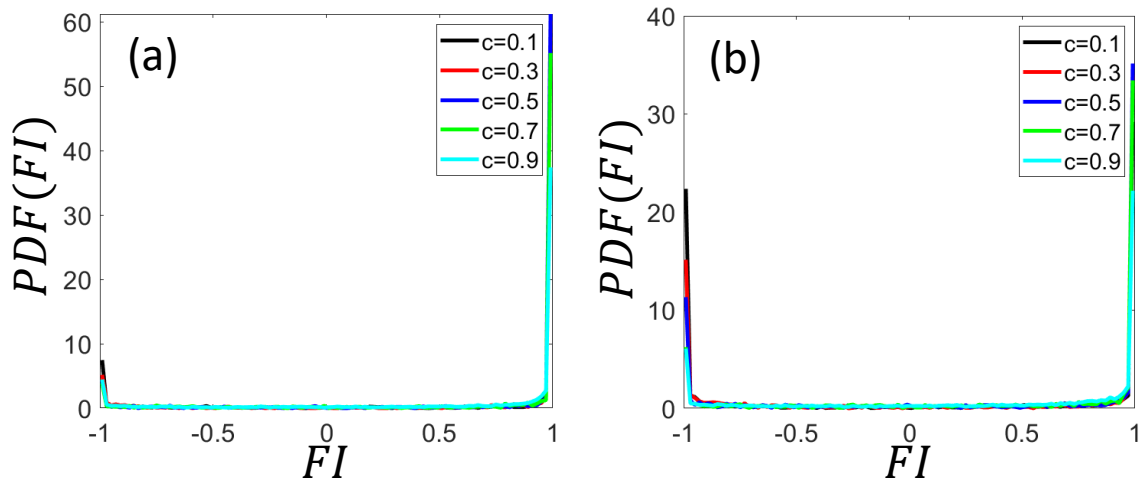
337 Y_{O_2} , and mixture fraction ξ are shown in Figs. 3b, c, d, e and f, respectively, along with green
 338 lines indicating the stoichiometric mixture fraction $\xi_{st} = 0.0914$. Figures 3b-f further
 339 demonstrate the large amounts of wrinkling of the jet flame, which is particularly evident
 340 further downstream of the jet exit (i.e. from $x/D_j \approx 5$ onwards). Figure 3b shows the injection
 341 of cold gaseous fuel in the inner jet whilst the temperature T increases as the inner jet mixes
 342 with the pilot stream, and droplet evaporation can be observed. The droplet evaporation
 343 happens relatively quickly for the small droplets, with small regions of gaseous fuel-rich
 344 mixture visible close to the jet exit (e.g. $x/D_j \approx 2$), as shown in Figs. 3d and 3f.

345

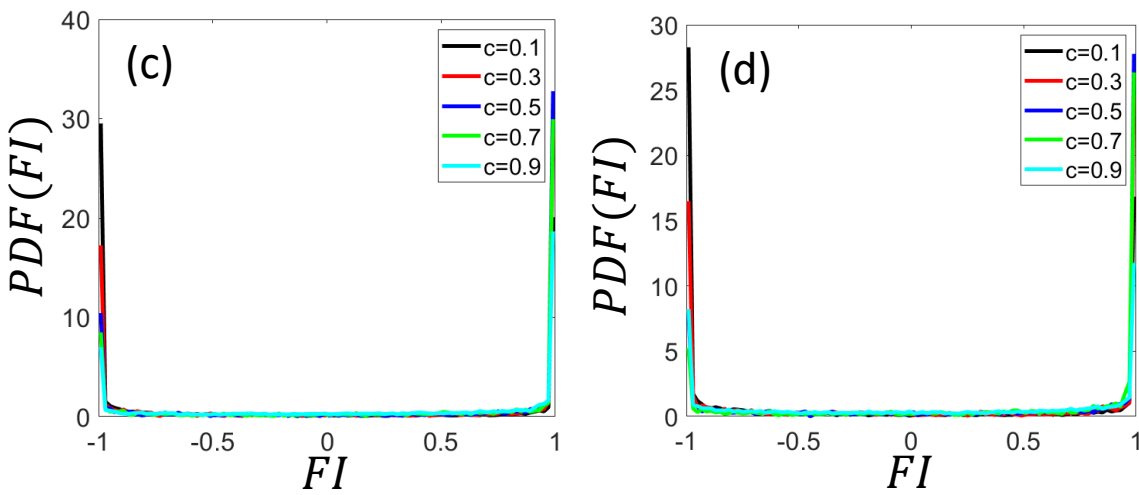


348 **Fig. 3:** Instantaneous plots of (a) reaction progress variable $c = 0.8$ iso-surface coloured with
 349 temperature T [K], (b) temperature T [K] on the central x-y plane, (c) reaction progress
 350 variable c , (d) Ethanol C_2H_6O mass fraction on the central x-y plane, (e) oxygen O_2 mass
 351 fraction at the central x-y plane, and (f) mixture fraction ξ at the central x-y plane. In (b)-(f)
 352 the green lines indicate the stoichiometric mixture fraction $\xi_{st} = 0.0914$ contours.

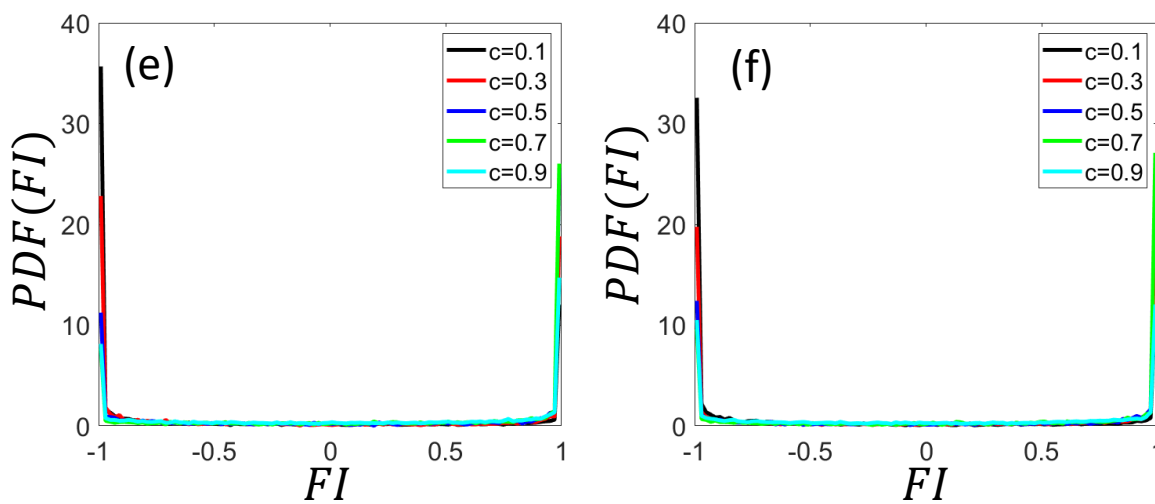
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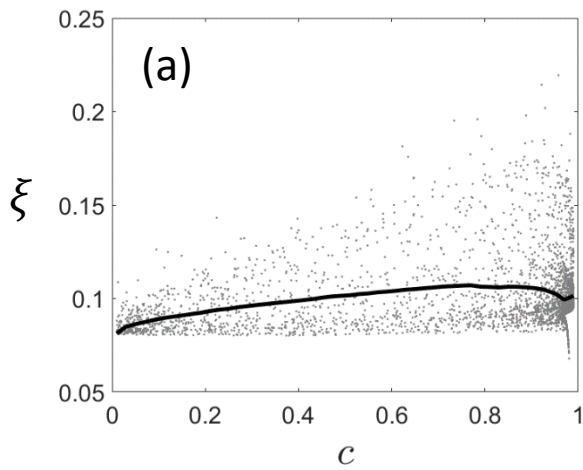


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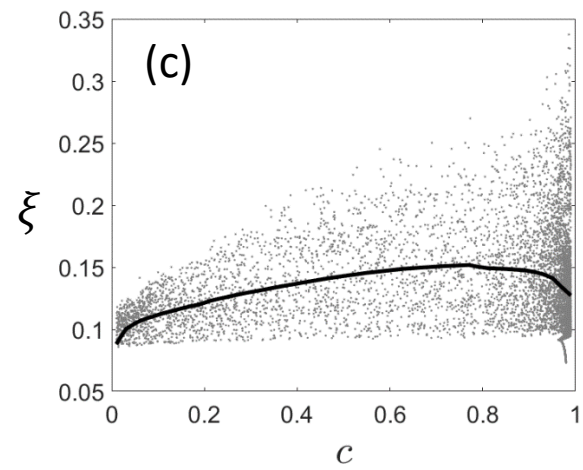
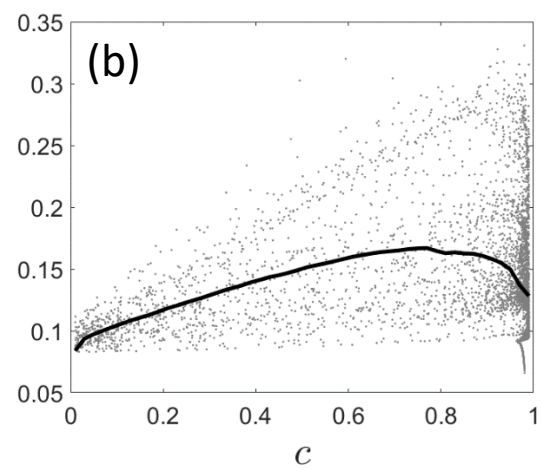
358 **Fig. 4:** Probability density functions of Flame Index (i.e. $FI = \nabla Y_{C_2H_6O} \cdot \nabla Y_{O_2} / [|\nabla Y_{C_2H_6O}| |\nabla Y_{O_2}|]$) for
 359 $c = 0.1, 0.3, 0.5, 0.7$ and 0.9 , at (a) $x = 2D_j$, (b) $x = 4D_j$, (c) $x = 6D_j$, (d) $x = 8D_j$, (e) $x =$
 360 $10D_j$, and (f) $x = 12D_j$.

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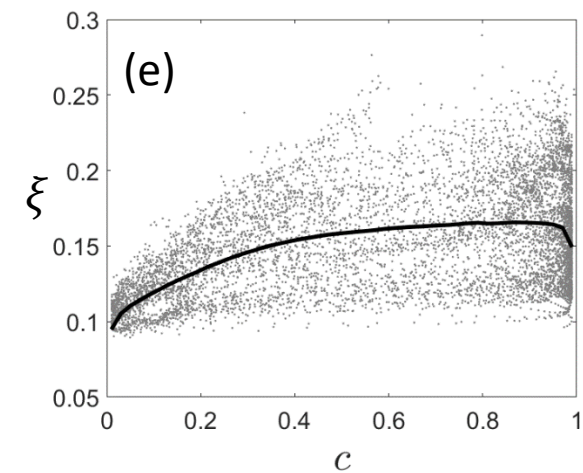
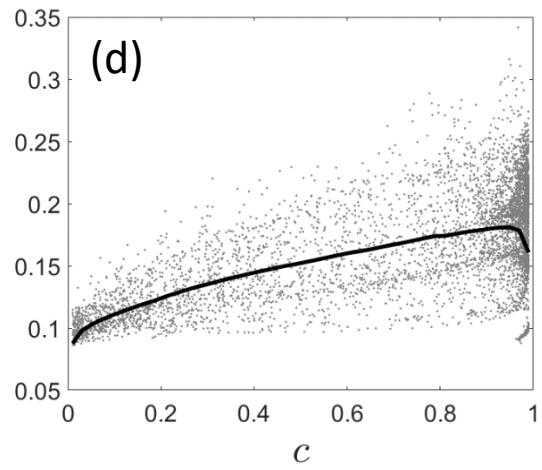
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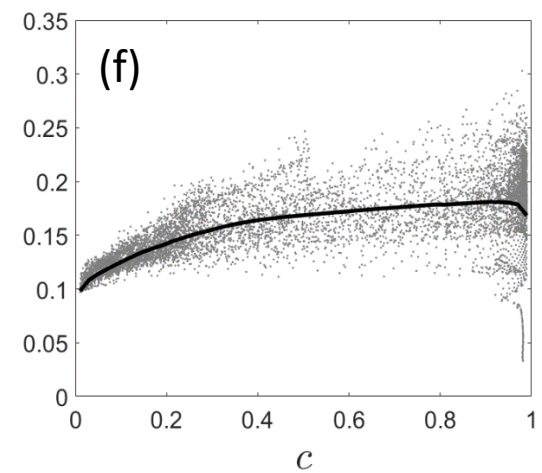
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365



366 **Fig. 5:** Scatter of mixture fraction ξ (grey dots) with reaction progress variable c and variations
367 of the mean value of mixture fraction ξ conditioned upon c (black line) at (a) $x = 2D_j$, (b)
368 $x = 4D_j$, (c) $x = 6D_j$, (d) $x = 8D_j$, (e) $x = 10D_j$, and (f) $x = 12D_j$.

369

370 Further downstream, large droplets also evaporate and give rise to larger hot regions of gaseous
 371 fuel-rich mixtures at approximately $x/D_j \approx 5$, as shown in Figs. 3d and 3f. The largest droplets
 372 do not evaporate until far downstream of the jet exit and the evidence of evaporation can be
 373 observed as far as $x/D_j \approx 20$, which is not shown here. The evaporation process occurring in
 374 the mixing layer is visible in Fig. 3f in the mixture ξ field, which increases continuously from
 375 the nozzle lip and shows large values of ξ/ξ_{st} up to $\xi/\xi_{st} = 2.0$ at $x/D_j = 15$ and $\xi/\xi_{st} = 2.5$
 376 at $x/D_j = 20$ before decreasing slowly due to mixing. Further downstream (i.e. $x/D_j > 10$),
 377 around the pockets of very high fuel content created by the droplet evaporation, as shown in
 378 Figs. 3d and 3f, the burning occurs increasingly in a non-premixed mode because the hot fuel
 379 does not have the time to fully mix with the surrounding air leading to partial-premixing, which
 380 is characteristic of spray flames. In addition, an animation has been provided in the
 381 supplementary material to accompany this paper which shows the mid-plane of the jet coloured
 382 with temperature with the spray particles on that plane coloured by the evaporation rate. The
 383 nature of the combustion (e.g. premixed, non-premixed) can be characterised by considering a
 384 Flame Index, FI , defined as $FI = \nabla Y_{C_2H_6O} \cdot \nabla Y_{O_2} / [|\nabla Y_{C_2H_6O}| |\nabla Y_{O_2}|]$ [60]. A Flame Index value
 385 of $FI = -1.0$ indicates non-premixed mode of combustion, whereas a Flame Index value of
 386 $FI = 1.0$ indicates premixed mode of combustion. The PDFs of Flame Index at different
 387 reaction progress variables (i.e. $c = 0.1, 0.3, 0.5, 0.7$ and 0.9) at $x/D_j = 2, 4, 6, 8, 10$ and 12
 388 are shown in Figs. 4a-f, respectively. It can be seen from Fig. 4 that close to the nozzle exit
 389 (e.g. $x/D_j = 2$) the premixed mode of combustion remains dominant across the flame.
 390 However, moving further downstream (i.e. $x/D_j = 4, 6, 8, 10, 12$) greater contributions of non-
 391 premixed mode of combustion can be seen towards the unburned gas side of the flame (i.e. $c =$
 392 0.1) due to the greater number of droplets beginning to evaporate downstream. The non-
 393 premixed mode of combustion decreases (i.e. the PDF peak at $FI = -1.0$ decreases) with
 394 increasing c , as mixing progressively takes place within the flame. This supports the
 395 observations made earlier in Figs. 3d and 3f.

396

397 The scatters of mixture fraction ξ with c as well as the conditional average at $x/D_j = 2, 4, 6,$
 398 $8, 10$ and 12 are shown in Figs. 5a-f, respectively. In the current and subsequent sections, the
 399 mean values conditional upon c are determined by considering the ensemble averaged value of
 400 the quantity being considered on a given c -isosurface. It is evident from Figs. 5a-f that close to
 401 the jet exit there are significant fuel-lean contributions (see Fig. 5a) whilst moving further

402 downstream (see Figs. 5b-f) significant fuel-rich contributions are found due to droplet
 403 evaporation. These observations are consistent with those found in Figs. 3d and f. It should be
 404 noted that if one examines Fig. 2a and b, at the axial locations considered (i.e. $x/D_j = 2, 4, 6,$
 405 $8, 10$ and 12) the Damköhler number would remain low under stoichiometric conditions (i.e.
 406 ranging from $Da = 0.9$ to 2.5). Therefore, considering Figs. 2a and b alongside combustion of
 407 either fuel-lean or fuel-rich mixtures, it is evident that low Damköhler number conditions are
 408 prevalent here. These conditions must be considered when investigating the behaviour of the
 409 displacement speed S_d and its components, particularly S_{ev} and S_c .

410

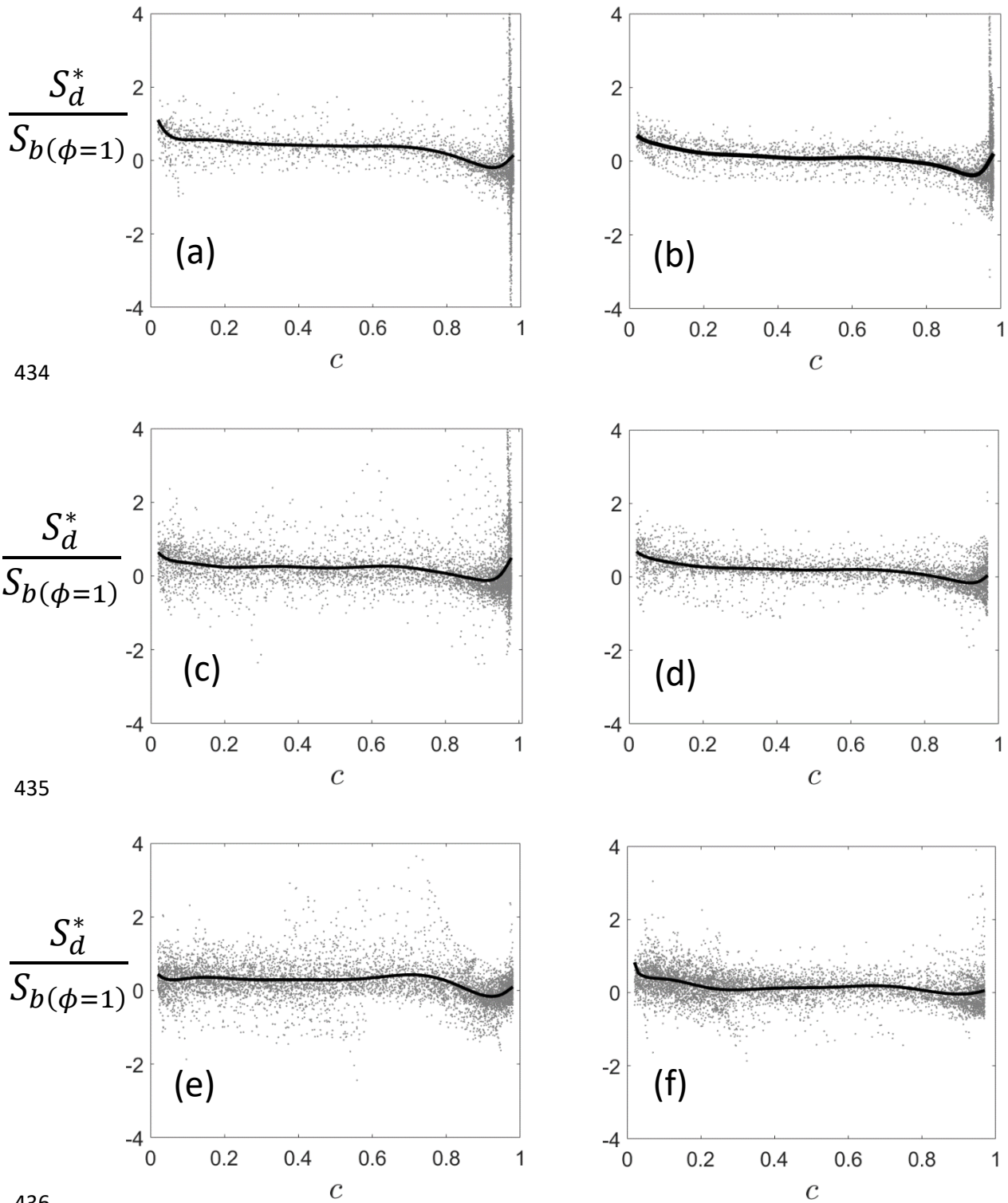
411 **3.2 Density-weighted displacement speed S_d^* behaviour**

412 The scatters of density-weighted displacement speed $S_d^*/S_{b(\phi=1)}$ (where $S_{b(\phi=1)}$ is the laminar
 413 burning speed of the stoichiometric mixture) with c as well as the conditional average at
 414 $x/D_j = 2, 4, 6, 8, 10$ and 12 are shown in Figs. 6a-f, respectively. It is evident from Figs. 6a-
 415 f that the density-weighted displacement speed S_d^* exhibits similar qualitative behaviour at all
 416 axial locations considered. It can be seen from the scatters in Figs. 6a-f that the density-
 417 weighted displacement speed S_d^* can exhibit both positive and negative values across c but is
 418 generally positive as shown by the variations of the mean values conditional upon c . Generally,
 419 larger positive values towards the unburned gas side falling towards the burned gas side and
 420 potentially exhibiting negative conditionally averaged values around $c = 0.9$ at all axial
 421 locations considered. A negative value of S_d^* indicates that the flame retreats into the burned
 422 gas instead of propagating into the unburned reactants.

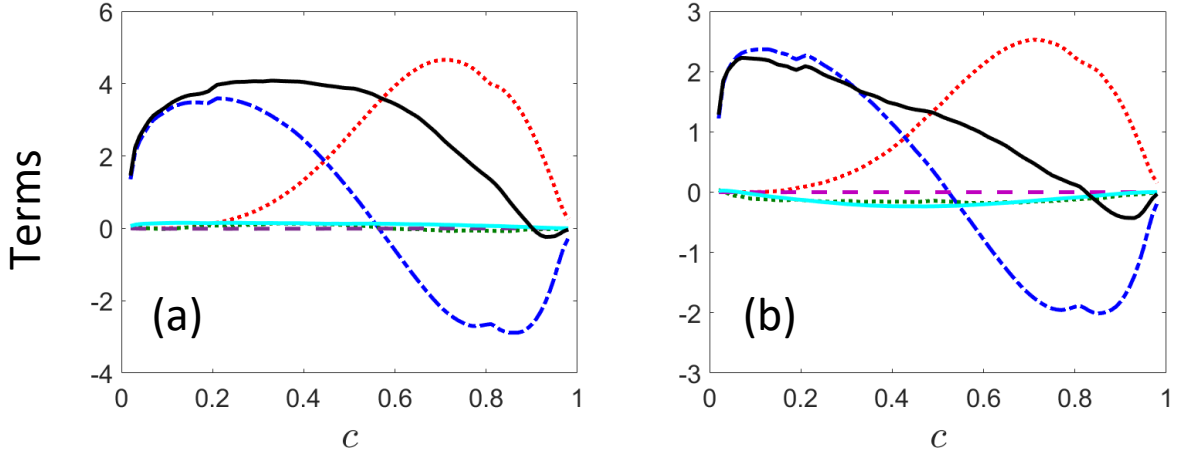
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424 The sign of the displacement speed S_d is same as that of S_d^* , and, therefore, these plots are not
 425 shown here for the sake of brevity. A combination of positive mean values of S_d^* towards the
 426 unburned gas side and negative mean values on the burned gas side suggests thickening of the
 427 flame, and instances of local flame thickening can be discerned from Fig. 3b. The observed
 428 behaviour here is consistent with observations previously made for low Damköhler number
 429 premixed and stratified gaseous flames [55]. The observed qualitative and quantitative
 430 behaviours of the displacement speed S_d and density-weighted displacement speed S_d^* at
 431 different axial locations can be explained in terms of the contributions of $\vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c)$,
 432 $(-2\rho D \kappa_m |\nabla c|)$, $\dot{\omega}_c$, \dot{S}_{ev} and \dot{S}_c .

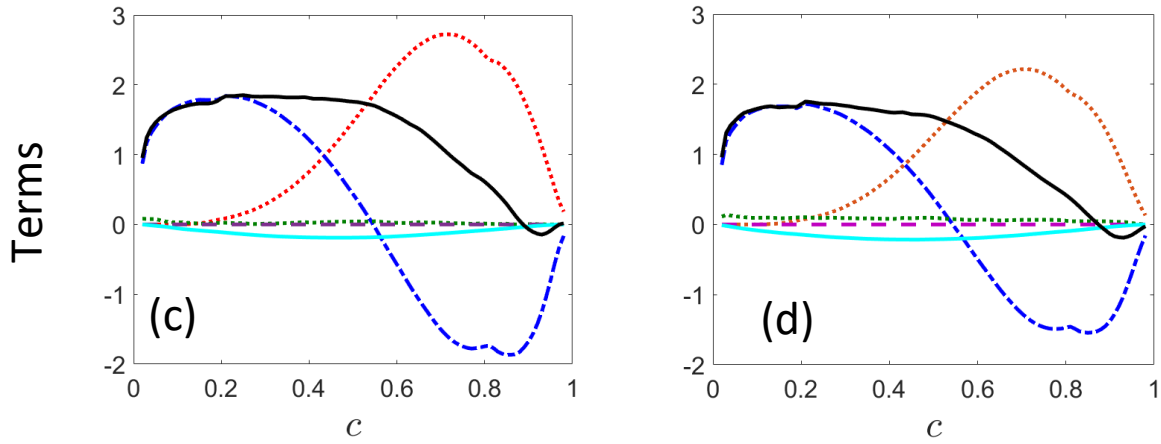
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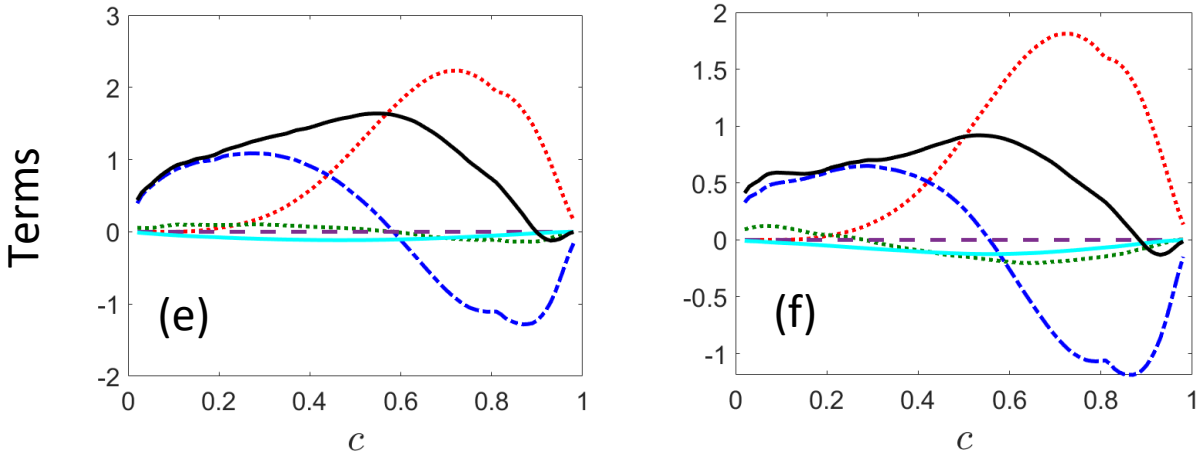
437 **Fig. 6:** Scatter of the variations of density-weighted displacement speed S_d^* (grey dots) with
 438 reaction progress variable c and mean values of density-weighted displacement speed
 439 conditioned upon c (black line) at (a) $x = 2D_j$, (b) $x = 4D_j$, (c) $x = 6D_j$, (d) $x =$
 440 $8D_j$, (e) $x = 10D_j$, and (f) $x = 12D_j$. All quantities are normalised by normalised
 441 using the unstrained laminar burning velocity of the stoichiometric mixture $S_{b(\phi=1)}$.



442



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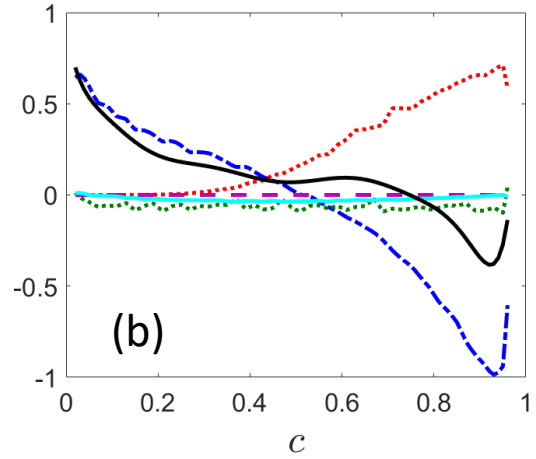
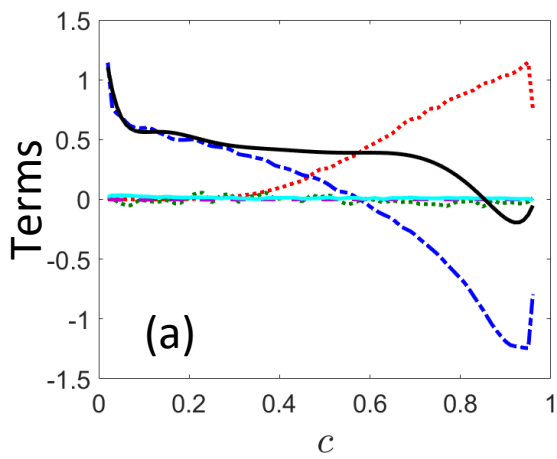


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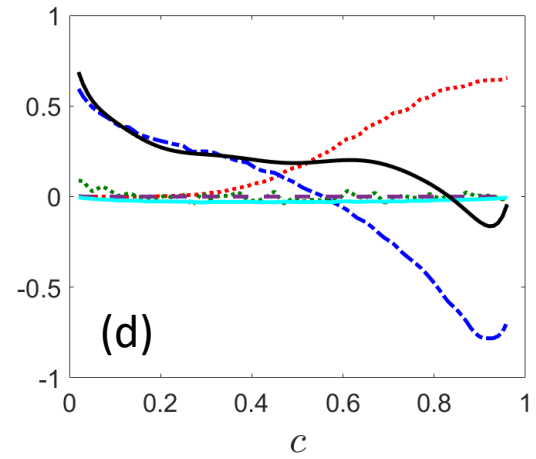
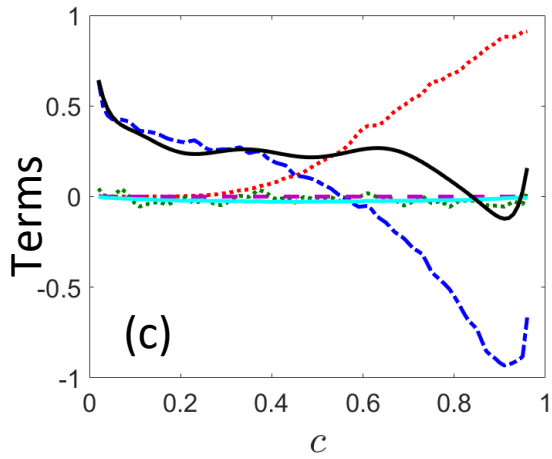
445 **Fig. 7:** Variations of the mean values of $\vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c)$ [- - -], $(-2\rho D \kappa_m |\nabla c|)$ [····], $\dot{\omega}_c$
 446 [····], \dot{S}_{ev} [- - -] and \dot{S}_c [—] as well as the combined contribution of the
 447 contributions [—] conditioned upon reaction progress variable c at (a) $x = 2D_j$,
 448 (b) $x = 4D_j$, (c) $x = 6D_j$, (d) $x = 8D_j$, (e) $x = 10D_j$, and (f) $x = 12D_j$. All quantities
 449 are normalised using $\rho_0 S_b(\phi=1)/\delta_{th(\phi=1)}$ where $\delta_{th(\phi=1)}$ and $S_b(\phi=1)$ are the thermal
 450 flame thickness and unstrained laminar burning velocity of the stoichiometric mixture,
 451 respectively.

452

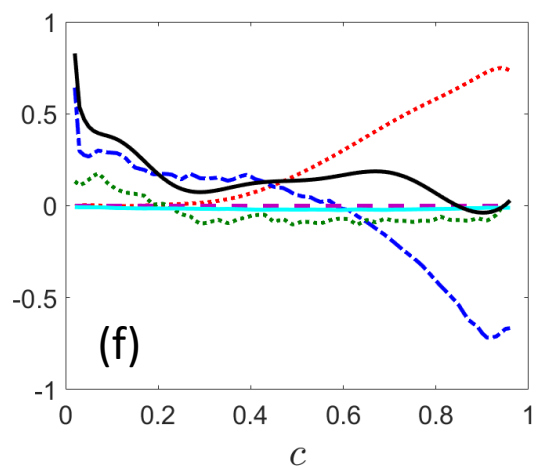
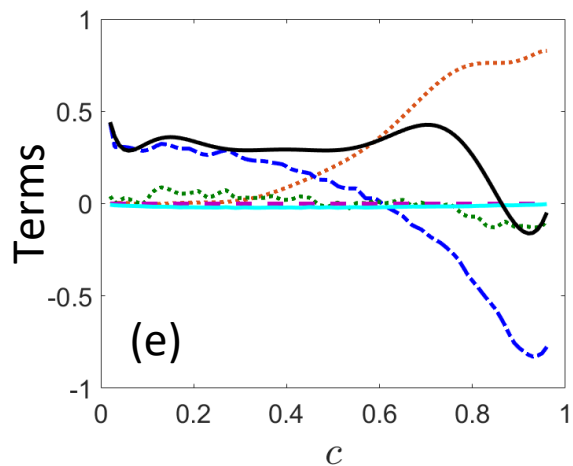
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457 **Fig. 8:** Variations of the mean values of density-weighted displacement speed S_d^* [—]
 458 and its components (i.e. S_r^* [····], S_n^* [- - -], S_t^* [····], S_{ev}^* [- - -] and S_c^* [
 459 —]) conditioned upon reaction progress variable c at (a) $x = 2D_j$, (b) $x = 4D_j$,
 460 (a) $x = 6D_j$, (a) $x = 8D_j$, (a) $x = 10D_j$, and (a) $x = 12D_j$. All quantities are
 461 normalised using $S_{b(\phi=1)}$.

462 The variations of the mean values of $\dot{\omega}_c$, $\vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c)$, $(-2\rho D \kappa_m |\nabla c|)$, \dot{S}_{ev} and \dot{S}_c
 463 (normalised by $\rho_0 S_{b(\phi=1)}/\delta_{th(\phi=1)}$ where $\delta_{th(\phi=1)}$ is the thermal flame thickness of the
 464 laminar stoichiometric mixture) conditional upon c as well as their combined contributions are
 465 shown for $x/D_j = 2, 4, 6, 8, 10$ and 12 in Figs. 7a-f, respectively. It is evident from Figs. 7a-
 466 f that the mean value of reaction rate of reaction progress variable $\dot{\omega}_c$ is deterministically
 467 positive across c at all axial locations considered here exhibiting similar qualitative behaviour
 468 - small values towards the unburned gas side with larger values towards the burned gas side
 469 and a peak value close to $c = 0.7$ in the reaction zone. The magnitude of the mean values of
 470 $\dot{\omega}_c$ has been found to decrease moving downstream of the jet exit which is due to the
 471 evaporation of larger droplets leading to fuel-rich conditions and thus giving rise to reduced
 472 burning rates. Furthermore, $\dot{\omega}_c$ acts as a leading order term for all axial locations considered
 473 here. The mean flame normal molecular diffusion contribution $\vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c)$ shows similar
 474 qualitative behaviour at all axial locations considered – exhibiting positive mean values
 475 towards the unburned gas side and negative mean values towards the burned gas side with a
 476 transition close to $c = 0.55$. The magnitudes of the mean values of $\vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c)$
 477 conditional upon c have been found to decrease moving downstream of the jet exit as a result
 478 of increased flame thickness for fuel-rich mixtures. However, $\vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c)$ acts as a
 479 leading order term for all axial locations considered here. The mean tangential molecular
 480 diffusion $(-2\rho D \kappa_m |\nabla c|)$ conditional upon c has been found to be small in comparison to the
 481 mean values of $\dot{\omega}_c$ and $\vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c)$ at all axial locations considered in the current analysis.
 482 The mean value of the term arising due to droplet evaporation \dot{S}_{ev} has been shown to be
 483 negligible across c for all axial locations considered in the current analysis. The mean value of
 484 the cross-scalar dissipation contribution \dot{S}_c has been shown to be small, but non-negligible,
 485 across c for all axial locations considered. It should be noted that the mean contribution of \dot{S}_c
 486 exhibits positive values across c at $x/D_j = 2$. However, at axial locations further
 487 downstream, \dot{S}_c exhibits negative mean values. It is evident, therefore, that the combined
 488 contribution of $\dot{\omega}_c$, $\vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c)$, $(-2\rho D \kappa_m |\nabla c|)$, \dot{S}_{ev} and \dot{S}_c is predominantly determined
 489 by the competition between the contributions of $\dot{\omega}_c$ and $\vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c)$. The observations
 490 made here are consistent with those previously made for the mean variations of $\dot{\omega}_c$, $\vec{N} \cdot$
 491 $\nabla(\rho D \vec{N} \cdot \nabla c)$, $(-2\rho D \kappa_m |\nabla c|)$, \dot{S}_{ev} and \dot{S}_c conditional upon c for turbulent spray flames in
 492 canonical configurations [20,27]. It can be seen from Figs. 7a-f that the mean value of the

493 combined contribution of $\dot{\omega}_c$, $\vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c)$, $(-2\rho D \kappa_m |\nabla c|)$, \dot{S}_{ev} and \dot{S}_c remains positive
 494 for the majority of the flame but small negative values are obtained towards the burned gas
 495 side, as the negative contribution of $\vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c)$ overcomes the positive contributions.

496 The variations of the mean values of the density-weighted displacement speed S_d^* and its
 497 contributions S_r^* , S_n^* , S_t^* , S_{ev}^* and S_c^* conditional upon c are shown for $x/D_j = 2, 4, 6, 8, 10$
 498 and 12 in Figs. 8a-f, respectively. It can be seen from Figs. 8a-f that the general behaviours of
 499 the contributions of S_r^* , S_n^* , S_t^* , S_{ev}^* and S_c^* are consistent with behaviours of $\dot{\omega}_c$, $\vec{N} \cdot$
 500 $\nabla(\rho D \vec{N} \cdot \nabla c)$, $(-2\rho D \kappa_m |\nabla c|)$, \dot{S}_{ev} and \dot{S}_c . Furthermore, the observed behaviour for the mean
 501 value of the combined contributions of $\dot{\omega}_c$, $\vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c)$, $(-2\rho D \kappa_m |\nabla c|)$, \dot{S}_{ev} and \dot{S}_c is
 502 consistent with the general behaviour observed for the mean values of density-weighted
 503 displacement speed S_d^* . Accordingly, the behaviour of the density-weighted displacement
 504 speed S_d^* is predominantly determined by the competition between the reaction rate component
 505 S_r^* and the normal molecular diffusion component S_n^* . Moreover, the mean contributions of S_t^* ,
 506 S_{ev}^* and S_c^* remain small in magnitude in comparison to the leading order contributions of S_r^*
 507 and S_n^* . These observations are, again, consistent with those previously made for turbulent
 508 spray flames in canonical configurations [20,27].

509 It should be noted that, from a modelling perspective, it is often useful to know the curvature
 510 (i.e. $\kappa_m = \nabla \cdot \vec{N}/2$) and tangential strain rate (i.e. $a_T = (\delta_{ij} - N_i N_j)(\partial u_i / \partial x_j)|_{c=c^*}$) dependencies
 511 of the density-weighted displacement speed S_d^* and its components. Table 3 shows the
 512 correlation coefficients for $\kappa_m - a_T$, $\kappa_m - S_d^*$, $\kappa_m - S_r^*$, $\kappa_m - S_n^*$, $\kappa_m - S_t^*$, $a_T - S_d^*$, $a_T - S_r^*$,
 513 $a_T - S_n^*$ and $a_T - S_t^*$ at $c = 0.7$ (i.e. the location within the flame of the maximum reaction
 514 rate) for all axial locations considered in the current study (i.e. $x/D_j = 2, 4, 6, 8, 10$ and 12).

515 It can be seen from Table 3 that $\kappa_m - a_T$ exhibits weak negative correlations at all axial
 516 locations considered but that the extent of the negative correlation is generally larger moving
 517 further downstream. These general observations are consistent with previous findings in
 518 turbulent stratified flames under canonical configurations [55] and turbulent premixed jet
 519 flames [61].

520

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523

524 **Table 3:** Correlation coefficients of $\kappa_m - a_T$, $\kappa_m - S_d^*$, $\kappa_m - S_r^*$, $\kappa_m - S_n^*$, $\kappa_m - S_t^*$, $a_T - S_d^*$, $a_T - S_r^*$,
525 $a_T - S_n^*$ and $a_T - S_t^*$ across the jet at $x/D_j = 2, 4, 6, 8, 10$ and 12 for $c = 0.7$ isosurface

| x/D_j | 2 | 4 | 6 | 8 | 10 | 12 |
|------------------------------|---------|---------|---------|---------|---------|---------|
| $\kappa_m - a_T$ | -0.2063 | -0.0673 | -0.4344 | -0.2533 | -0.2060 | -0.3291 |
| $\kappa_m - S_d^*$ | -0.7387 | -0.6670 | -0.7892 | -0.7598 | -0.7787 | -0.6699 |
| $\kappa_m - S_r^*$ | -0.0426 | 0.2227 | 0.0147 | 0.3181 | -0.3665 | 0.1866 |
| $\kappa_m - S_n^*$ | -0.0065 | 0.0326 | -0.2670 | -0.2359 | -0.3388 | -0.0830 |
| $\kappa_m - (S_r^* + S_n^*)$ | -0.0342 | 0.2366 | -0.1305 | 0.1752 | -0.4236 | 0.1083 |
| $\kappa_m - S_t^*$ | -0.9983 | -0.9861 | -0.9943 | -0.9962 | -0.9963 | -0.9971 |
| $a_T - S_d^*$ | -0.0120 | -0.1784 | 0.2709 | -0.0015 | 0.0563 | -0.0416 |
| $a_T - S_r^*$ | -0.2310 | -0.0820 | 0.0781 | -0.2036 | -0.0319 | -0.3804 |
| $a_T - S_n^*$ | -0.1941 | -0.4994 | -0.1958 | -0.2972 | -0.1358 | -0.2017 |
| $a_T - (S_r^* + S_n^*)$ | -0.2727 | -0.2840 | -0.0422 | -0.3379 | -0.0703 | -0.3917 |
| $a_T - S_t^*$ | 0.2023 | 0.0481 | 0.4128 | 0.2624 | 0.2224 | 0.3331 |

526
527 It is well-known that S_d^* in turbulent premixed and stratified flames exhibits considerable strain
528 rate and curvature dependences [55-57], and a qualitatively similar behaviour has been reported
529 for turbulent spray flames in canonical configurations. Therefore, it is worthwhile to examine
530 the curvature and strain rate dependences of S_d^* in the configuration considered here. The
531 correlation coefficients for S_d^* and its leading order components with local tangential strain
532 rate a_T and curvature κ_m at different axial locations are exemplarily shown for $c = 0.7$
533 isosurface in Table 3. The mean value of reaction rate $\dot{\omega}_c$ assumes its peak value close to $c =$
534 0.7 and thus the $c = 0.7$ isosurface can be taken to represent the flame surface for the following
535 discussion in accordance with previous analyses [55-57]. It can be seen from Table 3 that S_d^*
536 exhibits negative correlation with curvature κ_m at all axial locations considered. This negative
537 correlation is consistent with previous findings in turbulent premixed and stratified flames [55-
538 57]. To better understand the behaviour of the $\kappa_m - S_d^*$ correlation, it is necessary to examine
539 the curvature dependence of the leading components of the density-weighted displacement
540 speed. It is evident from Table 3 that S_r^* and κ_m are generally weakly, but predominantly
541 positively, correlated whereas S_n^* shows weak, but predominantly negative, correlation with
542 κ_m (both observations being consistent with previous studies on turbulent premixed and

543 stratified flames [55-57]). Table 3 also shows a weak correlation between κ_m and $(S_r^* + S_n^*)$
 544 (i.e. the two major contributors to S_d^*). However, S_t^* and κ_m are found to be (deterministically)
 545 negatively correlated, as the mass diffusivity D on a given c isosurface is not expected to
 546 exhibit any appreciable correlation with κ_m . This strong negative $\kappa_m - S_t^*$ correlation is
 547 principally responsible for the negative correlation between S_d^* and κ_m , which is consistent
 548 with previous findings based on turbulent premixed and stratified flames [55-57].

549 Table 3 shows that a_T and S_d^* exhibit weak correlations for all axial locations considered here,
 550 which is consistent with previous analyses on turbulent premixed and stratified flames [55-57].
 551 To better understand the behaviour of the correlation between a_T and S_d^* , it is useful to examine
 552 the tangential strain rate dependencies of the leading components of the density-weighted
 553 displacement speed. It is evident from Table 3 that a_T and S_r^* are weakly and predominantly
 554 negatively correlated, and that a_T and S_n^* are negatively correlated at all axial locations
 555 considered in the current study (both observations are consistent with previous analyses on
 556 turbulent premixed and stratified flames [55-57]). Table 3 also shows that the correlation
 557 between a_T and $(S_r^* + S_n^*)$ (i.e. the two major contributors to S_d^*) is weakly negatively
 558 correlated at all axial locations considered and this negative correlation is consistent with
 559 previous analyses on turbulent premixed and stratified flames [55-57]. As the mean curvature
 560 and tangential strain rate are negatively correlated, the tangential component of displacement
 561 speed $S_t = -2D\kappa_m$ and a_T are expected to be positively correlated in all cases which can be
 562 verified from weak positive $a_T - S_t^*$ correlation in Table 3 at all axial locations considered
 563 here. This has been found to be consistent with previous studies on turbulent premixed and
 564 stratified flames [55-57]. The positive $a_T - S_t^*$ correlation overcomes the negative correlation
 565 between a_T and $(S_r^* + S_n^*)$ to give rise to a weak correlation between a_T and S_d^* (see Table 3).
 566 It is worth noting that the statistics of S_d^* , and its local curvature and tangential strain rate
 567 dependences for the jet flame considered here are also found to be qualitatively similar to the
 568 spray flames in canonical configurations (e.g. statistically planar or spherical flames) [20,26].
 569 The physical explanations for the observed curvature and tangential strain rate dependences of
 570 S_d^* components have been provided elsewhere [20,26,55-57] in detail and thus will not be
 571 repeated here.

572

573 **3.3 Implications and further considerations**

574 The statistical behaviours of the mean contributions of $\dot{\omega}_c$, $\vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c)$, $(-2\rho D \kappa_m |\nabla c|)$,
 575 \dot{S}_{ev} and \dot{S}_c offer useful insights into the modelling aspects for turbulent spray flames. Firstly,

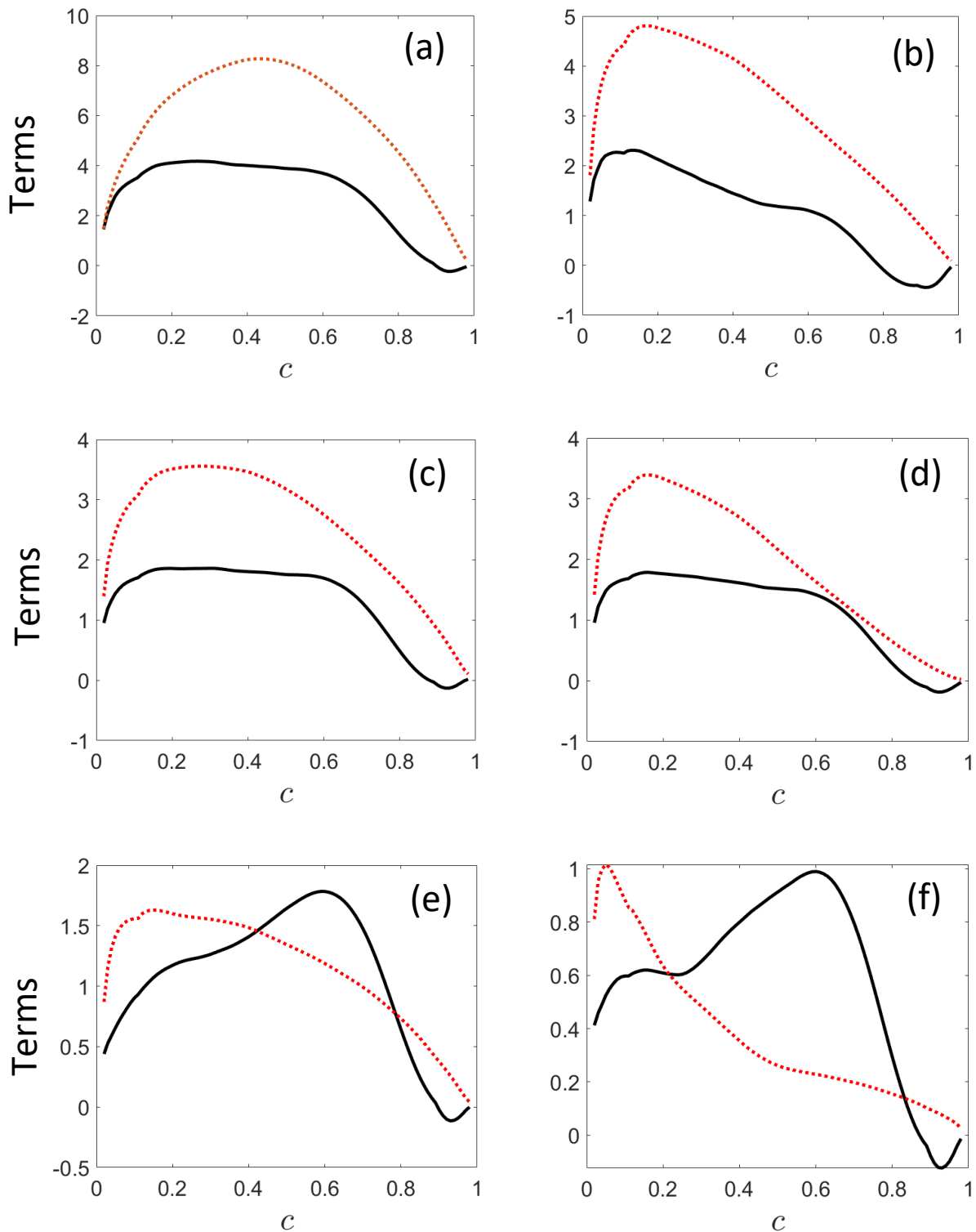
576 it should be noted that the qualitative nature of the mean variations of $\dot{\omega}_c$, $\vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c)$,
577 $(-2\rho D \kappa_m |\nabla c|)$, \dot{S}_{ev} and \dot{S}_c observed here is consistent with those observed in turbulent spray
578 flames for canonical configurations [20,27], which suggests that the flow geometry in the
579 absence of mean flame curvature might not be an important factor in the behaviour of these
580 terms. Moreover, the observed mean behaviours of $\dot{\omega}_c$, $\vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c)$, $(-2\rho D \kappa_m |\nabla c|)$ and
581 \dot{S}_c are consistent with observations made for these quantities in turbulent stratified gaseous
582 flames [55]. This suggests that the same modelling methodologies that have been employed
583 with respect to turbulent stratified flames might be possible to extend for turbulent spray
584 flames.

585 In the context of the FSD modelling approach [28,61,62], the following assumption is often
586 invoked:

$$587 \quad \rho S_d |\nabla c| = \dot{\omega}_c + \vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c) - 2\rho D \kappa_m |\nabla c| + \dot{S}_{ev} + \dot{S}_c \approx \rho_0 S_{b(\phi)} |\nabla c| \quad (45)$$

588 where ρ_0 is the unburned reactant density and $S_{b(\phi)}$ is the laminar burning speed as a function
589 of the local equivalence ratio ϕ . The variations of the mean values of $\rho_0 S_{b(\phi)} |\nabla c|$ and the
590 combined contribution of $\dot{\omega}_c + \vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c) - 2\rho D \kappa_m |\nabla c| + \dot{S}_{ev} + \dot{S}_c$ conditional upon c
591 are shown in Figs. 9a-f for $x/D_j = 2, 4, 6, 8, 10$ and 12 . It can be seen from Figs. 9a-f that the
592 approximation of $\dot{\omega}_c + \vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c) - 2\rho D \kappa_m |\nabla c| + \dot{S}_{ev} + \dot{S}_c$ using $\rho_0 S_{b(\phi)} |\nabla c|$
593 provides poor agreement across c for all axial locations considered in the current study. It is
594 evident that the mean values of $\rho_0 S_{b(\phi)} |\nabla c|$ largely overpredict the mean values of $\dot{\omega}_c + \vec{N} \cdot$
595 $\nabla(\rho D \vec{N} \cdot \nabla c) - 2\rho D \kappa_m |\nabla c| + \dot{S}_{ev} + \dot{S}_c$ for axial locations $x/D_j = 2, 4, 6, 8$. This finding is
596 consistent with previous analyses of low Damköhler number turbulent premixed and stratified
597 gaseous flames [55]. Whilst the extent of over-prediction is relatively small at $x/D_j = 10$ and
598 12 , the qualitative trends of $\dot{\omega}_c + \vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c) - 2\rho D \kappa_m |\nabla c| + \dot{S}_{ev} + \dot{S}_c$ are not captured
599 by $\rho_0 S_{b(\phi)} |\nabla c|$. Furthermore, it should be noted $\dot{\omega}_c + \vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c) - 2\rho D \kappa_m |\nabla c| + \dot{S}_{ev} +$
600 \dot{S}_c exhibits negative values which $\rho_0 S_{b(\phi)} |\nabla c|$ cannot adequately account for. On Reynolds
601 averaging/LES filtering Eq. 45 one obtains: $\overline{(\rho S_d)}_s = \rho_0 S_{b(\phi)}$ (where $\overline{(Q)}_s = \overline{Q |\nabla c|} / \overline{|\nabla c|}$ is
602 the surface-weighted value of a general quantity Q [61,62]), which is often used for the FSD
603 based closures in turbulent premixed and stratified flames [62-64]. However, the inequality of
604 left- and right-hand sides of Eq. 45 reveals that such modelling approaches might not be

605 appropriate for low Damköhler number spray flames in general and that alternative modelling
 606 approaches might need to be considered.



607

608

609

610 **Fig. 9:** Variations of the mean values of $\dot{\omega}_c + \vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c) - 2\rho D \kappa_m |\nabla c| + \dot{S}_{ev} + \dot{S}_c$ [
 611 —] and $\rho_0 S_{b(\phi)} |\nabla c|$ [\cdots] conditioned upon reaction progress variable c at (a)
 612 $x = 2D_j$, (b) $x = 4D_j$, (a) $x = 6D_j$, (a) $x = 8D_j$, (a) $x = 10D_j$, and (a) $x = 12D_j$. All
 613 quantities are normalised using $\rho_0 S_{b(\phi=1)} / \delta_{th(\phi=1)}$.

614 4. CONCLUSIONS

615 In the current study, a three-dimensional Direct Numerical Simulation of an open turbulent jet
616 spray flame representing a laboratory-scale burner configuration [14] has been considered to
617 investigate the behaviour of the density-weighted displacement speed S_d^* and its components.
618 The open turbulent jet spray flame has been found to exhibit fuel-lean conditions close to the
619 jet exit, but fuel-rich conditions have been observed further downstream due to the evaporation
620 of fuel droplets. It has been found that the displacement speed S_d and density-weighted
621 displacement speed S_d^* show qualitatively similar behaviour for all axial locations considered
622 – predominantly positive mean values across the flame but with small, potentially negative,
623 mean values towards the burned gas side. It is also found that the observed mean behaviours
624 of the displacement speed and density-weighted displacement speed are fundamentally
625 determined by the contributions of $\dot{\omega}_c$, $\vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c)$, $(-2\rho D \kappa_m |\nabla c|)$, \dot{S}_{ev} and \dot{S}_c . The
626 reaction rate $\dot{\omega}_c$ and normal molecular diffusion rate $\vec{N} \cdot \nabla(\rho D \vec{N} \cdot \nabla c)$ are found to be leading
627 order contributors and that the competition between their contributions determines the mean
628 behaviour of the density weighted displacement speed. These observations are consistent with
629 previous studies of turbulent spray flames in a canonical configuration and low Damköhler
630 number turbulent premixed and stratified gaseous flames. This suggests that the flow geometry
631 in the absence of mean curvature might not play an important role in deciding the general
632 behaviour of the displacement speed and its components. This further indicates that the
633 modelling methodologies, which are employed for turbulent stratified flames, might have the
634 potentials to be extended for turbulent spray flames. However, the surface-weighted value of
635 the product of displacement speed with local density cannot be approximated by the product
636 of unburned gas density and the local laminar burning velocity for the sampling locations
637 considered here. This is consistent with previous findings for low Damköhler number stratified
638 flames and thus the modelling methodologies in the context of turbulent spray flames need to
639 account for attributes of low Damköhler number combustion. These aspects will form the basis
640 of future investigations.

641

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