

**Radiolytic activation of cytarabine prodrug possessing 2-oxoalkyl group:
one-electron reduction and cytotoxicity characteristics**

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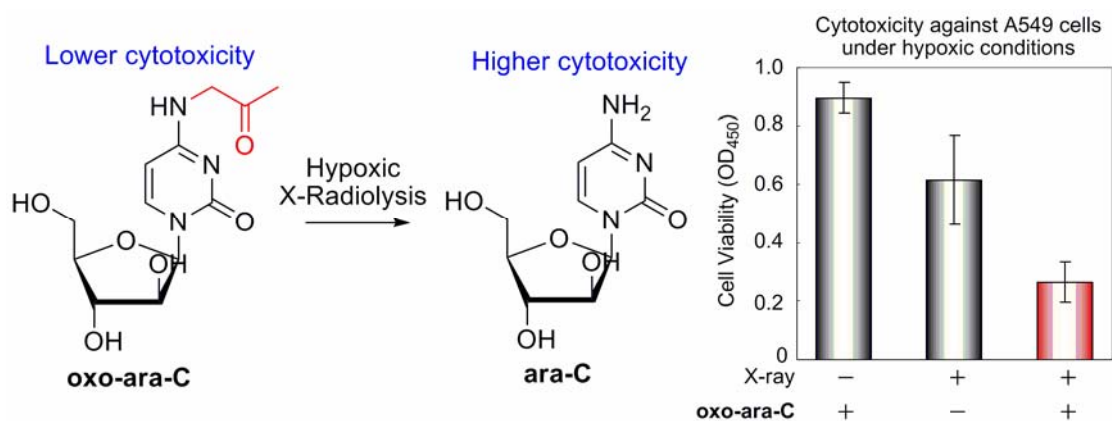
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GRAPHICAL ABSTRACT



An antitumor agent of cytarabine (**ara-C**) was conjugated with a 2-oxopropyl group at N(4) position to obtain a radiation activated prodrug (**oxo-ara-C**) that releases the toxic parent agent **ara-C** in hypoxic tumor cells *via* radiolytic one-electron reduction by hydrated electrons (e_{aq}^-).

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Abstract

An antitumor agent of cytarabine (**ara-C**) was conjugated with a 2-oxopropyl group at N(4) position to obtain a radiation activated prodrug (**oxo-ara-C**) that targets hypoxic tumor tissues to show a selective cytotoxicity. The parent antitumor agent, **ara-C**, was confirmed to be released from **oxo-ara-C** *via* one-electron reduction upon hypoxic X-irradiation. The prodrug **oxo-ara-C** was of dramatically reduced cytotoxicity against human lung adenocarcinoma A549 cells relative to **ara-C** because of the effect of 2-oxopropyl substituent. In contrast, X-irradiation of hypoxic A549 cells containing **oxo-ara-C** enhanced the cytotoxic effect, indicating that toxic **ara-C** was preferentially released in hypoxic cells *via* radiolytic one-electron reduction by hydrated electrons (e_{aq}^-).

Introduction

Cytarabine (1-*beta*-D-arabinofuranosyl-cytosine; **ara-C**) synthesized by Walwick and coworkers is a family of antimetabolites.¹ This deoxycytidine analog undergoes phosphorylation by intrinsic kinases, and is subsequently incorporated into DNA during chain elongation mediated by polymerase, thereby resulting in the inhibition of DNA replication and elongation.² Due to the highly cytotoxic effect, **ara-C** has been widely used as an antineoplastic and antiviral agent. However, **ara-C** has drawbacks in clinical applications: typically high dose of **ara-C** causes serious side effects such as encephalopathy and cerebellar dysfunction.³ Therefore, there has been a demand for giving a target specific feature to **ara-C** that could discriminate between tumor and normal cells for cytotoxicity. In view of a tumor-specific microenvironment of hypoxia as generated from an imbalance between supply and consumption of oxygen in solid tumor tissues,⁴ we have attempted to exploit a new class of antitumor prodrugs that targets a hypoxic microenvironment to show a selective antitumor effect.⁵

We have thus far identified a series of 2-oxoalkyl groups as the effective substituents for

conjugation to nucleic acid derivatives that is removable by X-irradiation under hypoxic conditions in aqueous solution.⁶ An activation mechanism has been proposed by which the 2-oxoalkyl group undergoes one-electron reduction by hydrated electrons (e_{aq}^-)⁷ generated *via* radiolysis of water to form the corresponding π^* anion radical, followed by thermal activation into the σ^* anion radical and is readily hydrolyzed to release the 2-oxoalkyl group.⁸ We have applied these characteristics of the 2-oxoalkyl group to develop prodrugs of antitumor agents, 1-(2'-oxopropyl)-5-fluorouracil and 2'-deoxy-5-fluoro-3-(2'-oxoalkyl)uridine, which are activated to release 5-fluorouracil (5-FU) or 5-fluoro-2'-deoxyuridine (5-FdUrd), respectively, upon hypoxic X-irradiation.⁶

In this study, we designed an **ara-C** derivative possessing a 2-oxopropyl group at N(4) position (**oxo-ara-C**) to obtain a novel radiation-activated prodrug. We performed the radiolytic one-electron reduction of **oxo-ara-C** and evaluated its cytotoxic effect under hypoxic conditions. The 2-oxopropyl group was readily removed from **oxo-ara-C** upon hypoxic X-irradiation. Analysis of cell viability with SF assay revealed that **oxo-ara-C** released the toxic parent **ara-C** *via* one-electron reduction as induced upon X-irradiation under hypoxic conditions, thereby resulting in

cytotoxic effects.

Results and discussion

The synthesis of **oxo-ara-C** is outlined in Scheme 1. Triazole group of **1**⁹ was substituted with 1-amino-2,2-dimethoxypropane¹⁰ to give an acetal derivative **2**. Hydrolysis of **2** furnished the desired **oxo-ara-C**. A control compound **3** without a carbonyl group at the N(4) position was prepared by the coupling of **1** with propylamine.

We examined the efficiency of radiolytic one-electron reduction of **oxo-ara-C** for controlled releasing of **ara-C** in an argon-purged aqueous solution containing 2-methyl-2-propanol (10 mM) as the scavenger of oxidizing hydroxyl radicals ($\bullet\text{OH}$).^{7,11} Under these radiolysis conditions, reducing hydrated electrons (e_{aq}^-) were generated as the major active species. Fig. 1 shows a representative reaction profile analyzed by HPLC of the radiolytic one-electron reduction of **oxo-ara-C** by the resulting e_{aq}^- under hypoxic conditions. The appearance of a single new peak in Fig. 1 was attributable to the formation of the parent **ara-C**, as confirmed by the overlap injection

of authentic samples in the HPLC analysis. The G values¹² were 151 nmol J⁻¹ for the decomposition of **oxo-ara-C** and 69 nmol J⁻¹ for the formation of corresponding **ara-C**, respectively. Thus, 46% of the decomposed **oxo-ara-C** was converted to **ara-C** upon hypoxic X-irradiation. In contrast, the radiolytic decomposition of **oxo-ara-C** was dramatically suppressed under aerobic conditions; the G values were evaluated as 27 nmol J⁻¹ for the decomposition of **oxo-ara-C** and 4 nmol J⁻¹ for the formation of the corresponding **ara-C**, respectively. These results indicate that **oxo-ara-C** is activated to release **ara-C** in a hypoxia selective manner, as were observed in the cases of the 5-FU and 5-FdUrd prodrugs possessing 2-oxoalkyl groups.^{6,11}

To confirm the mechanistic validity of one-electron reduction of **oxo-ara-C** by e_{aq}^- , we conducted a control X-irradiation of **oxo-ara-C** in an aqueous solution purged with nitrous oxide (N₂O)¹³ gas, which efficiently captures reducing species of e_{aq}^- to produce oxidizing species of \bullet OH. As shown in Table 1, the formation of **ara-C** was suppressed to greater extent in the presence of N₂O relative to the argon-purged aqueous solution, while efficient decomposition of **oxo-ara-C** occurred due to the \bullet OH reaction (see also Fig. S1, S2.). Thus, it is reasonable to conclude that

reducing e_{aq}^- is essential for the activation of **oxo-ara-C** and the concomitant release of **ara-C**.

In a separate experiment, we carried out radiolytic reduction of N(4)-substituted **ara-C** without a carbonyl group **3** in aqueous solution. As shown in Table 1, hypoxic irradiation of **3** failed to release **ara-C**, indicating that the carbonyl moiety is a key structural unit for the release of **ara-C** *via* one-electron reduction of **oxo-ara-C** by e_{aq}^- .

Understanding of the function of **oxo-ara-C** in living cells is important for its biological applications. We therefore assessed the cytotoxicity of **ara-C** and **oxo-ara-C** toward A549 cells (human lung adenocarcinoma). A549 cells were cultured with various concentrations of **ara-C** or **oxo-ara-C** under aerobic conditions for 72 h, and subjected to a cell viability assay (Fig. 2). The IC_{50} values were 20.7 and 0.38 μ M for **oxo-ara-C** and **ara-C**, respectively, indicating that a small modification of **ara-C** by a substituent such as the 2-oxopropyl group can effectively reduce the cytotoxicity of the parent antitumor agent. This result may be promising for further investigations to identify whether the dose of **oxo-ara-C** could be more increased without considerable side effects, compared to **ara-C**.

We subsequently exposed A549 cells to varying doses of X-ray in the presence or absence of **oxo-ara-C** under aerobic or hypoxic conditions, and characterized the hypoxia- and radiation-dependent cytotoxic effects of **oxo-ara-C** (Fig. 3). In accord with the suppression of cytotoxic effects as described above, the cells were viable even in the presence of 3.3 μ M **oxo-ara-C** without X-irradiation (*compare* entries 1 to 2).¹⁴ Although the A549 cells were practically resistant to radiation under hypoxic conditions (entry 7), **oxo-ara-C** showed a striking property to significantly enhance the radiation sensitivity of A549 cells (entry 8). In contrast, **oxo-ara-C** had little effect on the radiation sensitivity under aerobic conditions (*compare* entries 5 to 6). These results strongly suggest that **oxo-ara-C** preferentially released toxic **ara-C** *via* radiolytic one-electron reduction in hypoxic cells and thereby resulted in enhanced cytotoxicity, consistent with the radiation chemical reactivity.

Thus, **oxo-ara-C** has been demonstrated to be useful as a radiation-activated prodrug that targets hypoxic tumor cells to show a cytotoxic effect. Incorporation of a 2-oxoalkyl group into **ara-C** dramatically reduced its cytotoxicity, while the toxic activity was recovered upon hypoxic

irradiation. Although the suppression mechanism of intrinsic cytotoxicity of **oxo-ara-C** in the presence of 2-oxoalkyl group remains unclear, it is presumable that the steric hindrance and/or disturbance of hydrogen bonding at N(4) position may inhibit the recognition of intracellular DNA polymerases, which are key enzymes involving in appearance of the cytotoxic effects of **ara-C**. In order to confirm this hypothesis, further mechanistic studies using DNA polymerase and triphosphate of **oxo-ara-C** are in progress.

Conclusion

In summary, we designed and synthesized an **ara-C** derivative possessing a 2-oxopropyl group at N(4) position (**oxo-ara-C**) as a radiation-activated prodrug. **Oxo-ara-C** was activated to release **ara-C** by e_{aq}^- as generated in radiolysis of water under hypoxic conditions. Assessment of the viability of A549 cells revealed that incorporation of 2-oxoalkyl group into **ara-C** resulted in a dramatic suppression of the cytotoxic effect, while hypoxic X-irradiation recovered the cytotoxicity *via* removal of the 2-oxoalkyl group to re-generate active **ara-C**. Thus, the **ara-C** derivatives

possessing 2-oxoalkyl groups are promising candidates as a new class of radiation-activated prodrugs for the treatment of hypoxic tumor tissues.

Experimental

General

All reactions were carried out under a dry nitrogen atmosphere using freshly distilled solvents unless otherwise noted. Reagents were purchased from Aldrich, Wako pure chemical industries and Nacalai tesque, and used as received. Ultrapure water was obtained from YAMATO WR-600A. Precoated TLC (Merck silica gel 60 F₂₅₄) plates were used for monitoring the reactions. Column chromatography was carried out on Wakogel C-300 (Wako pure chemical industries). ¹H NMR spectra were measured with JOEL JMN-AL-300 (300 MHz) or JOEL JMN-AL-400 (400 MHz) spectrophotometers at ambient temperature. ¹³C NMR spectra were measured with JOEL JMN-AL-300 (75.5 MHz) or JOEL JMN-AL-400 (100 MHz) spectrophotometers at ambient temperature. Coupling constants (*J* values) are reported in Hertz. The chemical shifts are expressed in ppm downfield from tetramethylsilane, using residual methanol ($\delta = 3.30$ in ¹H NMR, $\delta = 49.0$ in ¹³C NMR) as an internal standard. Multiplicity is designed as singlet (s), doublet (d), triplet (t), doublet-doublet (dd), or multiplet (m). FAB Mass spectra were recorded on JOEL JMS-SX102A

spectrometer, using glycerol matrix. A Rigaku RADIOFLEX-350 was used for X-irradiation.

High-performance liquid chromatography (HPLC) was performed with Shimadzu LC-6A system.

Sample solutions were injected on a reversed phase column (Inertsil ODS-3, GL Science Inc., ϕ 4.6 \times 150 nm). The 0.1 M triethylamine (Et₃N) 5 vol% acetonitrile/water solution containing acetic acid, pH 7.0 was delivered as mobile phase at a flow rate of 0.6 ml/min at 40 °C. The elution peaks were monitored at 260 nm wavelength.

4-(2,2-Dimethoxypropylamino)-1-(β -D-arabinofuranosyl)pyrimidine-2-(1H)-one (2)

1-Amino-2,2-dimethoxypropane¹⁰ (796 mg 6.68 mmol) was added to a solution of **1**⁹ (4-(1,2,4-Triazol-yl)-1-(β -D-2,3,5-tri-O-acetyl-arabinofuranosyl)pyrimidine-2(1H)-one, 167 mg, 0.40 mmol) in dioxane (1.2 ml) and stirred for 9.5 h at 80 °C. The solvent was removed under reduced pressure. The crude product was purified by column chromatography (SiO₂ 6% methanol-chloroform) to give **2** (71 mg, 52%) as a white solid; mp 171–173 °C; ¹H NMR (CD₃OD, 400 MHz) δ 7.75 (d, 1H, J = 7.6 Hz), 6.16 (d, 1H, J = 4.0 Hz), 5.93 (d, 1H, J = 7.6 Hz), 4.16 (dd, 1H, J = 2.4, 3.6 Hz), 4.04 (dd, 1H, J = 2.4, 2.4 Hz), 3.92 (m, 1H), 3.79 (m, 2H), 3.58 (s, 2H), 3.23 (s,

6H), 1.28 (s, 3H); ^{13}C NMR (CD_3OD , 75.5 MHz) δ 165.9, 158.7, 143.0, 101.8, 95.7, 88.3, 86.4, 78.2, 76.8, 62.8, 48.8, 45.3, 20.5; FABMS (glycerol) m/z 346 $[(\text{M} + \text{H})^+]$; HRMS calcd. for $\text{C}_{14}\text{H}_{24}\text{N}_3\text{O}_7$ $[(\text{M} + \text{H})^+]$ 346.1614, found 346.1611.

4-propylamino-1-(β -D-arabinofuranosyl)pyrimidine-2-(1H)-one (3)

Propylamine (0.23 ml, 2.80 mmol) was added to a solution of **1**⁹ (4-(1,2,4-Triazol-yl)-1-(β -D-2,3,5-tri-O-acetyl arabinofuranosyl)pyrimidine-2(1H)-one, 43.5 mg, 0.103 mmol) in ethanol (0.3 ml) at room temperature. After 4.5 h the product was concentrated under reduced pressure. The crude product was purified by column chromatography (SiO_2 1% Et_3N , 9% methanol-chloroform) to give **3** (21 mg, 71%) as a white solid; mp 194–195 °C; ^1H NMR (CD_3OD , 300 MHz) δ 7.72 (d, 1H, $J = 7.5$ Hz), 6.17 (d, 1H, $J = 3.9$ Hz), 5.79 (d, 1H, $J = 7.8$ Hz), 4.16 (m, 1H), 4.07 (m, 1H), 3.93 (m, 1H), 3.80 (m, 2H), 3.32 (m, 2H), 1.60 (m, 2H, $J = 7.2$ Hz), 0.96 (t, 3H, $J = 7.2$ Hz); ^{13}C NMR (CD_3OD , 100 MHz) δ 165.4, 158.6, 142.5, 95.6, 88.2, 86.3, 78.2, 76.8, 62.8, 43.4, 23.3, 11.7; FABMS (glycerol) m/z 286 $[(\text{M} + \text{H})^+]$; HRMS calcd. for $\text{C}_{12}\text{H}_{20}\text{N}_3\text{O}_5$ $[(\text{M} + \text{H})^+]$ 286.1403, found 286.1394.

4-(2-oxopropylamino)-1-(β -D-arabinofuranosyl)pyrimidine-2-(1*H*)-one (**4**) (oxo-ara-C)

2 (20 mg 0.058 mmol) was hydrolyzed by treatment with aqueous 1 M oxalic acid in THF (0.3 ml) at room temperature for 5 h. The solvent was evaporated under reduced pressure and the crude product was purified by column chromatography (SiO₂ 1% Et₃N, 12% methanol-chloroform) to give **oxo-ara-C** (7.7 mg, 44%) as a white solid; mp 180–182 °C; ¹H NMR (CD₃OD, 300 MHz) δ 7.79 (d, 1H, *J* = 7.5 Hz), 6.15 (d, 1H, *J* = 3.8 Hz), 5.93 (d, 1H, *J* = 7.5 Hz), 4.26 (s, 2H), 4.16 (dd, 1H, *J* = 2.4, 3.6 Hz), 4.04 (dd, 1H, *J* = 2.8, 2.6 Hz), 3.92 (m, 1H), 3.79 (m, 2H), 2.18 (s, 3H); ¹³C NMR (CD₃OD, 75.5 MHz) δ 206.0, 165.6, 158.5, 143.4, 95.4, 88.4, 86.6, 78.2, 76.8, 62.8, 51.3, 27.1; FABMS (glycerol) *m/z* 300 [(M + H)⁺]; HRMS calcd. for C₁₂H₁₈N₃O₆ [(M + H)⁺] 300.1196, found 300.1189.

Radiolytic reduction

Aqueous solutions of **oxo-ara-C** and **3** (100 μ M), containing 2-methyl-2-propanol (10 mM), were purged with argon or nitrous oxide for 15 min and then irradiated in a sealed glass ampoule at ambient temperature with an X-ray source (4.0 Gy min⁻¹). After the X-irradiation, the solution was

immediately subjected to HPLC analysis.

Assessment of cytotoxicity toward A549 cells

A549 cells were cultured in Dulbecco's modified Eagle's minimum essential medium (DMEM) containing 10% fetal bovine serum (FBS). The cells were seeded into 96-well plates (2000 cells/well) and cultured at 37 °C in a well-humidified incubator with 5% CO₂ and 95% air (aerobic condition) for 24 hours. The cells were then incubated with the various concentrations of **ara-C** or **oxo-ara-C** under aerobic conditions for 72 hours, and added with 11 µL of Cell Count Reagent SF¹⁵ (nacalai, Japan). The plates were further incubated at 37 °C for 2 hours and the cell viability assay was performed using Microplate Reader (BIO-RAD).

Radiation-induced cytotoxicity of oxo-ara-C

A549 cells were seeded into 96-well plates (2000 cells/well) and incubated at 37 °C for 24 hours under aerobic or hypoxic conditions. For the hypoxic treatment (< 0.02% of oxygen), the cells were treated in a hypoxic chamber, BACTRON- II (Sheldon Manufacturing Inc., Cornelius, OR, USA). The plates kept under aerobic or hypoxic conditions using Anaeron Pack System

(Mitsubishi Gas Chemical Company Inc., Japan) were X-irradiated at a dose of 4 Gy and incubated for 72 hours under aerobic conditions. After adding 12 μ L of Cell Count Reagent SF solution (nacalai, Japan) to each well, and the cell viability assay was performed as described above.

References

- 1 (a) J. S. Evans, E. A. Musser, L. Bostwick and G. D. Mengel, *Cancer Res.*, 1964, **24**, 1285–1293;
(b) K. Ooi, *Yakugaku Zasshi*, 2002, **122**, 471–480.
- 2 (a) J. J. Furth and S. S. Cohen, *Cancer Res.*, 1968, **28**, 2061–2067; (b) P. P. Major, E. M. Egan, D. J. Herrick and D. W. Kufe, *Biochem. Pharmacol.*, 1982, **31**, 2937–2940; (c) D. Kufe, D. Spriggs, E. M. Egan and D. Munroe, *Blood*, 1984, **64**, 54–58.
- 3 D. R. Macdonald, *Neurol. Clin.*, 1991, **9**, 955–967.
- 4 (a) A. L. Harris, *Nat. Rev. Cancer*, 2002, **2**, 38–47; (b) S. Kizaka-Kondo, M. Inoue, H. Harada and M. Hiraoka, *Cancer Sci.*, 2003, **94**, 1021–1028.
- 5 K. Tanabe, Z. Zhang, T. Ito, H. Hatta and S. Nishimoto, *Org. Biomol. Chem.*, 2007, **5**, 3745–3757.
- 6 (a) K. Tanabe, H. Kanezaki, H. Ishii and S. Nishimoto, *Org. Biomol. Chem.*, 2007, **5**, 1242–1246.
(b) Y. Shibamoto, L. Zhou, H. Hatta, M. Mori and S. Nishimoto, *Int. J. Radiat. Oncol. Biol. Phys.*, 2001, **49**, 407–413; (c) Y. Shibamoto, Y. Tachi, K. Tanabe, H. Hatta and S. Nishimoto, *Int. J. Radiat.*

Oncol. Biol. Phys., 2004, **58**, 397–402; (d) K. Tanabe, Y. Mimasu, A. Eto, Y. Tachi, S. Sakakibara,

M. Mori, H. Hatta and S. Nishimoto, *Bioorg. Med. Chem.*, 2003, **11**, 4551–4556

7 Radiolysis of diluted aqueous solution at around pH 7.0 produces primary water radicals such as oxidizing hydroxyl radicals ($\bullet\text{OH}$), reducing hydrated electrons (e_{aq}^-) and reducing hydrogen atoms ($\bullet\text{H}$) with the G values of $G(\bullet\text{OH}) = 280 \text{ nmol J}^{-1}$, $G(e_{\text{aq}}^-) = 280 \text{ nmol J}^{-1}$, and $G(\bullet\text{H}) = 60 \text{ nmol J}^{-1}$, respectively.

8 M. Mori, T. Ito, S. Teshima, H. Hatta, S. Fujita and S. Nishimoto, *J. Phys. Chem.*, 2006, **110**, 12198–12204.

9 K. J. Divakar and C. B. Reese, *J. Chem. Soc. Perkin Trans.*, 1982, **5**, 1171–1176.

10 R. Calabretta, C. Giordano, C. Gallina, V. Morea, V. Consalvi and R. Scandurra, *Eur. J. Med. Chem.*, 1995, **30**, 931–941.

11 M. Mori, H. Hatta and S. Nishimoto, *J. Org. Chem.*, 2000, **65**, 4641–4647.

12 The number of molecules produced or changed per 1 J of radiation energy absorbed by the reaction system.

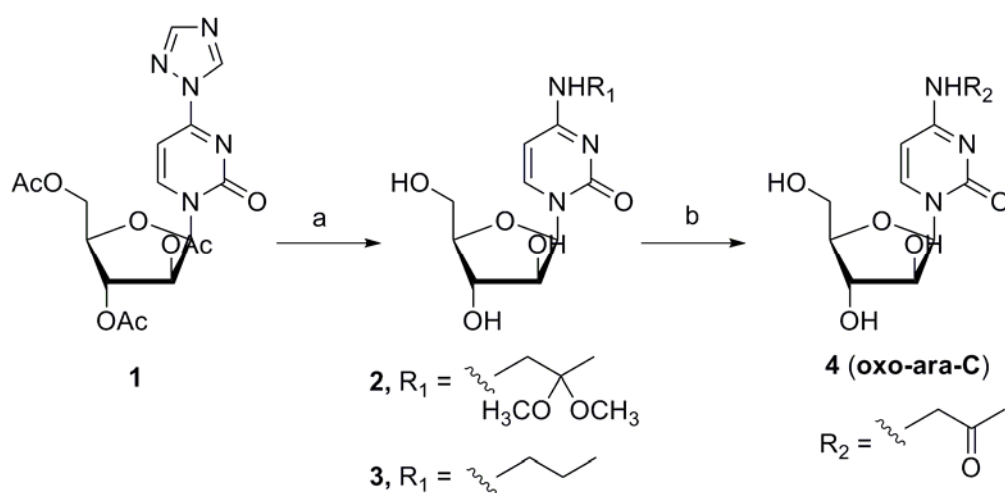
13 Dissolved N₂O exclusively scavenges reducing hydrated electrons to produce hydroxyl radicals,

as follows: $e_{aq}^- + N_2O \rightarrow \bullet OH + OH^- + N_2$ ($k = 9.1 \times 10^9 \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1}$)

14 Without X-irradiation, the effect of oxygen on the cell viability is negligible independent of the presence or the absence of **oxo-ara-C** (Fig. 3, entry 1–4).

15 M. Ishiyama, Y. Miyazono, K. Sasamoto, Y. Ohkura and K. Ueno, *Talanta*, 1997, **44**, 1299–1305.

FIGURES



Scheme 1. *Reagents and conditions:* (a) 1-amino-2,2-dimethoxypropane, dioxane, 80 °C, 9.5 h (for **2**, 52%); propylamine, ethanol, room temperature, 4.5 h (for **3**, 71%); (b) **2**, oxalic acid, THF, room temperature, 5 h, (44%)

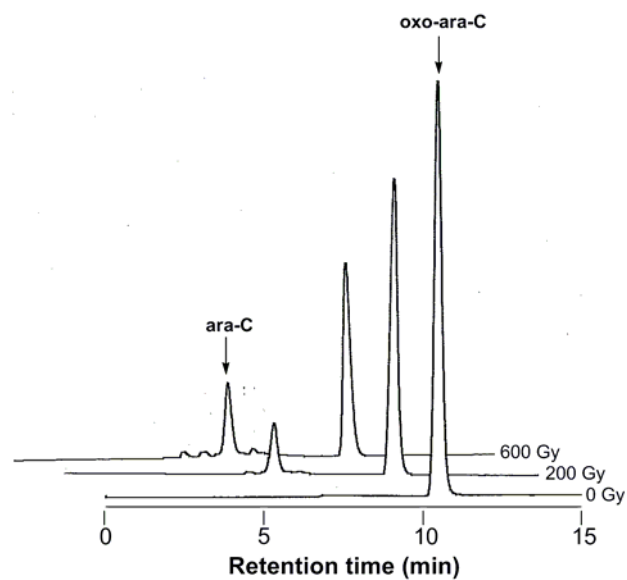


Fig. 1 HPLC profiles for the one-electron reduction of **oxo-ara-C** (100 μ M) in the hypoxic X-radiolysis (0, 200, and 600 Gy) of aqueous solution containing 2-methyl-2-propanol (10 mM).

Table 1. *G* values for the decomposition of **oxo-ara-C** and for the formation of **ara-C** in the X-radiolysis in several gas-saturated solutions.

Compound	(saturating gas)	<i>G</i> value / nmol J ⁻¹	
		Decomposition	Formation
oxo-ara-C	(argon)	151	69
oxo-ara-C	(air)	27	4
oxo-ara-C	(nitrous oxide)	141	5
3	(argon)	130	ND ^a

^a Not detected

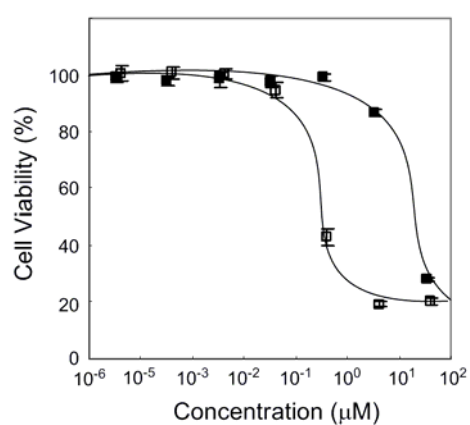


Fig. 2 Cytotoxicity of **ara-C** and **oxo-ara-C** against A549 tumor cells. A549 cells were incubated with indicated concentrations of **ara-C** or **oxo-ara-C** under aerobic conditions for 72 hours. To calculate the cell viability in each conditions, SF counts (OD₄₅₀) in each drug concentration were compared to those in minimal drug concentrations. **ara-C**: open square, **oxo-ara-C**: closed square.

Results are shown with the mean \pm S.D. n = 3.

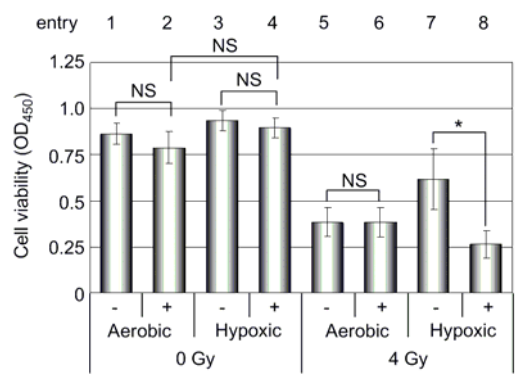


Fig. 3 Radiation-induced cytotoxicity of **oxo-ara-C** against A549 cells under hypoxic conditions.

A549 cells were cultured in the presence (+) or absence (-) of 3.3 μ M **oxo-ara-C**, and X-irradiated (4 Gy) under aerobic or hypoxic conditions. Results are shown with the mean \pm S.D. n = 3 (* P < 0.05. NS: not significant).

Electronic Supplementary Information

Radiolytic activation of cytarabine prodrug possessing 2-oxoalkyl group: one-electron reduction and cytotoxicity characteristics

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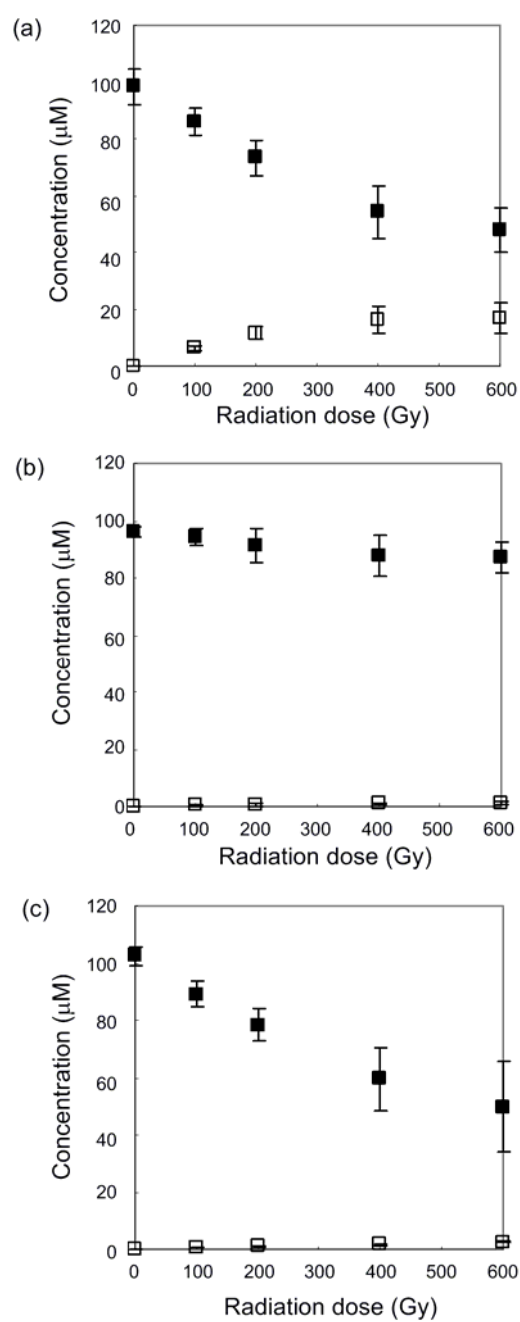


Fig. S1. Radiolytic reduction of 100 μM **oxo-ara-C** (closed square) to release **ara-C** (open square) in hypoxic (a), aerobic (b) and N_2O -saturated (c) aqueous solutions containing 2-methyl-2-propanol (10 mM)

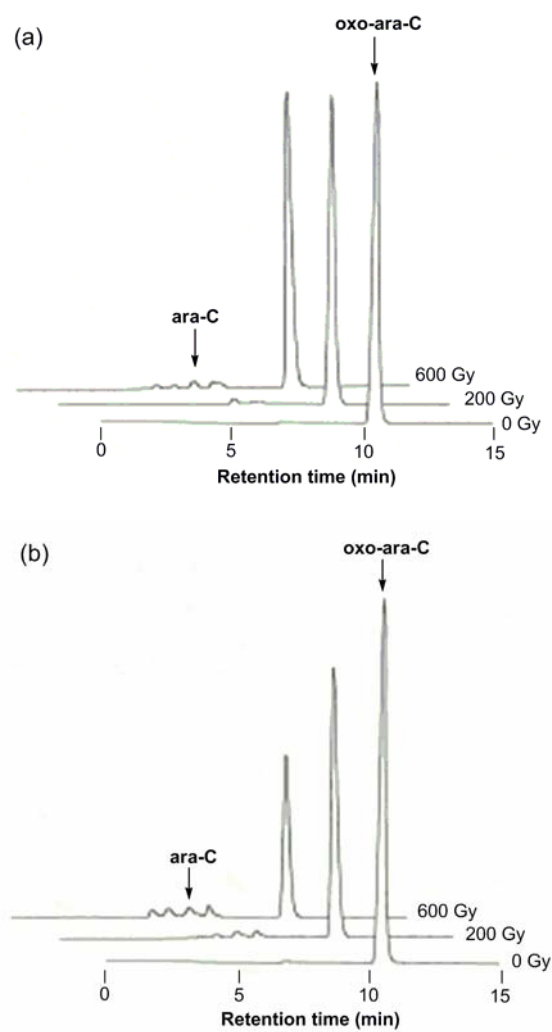


Fig. S2. HPLC profiles for the X-radiolysis (0, 200, and 600 Gy) of **oxo-ara-C** (100 μ M) in aerobic (a) and N_2O -saturated (b) aqueous solutions containing 2-methyl-2-propanol (10 mM)