with P, but reached to a constant value at higher value of P (Table 1). Unfractionated polymer of  $\bar{P}$ =800 showed lower value than fractionated one, although this was uncertain at higher value of  $\bar{P}$ .

Table 1. Relation between the degree of polymerization and tensile strength (Kg/mm<sup>2</sup>)

Degree P of drawing (%)	370 590	F00	1040	1970	11600	16000	Unfractionated	
		590					800	8900
0	6.4	7.2	7.4	6.7	6.9	7.0	4 8	7.0
100	8.5	10.0	9.8	9.8	10.3	10.0	7.7	11.1
200	9.9	12.3	12.2	12.3	14.2	13.2	8.7	17.0
400	12.9	17.2	. 18.8	18.8	21.0	18.5	12.6	29.0

The degree of polymerization has more remarkable influence on the flex-life (folding strength) than on tensile strength (Table 2). The flex-life of unfractionated polymer was inferior to fractionated one.

Table 2. Relation between the degree of polymerization and flex-life (cycles).

Degree P	070	F00	1970 3880	2000	Unfractionated		
ing (%)	370	590		3880	840	13000	
0	28	24	27	22	20	22	
100	89	426	680	712	115	97	
200	238	1496	3105	2683	188	189	
400	735	4010	18160	21780	669	2630	

## 23. The Elasto-viscous Behaviour of Plasticized Polyvinyl Chloride at the Elevated Temperature. (III)

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(Sakurada Laboratory)

The variation of the elongation of plasticized polyvinyl chloride films with time under constant load was estimated at various temperatures. Appling the four element mechanical model which consists of springs and dash-pots (W.M. Gearhart and W.D. Kennedy, Ind. Eng. Chem., 41, 695 (1949)), the spring constants and viscosity of dash-pots were calculated. As plasticizers, di-n-octyl phthalate (DOP) and tricresyl phosphate (TCP) were used.

The relation between the concentration of DOP and the logarithm of the principal viscosity  $\eta_3$  (the viscosity of the series connected dash-pot) was linear at every temperature except 150°C. Linear relation between ln  $\eta_3$  and 1/T (T;

absolute temperature) was observed and the activation energy of viscous flow  $E_3$  was calculated. The values of  $E_3$  for TCP, DOP and DBP (dibutyl phthalate) were 7.5, 7.0 and 5.8 Kcal., respectively (weight ratio, resin 100: plasticizer 100). TCP gave higher values of  $\eta_3$  than DOP at lower temperatures, but at an elevated temperature this order was reversed (Table).

Values of  $\eta_3$  of plasticized polyvinyl chloride at various temperatures (Weight ratio, resin 100: plasticizer 100). (unit of  $\eta_3$ : poise).

Temp. (°C)	16-28	65	100	150
DBP	1.53×10 <sup>10</sup>	4.34×109	$2.08{ imes}10^{9}$	3.61×10 <sup>8</sup>
DOP	$2.28 \times 10^{10}$	$5.28 \times 10^{9}$	$1.66{ imes}10^{9}$	$6.92 \times 10^{8}$
TCP	$3.55{ imes}10^{10}$	5.92×109	$1.82{ imes}10^{9}$	1.00×108

## 24. Elastic and Thermal Properties of Vinylon AN and Vinylon C

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Various mechanical and thermal properties of an ordinary Vinylon, Vinylon C and Vinylon AN, which were synthesized using the same heattreated polyvinyl alcohol fiber as the raw material, were measured.

Vinylon AN and Vinylon C have higher dry and wet tanacity and wet Young's modulus (Table 1).

		Dry			t.	Wet	Wet and	
Materials	Denier	Tenacity (g)	Elong.	Young's Modulus (Kg/mm²)	Tenacity (g)	Elong.	Young's Modulus (Kg/mm²)	Dry Tenacity Ratio (%)
Ordinary Vinylon	1.90	6.04	25.7	496	3.89	29.5	94	64.4
Vinylon C	2.24	7.56	38.4	_	5.59	44.0		73.9
Vinylon AN	2.26	6.52	31.2	524	5.12	36.9	264	78.5

Table 1. Mechanical properties (room temp., RH 60-70%).

Vinylon AN showed higher degree of elasticity than ordinary Vinylon, and more excellent properties at the elevated temperature (Table 2).