

Charge traps in lead-halide perovskites with different grain sizes

Hyung Do Kim¹ and Hideo Ohkita^{1*}

¹*Department of Polymer Chemistry, Graduate School of Engineering, Kyoto University, Katsura, Kyoto 615-8510, Japan*

*E-mail: ohkita@photo.polym.kyoto-u.ac.jp

It is crucially required to understand charge traps in $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskites, which severely affect charge recombination mechanism and hence have a critical impact on photovoltaic performance. Herein, we studied the photoexcited dynamics of $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskites with different grain sizes by measuring intensity dependence of time-resolved photoluminescence. As a result, we found that the photoluminescence (PL) decay for all the samples can be well analyzed by bimolecular radiative recombination and trap-assisted Shockley-Read-Hall recombination model. We discuss how the PL dynamics impact on the photovoltaic performance of $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskite solar cells on the basis of the recombination analysis.

1. Introduction

Solution processed organic–inorganic perovskites composed of ABX_3 compositions, such as $CH_3NH_3PbI_3$ and $CH_3NH_3PbI_{3-x}Cl_x$ have emerged as a promising material in application of photovoltaic devices because of their intriguing optoelectronic properties, including direct bandgap with excellent absorption coefficient, high charge carrier mobility, high long diffusion length, and Wannier–Mott exciton nature.^{1–7)} Consequently, organic–inorganic perovskite solar cells based on metal halides have been considered as a suitable candidate for alternative energy sources because they potentially provide flexible and light-weight devices with large-scale and low-cost production.^{5,8)} Owing to many research efforts, the power conversion efficiency (PCE) of perovskite solar cells has been rapidly improved in the past few years and recently exceeded 22%.^{8,9)}

The photovoltaic performance of inorganic solar cells is generally dependent upon grain sizes of the active layer.^{10–12)} We have recently studied how grain sizes affect the photovoltaic performance of planar heterojunction $CH_3NH_3PbI_3$ devices fabricated by the fast deposition–crystallization (FDC) procedure.^{13,14)} We found that all the photovoltaic parameters increase as the grain size of perovskites is increased. Consequently, the highest PCE of 19.4% was observed for the device with the largest grain size of ≈ 500 nm.¹³⁾ The enhancement in short-circuit current density (J_{SC}) is mainly ascribed to the increase in absorption efficiency of perovskites at longer wavelengths with increasing thickness of perovskite layers. As with the improvement in J_{SC} , the open-circuit voltage (V_{OC}) is also increased with increasing grain size of perovskites. By analyzing the intensity dependence of V_{OC} on the basis of bimolecular radiative recombination and trap-assisted Shockley–Read–Hall (SRH) recombination, we found that the smaller V_{OC} observed for the perovskite device with the smaller grain size is due to the higher trap density N_t . In other words, this finding suggests that the photovoltaic performance is limited by charge traps in $CH_3NH_3PbI_3$ perovskites. Such charge traps have been reported to affect photoluminescence (PL) dynamics.^{15–19)}

In this work, we study the PL dynamics in $CH_3NH_3PbI_3$ perovskites with different grain sizes by measuring the intensity dependence of time-resolved PL (TRPL) spectroscopy. By analyzing the PL decay on the basis of the bimolecular radiative recombination and the trap-assisted SRH recombination model, we have shown that trap density (N_t) is decreased with

increasing grain size of perovskites. Such trend is in good agreement with our previous report analyzed from the intensity-dependent V_{OC} .¹³⁾ We further discuss the primary loss mechanism on the basis of quantitative analyses.

2. Experimental methods

2.1 Materials and Sample Preparation

A methylammonium iodide ($\text{CH}_3\text{NH}_3\text{I}$) was prepared by dropwise addition of a methanol solution of methylamine (CH_3NH_2 , 90 mL, 40%, 0.882 M; Wako Pure Chemical Industries, Ltd.) over 10 min to an aqueous solution of hydrogen iodide (HI, 96.9 mL, 57 wt%, 1.29 M; Wako Pure Chemical Industries Ltd.) in a 500-mL round bottom flask at 0 °C. The precipitates were recovered by evaporation at 50 °C for 30 min after stirring for 2 h. The resultant yellowish raw products were dissolved in ethanol, recrystallized from diethyl ether, and then finally filtered. These steps were repeated three times. After the filtration, the white solid products $\text{CH}_3\text{NH}_3\text{I}$ were dried at 60 °C in a vacuum oven for 24 h. Perovskite stock solutions ($\text{CH}_3\text{NH}_3\text{PbI}_3$) were prepared by mixing $\text{CH}_3\text{NH}_3\text{I}$ with purified PbI_2 (Tokyo Chemical Industry with Co., Ltd., Japan) at a molar ratio of 1 to 1 in anhydrous *N,N*-dimethylformamide (DMF, 99.8%, Sigma–Aldrich) with different concentration of 25 wt% (316 mg mL^{-1}), 45 wt% (778 mg mL^{-1}), and 55 wt% (1160 mg mL^{-1}) and then stirred at 70 °C overnight in a nitrogen-filled glove box (H_2O and $\text{O}_2 < 1 \text{ ppm}$).

The glass substrates were washed with toluene, then acetone, and finally ethanol in an ultrasonic bath for 15 min each and then dried under nitrogen flow. The washed substrates were further treated with a UV–ozone cleaner (Nippon Laser & Electronics Lab., NL-UV253S) for 30 min. Perovskite films were prepared atop the UV–ozone-cleaned substrate by spin-coating at 5000 rpm for 30 s, and after 6 s, anhydrous chlorobenzene (99.8%, 0.3 mL, Sigma–Aldrich) was quickly dropped onto the center of the substrate. The instant color change of films from yellow to brown was observed upon dropping chlorobenzene solvent. The resulting dark brown films were dried at 100 °C for 10 min. The grain size of resultant films was simply controlled by the different concentration of perovskite stock solution.

2.2 Measurements

The surface morphology and thickness of perovskite films were measured with an atomic force microscope (AFM) (Shimadzu, SPM-9600) with a silicon probe in contact mode and a scanning electron microscope (SEM) (Keyence, VE-9800). Time-resolved photoluminescence (TRPL) were measured by a time-correlated single photon-counting (TCSPC) fluorescence lifetime system (Horiba Jobin Yvon, FluoroCube). The samples were excited by a pulse laser with a photon energy of 1.84 eV and a pulse width of <1 ns. The excitation intensity was reduced to $\sim \text{nJ cm}^{-2}$ by using the neutral density (ND) filters to avoid electron-hole annihilation processes. A pulse picker was used to reduce the pulse repetition rate to 1 MHz, allowing measurements of lifetimes in the hundreds of nanosecond range. The samples were excited at 1.84 eV and the emission was collected at 1.60 eV after passing a long-pass filter to prevent an incoming excitation laser reflected. The external quantum efficiency (EQE) spectra were measured with a spectral response measurement system (Bunkoukeiki, ECT-250D). The power of the incident monochromatic light was kept under 0.05 mW cm^{-2} , which was measured with a calibrated silicon reference cell (Bunkoukeiki, BS-520BK).

3. Results and discussion

3.1 Film Morphology

Figure 1 shows the SEM and AFM images of $\text{CH}_3\text{NH}_3\text{PbI}_3$ films prepared by the FDC method. As shown in the figure, $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskite layers are flat and pin-hole free, which are homogeneously covered with relatively uniform grain sizes. In addition, the grain size of perovskites increased as the concentration of perovskite stock solution increased. On the other hand, the higher concentration of stock solution was employed, the wider distribution of grain size was observed. The average grain size of $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskite was estimated to be $\sim 100 \text{ nm}$ for the 25 wt% stock solution, $\sim 300 \text{ nm}$ for the 45 wt% stock solution, and $\sim 500 \text{ nm}$ for the 55 wt% stock solution. More interestingly, as shown in Figure 1d, the average grain size is almost the same as the thickness of $\text{CH}_3\text{NH}_3\text{PbI}_3$ layers: $\sim 100 \text{ nm}$ for the 25 wt% stock solution, $\sim 300 \text{ nm}$ for the 45 wt% stock solution, and $\sim 500 \text{ nm}$ for the 55 wt% stock solution. This suggests that $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskite layers prepared by the FDC method are composed of monograins in the direction normal to the substrate. These results are in good agreement with our previous report.¹³⁾

3.2 Absorption and PL Spectra

Figure 2a shows the absorption and steady state PL spectra of the $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskite film with a grain size of ≈ 500 nm. As shown in the figure, the onset of absorption spectra is equal to the maximum peak position of the PL spectra, which is indicative of the direct bandgap semiconductors.^{1,2)} As shown in Figure 2b, a direct bandgap was evaluated from the x -axis intercept of $(E \times \text{EQE})^2$ plotted against the photon energy E .^{20–23)} The bandgap energy increased from 1.57 to 1.59 eV with decreasing grain size of perovskites. These results are consistent with previous reports.^{23–26)}

3.3 PL Dynamics

Figure 3 shows the intensity dependence of the TRPL dynamics of $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskite films with different grain sizes at room temperature. As shown in the figure, no single-exponential decay was observed for all the samples, suggesting that there are several emissive species with different lifetime. However, this is not the case because no spectral change was observed for the TRPL over the time range measured, as shown in Figure 4. This indicates that the radiative PL decay is attributed to one emissive species. In $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskites, photoexcited states can be easily dissociated into free electrons and holes because the exciton binding energy is less than the thermal energy at room temperature.^{4–7)} Thus, such radiative PL is ascribed to the bimolecular recombination of free electrons and holes as will be discussed below.

3.4 PL Decay Analyses

We here analyze the PL decay dynamics in more details to address the origin of the non-exponential kinetics. The PL decays were well fitted by a double-exponential decay function. As shown in Figure 5, the lifetime of the slow decay fraction decreased but the lifetime of the fast decay fraction increased with increasing excitation intensity, indicating that the PL decay is due to the bimolecular recombination. As discussed previously, we analyze this PL decay dynamics on the basis of the bimolecular recombination and the trap-assisted SRH recombination in $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskites. The slow fraction is ascribed to the radiative recombination lifetime (τ_{rad}) and the fast fraction is to the trap-assisted SRH

recombination lifetime (τ_{SRH}). The radiative recombination lifetime is inversely proportional to the density of free charge carriers, which is given by Equation (1)

$$\tau_{\text{rad}} = \frac{1}{B_{\text{rad}}(\Delta n + n_0 + p_0)} \quad (1)$$

where B_{rad} is the radiative recombination coefficient, Δn is the density of electrons photogenerated under the illumination, and n_0 and p_0 are the density of electrons and that of holes generated by unintentional doping, which are given by Equations (2) and (3), respectively.

$$n_0 = N_C \exp\left(\frac{E_f - E_C}{k_B T}\right) \quad (2)$$

$$p_0 = N_V \exp\left(\frac{E_V - E_f}{k_B T}\right) \quad (3)$$

Here, N_C and N_V are the effective density of states in conduction and valence bands, which are given by $N_C = 2(2\pi m_n^* k_B T / h^2)^{3/2}$ and $N_V = 2(2\pi m_p^* k_B T / h^2)^{3/2}$ where m_n^* and m_p^* are the effective mass for electrons and holes, respectively, k_B is the Boltzmann constant, T is the absolute temperature, and h is the Planck constant. On the other hand, the SRH recombination lifetime is given by Equation (4)

$$\tau_{\text{SRH}} = \frac{\tau_p(n + n_1) + \tau_n(p + p_1)}{n_0 + p_0 + \Delta n} \quad (4)$$

where n and p are given by $n = \Delta n + n_0$ and $p = \Delta p + p_0$ where Δp is the density of holes photogenerated under the illumination, n_1 and p_1 are the density of electrons and that of holes when the Fermi levels of electron and hole coincide with the trap level, which are expressed by Equation (5) and (6), respectively.

$$n_1 = N_C \exp\left(\frac{E_t - E_C}{k_B T}\right) \quad (5)$$

$$p_1 = N_V \exp\left(\frac{E_V - E_t}{k_B T}\right) \quad (6)$$

Here, τ_n and τ_p are the lifetime of electrons and holes related to the trapped sites, which are given by Equation (7) and (8), respectively,

$$\tau_n = \frac{1}{\sigma_n v_{\text{th}}^n N_t} \quad (7)$$

$$\tau_p = \frac{1}{\sigma_p v_{th}^p N_t} \quad (8)$$

where σ_n and σ_p are the capture cross sections with trap for electrons and holes, respectively, N_t is the density of trap states with an energy E_t in the bandgap, and V_{th}^n and V_{th}^p are the thermal velocity for electrons and holes, which are defined as $V_{th}^n = (3k_B T / m_n^*)^{1/2}$ and $V_{th}^p = (3k_B T / m_p^*)^{1/2}$, respectively. We estimated B_{rad} and N_t on the basis of Equation (1) and (4) by using the parameters taken from the literature, respectively: $\sigma_n = 10^{-15} \text{ cm}^2$ and $m_n^* = 0.23m_0$ where m_0 is the mass of stationary electron.^{27,28)} We assumed that the energy difference between the conduction band and the Fermi level is $E_C - E_f = 0.30 \text{ eV}$ because $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskites are considered to be n-type materials caused by an unintentional doping as reported previously.^{13,29)} As a result, as shown in Figure 5a, the intensity dependence of the slow fraction was well fitted by this model with a parameter of $B_{rad} = 8.0 \times 10^{-9} \text{ cm}^3 \text{ s}^{-1}$ regardless of the grain sizes, which is comparable to that of the previous report.^{13,30)} On the other hand, as shown in Figure 5b, the intensity dependence of the fast fraction was well fitted with $E_C - E_f = 0.30 \text{ eV}$ and $E_C - E_t = 0.38 \text{ eV}$. As summarized in Table 1, N_t was evaluated to be $4.58 \times 10^{16} \text{ cm}^{-3}$ for $\approx 100 \text{ nm}$, $2.33 \times 10^{16} \text{ cm}^{-3}$ for $\approx 300 \text{ nm}$, and $8.77 \times 10^{15} \text{ cm}^{-3}$ for $\approx 500 \text{ nm}$ grain size of $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskites, which are in good agreement with that obtained from the intensity-dependent V_{OC} analysis as reported previously.¹³⁾ This agreement would ensure the reliability of these parameters estimated in this study.

3.4 Primary Loss Mechanism

Finally, we discuss the primary loss mechanism of photoexcited states in $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskites. As shown in Table 1, B_{rad} was independent of the grain size of perovskites while N_t increased from 8.77×10^{15} to $4.58 \times 10^{16} \text{ cm}^{-3}$ with decreasing grain size of perovskites from ≈ 500 to $\approx 100 \text{ nm}$. This indicates that the primary loss mechanism in the excited state of $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskites is the trap-assisted SRH recombination, which is a nonradiative loss rather than the bimolecular radiative recombination. This result is consistent with our previous report of the primary charge recombination mechanism for $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskites solar cells analyzed by intensity dependence of V_{OC} . We further discuss how such charge traps have impact on the charge collection under the short-circuit condition in $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskite solar cells. The charge collection time is given by τ_{CC}

$= L^2/2\mu V_{OC}$ where L is the thickness of the active layer and μ is the charge carrier mobility. Recent studies have shown that μ in $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskites is as high as $1\text{--}100\text{ cm}^2\text{ V}^{-1}\text{ s}^{-1}$.^{3,5)} Here, we thus assumed that μ is $1\text{ cm}^2\text{ V}^{-1}\text{ s}^{-1}$ as a lower limit. As shown in Table 2, τ_{CC} was as short as $\leq 10^{-9}\text{ s}$, which is consistent with the charge injection time reported previously by considering the difference in the thickness.³¹⁾ This collection time was much shorter than the shorter PL lifetime τ_{fast} even under the lowest excitation intensity of 0.6 nJ cm^{-2} , indicating that the charge collection efficiency is as high as $\sim 100\%$. In other words, the shorter PL lifetime would have negligible impact on the photocurrent generation or collection in $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskite solar cells. In summary, these findings indicate that charge traps in perovskites cause the V_{OC} loss but have no impact on the J_{SC} in the devices.

4. Conclusions

We fabricated flat and dense $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskites with different grain sizes ranging from ≈ 100 to $\approx 500\text{ nm}$ by the FDC method using different concentration of $\text{CH}_3\text{NH}_3\text{PbI}_3$ stock solution. The photoexcited dynamics were studied for the corresponding samples by the TRPL measurements. As a result, we found that the PL decay can be well analyzed for all the samples by a double-exponential decay function. The two lifetimes were dependent on the excitation intensity, suggesting that bimolecular processes are involved. We also found that the TRPL spectra remain the same over the time range measured. We thus analyzed the PL decay dynamics on the basis of the bimolecular radiative recombination and the trap-assisted SRH recombination. As a result, we found that the B_{rad} is independent of the grain size of $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskites while the N_t increases with decreasing their grain size. This suggests that the photoexcited species in $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskites are primarily deactivated by the trap-assisted SRH recombination, which results in the V_{OC} loss as reported previously.¹³⁾ On the other hand, the charge traps have negligible impact on the charge collection or J_{SC} in the devices. We therefore conclude that of particular importance is to fabricate pure perovskites with low N_t for further enhancement in the photovoltaic performance of $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskite solar cells.

Acknowledgment

This work was supported by the JST Advanced Low Carbon Technology Research and

Development (ALCA) program (Solar Cell and Solar Energy Systems).

References

- 1) S. D. Wolf, J. Holovsky, S.-J. Moon, P. Löper, B. Niesen, M. Ledinsky, F.-J. Haug, J.-H. Yum, and C. Ballif, *J. Phys. Chem. Lett.* **5**, 1035 (2014).
- 2) C. Wehrenfennig, M. Liu, H. J. Snaith, M. B. Johnston, and L. M. Herz, *J. Phys. Chem. Lett.* **5**, 1300 (2014).
- 3) J. Ma and L.-W. Wang, *Nano Lett.* **17**, 3646 (2017).
- 4) Q. Lin, A. Armin, R. C. R. Nagiri, P. L. Burn, and P. Meredith, *Nat. Photonics* **9**, 106 (2015).
- 5) M. Ahmadi, T. Wu, and B. Hu, *Adv. Mater.* **29**, 1605242 (2017).
- 6) M. E. Ziffer, J. C. Mohammed, and D. S. Ginger, *ACS Photonics* **3**, 1060 (2016).
- 7) L. Q. Phuong, Y. Nakaike, A. Wakamiya, and Y. Kanemitsu, *J. Phys. Chem. Lett.* **7**, 4905 (2016).
- 8) N.-G. Park, M. Grätzel, T. Miyasaka, K. Zhu, and K. Emery, *Nat. Energy* **1**, 16152 (2016).
- 9) W. S. Yang, B.-W. Park, E. H. Jung, N. J. Jeon, Y. C. Kim, D. U. Lee, S. S. Shin, J. Seo, E. K. Kim, J. H. Noh, and S. I. Seok, *Science* **356**, 1376 (2017).
- 10) M. Imaizumi, T. Ito, M. Yamaguchi, and K. Kaneko, *J. Appl. Phys.* **81**, 7635 (1997).
- 11) B. Zebentout, Z. Benamara, and T. Mohammed-Brahim, *Thin Solid Films* **516**, 84 (2007).
- 12) M. K. Sharma and D. P. Joshi, *Indian J. Pure Appl. Phys.* **48**, 575 (2010).
- 13) H. D. Kim, H. Ohkita, H. Benten, and S. Ito, *Adv. Mater.* **28**, 917 (2016).
- 14) M. Xiao, F. Huang, W. Huang, Y. Dkhissi, Y. Zhu, J. Etheridge, A. Gray-Weale, U. Bach, Y.-B. Cheng, and L. Spiccia, *Angew. Chem.* **126**, 10056 (2014).
- 15) M. Maiberg and R. Scheer, *J. Appl. Phys.* **116**, 123710 (2014).
- 16) M. Maiberg, T. Hölscher, S. Zahedi-Azad, and R. Scheer, *J. Appl. Phys.* **118**, 105701 (2015).
- 17) A. Redinger, S. Levchenko, C. J. Hages, D. Greiner, C. A. Kaufmann, and T. Unold, *Appl. Phys. Lett.* **110**, 122104 (2017).
- 18) C. R. Haughn, K. J. Schmieder, J. M. O. Zide, A. Barnett, C. Ebert, R. Opila, and M. F. Doty, *Appl. Phys. Lett.* **102**, 182108 (2013).
- 19) S. D. Stranks, V. M. Burlakov, T. Leijtens, J. M. Ball, A. Goriely, and H. J. Snaith, *Phys. Rev. Appl.* **2**, 034007 (2014).
- 20) W. L. Leong, Z.-E. Ooi, D. Sabba, C. Yi, S. M. Zakeeruddin, M. Graetzel, J. M. Gordon, E. A. Katz, and N. Mathews, *Adv. Mater.* **28**, 2439 (2016).
- 21) T. K. Todorov, O. Gunawan, T. Gokmen, and D. B. Mitzi, *Prog. Photovoltaics: Res. Appl.*

- 21**, 82 (2013).
- 22) H. D. Kim, Y. Miyamoto, H. Kubota, T. Yamanari, and H. Ohkita, Chem. Lett. **46**, 253 (2017).
- 23) H. D. Kim, N. Yanagawa, A. Shimazaki, M. Endo, A. Wakamiya, H. Ohkita, H. Benten, and S. Ito, ACS Appl. Mater. Interfaces **9**, 19988 (2017).
- 24) A. R. bin M. Yusoff and M. K. Nazeeruddin, J. Phys. Chem. Lett. **7**, 851 (2016).
- 25) H. D. Kim and H. Ohkita, Solar RRL **1**, 1700027 (2017).
- 26) Y. Yamada, T. Nakamura, M. Endo, A. Wakamiya, and Y. Kanemitsu, Appl. Phys. Express **7**, 032302 (2014).
- 27) M. L. Agiorgousis, Y.-Y. Sun, H. Zeng, and S. Zhang, J. Am. Chem. Soc. **136**, 14570 (2014).
- 28) W.-J. Yin, J.-H. Yang, J. Kang, Y. Yan, and S.-H. Wei, J. Mater. Chem. A **3**, 8926 (2015).
- 29) Q. Wang, Y. Shao, H. Xie, L. Lyu, X. Liu, Y. Gao, and J. Huang, Appl. Phys. Lett. **105**, 163508 (2014).
- 30) A. R.S. Kandada and A. Petrozza, APL Mater. **4**, 091506 (2016).
- 31) S. Makuta, M. Liu, M. Endo, H. Nishimura, A. Wakamiya, and Y. Tachibana, Chem. Commun. **52**, 673 (2016).

Figure Captions

Fig. 1. (Black and white) SEM images of $\text{CH}_3\text{NH}_3\text{PbI}_3$ films fabricated by the FDC method using different concentrations of stock solution: a) 25 wt%, b) 45 wt%, and c) 55 wt%. The inset in each figure is AFM images of the same film. The scale bars correspond to 250 nm in length.

Fig. 2. (Black and white) a) Absorption (solid line) and PL (broken line) spectra of $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskite films with a grain size of ≈ 500 nm. b) $(E \times \text{EQE})^2$ plotted against a photon energy E for $\text{CH}_3\text{NH}_3\text{PbI}_3$ solar cells with different grain sizes: ≈ 100 nm (open circles), ≈ 300 nm (open triangles), and ≈ 500 nm (open squares). The broken lines show the x-intercept.

Fig. 3. (Black and white) Intensity dependence of TRPL dynamics for $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskite films with different grain sizes: a) ≈ 100 nm, b) ≈ 300 nm, and c) ≈ 500 nm. The samples were excited at 1.84 eV and the PL dynamics were monitored at 1.60 eV under different excitation intensities: 2.8 nJ cm^{-2} (close diamonds), 2.1 nJ cm^{-2} (close inverted triangles), 1.4 nJ cm^{-2} (close squares), 0.9 nJ cm^{-2} (close triangles), and 0.6 nJ cm^{-2} (close circles). The white broken lines represent the fits to the experimental data with a double-exponential decay function.

Fig. 4. (Black and white) TRPL spectra for $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskite films with different grain sizes: a) ≈ 100 nm, b) ≈ 300 nm, and c) ≈ 500 nm. The samples were excited at 1.84 eV and the PL dynamics were monitored at 1.60 eV under different delay times: 0 ns (close circles), 1 ns (close triangles), 10 ns (close squares), 50 ns (close inverted triangles), 100 ns (close diamonds), and 700 ns (open circles). The inset in each figure shows the normalized TRPL spectra at the different delay times.

Fig. 5. (Black and white) a) The longer lifetime, b) the shorter lifetime, and c) the average lifetime as a function of an excess carrier density for $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskite films with different grain sizes: ≈ 100 nm (open circles), ≈ 300 nm (open triangles), and ≈ 500 nm (open squares). The broken lines and the dashed–dotted lines were extracted by a fit to experimental data with Equation (1) and Equation (4), respectively. The open plots and error bars in the y-axis represent the average and the standard deviation of several values measured.

Table I. Recombination kinetic parameters for $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskites.

Grain size / nm	$B_{\text{rad}} / \text{cm}^3 \text{s}^{-1}$		$N_{\text{t}} / \text{cm}^{-3}$	
	Device ^{a)}	TRPL	Device ^{a)}	TRPL
≈ 100	3.0×10^{-9}	8.0×10^{-9}	5.30×10^{16}	4.58×10^{16}
≈ 300	3.0×10^{-9}	8.0×10^{-9}	1.05×10^{16}	2.55×10^{16}
≈ 500	3.0×10^{-9}	8.0×10^{-9}	2.84×10^{15}	8.77×10^{15}

^{a)} B_{rad} and N_{t} are taken from our previous literature.¹³⁾

Table II. Comparison between τ_{fast} , and τ_{CC} in $\text{CH}_3\text{NH}_3\text{PbI}_3$ perovskites.

Grain size (L) / nm	$V_{\text{OC}} / \text{V}^{\text{a)}}$	$\tau_{\text{CC}} / \text{ns}$	$\tau_{\text{fast}} / \text{ns}^{\text{b)}}$
≈ 100	1.00	5.0×10^{-11}	1.6×10^{-9}
≈ 300	1.04	4.3×10^{-10}	2.4×10^{-9}
≈ 500	1.08	1.2×10^{-9}	7.1×10^{-9}

^{a)} V_{OC} are taken from our previous literature.¹³⁾

^{b)} The values represent the shorter lifetime under an excitation intensity of 0.6 nJ cm^{-2} .

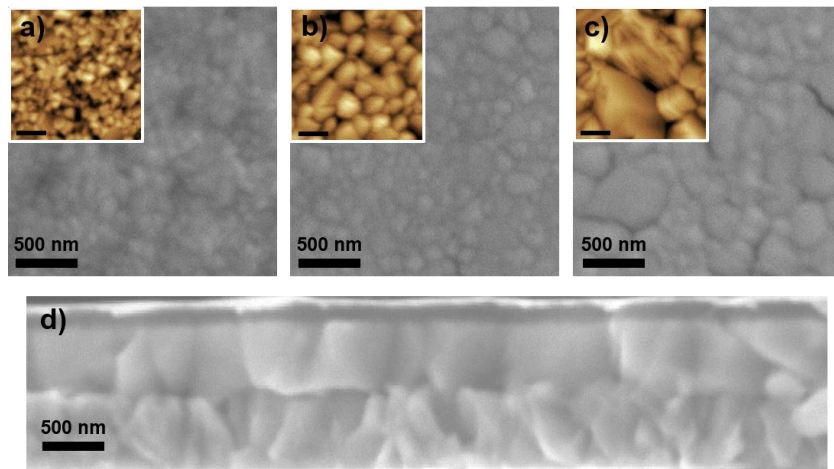


Fig.1. (Black and white)

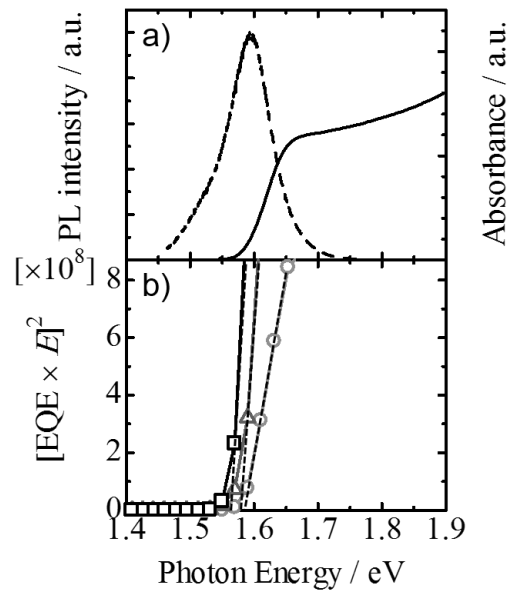


Fig. 2. (Black and white)

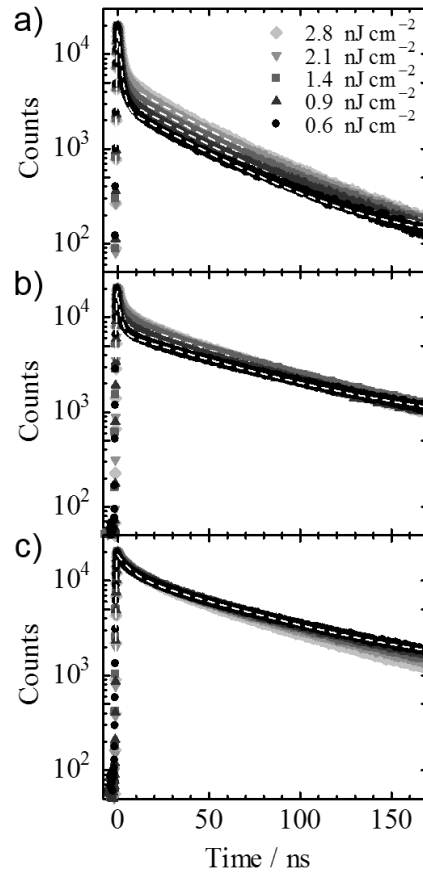


Fig. 3. (Black and white)

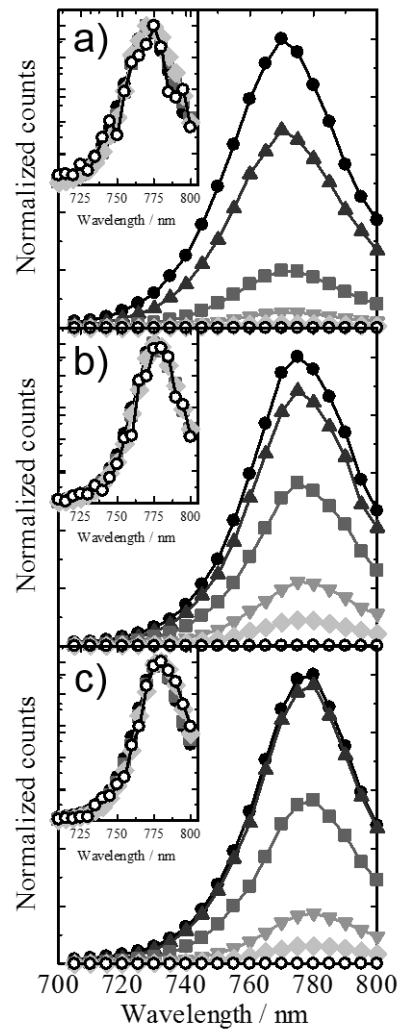


Fig. 4. (Black and white)

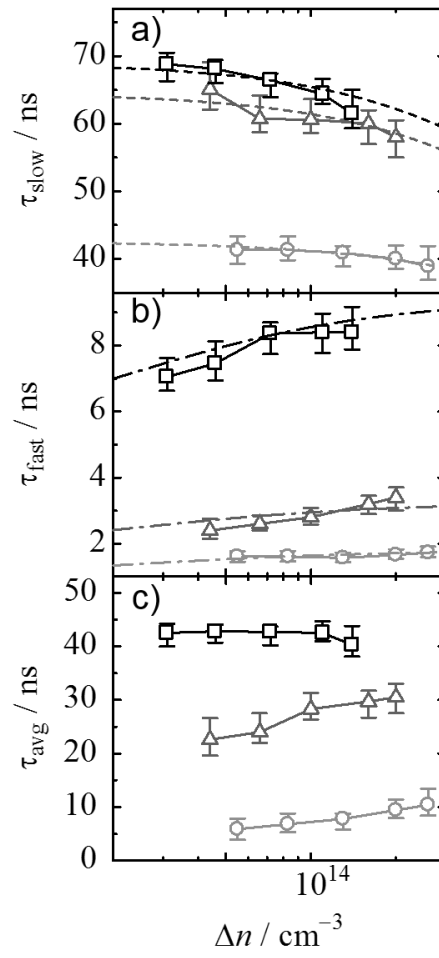


Fig. 5. (Black and white)