

**Molecular mechanisms of mutations induced by
environmental pollutants, 3-nitrobenzanthrone,
crotonaldehyde and acrolein**

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PREFACE

The study presented in this thesis has been carried out under the directions of Professor Saburo Matsui at Research Center for Environmental Quality Control, Kyoto University and Professor Hiraku Takebe* at Department of Radiation Genetics, Faculty of Medicine, Kyoto University during 1996 - 1998. Throughout the course of this research, the author has had the continuous good fortune to be in contact with people who provided useful suggestions, encouragement and aids of many kinds. The author would like to express his gratitude to all of them for their many contributions to the success of this work.

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GENERAL INTRODUCTION

Natural and synthetically-produced chemicals represent a major paradox in our modern world. Millions of different chemical substances are used daily around the world, and have proven to be very beneficial to mankind in many years. Unfortunately, there are also undesirable consequences associated with the use of these chemicals, often with hazardous or toxic effects on humans. In recent years, genetic toxicity of some of these chemicals has been revealed by the methods of molecular biology.

It has been believed that deoxyribonucleic acid (DNA) is the informationally active chemical component of essentially all genetic materials and it was assumed that this macromolecule must be extraordinarily stable maintaining high degree of fidelity required for the master blueprint. It has been revealed, however, that the primary structure of DNA is in fact quite dynamic and subject to constant change. Many of these changes arise as a consequence of errors during replication, recombination and repair of DNA. Base alterations may also arise from the inherent instability of specific chemical bonds that constitute the normal nucleotides under physiological temperature and pH conditions. DNA of living cells often reacts with various chemical compounds and physical agents (*e.g.*, X-rays), many of which are present in the environment. Some of these chemicals are products of the metabolism or decomposition of other living forms with which many organisms exist in intimate proximity. Others, particularly in recent decades, are man-made and contribute to genetic insult faced by individuals living in highly industrialized communities (1).

An agent that leads to an increase in the frequency of mutation, 'mutagen', either acts on DNA directly to change its template properties or in some way subvert replication so that a wrong base is inserted. DNA reacting chemicals act directly on DNA to change the bases into chemically distinct structures. These new bases often base pair in a different way from the original bases, in which case the effect of mutagen is to alter the genetic code directly (2). In recent years, an increasing public awareness of environmental mutagens has led to an intense interest in studying the mechanisms by which genotoxic chemicals interact with and damage DNA. One of the goals in this thesis is to illustrate the interaction of environmental chemicals with DNA.

Mutagen may produce mutagen-specific base alterations as fingerprints in DNA, with respect to the nature of the changes, locations of the changes, and frequencies of the alterations in the gene. Therefore, an analysis of the spectrum of mutation may provide an answer through the empirical approach to the fundamental question: What causes genetic changes in cells? It is now well established that changes in the specific DNA base sequences in the cancer-related genes lead to cancer. Some of the chemical carcinogens have been shown to yield the specific fingerprints of DNA damage. Therefore, to identify the spectra of mutation induced by carcinogenic chemicals which are present in the environment should contribute in evaluating their carcinogenic or mutagenic risks.

In this thesis the author focused on three environmental mutagens, 3-nitrobenzanthrone (NBA, 3-nitro-7*H*-benz[*d,e*]anthracen-7-one), crotonaldehyde and acrolein (Figure A). These compounds are air pollutants in urban area and

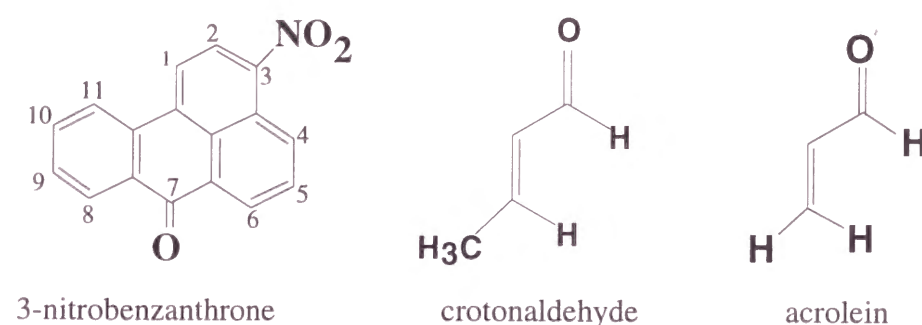


Figure A. Structures of 3-nitrobenzanthrone, crotonaldehyde and acrolein

virtually human population is exposed to them regularly.

NBA is a new class of powerful bacterial mutagen and suspected human carcinogen recently found in diesel exhaust and airborne particulates (3). Its mutagenicity by Ames *Salmonella* assay is very high (208,000 revertants / nmol in *Salmonella typhimurium* TA98 and 6,290,000 revertants / nmol in YG1024) and compares with that of 1,8-dinitropyrene (1,8-DNP) which is the direct mutagen of strongest activity so far reported (3). This new mutagen is also shown to induce micronuclei in mouse peripheral blood reticulocytes, suggesting its potential genotoxicity to mammals (3). However, any other study on biological effect of NBA has not been available.

Crotonaldehyde and acrolein are the simplest member of the class of α,β -unsaturated carbonyl compounds. These are products of combustion; automobile emissions, forest fires and urban fires, and cigarette smoke all contain significant quantities. These chemicals are mutagenic to bacteria *Salmonella typhimurium* (4,5). Although simple in structure, these compounds exhibit a rich chemistry of DNA modification. The structures of the DNA adducts produced by the chemicals has been well elucidated (6). Deoxyguanosine residues appear to be most reactive to the compounds and the chemicals produce hydroxy-1,*N*2-propanodeoxyguanosine (7). Despite the well understanding of their DNA damages, mutational specificities of them are still unknown.

For better understanding of mutagenicity of these chemicals, the author intends to donor the fundamental data of their genotoxicity. The aim of this research is to elucidate the molecular mechanisms of mutations induced by NBA, crotonaldehyde

and acrolein. The specific objectives are as follows:

1. Identify the chemical structures of DNA damage (adduct) induced by NBA
2. Reveal the mutational specificities of NBA, crotonaldehyde and acrolein

Survey of this thesis

Chapter 1 concerns with the structure of DNA adduct formed by NBA. By use of a chemically-synthesized reactive derivative of NBA, the structure of NBA-DNA adduct was identified as *N*-acetyl-3-amino-2-(2'-deoxyguanosin-8-yl)-benzanthrone.

Chapter 2 deals with the investigation of DNA adducts formed in human cells treated with NBA. The result of Chapter 1 is based on *in vitro* study. In order to construct a bridge linked *in vitro* to *in vivo*, the author examined the formation of the adduct in the human cells. ³²P-postlabeling analysis revealed that the adduct exists in the cells. This result indicates that the result of Chapter 1 is able to be applicable to phenomena *in vivo*.

Chapter 3 describes the mutational specificity of NBA in human cells using the chemically-synthesized reactive derivative of NBA. The derivative induced mainly G:C to T:A base substitutions, and its specificity was similar to those by other nit

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CHAPTER 1

An unusual DNA adduct derived from the powerfully mutagenic environmental contaminant

3-nitrobenzanthrone

Abstract

The covalent binding of an *N*-oxy metabolite of the powerfully mutagenic 3-nitrobenzanthrone (NBA) to 2'-deoxyguanosine (dG) and calf thymus DNA has been investigated *in vitro*. The major adduct obtained from the reaction of *N*-acetoxy-*N*-acetyl derivative (*N*-Aco-*N*-Ac-ABA) of 3-aminobenzanthrone (ABA) and dG was identified as *N*-acetyl-3-amino-2-(2'-deoxyguanosin-8-yl)benzanthrone (dG-*N*-Ac-ABA) by ¹H NMR and mass spectroscopies. The adduct also formed in the reaction of *N*-Aco-*N*-Ac-ABA with calf thymus DNA. The coupling with dG moiety occurred exclusively at C-2 of benzanthrone (BA), suggesting a significant contribution of a resonance-stabilized arenium ion intermediate derived from BA to the production of this new type of adduct.

Introduction

Nitrated polycyclic aromatic hydrocarbons and their derivatives (nitro-PAH) are widespread environmental pollutants (1-4). Some members of this class of compounds are known to be mutagenic in bacterial and mammalian cells and tumorigenic in rodents (3-5). Recently, 3-nitrobenzanthrone (3-nitro-7*H*-benz[*d*,*e*]anthracen-7-one; NBA) has been found to be a new class of powerful mutagen present in air borne particulates and/or diesel exhaust particles (6). The direct-acting mutagenicity of this compound in *Salmonella typhimurium* TA98 is comparable to that of 1,8-dinitropyrene, which is the strongest direct-acting mutagen so far reported in the literature. NBA has also been demonstrated to induce micronuclei in mouse peripheral blood reticulocytes after intraperitoneal administration, suggesting its potential as a genotoxin in the mammals (6).

The covalent binding of nitro-PAH metabolites to cellular DNA has been well recognized to play an important role for the initiation of mutagenesis and carcinogenesis (2,5,7). Several mutagenic nitro-PAH are shown both *in vivo* and *in vitro* to be reductively activated to form the DNA adducts predominantly at C-8 or *N*²-position of 2'-deoxyguanosine (dG) and to a lesser extent at C-8 and *N*⁶-positions of 2'-deoxyadenosine (dA) (7,8). Alike other known nitro-PAH mutagens, NBA would be subjected to the metabolic activation in order to form a DNA adduct and exert the genotoxic effect (7-9). In a nitroreductase-deficient *Salmonella* strain TA98NR, the direct-acting mutagenic activity of NBA was found to be lower than in TA98, while in strain YG1024 which overproduces acetyltransferase, the mutagenic activity was 30 times as high as in TA98 (6, 10). These results strongly suggest

that NBA, alike other nitro-PAH, undergoes initial metabolic reduction of the nitro group to form an *N*-arylhydroxylamine, which is then esterified to produce a reactive *N*-acetoxy or *N*-sulfonyloxy ester (7,11-14). These activated esters readily undergo the heterolytic N-O cleavage to form a highly reactive nitrenium ion, which combines with a DNA molecule to yield the corresponding DNA adduct (8,9).

The objectives of this Chapter are to elucidate the mechanism of the mutagenesis induced by NBA and also to obtain the NBA-DNA adduct for using as a marker for exposure of cells to NBA. In order to accomplish these objectives, *N*-acetoxy-*N*-acetyl-3-aminobenzanthrone (*N*-Aco-*N*-Ac-ABA) was synthesized and the formation and structure of the addition products from this compound and dG or DNA were investigated.

Materials and methods

Warning: *3-Nitrobenzanthrone, 3-aminobenzanthrone, and their derivatives are all potential carcinogens and therefore should be handled with care.*

General

Melting points were determined on a Yanagimoto hot stage apparatus and are uncorrected. Thin layer chromatography (TLC) was performed by using Merck precoated silica gel sheets 60F-254. Silica gel (Wakogel C-200) was used for column chromatography. IR spectra were recorded on a SHIMADZU FT-IR DR 8000/8100 infrared spectrophotometer and only prominent peaks in 2000-700 cm⁻¹ region were recorded. ¹H-NMR spectra were obtained in DMSO-*d*₆ with a Varian Gemini-200 (200 MHz) spectrometer. For the ¹H and ¹³C NMR characterization of the adducts, JEOL-JNM-A500 (500MHz for ¹H NMR) and JEOL-JNM-LA300 (75.45MHz for ¹³C NMR) spectrometers were used. Electron impact mass spectra (EI-MS) were recorded on a SHIMADZU GC-MS QP-2000A spectrometer at 70 eV and chemical ionization mass spectra (CI-MS) on a SHIMADZU GC-MS QP-5000 with DI-50 using isobutane as a reacting gas. Electrospray mass spectra (ES-MS) were determined on a Thermo Quest TSQ-7000 system equipped with a Michrom BioResources UMA model 600 HPLC system. Fast atom bombardment mass spectra (FAB-MS) were recorded with a JEOL MS-MP 7000.

Chromatography

High performance liquid chromatograph (HPLC) analysis and purification of the

adducts were performed using a Waters 600E HPLC system equipped with a Waters 991J photodiode array detector and a reversed-phase column of Waters μ bondosphere C₁₈(3.9 x 15 mm). Preparative low pressure liquid chromatograph (LPLC) for the purification of crude adducts was performed with a Yamazen Model 540 equipped with a Model SSC-3000AII UV detector and a FR 50N fraction collector, using a Yamazen ODS-S-40 B semi-preparative column (30 mm i.d. x 250 mm).

Chemicals and Reagents

All solvents were of analytical grade and used after distillation. Reagent grade benzanthrone was purchased from Tokyo Kasei Kogyo Co. Ltd. and used after recrystallization from ethanol (purity >99.9%). 2'-Deoxyguanosine (dG) was purchased from Wako Chemicals Co. Ltd. Calf thymus DNA (Type I), deoxyribon

N-Acetyl-N-hydroxy-3-aminobenzanthrone (N-Ac-N-OH-ABA)

This compound was obtained by the reported method (15). To a suspension of NBA (275 mg) in dimethylformamide (DMF; 20 ml) cooled to 0 °C, 5% palladium /carbon (50 mg) and hydrazine monohydrate (0.2 ml) was added with stirring. The color of the reaction mixture immediately changed from greenish yellow to red, and NBA disappeared gradually as the reaction proceeded. The progress of the reaction was monitored by TLC (hexane - ethyl acetate, 5:1). When the reduction was complete, triethylamine (0.3 ml) followed by acetyl chloride (1.5 ml) was added below 10 °C. After stirring at room temperature for 1 h, the reaction mixture was filtered, diluted with water, and neutralized with aq. sodium carbonate. The solution was extracted with dichloromethane and the resulting organic phase was reextracted with 0.1 M NaOH (20 x 3 ml). The combined aqueous extracts were acidified with concentrated HCl to give the expected product as a yellow solid, pure enough for subsequent purposes. ¹H NMR (DMSO-*d*₆) δ 2.3 (s, 3H), 7.6-7.9 (m, 4H), 8.3-8.4 (m, 2H), 8.6-8.7 (m, 2H), 8.8 (d, *J* = 8.0, 1H), 10.9 (s, 1H); IR (KBr) λ_{max} 2839, 1637, 1576, 1327 cm⁻¹; FAB-MS *m/z* 304.0 (M+1); HR-EI-MS C₁₉H₁₃NO₃, calcd 303.0895, found 303.0883.

N-Acetoxy-N-acetyl-3-aminobenzanthrone (N-Aco-N-Ac-ABA)

N-Ac-N-OH-ABA (200 mg) were dissolved in 5 mL of 2% NaOH. After acetic anhydride (1.2 equiv) was added to the solution, the precipitate was filtered and washed with water. *N-Aco-N-Ac-ABA* was obtained as a yellow solid. This solid was used without further purification for the reactions with dG and DNA. For the

mass and NMR inspections, the compound was further purified by column chromatography on silica gel using dichloromethane/ethanol as the eluent. Although most part of *N-Aco-N-Ac-ABA* decomposed during elution, a small amount of pure compound could be isolated as a yellow powder. Mp 196 °C (decomp); ¹H NMR (DMSO-*d*₆) δ 1.6 (s, 3H), 2.2 (s, 3H), 7.5 (t, *J* = 7.2, 1H), 7.7-7.9 (m, 3H), 8.2 (d, *J* = 8.0, 1H), 8.4-8.6 (m, 3H), 8.7 (d, *J* = 7.2

gradient to 100% methanol in 120 min at a flow rate of 0.8 ml/min. The fraction eluting at 67.37 min was collected and freeze-dried.

Reaction of N-acetoxy-N-acetyl-3-aminobenzanthrone (N-Aco-N-Ac-ABA) with calf thymus DNA

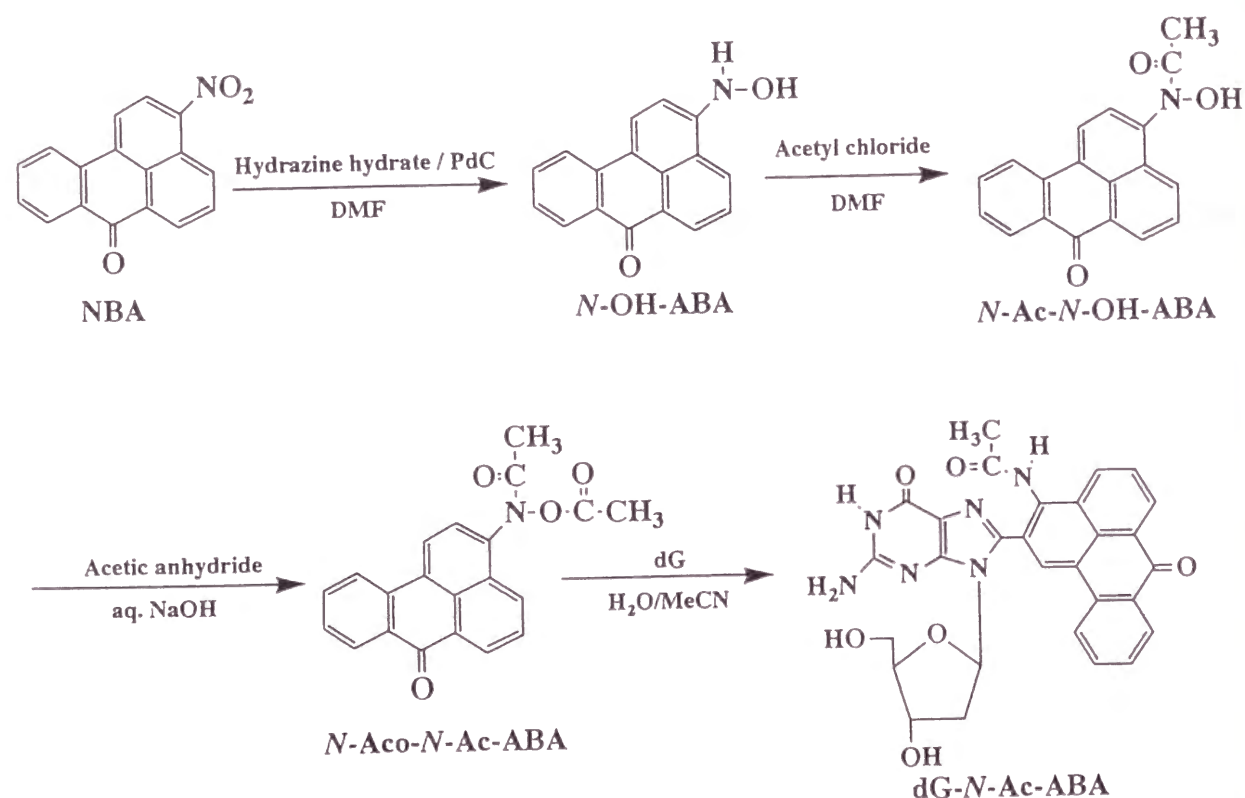
Calf thymus DNA (Type I, 1.5 mg) was incubated at 50 °C for 15 h in the presence of *N-Aco-N-Ac-ABA* (3 mg) in a solution of 10 mM sodium citrate buffer (1.2 ml, pH 5) containing 50% acetonitrile. The reaction mixture was then extracted with ethyl acetate (5 x 0.4 ml). After the addition of one tenth volume of 3 M sodium acetate (pH 5.2), the modified DNA was precipitated by diluting with ice-cold ethanol. The recovered DNA (0.75 mg) was redissolved in a mixture of 0.4 ml of 10 mM Tris-HCl (pH 8) and 1 mM EDTA and enzymatically digested to a mixture of nucleosides; initial incubation with deoxyribonuclease I (0.1 mg/mg of DNA) at 37 °C for 3 h in the presence of 5 mM MgCl₂ was followed by the second incubation with alkaline phosphatase (1.5 u/mg of DNA) and phosphodiesterase I (0.12 u/mg of DNA) at 37 °C for 24 h. The hydrolysate was extracted with water-saturated butanol and the extract was washed with butanol-saturated water. The butanol phase was taken up and evaporated to dryness under reduced pressure to leave a residue, which was redissolved in DMSO for HPLC analysis.

Results

Synthesis of N-Acetoxy Derivative of 3-Aminobenzanthrone (ABA)

For the *in vitro* preparation of nitro-PAH-DNA adducts, several successful approaches have previously been reported. They include the reactions of (i) dG with *N*-acetoxy-*N*-trifluoroacetyl or *N*-acetoxy derivatives of nitro-PAH metabolites (16-20), (ii) dG with *N*-acetyl-*N*-acyloxyamines (21-25), and (iii) DNA with hydroxylamine derivatives in acidic media (26-29). The author chooses the *N*-acetoxy-*N*-acetyl route, since *N*-acetoxy-*N*-acetyl and several related derivatives were successfully employed for the synthesis of dG adducts of several aromatic amines (Scheme 1-1) (21-25). Moreover, this compound is relatively stable and easy to handle for the present purpose.

The starting material *N-Ac-N-OH-ABA* was efficiently prepared by the Pd/C-catalyzed reduction of NBA with hydrazine hydrate in DMF, followed by selective *N*-acetylation of the resulting hydroxylamine according to the Weatra's method (15); the reduction of NBA to 3-(hydroxyamino)benzanthrone (*N-OH-ABA</*



Scheme 1-1. Synthesis and transformation of *N*-Aco-*N*-Ac-ABA

Ac-ABA as a yellow precipitate. Although the *O*-acetylation of *N*-Ac-*N*-OH-ABA at room temperature smoothly led to *N*-Aco-*N*-Ac-ABA, the same reaction at 0 °C led to an unidentified compound instead of the expected one.

Reaction of N-Acetoxy-N-acetyl-3-aminobenzanthrone (N-Aco-N-Ac-ABA) with dG and Characterization of dG-N-Ac-ABA Adduct

Among several ABA derivatives synthesized in the present work, only *N*-Aco-*N*-Ac-ABA reacted successfully with dG at pH 5 and 7. HPLC analysis of the reaction product at 60 °C showed new peaks, which were not observed in a control reaction

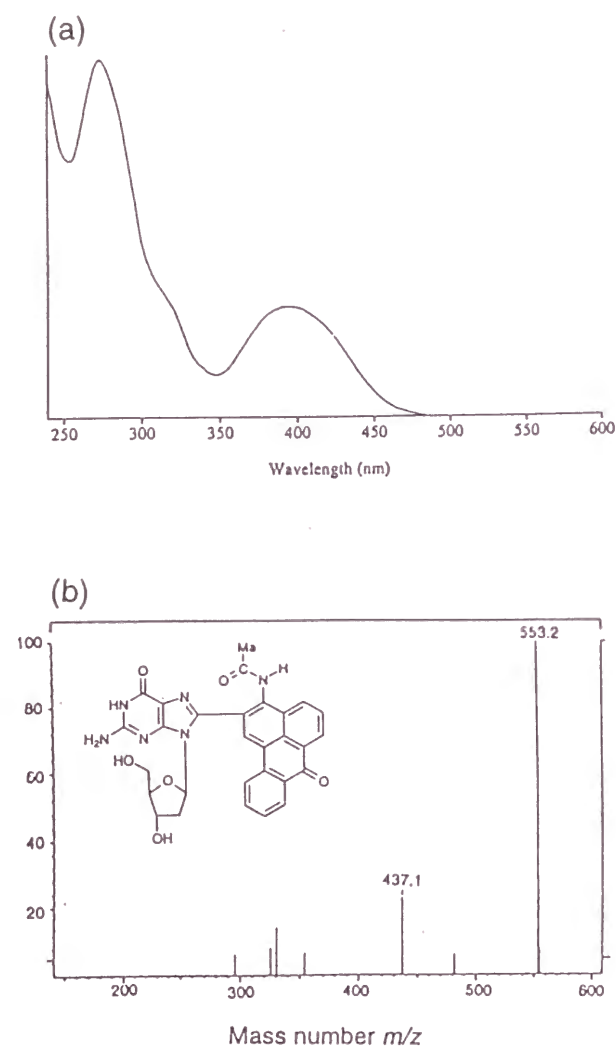


Figure 1-1. (a) On-line HPLC-UV spectrum and (b) ES-MS spectrum of dG-*N*-Ac-ABA

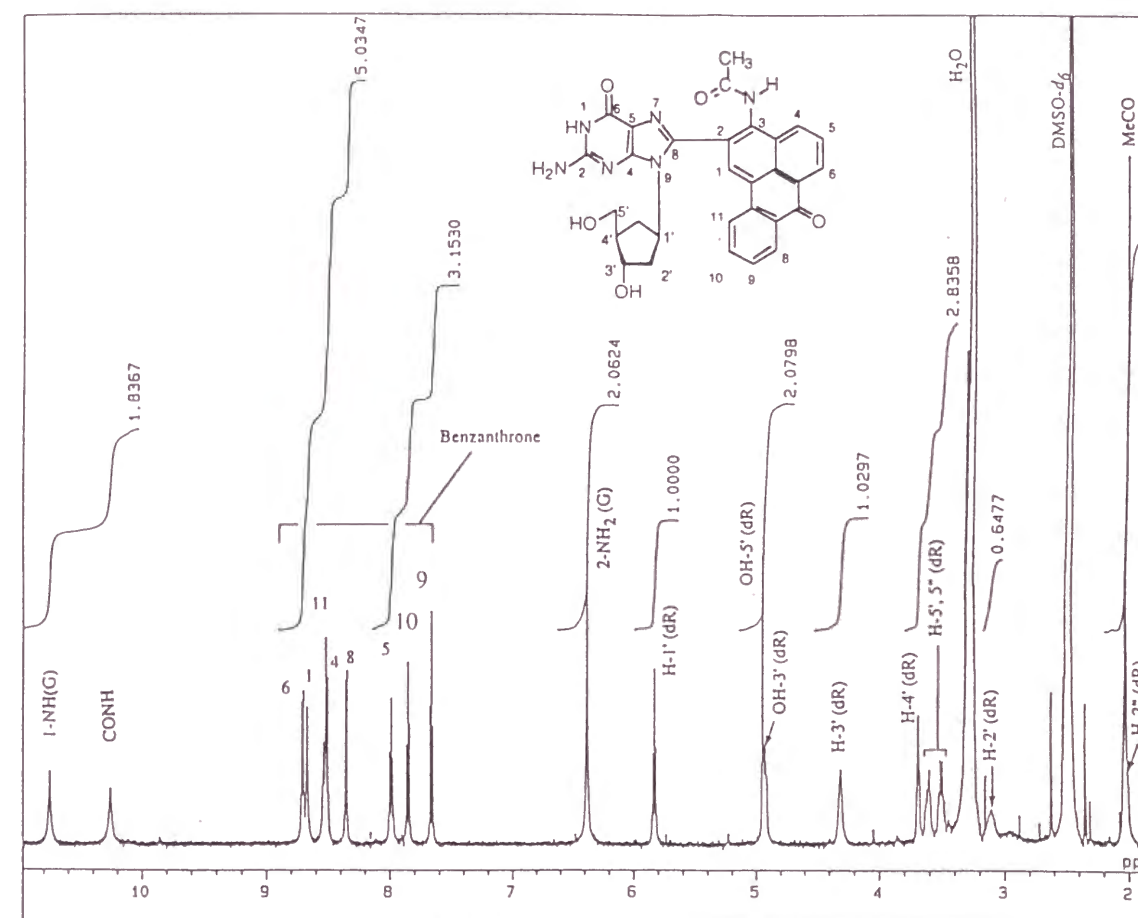


Figure 1-2. 500-MHz ^1H -NMR spectrum of dG-*N*-Ac-ABA in $\text{DMSO}-d_6$

Table 1-1 500 MHz ¹H NMR Spectral Parameters of dG-N-Ac-ABA

δ (ppm)	multiplicity		assignment
10.73	s		N1H (G)
10.16	s		CONH (amide)
8.71	d		H-6 (BA)
8.67	s		H-1 (BA)
8.54	d		H-4 (BA)
8.51	d		H-11 (BA)
8.36	d		H-8 (BA)
7.99	t		H-5 (BA)
7.86	dt		H-10 (BA)
7.67	t		H-9 (BA)
6.37	s (2)		2-NH ₂ (G)
5.84	t(1)		H-1'(dR)
4.95	d		OH-5'(dR)
4.93	t		OH-3'(dR)
4.36	m(br)		H-3'(dR)
3.70	ddd		H-4'(dR)
3.61	ddd		H-5'(dR) ^a
3.51	dd		H-5''(dR) ^a
3.11	m(br)		H-2'(dR)
2.03	s		H-2''(dR) and CH ₃ (amide)
(Hz)			
$J_{4,5}=8.0$	$J_{5,6}=8.0$	$J_{8,9}=7.2$	$J_{9,10}=7.1$
$J_{10,11}=8.0$	$J_{1'2'}=\sim 7.1$	$J_{1'2''}=\sim 6.9$	$J_{2'2''}=\sim 10.2$
$J_{2'3'}=\sim 5.$			

at C-8 through the amine or amide group. However, this is not the case with our adduct.

Table 1-2 ¹³C-NMR Chemical shift of dG-*N*-Ac-ABA

δ (ppm)	assignment	δ (ppm)	assignment
182.01	C7(BA)	127.63	C-11b (BA)
169.05	C=O (amide)	127.13*	C-8 (BA)
156.68	C6 (G)	127.01*	C-5 (BA)
153.16	C2 (G)	126.28*	C-1 (BA)
151.09	C4 (G)	125.56	C-11c (BA)
144.56	C8 (G)	123.82	C-2 (BA)
136.15	C3 (BA)	123.67*	C-11(BA)
134.71	C-11a (BA)	117.29	C-5 (G)
133.88*	C-10 (BA)	87.65*	C-4 (dR)
131.70*	C-4 (BA)	84.91*	C-1 (dR)
129.93*	C-6 (BA)	70.97*	C-3(dR)
129.87	C-7a (BA)	62.13	C-5 (dR)
129.16	C-6a (BA)	36.95	C-2 (dR)
128.60*	C-9 (BA)	22.71*	CH ₃ (amide)
127.65	C-3a (BA)		

Assignments are made on 2D-NMR experiments as well as by direct comparison with 2'-deoxyguanosine and benzanthrone ²⁷.

Peaks enhanced in DEPT are marked with an asterisk.

Reaction of *N*-Acetoxy-*N*-acetyl-3

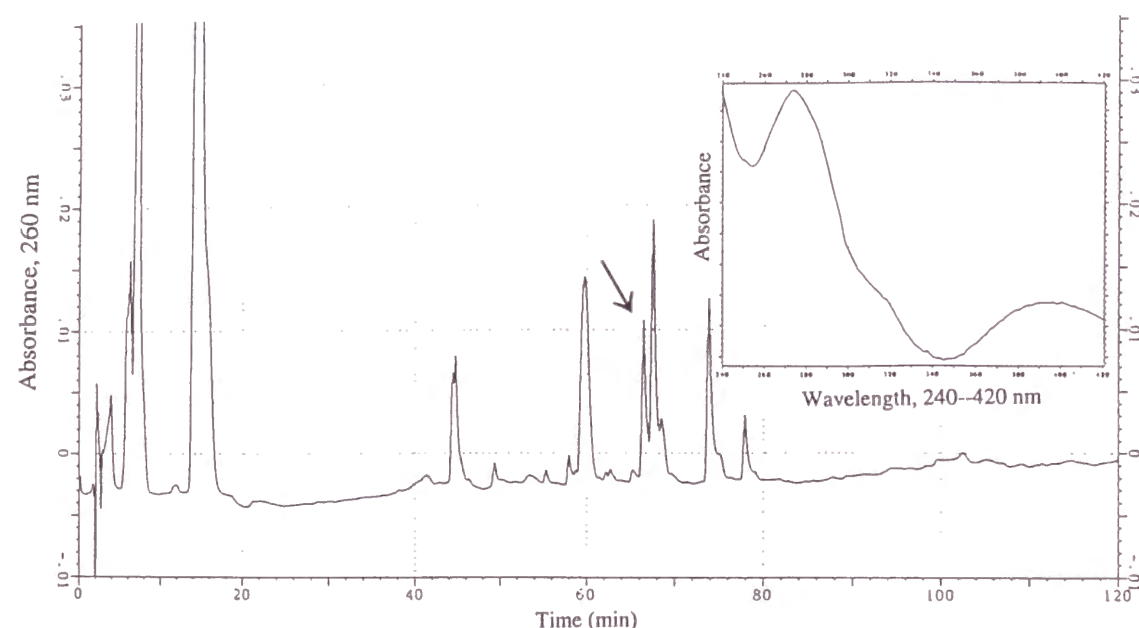


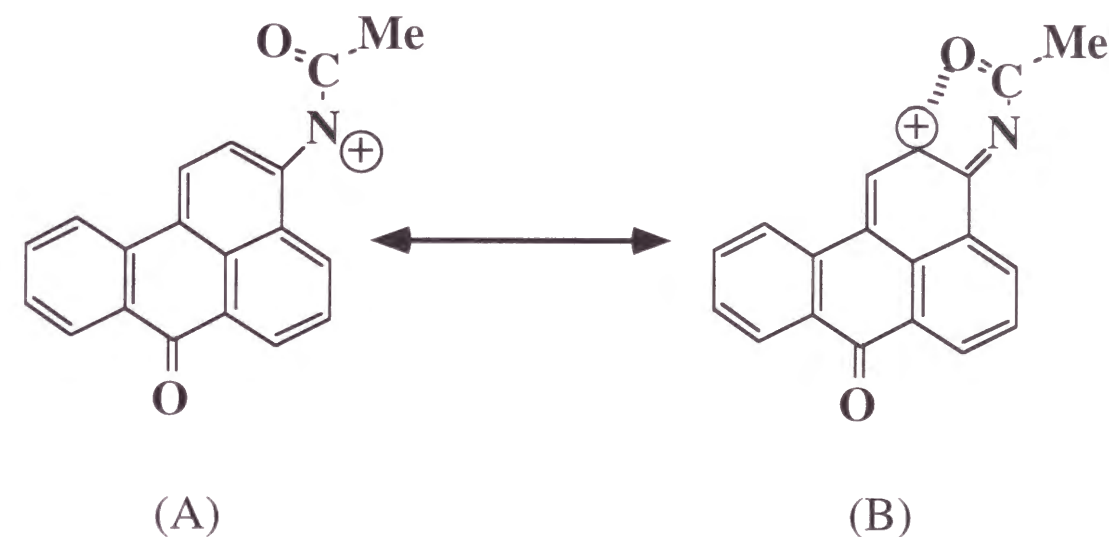
Figure 1-3. HPLC analysis and on-line UV spectra of the enzymatic digest of calf thymus DNA modified with *N-Aco-N-Ac-ABA*

Discussion

In this research, the author has synthesized *N*-acetoxy-*N*-acetyl-3-aminobenzanthrone (*N-Aco-N-Ac-ABA*) and made it to react with dG and calf thymus DNA, obtaining the dG-*N-Ac-ABA* adduct in a yield of around 10%. Synthesis and reactions of the DNA and dG adducts from *N*-acetoxy-*N*-acetyl amino derivatives of fluorene and biphenyl compounds have already been studied well. In those studies, *N*-acetoxy-*N*-acetyl-2-aminofluorene was found to react with DNA and dG to give several dG-adducts in good yield (24). However, the reaction of *N*-acetoxy-*N*-acetyl-4-aminobiphenyl with dG failed to produce similar dG-adducts due to facile hydrolytic cleavage of the *N*-acetoxy group, which makes it difficult to generate a nitrenium ion species (30). Underwood *et al.* showed that *N*-acetyl-*N*-(2,6-dichlorobenzoyloxy)-4-aminobiphenyl was reluctant to hydrolysis and reacted with dG to give the corresponding adduct in moderate to good yield (23). In the case of this research, however, replacement of the acetoxy group in *N-Aco-N-Ac-ABA* by the 2,6-dichlorobenzoyloxy group was not fruitful for the adduct formation. *N*-Acetyl-*N*-(2,6-dichlorobenzoyloxy)-3-aminobenzanthrone was found to be so stable that it neither reacted with dG nor went to hydrolysis even at temperatures as high as 100 °C. Acetyl salicylate was reported to be useful for the synthesis of dG-C8-aminofluorene adduct (31), but it did not work satisfactorily on *N*-OH-ABA to give the dG adduct. *N*-OH-ABA was mixed with a plasmid DNA at pH 5 in ethanol/sodium citrate buffer, but the author could not observe any increase in the frequency of mutation in *supF* shuttle vector system (32). This finding suggests that the DNA adduct formation by *N*-OH-ABA does not take place effectively. Although direct

acetoxylation of the hydroxylamino function using acetyl cyanide was reported (17-19), the author has not yet examined it.

The dG adduct obtained in this study has been shown to possess a unique chemical structure, quite different from those derived from other aromatic nitro or amino compounds as previously reported. The adduct formation of nitro-PAH metabolites generally occurs at C-8 of dG through the amino or amide group, or at *N*² of dG through the most cationic ring carbon resulting from the resonance-stabilization of the initial nitrenium ion intermediate (8,9,29). For nitropyrenes, nitrofluoranthenes, and nitrobiphenyls, the C-8 substituted dG adducts are known to make significant contribution to the total DNA adducts (7,25-28). For nitro-PAH of greater ring numbers, such as 3-nitrobenzo[*a*]pyrene and 6-nitrochrysene, the *N*²-substituted guanine adducts are formed to considerable extent owing to the increased contribution of the carbocationic resonance form of a putative nitrenium ion intermediate (7, 19). In the case of NBA, however, the coupling occurred at C-8 of dG through C-2 of benzanthrone. This type of C-C bond coupling in the reaction of dG with aromatic *N*-acetoxy-amino compounds has not previously been reported, though there are a few scattered papers that report the C-C bond formation of electronically oxidized benzopyrene and dibenzocarbazole with dG or dA (33-35). The new type of adduct obtained in this research would plausibly be formed via the isomerization of the initially formed nitrenium ion (Scheme 1-2, (A)) to an energetically favored arenium ion (Scheme 1-2, (B)) with a localized positive charge at C-2 position, followed by attack of this carbocation on C-8 position of dG. However, it is also possible for this carbocation to attack the *N*² position of dG to



Scheme 1-2. Resonance-stabilization of a putative nitrenium ion intermediates from *N*-Aco-*N*-Ac-ABA

give an *N*²-adduct. HPLC-MS analysis of the product mixture from the reaction of *N*-Aco-*N*-Ac-ABA with dG showed the presence of an additional adduct of a molecular ion peak at *m/z* 553, but its amount was too small to identify the product.

The conformation of the glycoside linkage appears to play an important role in adduct persistence *in vivo* and consequently it may influence the toxicological properties of DNA adducts (29,36-41). The *syn* type adduct appears to induce a greater distortion of the DNA helix than the *anti* type one, resulting in rapid disappearance of the former adduct and greater persistence of the latter. Chemical shift of the sugar H-2' is generally a good probe for the preferred conformation of the glycoside linkage in dG, appearing at δ 2.5 ~ 3.1 due to the deshielding effect of nearby guanine N-3 atom in the *syn* conformation (36-41). The H-2' atom of dG-*N*-Ac-ABA adduct was found to absorb at δ 3.1, which was considerably downfield as compared with H-2' of the parent dG (δ 2.5), suggesting the conformation of this adduct to be *syn*. Upfield shift of the sugar C-2' absorption at δ 36.9 of dG-*N*-Ac-ABA in ¹³C NMR spectrum (generally δ 40) also supports this conformation (36-41).

Although the metabolic activation and *in vivo* DNA adduct formation of NBA have not been studied yet, alike other nitro arenes the *N*-hydroxy derivatives of ABA as well as the ultimate mutagenic species like *N*-acetoxy, *N*-sulfonyloxy, or *N*-propyloxy derivatives of ABA are all highly likely to be probable mutagenic metabolites of NBA. In order to examine whether they form the unique C-2 adduct like dG-*N*-Aco-ABA or ordinary C-8 and *N*²-adducts *in vivo*, further study is ongoing.

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CHAPTER 2

A ³²P-postlabeling analysis of DNA adducts formed in human hepatoma HepG2 cells treated with 3-nitrobenzanthrone

Abstract

3-Nitrobenzanthrone (NBA) is a mutagenic aromatic compound which is found in diesel exhaust and airborne particles. In this Chapter, the ³²P-postlabeling analysis was used to examine the adducts in DNA from human hepatoma HepG2 cells treated with NBA. Two major and two minor adduct spots were obtained in the analysis. The structure of the compound in an obtained minor adduct spot was identified to be *N*-acetyl-3-amino-2-(2'-deoxyguanosin-3',5'-bisphosphate-8-yl)-benzanthrone, based on TLC mobility of the compound in comparison with that of synthetic standard.

Introduction

3-Nitrobenzanthrone (NBA, 3-nitro-7*H*-benz[*d,e*]anthracen-7-one) is a powerful mutagenic aromatic nitroketone which is found in diesel exhaust and airborne particles (1). Its mutagenicity by Ames *Salmonella typhimurium* (TA98) assay is as high as that of 1,8-dinitropyrene (1,8-DNP) which is the strongest mutagen so far reported (1). The compound also induces micronuclei in mouse peripheral blood reticulocytes and, therefore, is thought to be genotoxic to mammals and a suspected human carcinogen (1).

Metabolic activation of nitroaromatic hydrocarbons occurs through *N*-reduction to form the *N*-hydroxy arylamine, which can bind to DNA, or may undergo further activation by esterification to produce highly reactive species which react with DNA (2-5). The covalent binding of genotoxic mutagens to DNA is regarded as a critical event in cancer initiation. In the case of NBA, reaction of a chemically synthesized reactive form, *N*-acetoxy-*N*-acetyl-3-aminobenzanthrone (*N*-Aco-*N*-Ac-ABA), with calf thymus DNA results in the formation of DNA adducts: *N*-acetyl-3-amino-2-(2'-deoxyguanosin-8-yl)-benzanthrone (dG-*N*-Ac-ABA) and a structure unidentified dA adduct (dA-*N*-Ac-ABA) *in vitro* (Chapter 1)(6). The dG-*N*-Ac-ABA possesses a unique chemical structure, that is quite different from those derived from other aromatic nitro or amino compounds as previously reported. The adduct formation of nitro-PAH metabolites generally occurs at C-8 of dG through the amino or amide group, or at *N*² of dG through the most cationic ring carbon resulting from the resonance-stabilization of the initial nitrenium ion intermediate (2,7). In the case of dG-*N*-Ac-ABA, the coupling between nucleoside and the aromatic compound

occurred at C-8 of dG through C-2 of benzanthrone. This type of C-C bond coupling in the reaction of dG with aromatic *N*-acetoxy-amino compounds has not previously been reported. In this study, DNA samples from human hepatoma HepG2 cells treated with NBA were analyzed by ³²P-postlabeling to examine the formation of these adducts in the cells.

Analysis of DNA adduct by HPLC

The ^{32}P -labeled sample with an additional treatment using PDE I was developed on TLC plate. The adduct spot corresponding to 5'-[^{32}P]pdG-*N*-Ac-ABA on TLC plate was cut out and extracted with 0.5 mL of 4 M pyridinium formate (pH 4.5) solution twice with shaking for 30 min. The mixture was centrifuged, and supernatant was collected by passing through a 0.2 μm filter and evaporated. The authentic sample of 5'-pdG-*N*-Ac-ABA was synthesized by the same procedure for 3'-dGp-*N*-Ac-ABA as described above. An aliquot of the radioactive sample was mixed with appropriate amount of the authentic sample and subjected to HPLC on a Waters μ bondosphere C_{18} column (3.9 mm x 15 mm) with the following solvent system at a flow rate 1.0 mL / min at 40 $^{\circ}\text{C}$: 0 - 10 min, 10% acetonitrile in 0.05 M TEAA (pH 7.0), 10 - 45 min, a linear gradient of 10 - 50 % acetonitrile in 0.05 M TEAA (pH 7.0), 45 - 50 min, 50 % acetonitrile in 0.05 M TEAA (pH 7.0). The eluent was monitored by measuring its absorbance at 395 nm and collected at 2 min intervals. Radioactivity of each fraction was measured by a liquid scintillation analyzer.

Results and discussion

The ^{32}P -postlabeling TLC profiles of synthetic 3',5'-[^{32}P]pdGp-*N*-Ac-ABA and 3',5'-[^{32}P]pdAp-*N*-Ac-ABA adduct standards, DNA from untreated human hepatoma HepG2 cells, DNA from the cells treated with 9 μM NBA for 8 hours, DNA from treated the cells spiked with the adduct standards are shown in Figure 2-1. Spots co-migrating with the adduct standards 3',5'-[^{32}P]pdGp-*N*-Ac-ABA and 3',5'-[^{32}P]pdAp-*N*-Ac-ABA appeared after DNA from the cells treated with NBA was developed (Figure 2-1 C, D). No spot was observed after DNA from non-treated cells was developed (Figure 2-1 E).

Several other spots were also observed around the origin on TLC plate after DNA from treated cells was developed. For analysis of these spots, further development were performed using solvents with higher salt and urea concentrations (Figure 2-2). The spots consisted of two spots (Spot 3, Spot 4). As in the case of PhIP-DNA adducts (11), spots 3 and 4 might be undigested dinucleotides or oligonucleotides containing the dG-adduct or the dA-adduct. Therefore, the author examined a ^{32}P -labeled sample further treated with phosphodi

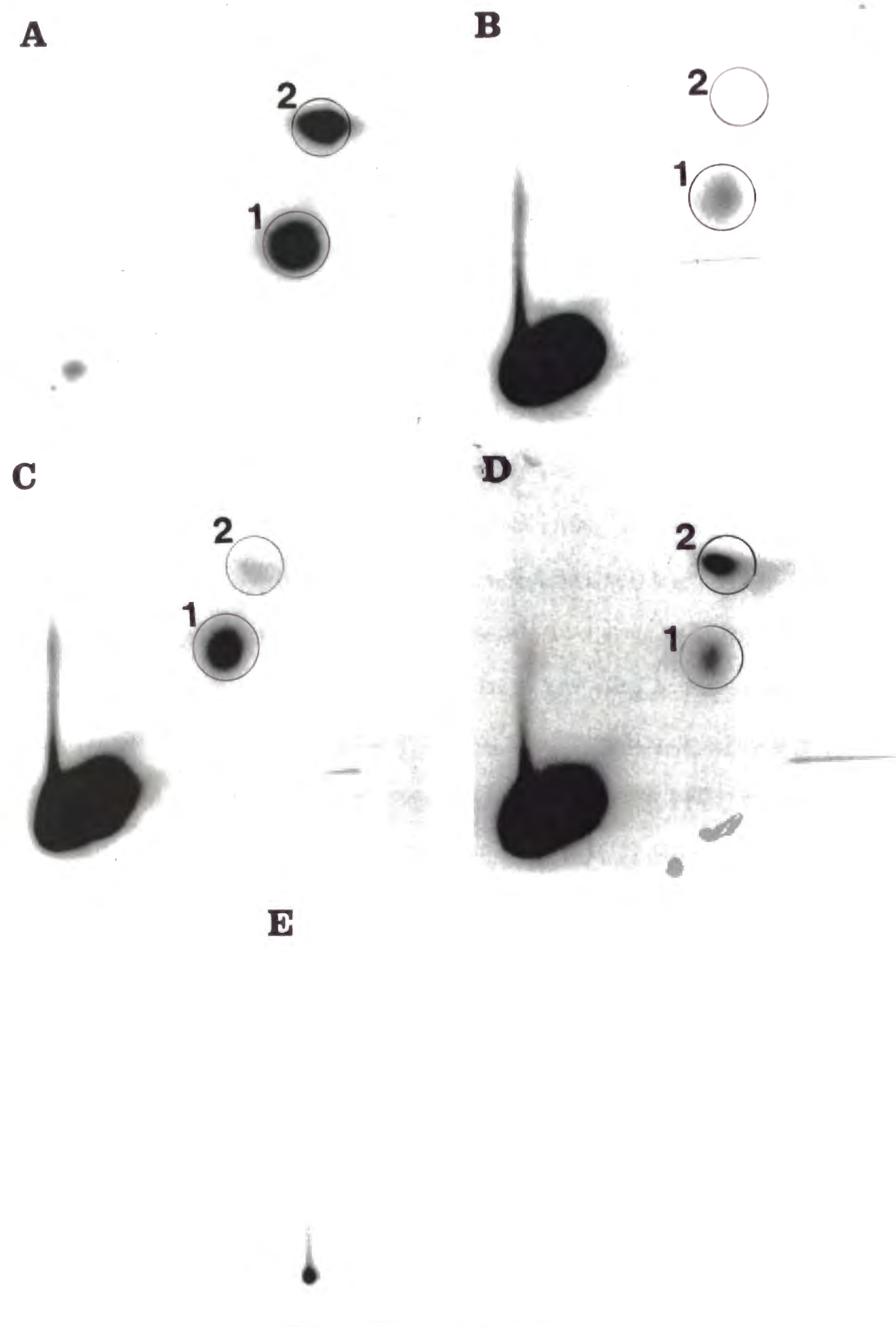
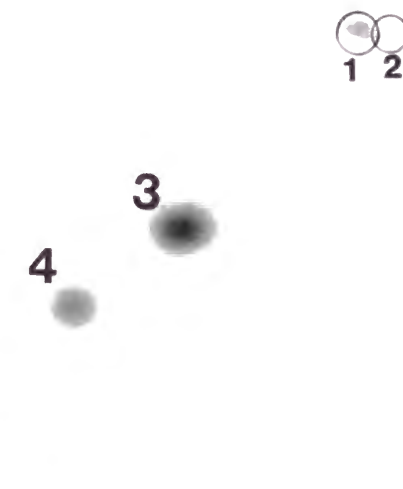


Figure 2-1. ³²P-Postlabeling analysis of (A) standard DNA adducts, 1: 3',5'-[³²P]pdGp-*N*-Ac-ABA, 2: 3',5'-[³²P]pdAp-*N*-Ac-ABA; (B) DNA from an NBA treated cells (9 μM for 8 hours); (C) DNA from NBA treated cells spiked with the adduct standard 3',5'-[³²P]pdGp-*N*-Ac-ABA; (D) DNA from NBA treated cells spiked with the adduct standard 3',5'-[³²P]pdAp-*N*-Ac-ABA; (E) DNA of control cells.

Figure 2-2. ³²P-Postlabeling analysis for development of spots observed around the origin in Figure 2-1. The TLC plate of Figure 2-1 (B) was further developed using higher salt and urea concentration solvents. Time of exposure to X-ray film was shorter than that of Figure 2-1.



HPLC analysis with the unlabeled synthetic 5'-pdG-*N*-Ac-ABA adduct standard as a visual marker. The elution position of the major radioactivity on HPLC coincided with that of the UV absorption peak of authentic 5'-pdG-*N*-Ac-ABA at retention time 31-33 min as shown in Figure 2-3. The HPLC analysis of 5'-pdA-*N*-Ac-ABA could not be performed because of weak radioactivity of the spot corresponding to 5'-[³²P]pdA-*N*-Ac-ABA.

Levels of adducts in the cells treated with NBA for 8 hours are shown in Table 2-1. Levels of all four spots increased with an increase in concentration of NBA. Spots 3 and 4 were the predominant adducts. Levels of 3',5'-[³²P]pdGp-*N*-Ac-ABA and 3',5'-[³²P]pdAp-*N*-Ac-ABA were less than 1/100 of those of spots 3 and 4.

In this study, the author found that four adducts (two major adducts and two minor adducts) exist in the DNA from the cells treated with NBA. Two minor adducts were identified as 3',5'-pdGp-*N*-Ac-ABA and 3',5'-pdAp-*N*-Ac-ABA, based on TLC mobility in comparison with synthetic standards. However, the structures of major DNA adducts are still unknown. Of known adducts between DNA and aromatic amines or nitro-PAH, the adducts with the *N*-acetyl group are generally found as minor adducts and those without *N*-acetyl group are found as major adducts *in vivo* (7). Therefore, spot 3 and 4 might be adducts without *N*-acetyl group. Further study should be done.

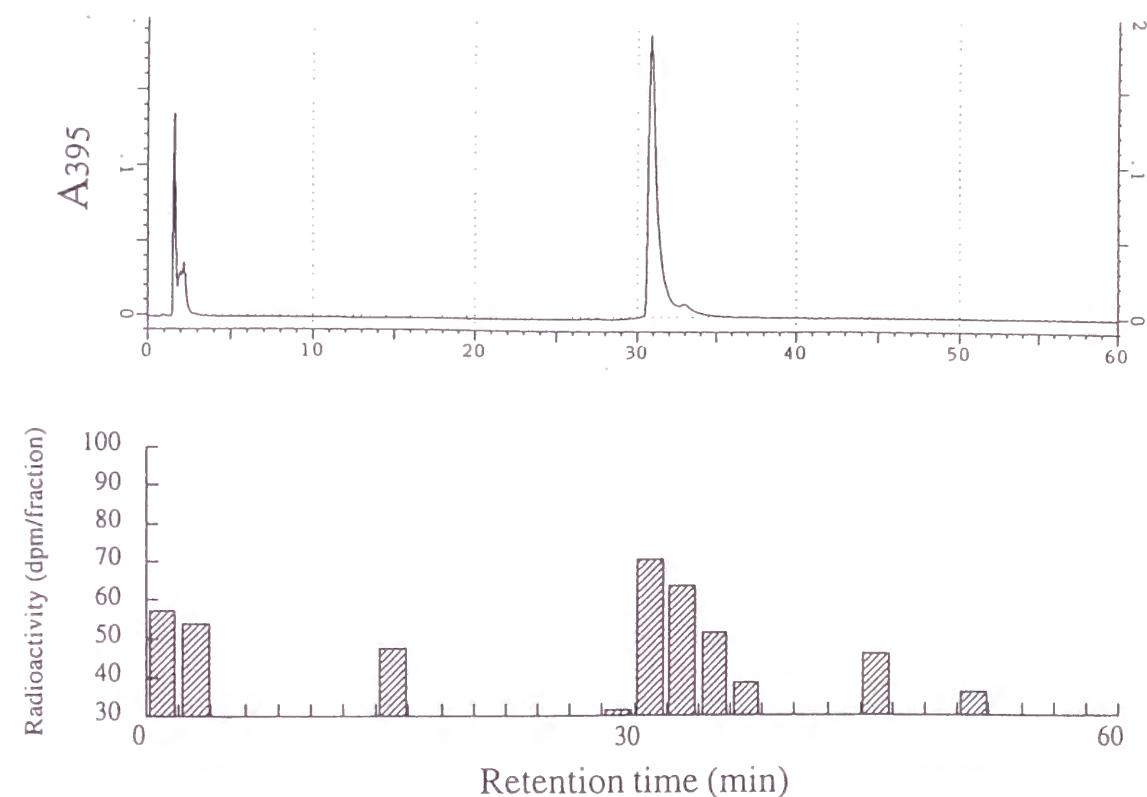


Table 2-1 DNA adduct levels in human hepatoma HepG2 cells treated with NBA

DNA adduct levels (RAL ^a x 10 ⁸)					
Cell		Spot 1:	Spot 2:	Spot 3	Spot 4
NBA	viability	3',5'-[³² P]pdGp-	3',5'-[³² P]pdAp-		
(μM)	(%)	N-Ac-ABA	N-Ac-ABA		
0	100	N.D.	N.D.	N.D.	N.D.
0.9	67±1.8	0.95±0.32	0.87±0.18	106±3.3	185±3.1
9	56±3.5	2.8±0.59	1.2±0.17	254±29	364±14

The values are mean ± SD of at least three assays.

N.D.: not detected (RAL < 1 x 10⁻⁹)

^aRAL: relative adduct labeling

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CHAPTER 3

Mutagenic specificity of a derivative of 3-nitrobenzanthrone in the *supF* shuttle vector plasmids

Abstract

3-Nitrobenzanthrone (NBA) is a powerful bacterial mutagen and a suspected human carcinogen present in diesel exhaust and airborne particulates (1). In the Chapter 1 (2), *N*-acetoxy-*N*-acetyl-3-aminobenzanthrone (*N*-Aco-*N*-Ac-ABA) was synthesized to yield the DNA adducts of NBA. In this Chapter, to investigate the mutagenic specificity of NBA in human cells, the author analyzed mutations induced by *N*-Aco-*N*-Ac-ABA using the *supF* shuttle vector plasmids. Base sequence analysis of 110 and 100 plasmids with mutations in the *supF* gene propagated in normal cells (WI38-VA13) and nucleotide excision repair deficient cells (XP2OS (SV)), respectively, revealed that the majority of the mutations was base substitutions (85 % and 90 %) and the rest were deletions and insertions (10 % and 15 %) in both cell lines. About half of the mutant plasmids had a single base substitution. Of the base substitutions, the most frequent mutation was G:C to T:A transversion (41 % and 51 %), followed by G:C to A:T transitions (18 % and 24 %) in

either cell. The mutations were distributed not randomly but located at several hot spots, and almost all (9 of 10) hot spots were at the sites of G:C base pairs. The polymerase-stop assay in the *supF* gene revealed that *N*-Aco-*N*-Ac-ABA preferentially bound to guanine residues, and mutation sites were generally consistent to the sites where the guanine adducts were formed.

Introduction

3-Nitrobenzanthrone (NBA¹, 3-nitro-7*H*-benz[*d,e*]anthracene-7-one) is a powerful mutagenic aromatic nitroketone present in diesel exhaust and airborne particles (1). Mutagenicity of NBA by the Ames *Salmonella typhimurium* (TA98) assay is as high as that of 1,8-dinitropyrene (1,8-DNP), the strongest mutagen so far reported (1). The compound also induces micronuclei in mouse peripheral blood reticulocytes and, therefore, is thought to be genotoxic to mammals and a suspected to be a human carcinogen (1).

Most carcinogens are metabolically activated in cells and bind covalently to DNA (3). Nitroaromatic hydrocarbons need to be metabolized into reactive electrophiles to exert their carcinogenic activity (4). The initial activation of nitroaromatic hydrocarbons is through the reduction catalyzed by both microsomal and cytosolic enzymes to form *N*-hydroxy arylamines (4). *N*-Hydroxy arylamines are further metabolized to *N*-hydroxy arylamids, *N*-sulfonyloxy arylamines, *N*-acetoxy arylamines and *N*-sulfonyloxy arylamids, by several transferases (4)(5). These esters except for *N*-hydroxy arylamids undergo heterolysis of the N-O bond to produce strongly electrophilic arylnitrenium / carbonium ions and react with DNA to form adducts (6)(7). In the Chapter 1 (2), the author synthesized an *N*-acetoxy-*N*-acetyl derivative of NBA, *N*-acetoxy-*N*-acetyl-3-aminobenzanthrone (*N*-Aco-*N*-Ac-ABA), which produces reactive arylnitrenium / carbonium ions of NBA and showed that this derivative reacted with DNA and formed the adduct *N*-acetyl-3-amino-2-(2'-deoxyguanosine-8-yl)-benzanthrone (dG-*N*-Ac-ABA) *in vitro*. The author has also confirmed the presence of dG-*N*-Ac-ABA in DNA in human

hepatoma HepG2 cells treated with NBA by the ^{32}P -postlabeling analysis (Chapter 2).

Shuttle vector plasmids have been used to examine the mutations caused by environmental mutagens and carcinogens (8). Mutations induced in the plasmids by mutagens reflect mutations in endogenous genes of mammalian cells in which plasmids are propagated (9). In this study, the author adopted the pMY189 shuttle vector plasmids to analyze the mutational specificity of NBA in human cells. The plasmids were treated with *N*-Aco-*N*-Ac-ABA, and frequencies and types of mutations in the *supF* gene of the plasmids induced by the derivative in normal and repair-deficient human cells were investigated.

Materials and methods

Chemicals

N-Aco-*N*-Ac-ABA was synthesized and purified on column chromatography as described (Chapter 1)(2). Ampicillin, chloramphenicol, nalidixic acid, isopropyl- β -D-thiogalactoside (IPTG) and 5-bromo-4-chloro-3-indoyl- β -D-galactoside (X-gal) were obtained from Wako Chemicals (Osaka, Japan). Restriction endonuclease *Dpn*I and T4 polynucleotide kinase were purchased from New England Biolabs (Beverly, MA) and Toyobo Biochemicals (Tokyo, Japan), respectively. Primer 1 (pBR322 *Eco*RI site primer: 5'-GTATCACGAGGCCCTT-3') was obtained from Takara Shuzo Biochemicals (Tokyo, Japan). Primer 2 (5'-TTCTACGGGGTCTGACGC-3') were synthesized by Nippon Seihun Co. (Tokyo, Japan).

Cells

An SV40-transformed normal human fibroblast cell line WI38-VA13 (10) was obtained from the American Type Culture Collection (Rockville, MD). DNA repair deficient XP2OS(SV) cells were established from a Japanese group A xeroderma pigmentosum (XP) patient (11). All cells were cultured in Dulbecco's modified minimum essential medium (Nikken, Kyoto, Japan) supplemented with 10% fetal bovine serum (Hyclone, Logan, UT).

Shuttle vector plasmid and Bacterial strains

The shuttle vector plasmid pMY189, constructed by Matsuda *et al.* (12), was

electrophoresis. The base sequences of the *supF* gene of the plasmids which were not aberrant in size were determined with the ABI PRISM™ Dye Primer (-21M13) Cycle Sequencing Ready Reaction Kit (Perkin-Elmer Corporation, Foster City, CA) using the 370A automatic DNA sequencer (Perkin-Elmer Corporation, Foster City, CA). The mutants having the identical base changes derived from the same transfection plate were not scored to exclude clusters of the same clones.

The χ^2 -goodness of fit test was used to determine if the *N*-Aco-*N*-Ac-ABA-induced base substitution mutations were distributed randomly or non-randomly over the coding region of the *supF* tRNA gene.

Polymerase-stop assay of N-Aco-N-Ac-ABA-treated template

Positions of DNA adducts in the *supF* gene of *N*-Aco-*N*-Ac-ABA-treated plasmids was determined by the *in vitro* DNA polymerase-stop assay (15) as described previously (16)(17). Briefly, double-stranded pMY189 plasmids treated with *N*-Aco-*N*-Ac-ABA (0, 0.2, 2 and 4 mM, at 37 °C for 1 hour in 10 mM sodium citrate buffer, pH 6.7) were denatured and annealed with ³²P end-labeled primer 1 (complementary to the sequence of non-coding strand at sites from -46 to -31 in the pMY189) or primer 2 (complementary to the sequence of coding strand at sites from 211 to 228 in the plasmid). Primers were labeled with [γ -³²P] ATP (3000 Ci/mmol) using T4 polynucleotide kinase. The polymerization reaction was performed in a manner similar to the sequencing reactions using a Tth Polymerase DNA Sequencing kit "Sequencing PRO" (Toyobo Biochemicals, Tokyo, Japan), except that dideoxynucleotides were omitted, and DNA adducts were allowed to terminate

polymerization. DNA from the four dideoxy sequencing reactions with an untreated template and a ³²P-labeled primer was electrophoresed on the same gel to serve as position makers of DNA sequences. The relative intensities of the bands on the autoradiogram of the gel were determined by the NIH Image software.

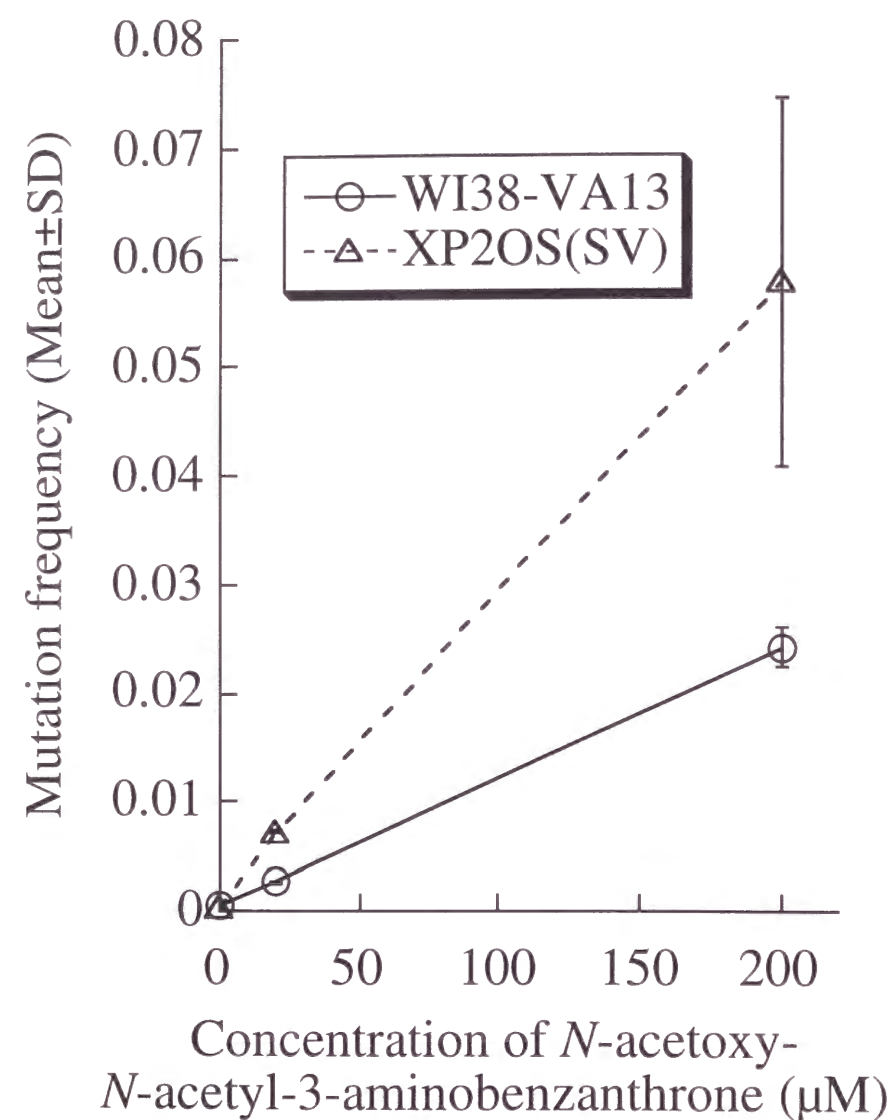


Figure 3-2. Mutation frequency of the *N-Aco-N-Ac-ABA*-treated pMY189 plasmids propagated in repair-proficient WI38-VA13 (○) and repair-deficient XP2OS(SV) (△). Average numbers of three independent experiments are plotted with the S.D.

Table 3-1 Types of sequence alternations in *supF* gene in *N-Aco-N-Ac-ABA*-treated plasmids pMY189 replicated in normal (WI38-VA13) or repair deficient (XP2OS(SV)) human fibroblasts

Types	No. of plasmids with mutations (%)	
	WI38-VA13	XP2OS(SV)
Base substitutions	99 (90)	86 (86)
Single	45 (41)	61 (61)
Tandem	10 (9)	3 (3)
Multiple*	44 (40)	22 (22)
Deletions and insertions	11 (10)	14 (14)
Single base deletion **	6 (5)	11 (11)
≥		

than those with base substitutions and most of them accompanied base substitutions. (Footnotes of the table).

Types of the base substitutions are shown in Table 3-2. The majority of the base substitutions (78-83 %) occurred at guanine or cytosine in both cells. The most frequent mutations were G:C to T:A transversions and the next frequent mutations were G:C to A:T transitions in either cell line.

The author selected *supF* mutants from experiments in which mutation frequency was at least 30-fold of the control, and DNA sequence analysis of the mutants was performed. Therefore, almost all the mutations shown here do not contain spontaneous mutations.

Mutation spectrum

Locations of the *N*-Aco-*N*-Ac-ABA-induced base substitutions over the coding region of the *supF* tRNA gene are shown in Figure 3-3. The distribution was similar in normal and XP cells. Mutations were not distributed randomly ($P<0.0001$ in both cells lines) but were located mostly at specific sites. There were several hot spots (≥ 7 base changes in normal cells and ≥ 6 base changes in XP cells), at sites 109, 115, 123, 134, 144 and 159 in normal cells, and 115, 123, 127 and 159 in XP cells, where the number of mutations observed was ≥ 4 -fold greater than the number expected for random distribution. The most prominent hot spot was at base pair 159 in both cell lines. Almost all (9 of 10) hot spots were located at the sites of G:C base pairs. Seven of 10 hot spots were at 5'-GGG-3' (or 5'-CCC-3') and 5'-GG-3' (or 5'-CC-3') sites.

Table 3-2 Types of base substitutions in *supF* gene in *N*-Aco-*N*-Ac-ABA-treated plasmids replicated in normal (WI38-VA13) or repair deficient (XP2OS(SV)) human fibroblasts

Base substitutions	No. of base changes (%)	
	WI38-VA13	XP2OS(SV)
G:C to T:A	94 (51)	58 (41)
G:C to A:T	32 (18)	34 (24)
G:C to C:G	17 (9)	24 (17)
A:T to G:C	26 (14)	

CHAPTER 4

A spectrum of mutations induced by crotonaldehyde in shuttle vector plasmids propagated in human cells

Abstract

A spectrum of crotonaldehyde-induced mutations in the *supF* gene of the shuttle vector plasmid pMY189 replicated in human fibroblast cells was examined. Base sequence analysis of 104 plasmids with mutations in the *supF* gene revealed that the majority of the mutations were base substitutions (85%) and the rest were frameshifts (15%). A single base substitution was most frequently found (47%), while 25% had multiple base substitutions and interestingly 13% had tandem (adjacent two) base substitutions. Of the base substitution mutations, 50% were G:C to T:A transversions and 23% were G:C to A:T transitions. The mutations were not distributed randomly but were located at several hotspots, most of which were G:C base pair in 5'-AAGG-3' (or 5'-CCTT-3') sequences. Production of propanodeoxyguanosine adducts may be related to such specificity in the mutation spectrum.

Plasmid kit. The purified plasmids were digested with the restriction endonuclease *DpnI* to eliminate the non-replicated plasmids which retain the bacterial methylation pattern.

Selection of mutated supF, and determination of DNA base sequences

Plasmid DNA was introduced into the indicator bacteria KS40/pKY241 by the electroporation apparatus *E.coli* Pulser (Bio-Rad Laboratories, Hercules, CA). The bacteria cells were plated on LB agar containing nalidixic acid, ampicillin and chloramphenicol at concentrations of 50 µg/mL, 150 µg/mL and 30 µg/mL, respectively, supplemented with IPTG and X-gal to select the plasmids containing the mutated *supF* genes. A portion of the cells was plated on LB agar containing ampicillin and chloramphenicol to measure the total number of transformants. After incubation for 24 hours at 37°C colonies were counted and mutation frequencies were calculated.

Mutated plasmids were extracted and purified from the overnight culture with the QIAprep-spin Plasmid kit and the size of each mutated plasmid was checked by agarose gel electrophoresis. The base sequences of the *supF* gene of the plasmids which were not aberrant in size were determined with the ABI PRISM™ Dye Primer (-21M13) Cycle Sequencing Ready Reaction Kit (Perkin-Elmer Corporation, Foster, CA) using the 370A automatic DNA sequencer (Perkin-Elmer Corporation, Foster, CA).

The χ^2 -goodness of fit test was used to determine if crotonaldehyde-induced base substitution mutations were distributed randomly or non-randomly over the

coding region of the *supF* tRNA gene.

Table 4-1 Types of sequence alternations in *supF* gene in crotonaldehyde-treated plasmids pMY189 replicated in normal human fibroblasts

Types	No. of plasmids with mutations (%)
Base substitutions	88 (85)
Single	49 (47)
Tandem	12 (12)
Multiple*	27 (26)
Frameshifts	16 (15)
Single base deletion **	5 (5)
≥ 2 bases deletion***	9 (9)
Single base insertion ****	1 (1)
≥ 2 bases insertion	1 (1)
Total plasmids sequenced	104 (100)

Plasmids with frameshifts are listed in "Frameshifts" only.

* Three had accompanying tandem base substitutions.

** Four had other base substitutions.

*** One had a base substitution.

**** This plasmid had base substitutions.

Table 4-2 Types of base substitutions in *supF* gene in crotonaldehyde-treated plasmids replicated in normal human fibroblasts

Base substitutions	No. of base changes (%)
G:C to T:A	70 (50)
G:C to A:T	32 (23)
G:C to C:G	19 (13)
A:T to G:C	13 (9)
A:T to T:A	5 (4)
A:T to C:G	3 (2)
Total	142 (100)

mechanisms responsible for the specificity at the 4 base sequences are not known. DNA damage may be induced or repair may be blocked selectively on these sequences by unknown mechanisms.

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CHAPTER 5

Spectrum of mutations in shuttle vectors and DNA intra-strand crosslinks induced by acrolein

Abstract

Types of mutations induced by acrolein in the *supF* gene on the shuttle vector plasmid pMY189 replicated in normal human fibroblast cells were examined. Base sequence analysis of 92 plasmids with mutations in the *supF* gene revealed that the majority of the mutations were base substitutions (76%) and the rest were deletions and insertions (24%). Single base substitutions were most frequently found (46%), while multiple base substitutions were 18% and tandem (two adjacent) base substitutions were 12% of the mutations. Of the base substitution mutations, G:C to T:A transversions were 44% and G:C to A:T transitions were 24%. The mutations were distributed not randomly but located at several hotspots. Acrolein produced DNA intra-strand crosslinks between guanine residues, which might be responsible for the induction of the tandem base substitution mutations.

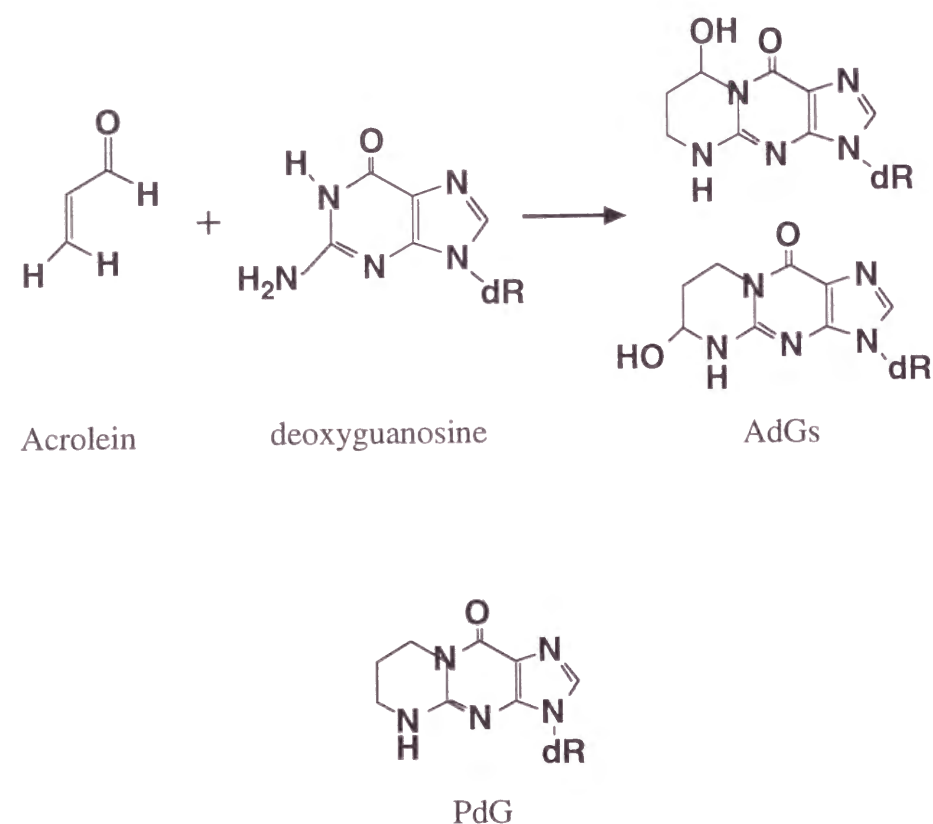


Figure 5-1. Acrolein, two forms of acrolein-derived hydroxy 1, *N*²-propanodeoxyguanosine (AdGs) and propanodeoxyguanosine (PdG).

Materials and methods

Chemicals

Acrolein (purity > 99.9 %), ampicillin, chloramphenicol, nalidixic acid, isopropyl- β -D-thiogalactoside (IPTG) and 5-bromo-4-chloro-3-indoyl- β -D-galactoside (X-gal) were obtained from Wako Chemicals (Osaka, Japan). Restriction endonuclease *Dpn*I was purchased from New England Biolabs, Inc (Beverly, MA). QIAGEN plasmid-kit, QIAprep-spin Plasmid kit and QIAquick-spin PCR Purification kit were obtained from QIAGEN GmbH (Hilden, Germany). Three kinds of 20 mer single-strand DNA (*i.e.*, 5'-TCGTGACTGGGAAAACCCTG-3', 5'-GCGTTACCCAACTTAATCGC-3', 5'-CTTGCAGCACATCCCCCTTT-3') and 5'-Biotinated 20 mer single-strand DNA (*i.e.*, 5'-TTAACGCGAATTTTAACAAA-3') were synthesized by Nippon Seihun Corp. (Tokyo, Japan). pBluescript KS(-) was purchased from Toyobo Corp. (Tokyo, Japan). Taq polymerase was obtained from Takara Shuzo Corp. (Kyoto, Japan). Dynabeads M-280 streptavidin was obtained from DYNAL A.S. (Oslo, Norway). [γ -³²P] ATP was purchased from Amersham Corp. (Chiba, Japan).</

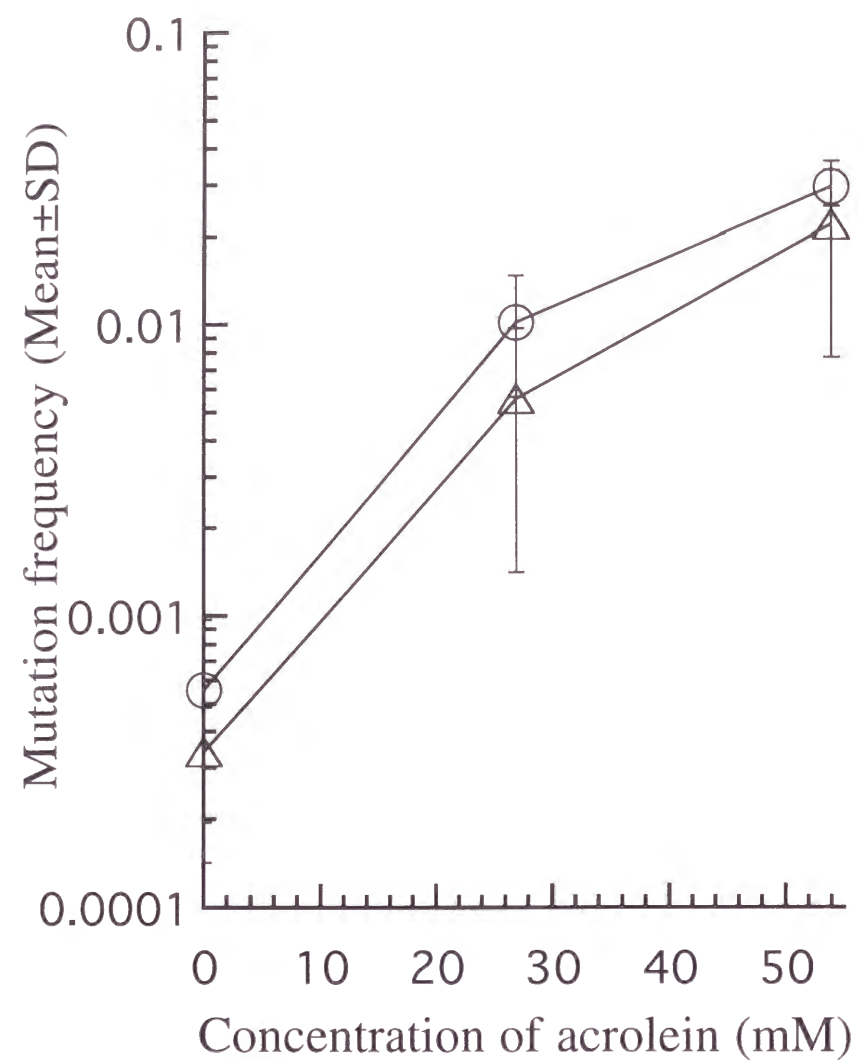


Figure 5-3. Mutation frequency of the acrolein-treated pMY189 plasmids propagated in repair-proficient WI38-VA13 (○) and repair-deficient XP2OS(SV) (△). Average numbers of three independent experiments are plotted with the S.D.

Table 5-1 Types of sequence alternations in *supF* gene in acrolein-treated plasmids pMY189 replicated in normal human fibroblasts

Types	No. of plasmids with mutations (%)
Base substitutions	70 (76)
Single	42 (46)
Tandem	11 (12)
Multiple*	17 (18)
Deletions and insertions	22 (24)
Single base deletion **	4 (4)
≥ 2 bases deletion***	16 (17)
Single base insertion	0 (0)
≥ 2 bases insertion***	2 (2)
Total plasmids sequenced	92 (100)

Plasmids having base substitutions and deletions or

Table 5-2 Types of base substitutions in *supF* gene in acrolein-treated plasmids replicated in normal human fibroblasts

Base substitutions	No. of base changes (%)
G:C to T:A	48 (44)
G:C to A:T	26 (24)
G:C to C:G	13 (12)
A:T to T:A	13 (12)
A:T to C:G	5 (5)
A:T to G:C	4 (4)
Total	108 (100)

The author selected *supF* mutants from experiments in which mutation frequency was at least 20-fold of the control, and DNA sequence analysis of the mutants was performed. Therefore, almost all the mutations shown here do not contain spontaneous mutations.

Mutation spectrum

Distribution of the acrolein-induced base substitutions over the coding region of the *supF* tRNA gene is shown in Figure 5-4. Mutations were not distributed randomly ($P < 0.0005$) but were located mostly at specific sites. There were three hotspots (>4 base changes), at sites 133, 159 and 160, where the number of mutations observed was 4-fold or more greater than the number expected for random distribution. The most prominent hotspot was at base pair 160, where almost all base substitutions (7 of 9) were transversions. All hotspots were located at the sites of G:C base pairs.

There were 13 tandem base substitutions; five were GG to AT (or CC to AT), four were GG to TT (CC to AA). Others were one each of GA to TT (TC to AA), GC to AT, AT to TG (AT to CA) and GT to TA (AC to TA). Most of the tandem base substitutions (9 of 13) occurred at GG (or CC) sites.

