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<th>Unoccupied electronic states of 3d-transition metal phthalocyanines (MPc: M = Mn, Fe, Co, Ni, Cu and Zn) studied by inverse photo-emission spectroscopy / Ion channels of alamethicin dimers N-termimnally linked by disulfide bond</th>
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Kyoto University
Interface Science - Molecular Aggregates -

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Prof ZHAO, Kongshuang South China Normal University, China, 26 December 2000-25 March 2001.

Scope of Research
The research at this subdivision is devoted to correlation studies on structures and properties of both natural and artificial molecular aggregates from two main standpoints: photoelectric and dielectric properties. The electronic structure of organic thin films is studied using photoemission and inverse photoemission spectroscopies in connection with the former, and its results are applied to create novel molecular systems with characteristic electronic functions. The latter is concerned with heterogeneous structures in microcapsules, biopolymers, biological membranes and biological cells, and the nonlinearity in their dielectric properties is also studied in relation to molecular motions.

Research Activities (Year 2001)

Presentations
Characterization of biological cells by dielectric spectroscopy, Asami K, 1st International Conference on Dielectric spectroscopy in Physical, Chemical and Biological Applications (Jerusalem, Israel), 13-15 March.
Unoccupied states of organic semiconductor thin films examined by inverse photoemission spectroscopy, Sato N, Workshop on Advanced Spectroscopy of Organic Materials for Electronic Applications (Glumslöv, Sweden), 4-7 June.

Grants
Unoccupied electronic states of 3d-transition metal phthalocyanines (MPc: M = Mn, Fe, Co, Ni, Cu and Zn) studied by inverse photo-emission spectroscopy

Unoccupied electronic states of thin films of 3d-metal phthalocyanines (MPc: M = Mn, Fe, Co, Ni, Cu and Zn) were systematically investigated in the energy range from the Fermi level to the vacuum levels using inverse photo-emission spectroscopy: obtained spectra are shown in Fig. 1. The spectra obtained for NiPc and CuPc were in good agreement with spectra previously reported. Overall the features of the observed spectra were similar to each other, regardless of the substitution of the central metals. This is explained by the dominant contribution to the spectra from the macrocyclic π orbitals of the Pc framework.

Changes in the contribution of the central metal, however, cause small variations in the spectra. In our analysis of the metal contribution, we examined the difference spectra obtained by subtracting a ZnPc reference spectrum (as shown as purple-colored areas in Fig. 1). The choice of ZnPc as a reference was made because the d orbitals of the zinc ion in ZnPc are fully occupied.

The difference spectra of MnPc, FePc, NiPc and CuPc were found to agree well with the reported X-ray absorption spectra. It is thus concluded that the difference spectra reflect the density of unoccupied states derived from the central metals. The difference spectra of MnPc, FePc, CoPc, NiPc and CuPc were further tentatively assigned with the help of reported results from extended Hückel and density functional calculations.

Ion channels of alamethicin dimers N-terminally linked by disulfide bond

Because of structural resemblance, the helix-bundle type channels formed by alamethicin, 20-residue antibiotic peptide, in lipid bilayers are available as a model of biological channels. The channels, however, have drawbacks in single channel analysis: the multi-conductance behavior that are due to transient changes in molecularity of peptides in a helix bundle, and the voltage-gated channel formation preventing characterization of the current-voltage relation.

To overcome the drawbacks we adopted covalent alamethicin dimers, that restrict the peptide molecularity and form long lasting channels that allow us to measure the current-voltage relations by applying a fast voltage ramp during channel opening. The dimers, N-terminally linked by a disulfide bond, revealed the relationship between the channel conductance and the peptide molecularity, and demonstrated that negative fixed charges at the narrowest part of the pore much more enhanced cation-selectivity in ion permeation than those at the pore mouth.