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Title: Crystal structures of EfeB and EfeO in a bacterial siderophore-independent iron transport

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1 Abstract

2 EfeUOB is a siderophore-independent iron uptake mechanism in bacteria. EfeU, EfeO, and 3 EfeB are a permease, an iron-binding or electron-transfer protein, and a peroxidase, respectively. 4 A Gram-negative bacterium, Sphingomonas sp. strain A1, encodes EfeU, EfeO, EfeB together with alginate-binding protein Algp7, a truncated EfeO-like protein (EfeO_{II}), in the genome. The 5 typical EfeO (EfeO_I) consists of N-terminal cupredoxin and C-terminal M75 peptidase domains. 6 7 Here, we detail the structure and function of bacterial EfeB and EfeO. Crystal structures of strain A1 EfeB and Escherichia coli EfeO1 were determined at 2.30 Å and 1.85 Å resolutions, 8 9 respectively. A molecule of heme involved in oxidase activity was bound to the C-terminal Dyp 10 peroxidase domain of EfeB. Two domains of EfeOI were connected by a short loop, and a zinc 11 ion was bound to four residues, Glu156, Glu159, Asp173, and Glu255, in the C-terminal M75 12 peptidase domain. These residues formed tetrahedron geometry suitable for metal binding and are well conserved among various EfeO proteins including Algp7 (EfeO_{II}), although the metal-13

on structure of a typical EfeO with two domains, postulating a novel metal-binding motif
"ExxE-//-D-//-E" in the EfeO C-terminal M75 peptidase domain.

binding site (HxxE) is proposed in the C-terminal M75 peptidase domain. This is the first report

5 Keywords: EfeB, EfeO, iron transport, X-ray crystallography

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7 1. Introduction

8 Iron is essential for various enzymatic reactions, protein stabilization, and signal

9 transduction in vivo. A typical mechanism used by microorganisms to acquire iron from the

10 external environment is the use of siderophores, which are metal chelators. Microorganisms

11 secrete siderophores, which form Fe³⁺–siderophore complexes extracellularly and solubilize

12 Fe^{3+} , and subsequently acquire iron by importing the Fe^{3+} -siderophore complex into the cell

13 [1]. Another mechanism for iron import is the Efe system [2]. The Efe system functions

independently of siderophores and consists of three proteins, namely, EfeU, EfeO, and EfeB, 1 2 and it was first identified in Escherichia coli strain O157:H7 [3]. The expression of Efe proteins 3 is induced by low pH or iron concentration. EfeU probably functions as a permease and 4 localizes in the inner membrane. The periplasmic protein EfeB containing heme shows peroxidase activity to oxidize Fe²⁺ to Fe³⁺ [4]. EfeO shows iron-binding or electron-transfer 5 activity and is located in the periplasm. EfeO has variations in its domain topology [5]. A typical 6 7 EfeO is classified as EfeO_I, composed of an N-terminal cupredoxin and C-terminal M75 peptidase domains. The cupredoxin is a copper-binding protein involved in stabilizing Cu²⁺ 8 9 during copper uptake [6], and the M75 peptidase binds zinc ions [7]. The cupredoxin and M75 10 peptidase domains in EfeO contain two and one metal-binding motifs, respectively [5]. Distinct 11 from EfeO_I, EfeO_I, possessing only M75 peptidase domain, forms another group of EfeO. 12 Nevertheless the tertiary structure of EfeO_{II} is available, that of EfeO_I remains to be clarified, obstructing understanding of molecular mechanism of Efe system. 13

The Gram-negative and alginate-assimilating Sphingomonas sp. strain A1 possesses an Efe 1 2 system consisting of EfeUO_IB and one additional EfeO_{II} protein, Algp7 (SPH726) (Fig. 1A). 3 Another EfeO₁ protein, SPH728, has a conserved N-terminal cupredoxin and C-terminal M75 4 peptidase domains. Algp7 (EfeO_{II}) exhibits a high alginate-binding ability with a dissociation constant (K_d value) of 3.6×10^{-8} M and functions to concentrate alginate on the cell surface [8-5 9]. Algp7 (EfeO_{II}) binds Cu²⁺, Fe²⁺, and Zn²⁺ as well as alginate [10]. Strain A1 incorporates 6 7 alginate as a polysaccharide into the cytoplasm through a cell surface pit containing Algp7 8 (EfeO_{II}) and an ATP-binding cassette transporter [9]. 9 Alginate polysaccharide constituents are broadly classified into three types: 10 polymannuronic acid blocks (M-blocks), polyguluronic acid blocks (G-blocks), and random 11 blocks. Various metals, including iron, are chelated into alginate G-blocks [10]. Therefore, 12strain A1 probably acquires metals by binding extracellular alginate with Algp7 (EfeO_{II}), which binds both alginate and metals. The elucidation of the mechanism of metal import using alginate 13

1	in strain A1 should provide important insights into alginate-dependent metal physiology in
2	bacteria. Furthermore, bacterial metal import machinery can be applied to bioremediation and
3	recovery of useful metals, including rare earth elements. This article describes the structure and
4	function of the Efe system component proteins, namely, EfeB and EfeO.
5	
6	2. Materials and methods
7	2.1. Reagents
8	General reagents of special grade were purchased from Fujifilm Wako Pure Chemicals and
9	Nacalai Tesque. Samarium (III) sulfate octahydrate was obtained from Sigma-Aldrich.
10	
11	2.2. Strains and media

- 12 E. coli strains BL21(DE3), Rosetta-gami 2(DE3), and MG1655 were used in this study.
- 13 Luria Bertani (LB) medium (1% tryptone, 0.5% yeast extract, and 1% NaCl) was used to culture

1 the *E. coli* strains.

2

3 2.3. Protein expression and purification

4 (i) EfeB

Recombinant EfeB was expressed in E. coli strain Rosetta-gami 2(DE3) harboring the 5 plasmid pET21b-EfeB that encoded EfeB with a hexahistidine tag at the C-terminus. The EfeB 6 7 gene was amplified from the strain A1 genome by PCR using synthetic oligo DNA primers 5'-8 (forward: 5'-GGCATATGGATGAGATCGAAAACGGCAAGG-3', reverse: 9 10 pET21b and treated with restriction enzymes NdeI and NotI. The transformed cells with 11 pET21b-EfeB were cultured in LB medium containing 0.1 mg/mL sodium ampicillin at 37°C 12 and 100 spm for 3 h. At the exponential growth phase, isopropyl β -D-1-thiogalactopyranoside (IPTG) was added to the medium to a final concentration of 0.1 mM, and then, the culture was 13

incubated for another 44 h at 16°C. The culture was centrifuged (14,450 \times g, 4°C, 10 min), the 1 2 supernatant was removed, and the pellet was suspended in 20 mM Tris-HCl (pH 7.5). The 3 suspension was treated with an ultrasonic disrupter (Insonator Model 201M, Kubota) at 9 kHz 4 for 20 min, and cell extracts were obtained by centrifugation (18,000 \times g, 4°C, 20 min). EfeB was purified from the cell extracts using TALON metal affinity resin (bed volume, 10 mL; 5 6 Takara Bio), and Hi Load 16/60 Superdex 75 pg column (1.6 × 10 cm; GE Healthcare). Purified 7 EfeB was then dialyzed using 20 mM Tris-HCl (pH 7.5). All purifications were conducted at 4°C. The concentration of purified EfeB was calculated by measuring Abs₂₈₀ with an absorption 8 coefficient for EfeB of 34,045 M⁻¹cm⁻¹. 9 10 (ii) Algp7 (EfeO_{II}) E. coli strain BL21(DE3) was transformed with the expression plasmid pET44a-Algp7 [8]. 11 12 The plasmid encoded the Algp7 (EfeOII) fusion protein that had a hexahistidine tag on the C-

13 terminus and a NusA tag on the N-terminus. Expression and purification of Algp7 (EfeO_{II}) were

conducted as described previously [9]. The concentration of purified Algp7 (EfeO_{II}) was 1 2 determined by measuring the absorbance at a wavelength of 280 nm (Abs280) with an absorbance coefficient for Algp7 (EfeO_{II}) of 17,420 M⁻¹cm⁻¹. 3 4 (iii) EfeO_I The plasmid pET21b-EfeO encoding E. coli EfeO without signal sequence was constructed 5 in the same way with EfeB except for the gene and primers. The EfeO_I gene was amplified from 6 7 the strain MG1655 genome by PCR using DNA primers (forward: 5'-AAGGAGATATACATATGGCTGATGTGCCGCAGGTCAAAGTGACC-3', 5'-8 reverse: 9 GGTGGTGGTGCTCGAGATCCAGTCCCAGCACACCGCGAAGTTG-3'). E. coli strain 10 BL21(DE3) cells were transformed with the resultant plasmid and cultured in 13.5 L of LB 11 medium containing 0.1 mg/mL sodium ampicillin at 37°C and 108 spm. At the exponential 12 growth phase, IPTG was added at a final concentration of 0.4 mM followed by further incubation at 16°C for 48 h. Cell collection and disruption were conducted in the same way 13

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with EfeB followed by purification using TALON metal affinity resin (bed volume; 30 mL),

- 2 HiLoad 26/10 Q-Sepharose HP column (2.6 × 10 cm), HiLoad 16/10 Phenyl Sepharose column
- 3 (1.6 \times 10 cm; GE healthcare), HiLoad 16/60 Superdex 200 pg column (1.6 \times 60 cm; GE
- 4 healthcare), and Mono Q HR 5/5 column (0.5×5 cm; GE healthcare).
- 5
- 6 2.4. Crystallization and structure determination
- 7 Each of purified EfeB, Algp7 (EfeO_{II}), and EfeO_I was crystallized. Crystallization was
- 8 performed in a 96-well plate using the sitting-drop vapor diffusion equilibrium method. One
- 9 microliter of 10 mg/mL EfeB solution was mixed with the same amount of crystallization liquid
- 10 [15% polyethylene glycol 400, 100 mM sodium acetate (pH 4.6), and 100 mM CaCl₂]. One
- 11 microliter of crystallization liquid [12.5% polyethylene glycol 8,000, 100 mM sodium citrate
- 12 (pH 3.9), and 200 mM NaCl] was added to 1 μ L of 8 mg/mL Algp7 (EfeO_{II}) solution. One
- 13 microliter of 20 mg/mL EfeO_I solution was mixed with the same amount of crystallization

liquid [20% polyethylene glycol 4,000, 160 mM ammonium sulfate, 80 mM sodium acetate 2 (pH 4.6), and 20% glycerol]. 3 Algp7 (EfeO_{II}) crystals were soaked in the crystallization parent liquid with 5 mM 4 Sm₂(SO₄)₃ for 15 min. The crystals were frozen in liquid nitrogen and then were analyzed at 5 the BL38B1 beamline at SPring-8 (Hyogo, Japan). Similarly, EfeB and EfeOI crystals were 6 soaked in the crystallization liquid with 20% ethylene glycol and frozen with liquid nitrogen, 7 and then were analyzed at the BL26B1 and BL38B1 beamlines, respectively, at SPring-8. 8 Each frozen crystal was irradiated with X-rays at a wavelength of 1.00 Å. The diffraction 9 data were collected using detector MX225-HS CCD (Rayonix) at BL26B1 or PILATUS3X 2M 10 (Dectris) at BL38B1 and then processed by HKL2000 [11] or XDS [12]. The software Molrep 11 [13] included in the CCP4interface (CCP4i) package was used for molecular replacement. 12 Structural refinement was performed using *Phenix.refine* [14] and *Refmac5* [15], the model was manually modified using WinCoot [16], and the figure was generated using PyMol [17]. The 13

11

1	structure of cupredoxin domain in EfeO _I was constructed manually referring to the electron
2	density map using WinCoot because of the lack of the coordinate model in the Protein Data
3	Bank (PDB) database. The atomic coordinates and structure factors (PDB ID: 6JBN for EfeB,
4	6JBO for Algp7 (EfeO _{II}), and 7WGU for EfeO _I) were deposited in PDB, Research
5	Collaboratory for Structural Bioinformatics, Rutgers University, New Brunswick, NJ
6	(http://www.rcsb.org/).
7	
8	2.5. Inductively coupled plasma-mass spectrometry (ICP-MS) analysis
9	The purified EfeO _I in 20 mM Tris-HCl (pH 7.5) was concentrated using VIVA SPIN 500
10	(Sartorius) to a final concentration of 48.7 mM. The solution (900 μ L) was subjected to the
11	ICP-MS analysis. Data collection and analysis was conducted by Tokai Technology Center

12 (Aichi, Japan).

- 1 2.6. Differential scanning fluorimetry (DSF)
- 2 Algp7 (EfeO_{II}) binding affinity for a rare earth element (Sm^{3+}) was evaluated using DSF in
- 3 the presence or absence of the element as described previously [18].
- 4
- 5 2.7. Measuring EfeB peroxidase activity

6 Nitrogen gas was injected into the glove box S-GBC type (Samplatec) to create an anaerobic

7 environment. In the glove box, 20 µM EfeB was added to the reaction solution containing 3

8 mM FeSO₄·7H₂O and 1 mM H₂O₂. The concentration of Fe^{2+} in the reaction solution was

9 determined by the nitroso-PSAP method using a metallo assay iron measurement kit

10 (Funakoshi). Nitroso-PSAP forms a complex with Fe^{2+} and exhibits coloration with a peak at a

11 wavelength of 750 nm [19]. Fifteen microliters of a 10-fold diluted sample was mixed with 160

- 12 μ L of ultrapure water or 30% guanidine hydrochloride as a surfactant and 160 μ L of 1% sodium
- 13 ascorbate as the reducing agent, followed by incubation for 10 min. Seventy-five microliters of

1 0.1% nitroso-PSAP was added to the solution and incubated for 5 min. The peroxidase activity

2 of EfeB was determined by measuring the absorbance of the solution at a wavelength of 750

3 nm to determine the concentration of residual Fe^{2+} .

4

5 3. Results and discussion

6 *3.1. EfeB includes heme in the crystal structure*

7 Strain A1 EfeB was crystallized, and its tertiary structure was determined by X-ray

8 crystallography. The statistics of X-ray diffraction and structure refinement data are shown in

9 Table S1. His231 and Ala269 were the anomalous regions in the Ramachandran plot for chains

10 A and B, respectively. The crystal structure of EfeB from E. coli strain O157:H7 has been

11 previously reported (PDB ID: 3O72) [4]. EfeB from *E. coli* and from strain A1 showed high

12 sequence similarity (identity: 60%), and both had a Dyp peroxidase domain on the C-terminus.

13 The tertiary structure of EfeB from strain A1 was, therefore, similar to that of EfeB from *E. coli*.

2 of four β -strands was sandwiched by α -helices (Fig. 1B). A heme molecule was bound to the 3 C-terminal Dyp peroxidase domain (Fig. 1C). 4 3.2. Oxidation activity of EfeB against Fe^{2+} 5 Because nitroso-PSAP is inactive in the presence of Fe^{3+} , Fe^{2+} concentration is measured in 6 the absence of a reducing agent. If a reducing agent is present, total concentrations of Fe³⁺ and 7 Fe^{2+} are measured to account for the fact that Fe^{3+} is reduced to Fe^{2+} by the reducing reagent. 8 In the presence of EfeB but in the absence of a reducing agent, the Fe²⁺ concentration decreased 9 significantly after 10 min of reaction time (Fig. 1D). Conversely, in the presence of Algp7 10 (EfeO_{II}) or in the negative control, there was no decrease of the Fe^{2+} concentration without the 11 reducing agent (Fig. 1D). These results indicated that EfeB exhibits peroxidase activity and 12oxidizes Fe²⁺ to Fe³⁺. EfeB peroxidase activity also occurred in the absence of H₂O₂, suggesting 13 15

Strain A1 EfeB contained 14 α-helices and 13 β-strands, and the antiparallel β-sheet consisting

1 that EfeB oxidized using dissolved oxygen in the reaction solution (Fig. 1D).

2

3 3.3. Rare earth element-binding ability of Algp7 (Efe O_{II})

Interestingly, in addition to Cu²⁺, Fe²⁺, and Zn²⁺, Algp7 (EfeO_{II}) was found to be stabilized 4 in the presence of the rare earth element (Sm³⁺) by DSF (Fig. 2A), suggesting that Algp7 5 (EfeO_{II}) bound to Sm^{3+} . 6 To clarify the binding mode of Sm³⁺ toward Algp7 (EfeO_{II}), the crystal of Algp7 (EfeO_{II}) 7 was prepared and soaked into the solution containing Sm³⁺ followed by X-ray crystallography. 8 9 The resulting diffractions showed up to 1.88 Å resolution (Table S1). The resolution obtained herein (1.88 Å) is higher than that previously reported (1.99 Å, PDB ID: 3WSC) [9]. The whole 10 structure is similar to that containing Cu^{2+} (Fig. 2B), however, there is no metal bound to Algp7 11 12(EfeO_{II}). A citrate molecule formed two hydrogen bonds with Arg204 and Arg212. To clarify one of the reasons for no metal binding, the Algp7 (EfeO_{II}) structure determined in this 13

experiment was superimposed on that of Algp7 (EfeO_{II}) complexed with Cu²⁺ (PDB ID: 5Y4C) 1 based on Cα. Among residues interacting directly with Cu²⁺, Glu82 and Glu177 were oriented 2 3 differently than those in the metal-unbound state (Fig. 2C). This suggested that Algp7 (EfeO_{II}) 4 residues responsible for binding to metal ions lose their flexibility during crystallization and that samarium hardly bound to the Cu^{2+} -binding pocket. 5 6 3.4. X-ray crystallography of EfeO₁ 7 8 Because structure of EfeOI remains to be determined, we attempted to express the strain A1 9 SPH728 (EfeO₁) in *E. coli* cells, but failed. Thus, EfeO₁ with a lack of signal peptide (Thr2-10 Ala26) from E. coli was expressed and purified in this study, because SPH728 (EfeO_I) exhibits a high sequence homology (identity, 41.7%, similarity, 56.9%) with E. coli EfeOI. The residues 11 12 in E. coli EfeOI are renumbered based on the lack of signal peptide. The X-ray crystal structure of E. coli EfeOI containing both cupredoxin and M75 peptidase domains at N- and C-terminus, 13 17

1	respectively, was determined at 1.85 Å resolution (Table S1, Fig. 3A left). This is the first
2	structure of EfeO with both domains. Two domains were connected by a loop consisting of 4
3	residues. Although the primary structure of cupredoxin domain of EfeO _I shows little similarity
4	with those of other cupredoxin proteins (e.g. CupA), the tertiary structure of the domain formed
5	Greek key β -barrel structure conserved among other cupredoxin proteins (Fig. 3A red). The
6	database Interpro (https://www.ebi.ac.uk/interpro/) was used to find the same family proteins
7	with cupredoxin domain of EfeOI. As a result, 12 proteins were found as the same family. The
8	tertiary structure of the cupredoxin domain of EfeO1 was compared with CupA from
9	Streptococcus pneumoniae (PDB ID: 4F2E) [20]. The identity of primary structure between two
10	proteins was low, 14.4%. However, root mean square deviation (1.404 Å) of superimposed two
11	proteins based on Ca showed similar structure (Fig. 3B). This result indicated that cupredoxin
12	domain of EfeO _I is involved in binding to cupper or electron transfer as seen in other CupA
13	proteins. The M75 peptidase domain was composed of 10 α -helices as seen in other M75

1 peptidase domains (Fig. 3A yellow), and was structurally similar to Algp7 (EfeO_{II}) (Fig. 3C).

2

3 3.5 Structural insights into metal binding by EfeO₁

4 EfeOI has been suggested to have three metal-binding sites (I, II, and III) based on conservation of amino acid residues and tertiary structure built by homology modelling [5]. Site 5 I (Cys16, Glu41, Met75, and Cys78) and site II (Glu43, Glu52, Glu53, and Glu55) located in 6 the cupredoxin domain have also been predicted to bind to Cu²⁺ and Fe³⁺, respectively. Site III 7 8 (His192, x193, x194, and Glu 195) was in the M75 peptidase domain. Contrary to the predicted 9 model, tertiary structure determined here revealed that four residues at the site I were 10 structurally distant, and two Cys residues formed a disulfide bridge (Fig. 4A, green colored 11 residues). At the site II, four residues also seemed to form incomplete metal-binding site 12 because Glu52 faced opposite from the other residues (Fig. 4A, yellow colored residues). Multiple sequence alignment of cupredoxin proteins including EfeO indicated that Cys78 was 13

conserved and possibly involved in metal binding (Fig. 4B). Cys78 presumably binds to metals 1 2 only in the reducing conditions and not in the oxidized conditions because of disulfide bridge 3 formation. Metal binding at site I was considered to be regulated by redox balance in the 4 periplasm. The site III has been predicted to form zinc-binding motif "HxxE". However, three-5 6 dimensional arrangement of the four residues hardly formed tetrahedron geometry suitable for 7 metal binding (Fig. 4C left, pink colored residues). Electron density map demonstrated that four other residues (Glu156, Glu159, Asp173, and Glu255) coordinated to a metal (Fig. 3A right). 8 ICP-MS analysis revealed that most molecules (60%) of EfeO_I possessed Zn^{2+} , indicating that 9 Zn^{2+} was bound to these four residues. Because no zinc ion was added through the purification 10 and crystallization steps, Zn²⁺ was probably bound to EfeO_I in *E. coli* cells. The zinc-binding 11 12site in EfeO_I completely corresponded to the copper-binding site of Algp7 (EfeO_{II}) (PDB ID: 5Y4C) (Fig. 4C right, blue colored residues). The four residues at the zinc-binding site are 13

1	highly conserved in EfeO proteins except for Pss1 (Fig. 4D). These results suggested that EfeO
2	has a metal-binding site different from the HxxE motif (site III). This novel metal-binding site
3	(Glu156, Glu159, Asp173, and Glu255) was proposed as "ExxE-//-D-//-E" motif.
4	In conclusion, this is the first report on crystal structure of EfeO with N-terminal cupredoxin
5	and C-terminal M75 peptidase domains, and proposal of the novel metal-binding motif "ExxE-
6	//-D-//-E" in EfeO C-terminal M75 peptidase domain.
7	
8	Declaration of competing interest
9	The authors declare that they have no known competing financial interests or personal
10	relationships that could have appeared to influence the work reported in this paper.
11	
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7	
8	Appendix A. Supplementary data
9	Supplementary data to this article can be found online athttps://XXXXXXXXX.
10	
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1 Figure legends

2 Fig. 1. EfeUOB cluster and structure/function of EfeB. (A) EfeUOB gene cluster in 3 Sphingomonas sp. strain A1. Cup and M75 represent cupredoxin and M75 peptidase domains, 4 respectively. (B) Crystal structure of strain A1 EfeB (cyan) was determined through molecular 5 replacement using the structure of E. coli EfeB (magenta) (PDB ID: 3072). Green, blue, and 6 red in the stick model represent carbon, nitrogen, and oxygen atoms in a heme molecule. (C) 7 Heme-binding site in strain A1 EfeB. Green, cyan, blue, and red represent carbon in heme, 8 carbon in EfeB, nitrogen, and oxygen atoms, respectively. Gray and cyan spheres show O₂ and water molecules, respectively. Dotted lines show hydrogen bonds. (D) Peroxidase activity of 9 10 strain A1 EfeB in the presence or absence of reducing agents. 11

12 Fig. 2. Binding of rare earth element by and crystal structure of Algp7 (EfeO_{II}). (A) DSF

13 analysis of Algp7 (EfeO_{II}) in the absence or presence of rare earth element (Sm^{3+}). The curves

1	show the negative derivative plot obtained from the fluorescence profile. Thermal shift to higher
2	temperature was observed in the presence of the rare earth element (Sm ³⁺). (B) Structure of
3	Algp7 (EfeO _{II}) was determined through molecular replacement using the previously clarified
4	structure of Algp7 (EfeO _{II}) (PDB ID: 3AT7). Pink, Algp7 (EfeO _{II}) determined here; blue, Algp7
5	(EfeO _{II}) complexed with Cu^{2+} (PDB ID: 5Y4C). The right structure is rotated 90° toward the
6	reader relative to that of left. (C) Copper-binding site in Algp7 (EfeO _{II}). Pink, carbon atoms of
7	Algp7 (EfeO _{II}) determined here; blue, carbon atoms of Algp7 (EfeO _{II}) complexed with Cu ²⁺
8	(PDB ID: 5Y4C).
9	
10	Fig. 3. Crystal structure of EfeO _I . (A) Left, overall structure of <i>E. coli</i> EfeO _I . Red and yellow
11	represent cupredoxin and M75 peptidase domains, respectively. Right, zinc bound (magenta) at
12	the metal-binding site in the M75 peptidase domain of EfeOI. Water molecules are colored blue.

13 Glu159 and Glu255 are disordered. The *Phenix Polder* omit map for zinc and ligated residues

1	is shown with more than 3.5 σ . (B) Superimposition of the cupredoxin domain (red) of EfeOI
2	and CupA (cyan). (C) Superimposition of the M75 peptidase domain (yellow) of EfeO _I and
3	Algp7 (EfeO _{II}) (blue). Zinc bound to EfeO _I and copper bound to Algp7 (EfeO _{II}) are shown as
4	magenta and green spheres, respectively.
5	
6	Fig. 4. Structural alignment of metal-binding sites in EfeO. (A) Three-dimensional arrangement
7	of site I (green) and II (yellow) residues in the cupredoxin domain (red) of EfeO _I . (B) Multiple
8	sequence alignment of the EfeOI cupredoxin domain and other cupredoxin proteins. The residue
9	numberings for EfeO _I are shown at the top. Site I (green), site II (yellow), copper binding site
10	(cyan) residues. (C) Left, close-up site III (left, magenta colored) in EfeOI. Right, close-up
11	metal-binding site in the M75 peptidase domain of EfeO _I (yellow) and Algp7 (EfeO _{II}) (blue).
12	Zinc bound to EfeO _I and copper bound to Algp7 (EfeO _{II}) are shown as magenta and green
13	spheres, respectively. (D) Multiple sequence alignment of M75 peptidase domain proteins

- 1 (orange, EfeO_I; cyan, EfeO_{II}). The residue numberings for EfeO_I are shown at the top. Site III
- 2 (magenta), metal-binding site residues (blue).
- 3
- 4

1 Figure 1



1 Figure 2





1 Figure 4





D						
-	152	160	170	180 19	0	200
EfeO	RQHYERI	PIAELF -	SDLDGS I 🗗	AREDDYEQKAADPKFTGF	HRL	KALFGDN 51
Acc	RWHYEAI	PVAESF -	GDLDPA I DA	AR PAEWTGF	HRL	KALWQDD 48
Pss1	Raayori	APAAQRL -	AELDNA I NA	AR ADYYEKREQDPGFSGF	HR I	YGLFEQH 51
Algp7	RMSYERI	PIAELF -	SDLDAS I DS	SR ADDHEKAEKDPAFFGF	IR	YGLFAQN 51
EfeM	RVYYESV	PIAELF -	SDLDAS IDS	SR VDDHEQGVAAEDFTGF	HRL	YALFSQN 51
Myc2	RLPWERI	EPLAGLV -	EEIDGKVDA	AR VDDFAG - VDDPAFTGW	HRL	YLLFSQN 50
Pad2	RTSY KI	PIAELF -	SDLDVSIDA	AR ADDYEQAEQDPDFTGF	HR I	YGLWEQG 51

	254	260	270	280	290	300
EfeO	-EDRY	SHTDLWDFQA	NVEGSOKIV	-DLLRPQLQKAN	PELLAKVDAN	FKKV 49
Acc	-EERY	SHTDLYDIDA	NLAGSKAAF	-QAVEPILRERD	PALATTIDARI	FADV 49
Pss1	-EERY	SHSDLNGFAA	NLDGTRKIV	-DLLRPLLARSA	GDLLQKIDAAN	WADL 49
Algp7	-EEDRY	Shtdlwdfqa	NFEGAKKIV	-DLFRPLVVKDN	RAFADKVDANI	FDTV 49
EfeM	-EDRY	SHTDLYDFQG	NIDGAKKIV	-DLFRPQIEQQD	KAFSSKVDKNI	FATV 49
Myc2	-EDRY	SKTDLNDFEA	NLQGSEAAV	-NRLSPALVKAD	PALLGKIEAG	FSEI 49
Pad2	-EDRY	SRTDLWDFDA	NFEGAEKIY	-ELLRPLIADDE	ADFTAKVDGNI	FAAV 49

	EfeB	Algp7 (EfeO _{II})	EfeO _I		
Data collection					
Wavelength (Å)	1.0000	1.0000	1.0000		
Space group	$P2_{1}2_{1}2_{1}$	$P2_{1}2_{1}2_{1}$	<i>C</i> 2		
Unit cell parameters (Å, °)	<i>a</i> = 100.0, <i>b</i> = 105.0,	<i>a</i> = 52.8, <i>b</i> = 97.5,	<i>a</i> = 139.9, <i>b</i> = 51.9,		
	<i>c</i> = 83.8	<i>c</i> = 104.9	<i>c</i> = 117.4		
			$\alpha = 90.0, \beta = 112.4, \gamma = 90.0$		
Resolution limit (Å)	$50.0 - 2.30 (2.44 - 2.30)^{a}$	$50.0 - 1.88 (1.91 - 1.88)^{a}$	$50.0 - 1.85 (1.96 - 1.85)^{a}$		
Total reflections	307,536 (50,278) ^a	307,493 (14,878) ^a	249,772 (33,933) ^a		
Unique reflections	38,892 (6,217) ^a	44,838 (2,188) ^a	65,646 (9,994) ^a		
Completeness (%)	97.5 (97.9) ^a	99.8 (99.3) ^a	97.7 (92.8) ^a		
Ι/σ (Ι)	$19.4(10.2)^{a}$	$36.5(5.3)^{a}$	$18.5(2.48)^{a}$		
$R_{\rm merge}$ (%)	6.1 (14.8) ^a	$7.2 (48.3)^{a}$	$3.9 (40.0)^{a}$		
CC(1/2)	99.9 (99.7) ^a	99.9 (95.7) ^a	99.9 (91.8)		
Wilson B (Å ²)	31.2	28.0	37.9		
Refinement					
Resolution limit (Å)	$44.5 - 2.30 (2.36 - 2.30)^{a}$	$46.5 - 1.88 (1.92 - 1.88)^{a}$	$48.2 - 1.85 (1.87 - 1.85)^{a}$		
<i>R</i> -factor (%)	$23.4(24.8)^{a}$	$18.2(25.1)^{a}$	20.1 (34.0) ^a		
$R_{\rm free}$ (%)	29.8 (36.0) ^a	$21.8(33.0)^{a}$	24.4 (37.3) ^a		
Final model	781 residues, 257 waters,	502 residues, 421 waters,	697 residues, 322 waters,		
	24 1,2-ethanediol,	9 1,2-ethanediol, 1 citrate	30 1,2-ethanediol,		
	1 oxygen molecule,		1 acetate, 1 sulfate,		
	5 di (hydroxyethyl) ether,		1 triethylene glycol,		
	3 triethylene glycol, 2		1 zinc ion		
	heme				
r.m.s.d.					
Bond (Å)	0.008	0.01	0.006		
Angle (°)	0.971	1.47	0.806		
Ramachandran plot (%)					
Favored regions	96.0	98.0	98.1		
Allowed regions	3.5	1.8	1.7		
Outliers	0.5	0.2	0.14		
Clashscore	7	8	5.85		
Rotamer ouliers (%)	0.63	0	2.5		
PDB ID	6JBN	6JBO	7WGU		

 Table S1. X-ray data collection and structure refinement statistics.

^a Data on highest shells are given in parenthesis.