

## A geminal method based on the generalized electron pairing applied to the Heisenberg model of hydrocarbons

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1 In the approximate valence bond (VB) description of  
2 molecular electronic structures, the resonating VB effect  
3 might be incorporated in an efficient manner by mixing the  
4 triplet component into conventional singlet geminals. We  
5 developed a variational optimization scheme for this  
6 *generalized pairing* type wave function in the framework of  
7 the spin Hamiltonian model. With numerical verifications,  
8 we found the resonance stabilization is partially described  
9 through the contamination of higher spin states for molecules  
10 such as non-Kekulé hydrocarbons.

11 **Keywords:** Valence bond theory, Geminal theory,  
12 **Heisenberg spin Hamiltonian**

13 The valence bond (VB) theory, based on the Lewis's  
14 electron pair concept, is an appealing chemical theory  
15 providing insights into molecular electronic structures from a  
16 perspective different from delocalized molecular orbitals.<sup>1-3</sup>  
17 The Heisenberg model, or classical VB model, is quite useful  
18 to qualitatively understand electronic structures from a  
19 viewpoint of localized chemical bonds. Notably, the  
20 Ovchinnikov's rule,<sup>4</sup> which predicts the ground state spin  
21 multiplicity of hydrocarbons, is derived from the Heisenberg  
22 model. This spin Hamiltonian theory was extensively applied  
23 to conjugated hydrocarbons.<sup>5-14</sup> The accumulated results  
24 show that the neutral ground and low-lying excited states of  
25 many conjugated hydrocarbons are well described by this  
26 simple spin Hamiltonian model. Theoretical developments in  
27 the classical VB theory were reviewed by Klein et al.<sup>15</sup>

28 In the rigorous VB model, however, the number of  
29 resonance structures combinatorially increases as the  
30 number of electrons in the system increases; we need to  
31 consider a huge number of VB basis functions even for  
32 medium-sized molecules. This fact complicates the  
33 computation and interpretation of VB wave functions. The  
34 situation is essentially the same in both *ab initio*<sup>16-18</sup> and  
35 qualitative VB theories. Due to these circumstances, one  
36 representative resonance structure is often used to describe  
37 the electronic state in simple terms. (Chemists may  
38 unconsciously make a selection for the chemical structure  
39 formula of a conjugated hydrocarbon.) This approximation in  
40 the VB method is called *perfect pairing* (PP). If a simple  
41 method to correct the resonating VB effect ignored in this  
42 approximation is established, it is expected to efficiently  
43 improve the numerical accuracy of the compact trial wave  
44 function while maintaining a clear chemical picture.

45 We mention the theory of many-electron wave functions  
46 based on the two-electron unit or *geminal*,<sup>19-21</sup> a fundamental  
47 basis of our present study. Several researchers devised  
48 methods to efficiently correct the geminal product type wave  
49 functions by partially relaxing the spin symmetry of

50 geminals.<sup>22-25</sup> Furthermore, Surján, Szabados and their  
51 coworkers developed a perturbation theory<sup>26</sup> and spin-  
52 projection techniques<sup>27-30</sup> to improve spin-unrestricted  
53 geminals. Some other directions to extend geminal product  
54 wave functions can be found in refs [31-33]. Notably, in the  
55 field of quantum Monte Carlo, the idea of *generalized pairing*  
56 (GP),<sup>34,35</sup> or fully extended spin functions in geminals, was  
57 proposed to capture the electron correlation. By mixing the  
58 triplet functions into standard singlet-type geminals, inter-  
59 geminal correlation effect is expected to be incorporated in  
60 an efficient manner. To the best of our knowledge, however,  
61 the GP concept has not been well tested especially for  
62 polyatomic molecular systems. In addition, its chemical  
63 meaning does not seem to be fully understood.

64 In this work, a new method to incorporate the resonating  
65 VB effect into the simple PP wave function was devised and  
66 numerically tested in the framework of the Heisenberg model.  
67 We propose a variational optimization scheme for GP type  
68 wave functions and the chemical meaning is clarified for  
69 unsaturated hydrocarbons.

70 In what follows, we present the formulation and method  
71 for the computation of GP type VB wave functions based on  
72 the variational principle. The Heisenberg spin Hamiltonian  
73 considered here is

$$74 \quad H = \sum_{\mu < \nu} J_{\mu\nu} \mathbf{S}_{\mu} \cdot \mathbf{S}_{\nu} \quad (1)$$

75 where  $J_{\mu\nu}$  is the exchange interaction constant and  $J_{\nu\mu} = J_{\mu\nu}$  ( $\mu$   
76  $< \nu$ ) is satisfied for convenience. The spin operator acts on the  
77 orthogonal basis functions  $\chi_{\mu}$  placed on the atomic sites. We  
78 assume  $J_{\mu\nu} > 0$ , i.e., antiferromagnetism. The constant  
79 parameters are assumed to include some non-orthogonality  
80 effects, which are generally important for the VB description.  
81 The spin operator can be represented as

$$82 \quad \mathbf{S}_{\mu} \cdot \mathbf{S}_{\nu} = \frac{1}{2} \left( a_{\nu}^{\beta+} a_{\mu}^{\alpha+} a_{\mu}^{\beta-} a_{\nu}^{\alpha-} + a_{\mu}^{\beta+} a_{\nu}^{\alpha+} a_{\nu}^{\beta-} a_{\mu}^{\alpha-} \right) \\ + \frac{1}{4} \left( a_{\mu}^{\alpha+} a_{\nu}^{\alpha+} a_{\nu}^{\alpha-} a_{\mu}^{\alpha-} - a_{\mu}^{\alpha+} a_{\nu}^{\beta+} a_{\nu}^{\beta-} a_{\mu}^{\alpha-} \right. \\ \left. - a_{\mu}^{\beta+} a_{\nu}^{\alpha+} a_{\nu}^{\alpha-} a_{\mu}^{\beta-} + a_{\mu}^{\beta+} a_{\nu}^{\beta+} a_{\nu}^{\beta-} a_{\mu}^{\beta-} \right) \quad (2)$$

83 where  $a_{\mu}^{\sigma+}$  and  $a_{\mu}^{\sigma}$  are creation and annihilation operators for  
84 spinorbital  $\chi_{\mu}^{\sigma}$  (having spatial part  $\chi_{\mu}$  and spin part  $\sigma$ ),  
85 respectively.

86 The tested trial wave functions for  $2N$ -electron systems are  
87 written as

$$88 \quad \Phi = \hat{A}[\psi_1(x_1, x_2)\psi_2(x_3, x_4)\cdots\psi_N(x_{2N-1}, x_{2N})] \quad (3)$$

89 where  $\psi_i$  is an antisymmetrized geminal,  $\hat{A}$  is the  
90 antisymmetrizer between geminals, and  $x_n$  shows the spatial

1 coordinate  $r_n$  and spin coordinate. The PP wave function  $\Phi_{\text{PP}}$   
 2 for the singlet state is simply constructed using normalized  
 3 singlet geminals as  $\psi_i$  in eq (3),

$$4 \quad \psi_i^{\text{PP}} = |\chi_\mu^\alpha \chi_\nu^\beta| - |\chi_\mu^\beta \chi_\nu^\alpha| \quad (\mu, \nu \in i). \quad (4)$$

5 The notation  $\mu, \nu \in i$  means  $\chi_\mu$  and  $\chi_\nu$  are exclusively involved  
 6 in the geminal labeled  $i$ . Higher spin states can be described  
 7 by replacing the singlet geminals with proper spin functions.  
 8 Correspondence between  $\Phi_{\text{PP}}$  and a chemical structure is  
 9 obvious by linking the atomic sites where  $\chi_\mu$  and  $\chi_\nu$  in each  
 10 geminal are placed on. The exact VB wave function can be  
 11 represented as a superposition of all the linearly independent  
 12 PP functions.

13 By extending the geminal to include the general spin  
 14 functions,

$$15 \quad \psi_i^{\text{GP}} = \sum_{\sigma_1 \sigma_2} C_{\mu\nu}^{\sigma_1 \sigma_2} |\chi_\mu^{\sigma_1} \chi_\nu^{\sigma_2}| \quad (\mu, \nu \in i), \quad (5)$$

16 we give the GP wave function  $\Phi_{\text{GP}}$  as an extension of the  
 17 corresponding  $\Phi_{\text{PP}}$ . The geminal expansion coefficients  
 18  $C_{\mu\nu}^{\sigma_1 \sigma_2}$  are variational parameters. Note that the so-called  
 19 strong orthogonality condition,

$$20 \quad \int \psi_i(r_1, r_2) \psi_k(r_1, r_3) dr_1 = 0 \quad (i \neq k), \quad (6)$$

21 which makes the variational solution tractable, is already  
 22 satisfied for  $\Phi_{\text{PP}}$  and  $\Phi_{\text{GP}}$ . Due to the strong orthogonality, the  
 23 structures of one- and two-electron density matrices become  
 24 quite simple. Non-zero elements of the density matrices for  
 25  $\Phi_{\text{GP}}$  are written as

$$26 \quad P_{\mu\mu}^{\sigma_1 \sigma_2} = \langle \Phi_{\text{GP}} | a_\mu^{\sigma_2+} a_\mu^{\sigma_1} | \Phi_{\text{GP}} \rangle, \quad (7)$$

$$= \sum_{\sigma_3} C_{\mu\nu}^{\sigma_1 \sigma_3} C_{\mu\nu}^{\sigma_2 \sigma_3} \quad (\mu, \nu \in i)$$

$$27 \quad \Gamma_{\mu\nu, \mu\nu}^{\sigma_1 \sigma_2, \sigma_3 \sigma_4} = \langle \Phi_{\text{GP}} | a_\mu^{\sigma_3+} a_\nu^{\sigma_4+} a_\nu^{\sigma_2} a_\mu^{\sigma_1} | \Phi_{\text{GP}} \rangle$$

$$= \begin{cases} C_{\mu\nu}^{\sigma_1 \sigma_2} C_{\mu\nu}^{\sigma_3 \sigma_4} & (\mu, \nu \in i) \\ P_{\mu\mu}^{\sigma_3 \sigma_1} P_{\nu\nu}^{\sigma_4 \sigma_2} & (\mu \in i, \nu \in k, i \neq k) \end{cases}, \quad (8)$$

$$28 \quad \Gamma_{\mu\nu, \nu\mu}^{\sigma_1 \sigma_2, \sigma_3 \sigma_4} = \langle \Phi_{\text{GP}} | a_\nu^{\sigma_3+} a_\mu^{\sigma_4+} a_\nu^{\sigma_2} a_\mu^{\sigma_1} | \Phi_{\text{GP}} \rangle$$

$$= \begin{cases} -C_{\mu\nu}^{\sigma_1 \sigma_2} C_{\mu\nu}^{\sigma_4 \sigma_3} & (\mu, \nu \in i) \\ -P_{\mu\mu}^{\sigma_4 \sigma_1} P_{\nu\nu}^{\sigma_3 \sigma_2} & (\mu \in i, \nu \in k, i \neq k) \end{cases}. \quad (9)$$

29 We note ionic terms do not appear in the trial wave function  
 30 because the present formulation is based on the spin  
 31 Hamiltonian in eq (1). If ionic terms appear in *ab initio* cases,  
 32 the non-diagonal and other elements of density matrices  
 33 might have non-zero values as shown, e.g., in ref [21]. In the  
 34 right-hand-side of eq (7),  $\nu$  comes from the same geminal  $\mu$   
 35 belongs to.

36 Then, the energy expectation value is expressed as follows:

$$E = \langle \Phi_{\text{GP}} | H | \Phi_{\text{GP}} \rangle$$

$$= \sum_{\mu < \nu} J_{\mu\nu} \left( -\frac{1}{2} \Gamma_{\nu\mu, \mu\nu}^{\alpha\beta, \alpha\beta} - \frac{1}{2} \Gamma_{\mu\nu, \nu\mu}^{\beta\alpha, \beta\alpha} - \frac{1}{4} \Gamma_{\nu\mu, \mu\nu}^{\alpha\alpha, \alpha\alpha} \right.$$

$$\left. - \frac{1}{4} \Gamma_{\nu\mu, \mu\nu}^{\beta\beta, \beta\beta} + \frac{1}{4} \Gamma_{\nu\mu, \mu\nu}^{\beta\alpha, \alpha\beta} + \frac{1}{4} \Gamma_{\nu\mu, \mu\nu}^{\alpha\beta, \beta\alpha} \right)$$

$$= \sum_i \sum_{\mu < \nu \in i} J_{\mu\nu} \left\{ C_{\mu\nu}^{\alpha\beta} C_{\mu\nu}^{\beta\alpha} + \frac{1}{4} (C_{\mu\nu}^{\alpha\alpha})^2 + \frac{1}{4} (C_{\mu\nu}^{\beta\beta})^2 \right.$$

$$\left. - \frac{1}{4} (C_{\mu\nu}^{\alpha\beta})^2 - \frac{1}{4} (C_{\mu\nu}^{\beta\alpha})^2 \right\} + \frac{1}{2} \sum_{\mu} \sum_{\sigma_1 \sigma_2} \tilde{h}_{\mu\mu}^{\sigma_1 \sigma_2} P_{\mu\mu}^{\sigma_2 \sigma_1} \quad (10)$$

37 where the one-electron effective potential is written as

$$39 \quad \tilde{h}_{\mu\mu}^{\sigma_1 \sigma_2} = \begin{cases} \frac{1}{4} \sum_{k(\neq i)} \sum_{\lambda \in k} J_{\mu\lambda} (P_{\lambda\lambda}^{\alpha\alpha} - P_{\lambda\lambda}^{\beta\beta}) & (\sigma_1 = \sigma_2 = \alpha) \\ \frac{1}{4} \sum_{k(\neq i)} \sum_{\lambda \in k} J_{\mu\lambda} (P_{\lambda\lambda}^{\beta\beta} - P_{\lambda\lambda}^{\alpha\alpha}) & (\sigma_1 = \sigma_2 = \beta) \\ \frac{1}{2} \sum_{k(\neq i)} \sum_{\lambda \in k} J_{\mu\lambda} P_{\lambda\lambda}^{\sigma_1 \sigma_2} & (\sigma_1 \neq \sigma_2) \end{cases}. \quad (11)$$

40 Since the total energy is represented as the sum of the energy  
 41 attributed to each geminal and interaction energies between  
 42 two geminals, variational optimization of the wave function  
 43 is accomplished by the self-consistent optimization of each  
 44 geminal under the field created by other geminals.<sup>20, 21, 36</sup> The  
 45 two-electron effective Hamiltonian to determine the  
 46 expansion coefficients in the geminal  $i$  constructed with the  
 47 orbitals  $\chi_\mu$  and  $\chi_\nu$  is given as

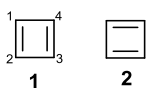
$$48 \quad H^i = \sum_{\sigma_1 \sigma_2} (\tilde{h}_{\mu\mu}^{\sigma_1 \sigma_2} a_\mu^{\sigma_1+} a_\mu^{\sigma_2} + \tilde{h}_{\nu\nu}^{\sigma_1 \sigma_2} a_\nu^{\sigma_1+} a_\nu^{\sigma_2}) + J_{\mu\nu} \mathbf{S}_\mu \cdot \mathbf{S}_\nu. \quad (12)$$

49 We evaluated the effective Hamiltonian matrix elements  
 50  $H_{\mu\nu, \mu\nu}^{\sigma_1 \sigma_2, \sigma_3 \sigma_4}$  for the antisymmetrized two-electron functions  
 51  $|\chi_\mu^{\sigma_1} \chi_\nu^{\sigma_2}|$  and  $|\chi_\mu^{\sigma_3} \chi_\nu^{\sigma_4}|$ . These four-by-four matrix  
 52 elements are gathered in **Table 1**. It can be observed that  
 53 different spin parts are allowed to mix in each geminal  
 54 through the effective one-electron potential. The sequential  
 55 optimization of each geminal is performed as follows. Firstly,  
 56 we give initial guess values for the geminal expansion  
 57 coefficients and compute the density matrices based on eqs  
 58 (7), (8) and (9). Since the effective one electron potential is  
 59 evaluated with the density matrices, we can also compute the  
 60 elements of the two-electron effective Hamiltonian matrix.  
 61 By sequentially diagonalizing the Hamiltonian matrices,  
 62 geminal expansion coefficients are updated. This process is  
 63 repeated until the self-consistency is achieved. For the  
 64 present purpose, optimization of one-electron orbitals  
 65 constructing each geminal is not performed.

66 We demonstrate several systems where the instability is  
 67 caused in  $\Phi_{\text{PP}}$ , and GP solutions were obtained. The  
 68 Heisenberg model requires only the information of the  
 69 linkage between the atomic sites, or molecular graph. Since a  
 70 molecular graph is identified with the carbon skeleton of a  
 71 conjugated hydrocarbon, we refer to the graph by the name

1 of the corresponding compound. Through our preliminary  
 2 calculations, we found the instability does not appear in  $\Phi_{PP}$   
 3 of alternant hydrocarbons such as ethylene, butadiene,  
 4 benzene, anthracene, and so on. This made us realize again  
 5 that the PP is a good approximation for many conjugated  
 6 hydrocarbons.

7 Firstly, we consider four atomic sites arranged on four  
 8 vertices of a square, representing cyclobutadiene (CBD)  
 9 molecule. Two PP functions schematically shown in **Figure**  
 10 **1** construct a complete basis for this four-electron system.  
 11 These correspond to the Kekulé structures. We write  $\Phi_{PP}$  and  
 12  $\Phi_{GP}$  corresponding to the structure  $\mathbf{X}$  as PP(X) and GP(X),  
 13 respectively. Non-zero values of the exchange interactions  
 14 were assumed only between adjacent sites;  $J_{12} = J_{34} = \varepsilon J$  and  
 15  $J_{14} = J_{23} = J$  where  $J$  is a constant and  $\varepsilon$  is a real parameter.  
 16

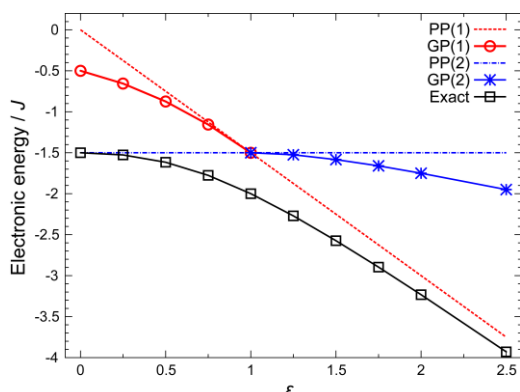


17 **Figure 1.** Two perfect pairing structures of CBD. Numbering for the  
 18 atomic sites is also shown.

21 Electronic energy changes of this system for the scaling  
 22 parameter  $\varepsilon$  are shown in **Figure 2**. When  $\varepsilon$  is zero, the four-  
 23 site model can be considered as two isolated ethylene  
 24 molecules forming bonds 1-4 and 2-3. The PP(2) gave the

51 **Table 1.** Matrix elements  $H_{\mu\nu,\mu\nu}^{\sigma_1\sigma_2,\sigma_3\sigma_4}$  for the effective two-electron Hamiltonian. The rows and columns represent the types of spin functions  $\sigma_1\sigma_2$  and  $\sigma_3\sigma_4$ , respectively. Only the upper triangular part is shown.

	$\alpha\beta$	$\beta\alpha$	$\alpha\alpha$	$\beta\beta$
$\alpha\beta$	$\tilde{h}_{\mu\mu}^{\alpha\alpha} + \tilde{h}_{\nu\nu}^{\beta\beta} - \frac{1}{4} J_{\mu\nu}$	$\frac{1}{2} J_{\mu\nu}$	$\tilde{h}_{\nu\nu}^{\beta\alpha}$	$\tilde{h}_{\mu\mu}^{\alpha\beta}$
$\beta\alpha$		$\tilde{h}_{\mu\mu}^{\beta\beta} + \tilde{h}_{\nu\nu}^{\alpha\alpha} - \frac{1}{4} J_{\mu\nu}$	$\tilde{h}_{\mu\mu}^{\beta\alpha}$	$\tilde{h}_{\nu\nu}^{\alpha\beta}$
$\alpha\alpha$			$\tilde{h}_{\mu\mu}^{\alpha\alpha} + \tilde{h}_{\nu\nu}^{\alpha\alpha} + \frac{1}{4} J_{\mu\nu}$	0
$\beta\beta$				$\tilde{h}_{\mu\mu}^{\beta\beta} + \tilde{h}_{\nu\nu}^{\beta\beta} + \frac{1}{4} J_{\mu\nu}$

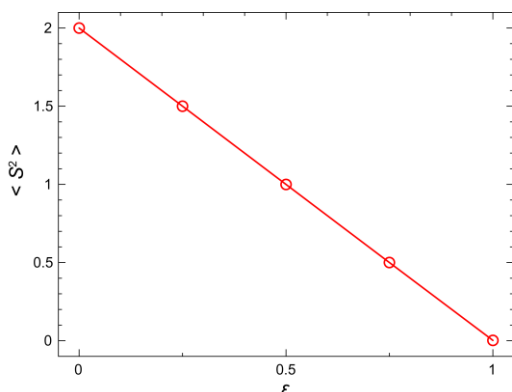


57 **Figure 2.** The electronic energies of the four-site model system in the  
 58 unit of  $J$  for the scaling parameter  $\varepsilon$ . See text for the notations. The exact  
 59 energies are also shown.  
 60

25 exact energy while the PP(1) gave the energy value of zero.  
 26 In this context, as  $\varepsilon$  is increased from zero, the two ethylene  
 27 molecules get closer. When  $\varepsilon$  becomes larger than one, two  
 28 bonds are recoupled to form stronger bonds 1-2 and 3-4. The  
 29 PP(1) becomes more stable than PP(2) in this region. Namely,  
 30 this four-site system can be considered as a model for the  
 31 bond recombination process. Because the more stable PP  
 32 wave function exists, PP(1) and PP(2) are not proper  
 33 representations of the system in the regions  $0 \leq \varepsilon \leq 1$  and  $1 \leq$   
 34  $\varepsilon$ , respectively. Notably, the GP type solutions of GP(1) and  
 35 GP(2) appeared in these regions. In the present case, there  
 36 exists the PP wave function more stable than the GP solutions.  
 37 The appearance of a GP solution means the instability of a  
 38 certain PP function, not always indicating the existence of the  
 39 globally stable geminal product wave function. We mention  
 40 the ground and low-lying excited states of CBD were  
 41 described by Voter and Goddard with their generalized  
 42 resonating valence bond (GRVB) wave function.<sup>37</sup> Mixing of  
 43 the two PP structures leads to the resonant and anti-resonant  
 44 singlet states. They showed the rectangular distortion of CBD  
 45 stabilizes the resonant state and destabilizes the anti-resonant  
 46 state. As the structure distorts, each state becomes similar to  
 47 a different PP function. The instability of the present PP wave  
 48 function, related to the emergence of GP solution, can also be  
 49 understood from the putative behaviors of resonant and anti-  
 50 resonant states.

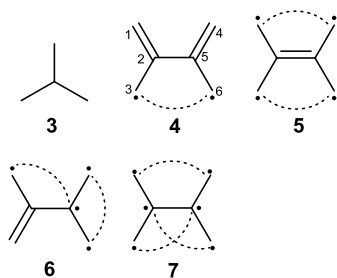
61 As  $\varepsilon$  is moved away from the value of one, the difference  
 62 between the PP and GP energies becomes large, while the  
 63 difference between the GP and exact energies also becomes  
 64 large. This behavior is due to the increasing spin  
 65 contamination in the GP wave function. The expectation  
 66 values of the total  $S^2$  operator for GP(1) are shown in **Figure**  
 67 **3**. Due to the symmetry of the system, it is sufficient to  
 68 consider only the behavior at  $0 \leq \varepsilon \leq 1$ . At  $\varepsilon = 0$ , the GP wave  
 69 function is the spin alternant (SA) determinant (i.e., the Néel  
 70 state in the literature of solid state physics), giving the value  
 71 of two for  $\langle S^2 \rangle$ . At  $\varepsilon = 1$ , the GP wave function degenerates  
 72 to the PP wave function. The expectation value  $\langle S^2 \rangle$  changes  
 73 monotonically as  $\varepsilon$  increases. We found the numerically  
 74 computed  $\langle S^2 \rangle$  is proportional to  $\varepsilon$  at  $0 \leq \varepsilon \leq 1$ . The result  
 75 implies that the proportion of the triplet component contained  
 76 with different signs in each geminal is proportional to  $\varepsilon^{1/4}$ . In  
 77

1 our present formulation, the  $S_z$  spin-symmetry is not  
 2 constrained. Thus, we obtained degenerate GP solutions in  
 3 the  $S_z$  space, depending on the initial guess for the geminal  
 4 expansion coefficients.



6  
 7 **Figure 3.** The expectation value of the total  $S^2$  operator for the  
 8 generalized pairing solution of GP(1). See text for the notations.

9  
 10 Secondly, we discuss simple non-Kekulé molecules,  
 11 tetramethylenemethane (TMM) and tetramethyleneethane  
 12 (TME), as examples. The molecular graph of TMM (3) and  
 13 four PP structures of TMM (4-7) are shown in **Figure 4**. We  
 14 assumed the non-zero value of the exchange interaction  
 15 constant  $J$  only between the adjacent atomic sites. Although  
 16 TME has two types of bonds, our current interest lies in its  
 17 non-Kekulé molecular topology, so we used a single constant  
 18 parameter for simplicity. The ground electronic state of TMM  
 19 is triplet, while that of TME is singlet. The PP wave function  
 20 of the triplet TMM is constructed with the singlet and triplet  
 21 geminals. We found  $\Phi_{PP}$  of these non-Kekulé molecules are  
 22 not stable and broken symmetry solutions are obtained by the  
 23 variational calculations.



26  
 27 **Figure 4.** The molecular graph of TMM (3) and four perfect pairing  
 28 structures of TME (4-7). Dashed lines show the singlet coupling between  
 29 non-adjacent atomic sites. Numbering for the atomic sites is also shown.

30  
 31 The electronic energies of TMM calculated with the SA,  
 32 PP, GP and exact wave functions are gathered in **Table 2**. We  
 33 found  $\Phi_{GP}$  gives lower energy than the SA determinant and  
 34  $\Phi_{PP}$ . The expectation value  $\langle S^2 \rangle$  of  $\Phi_{GP}$  was the intermediate  
 35 of those for SA and  $\Phi_{PP}$ . The GP energy accounts for 58.6 %  
 36 of the resonance energy, which is here defined as the  
 37 difference between the energies of the exact and the most  
 38 stable PP wave function. This implies that the resonating VB

39 effect is partially incorporated in the GP wave function while  
 40 the spin contamination cannot be ignored in terms of energy.

41 The energies of four PP structures of TME (4-7) and  
 42 those of corresponding GP wave functions are shown in  
 43 **Table 3**. The expectation value  $\langle S^2 \rangle$  of the four GP are also  
 44 given. Among the PP wave functions, PP(4) gives the lowest  
 45 energy because there are two bonds 1-2 and 4-5 in the  
 46 structure. Concerning the GP wave functions, GP(4) was the  
 47 most stable; it is considered as the globally stable geminal  
 48 product. The exact energy was  $-2.1642J$ ; GP(4) accounts for  
 49 71.3% of the resonance energy. Projection techniques for  
 50 restoring of the spin symmetry will be one promising  
 51 approach to further improve the GP wave functions.

52  
 53  
 54 **Table 2.** The electronic energies in the unit of  $J$  for TMM (3)  
 55 described with the SA, PP, GP and exact wave functions. The expectation  
 56 values of the total  $S^2$  operator are also shown.

	Energy / $J$	$\langle S^2 \rangle$
SA	-0.75	3
PP	-0.75	2
GP	-0.9571	2.2929
Exact	-1.25	2

57  
 58 **Table 3.** The electronic energies in the unit of  $J$  for TME described  
 59 with PP (4-7) and corresponding GP wave functions. The expectation  
 60 values  $\langle S^2 \rangle$  for the GP wave functions are also shown.

	PP	GP	$\langle S^2 \rangle$
4	-1.5	-1.6908	1.4505
5	-0.75	-1.3680	2.5528
6	-0.75	-1.4571	2.2929
7	0	-1.2500	3.0000

61  
 62 In summary, we developed a new method to  
 63 variationally optimize the generalized electron pairing type  
 64 VB wave function, and verified it in the Heisenberg model.  
 65 By numerical computations, chemical implications of the  
 66 spin-symmetry broken wave function were explored. We  
 67 found the mixing of higher spin states stabilizes the wave  
 68 function in the case when a certain PP wave function is not a  
 69 proper representation of the system; non-Kekulé molecule is  
 70 a typical example. This instability of the perfect pairing is  
 71 consistent with our empirical knowledge in chemistry. The  
 72 present method might also be applied to chemically  
 73 interesting open-shell conjugated molecules.<sup>37</sup> And lastly, the  
 74 tested trial wave function will also be useful as an *ab initio*  
 75 model and a reference to describe complex molecular  
 76 electronic structures.

77  
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